No. of printed pages: 2

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## SARDAR PATEL UNIVERSITY

M.Sc. (Polymer Science & Technology) Semester-II Examination-2015

## Monday, 27<sup>th</sup> April, 2015

## 10.30 a.m. to 1.30 p.m.

## **PS02EPST04: Industrial Process Chemistry**

**Total Marks: 70** 

Note: (1) Attempt all questions. (2) Figures to the right indicate full marks.

Write appropriate choice for the following.

 The rate of substitution chlorination of hydrogen on saturated carbon is the rate substitution of hydrogen on unsaturated carbons.

(a) Greater than (b) Less than (c) equal to (d) None of these

(2) \_\_\_\_\_\_hydrolyzing agents use for hydrolysis.

(a) Acid (b) alkali (c) water (d) all

- (3) Decomposition can be done at elevated temperature called :(a) Hydrolysis (b) Pyrolysis (c) Decomposition (d) Replacement
- (4) \_\_\_\_\_ Catalyst is used in hydration process.

(a) HCL (b)  $H_2SO_4$  (c)  $H_3PO_4$  (d) all of these

(5) In thermodynamic hydrolysis if free energy is negative the reaction is

(a) Spontaneous (b) favorable (c) a & b both (d) not suitable for practical application

- (6) Alkylation consist of replacement of Hydrogen atom present in ammonia by alkyl group.
  (a) C- alkylation (b) N- alkylation (c) Si-Pb-Al alkylation (d) none of these
- (7) \_\_\_\_\_ can be produced by esterification of terpthalic acid with methanol.

(a) ethylene terpthalate (b) di-methyl terpthalate (c) polyethylene terpthalate(d) ethylene glycol

(8) The addition of water molecule to olefin double bond will produced alcohol this reaction is known as :

(a) hydrolysis (b) hydration (c) hydrogenation (d) oxidation

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(8)

Q.2		Attempt any seven of the following	(14)
	(1)	Define: (1) Halogenation & hydration (2) Esterification & oxidation	
	(2)	Explain oxidation without rupture of carbon chain.	
	(3)	Write a note on oxinitration.	
	(4)	Give reaction mechanism for neutral ester & full ester.	
	(5)	Write variety of nitro agent can be used depending upon compound to be	
		nitrated.	
	(6)	Enlist oxidizing agent. Explain any two.	
	(7)	Which equipment used in hydration process. Explain any one.	
	(8)	Write a note on hydrolysis.	
	(9)	Write classification of alkylation process – based on structure of alkyl group.	
Q.3	(a)	Give an account on hydrocarbon with its classification.	(6)
	(b)	Classify alkylating agents and explain chloroderivatives.	(6)
		OR	
	(b)	Write a note on following.	(6)
		1. Chlorination & Iodination	
		2. Cascade of stirred reactor	
Q.4	(a)	Explain in detail manufacture process of phthalic anhydride.	(6)
	(b)	Write types and process for hydrogenating oil by drawing diagram.	(6)
		OR	
	(b)	Explain manufacture process of mono sulfonation of benzene.	(6)
Q.5	(a)	Which products obtain by esterification process? Explain any six.	(6)
	(b)	Explain manufacture process of ethanol	(6)
		OR	
	(b)	Explain mechanism of esterification process.	(6)
Q.6	(a)	Draw a neat labeled flow chart of manufacture process of methanol. Explain	(6)
		in detail.	
	(b)	Write a note on a side reaction of C-Alkylation process.	(6)
		OR	
	(b)	Write a note on reaction type and unit of liquid-liquid chlorination.	(6)

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