

[A-88]

SARDAR PATEL UNIVERSITY

M.Sc. (Physics)(3rd Semester) Examination

Day : Saturday, Date : 25/04/2015, Time : 2:30 p.m. to 5:30 p.m.

Subject : Crystallography and Materials Science Paper No. PS03EPHY01

CBCS(choice based credit system)

Important Note : Q.1 : Multiple choice questions (MCQ) carries one mark each.

Q.2 : Short questions carries two marks each (attempt any seven out of nine)

Q.3 to Q.7 : Long questions carries 12 marks .

Total Marks : 70

Choose the appropriate options from the following s.

Q.1

- i) If g is reciprocal lattice vector , the Bragg's law can be written as
(a) $K+g=0$ (b) $2k.g+g^2=0$ (c) $2k.g+k^2=0$ (d) $k.g=0$
- ii) The fundamental building block whose repetition generates a crystal is
a. Unit cell b. lattice cell c. primitive cell d. none of these.
- iii) Aluminium has f.c.c lattice with interatomic spacing equal to 4.01 Å. The value of lattice constant is
(a) 1.27 (b) 4.01 (c) 2.005 (d) 1.05
- iv) Basis is defined as
(a) An atom (b). a group of atoms
(c) One or group of atoms about a point (d). none of these
- v) The number of basic crystal system is
(a) 2 (b). 4. (c) . 7 (d) 9
- vi) Which of the following Bragg reflections are absent for an f.c.c lattice
(a) 100 (b) 200 (c) 220 (d) 111
- vii) Camera constant is defined as
(a) $\lambda L = rd$ (b) $\lambda R = DL$ (c) $\lambda L = RD$ (d) $\lambda L = rD$
- viii) $BaTiO_3$ is an example of
(a) ferroelectric (b) paramagnetic (c) antiferromagnetic
(d) ferromagnetic

Q.2

Answer any seven out of nine:

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- i) Mention the characteristics of ferroelectric materials?
- ii) What is an Ewald sphere?
- iii) Define single crystal , polycrystalline and amorphous materials .
- iv) What are atomic packing factor and coordination number?
- v) Differentiate between atomic scattering factor and structure factor.
- vi) Differentiate between ferroelectricity and antiferroelectricity
- vii) Explain the terms annealing and quenching
- viii) What are polariton and polaron? .
- ix) Write LST relation. What is Peierls instability?

P.T.O.

Q.3(a) Justify the statement : as and when reciprocal lattice touches the sphere there is a diffraction otherwise not. 6
Discuss spherical projection, stereographic projection and gnomonic projection .

Q.3(b) Construct the reciprocal lattice graphically. Prove that the reciprocal lattice vector is normal to crystal plane (hkl) and of length $1/d$. A unit cell has lattice parameters $a= 10.50$, $b= 8.50$ and $c= 5.25 \text{ \AA}$ with orthogonal axes . Display and calculate the reciprocal lattice parameters and the volume of both the cells . 6

OR

Q.3(b) What is camera constant and give its significance. Discuss a suitable technique to record an electron diffraction pattern from a polycrystalline specimen . How do you index such pattern. 6

Q.4(a) Obtain the structure factor equation for scattering of X rays from all the atoms of a unit cell of orthorhombic lattice. Interpret the result. How does it help in understanding the diffraction pattern. 6

Q.4(b) Derive the required formula for the conduction mobilities for localized and delocalized states for amorphous semiconductors.. 6

OR

Q.4(b) Explain ferro electric domain. What are the main characteristics of ferroelectric materials. Give its classification. Compare ferroelectricity and piezoelectricity. 6

Q.5(a) Discuss the electron –electron interaction in metal based on electrostatic screening . 6

Q.5(b) What are polymers? Give a broad classification of polymer. Discuss the structure of long chain polymers. 6

OR

Q.5(b) Describe the interaction of photons and transverse optical phonons while deriving the expression $\epsilon(\omega) = \epsilon(\infty) + [\epsilon(0) - \epsilon(\infty)] \cdot \frac{\omega_r^2}{\omega_r^2 - \omega^2}$. 6

Q.6(a) Write short notes on (1) long range order (2) Kondo effect 6

Q.6(b) What are liquid crystals? Give a detailed classification of liquid crystals. Describe in brief the applications of liquid crystals. 6

OR

Q.6(b) Define ordered and disordered alloys. Describe the effect of alloying on the residual resistivity of an alloy. 6
