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Ī	SEAT	' No.		No. of Pr	inted Pages : 02		
L.		'No 9] м.					
	L2	J M. Dav: Thursd	Sc. (Physics) (IV''' Se av. Date: 02/11/2017.	emester) Examination Time: 10:00 a.m. to 01:0	0 p.m.		
		Course N	No. PS04CPHY01 (<u>N1</u>	<u>iclear and Particle Physic</u>	<u>es</u>)		
¥			CBCS (choice base	ed credit system)			
Imp	ortant	Note: Signs have thei O.1: Eight mult	iple choice questions (MCQ) carry one mark eac	h.		
		O.2: Short answ	wer questions carrying	two marks each (attempt a	ny seven out of nine).		
		Q.3 to Q.6: Lon	ng answer questions ca	rrying 12 marks each.	Total Marks: 70		
Q1	(i)			state of the deuteron is;			
		(a) 1 ⁺	(b) 2 ⁻	(c) 0 ⁺	(d) 3 ⁺		
	(ii)	The magic nuclei ha	ving zero electric quad	lrupole moment has	shape.		
	~ /	(a) ellipsoidal		(b) spherical (d) toroidal			
		(c) cylindrical		(u) toronaai			
	(iii)	The three β - decay	process are	transitions			
		(a) Isobaric	(b) Isotopic	(c) Isomeric	(d) Isotonic		
	(iv)	The leptonic number	r for positron and anti-	neutrino is			
		(a) l	(b) -1	(c) 0	(d) 2		
	(v)	An electron collidin	g with the atomic nucl	eus experiences Coulomb :	scattering leading to change		
		in its	in its acceleration and emit electromagne				
		Bremsstrahlung radi (a) radial	(b) linear	(c) circular	(d) rotational		
	<i>(</i>)	· · ·			he particle motion		
	(vi)	(a) 1	(b) -2		(d) 2		
		. ,					
	(vii)	The gauge bosons responsible for the strong interaction are known as.(a) gluons(b) gravitons(c) neutrinos(d) mesons					
		(a) gluons					
	(viii)			ne hydrogen nucleus in the	PPI cycle is about (d) 8.5 MeV		
		(a) 931 MeV	(b) 6.55 MeV	(c) 26 MeV	(u) 8.5 Me v		
Q2	(i)	Explain Nordhiem rule.					
	(ii)	Explain experimental properties of the deuteron.					
	(iii)	Discuss Geiger Nut	tal law.				
	(iv)	Derive an expression	of neutrons in matter.				
	(v)	What is compound	nucleus? Write a nucle	ear reaction where in comp	ound nucleus is formed.		
	(vi)	Describe classificat	ion of nuclear reactors				
	(vii)	Explain why fusion	of light nuclei and fis	sion of heavy nuclei releas	es energy?		
(viii) What are breeder reactors? Describe in short its working.(ix) Explain briefly the idea of grand unification theory.							

Q3	(a)	Using necessary equation derive magnetic moment of deuteron.				
¥-	(b)	Discuss various exchange forces in nuclear interaction.	6			
	OR					
	(b)	In the collective nuclear model discuss in brief the rotational motion of the nucleus.	6			
Q4	(a)	Deriving the equation of disintegration energy of α - particles and discuss reason of emission of the particles.	6			
	(b)	Explain the Fermi theory of β - decay by deriving and discussing the equation of transition probability.	6			
	OR					
	(b)	Discuss by deriving equation for the energy loss of heavy charged particle interacting in matter.	6			
Q5	(a)	Write the main components of a nuclear reactor and discuss each of them in short.	6			
(b) Discuss nucleosynthesis in stars.						
	OR					
	(b)	Discuss the medical diagnostic and therapeutic applications of nuclear physics.	6			
Q6	(a)	What are the different conservation laws obeyed by the fundamental particles? With the help of these conservation laws verify whether the following decay/ reaction are allowed or forbidden	6			
		1) $P \rightarrow \pi^+ \pi^- \pi^+$ 2) $\pi^- + n \rightarrow \Lambda^\circ + K^-$	6			
	(b)	Discuss Gell-Mann's SU(3) Quark Model for hadrons. Explain its successes and failures.				
		OR				
	(b)	Discuss the antiproton detection experiment.	6			

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Ĺ	OLA.		No. of Printed Pages : (32			
	[2	3] SARDAR PATEL UI Vallabh Vidya					
		M. Sc. (Physics) 4 th Seme Monday, 6 th Noven	er Examination				
		Monday, 6 th Noven	per, 2017				
		Time: 10:00 am to Subject: PS04CPHY02 [Theoret					
		5 -	Total Marks	: 70			
Note		Figures to the right indicate marks. Symbols have their traditional meaning.					
Q:1	Atte	empt all of the following Multiple choic	type questions. [01 mark each]	[08]			
(1)		The energy corresponding to a free elec					
	(a)	$h^2k^2/8\pi^2m$	(c) $\frac{h^2k}{8\pi^2m}$				
	(b)	$h k^2 / \frac{8\pi^2 m}{8\pi^2 m}$	(d) $h^2 k^2 / 4\pi^2 m$				
		$\sqrt{8\pi^2m}$	$(4\pi^2 m)$				
(2)		A plasma oscillation in a metal is a	excitation of the conduction electro	ns.			
~ /	(a)	collective transverse	(c) transverse				
	(b)	collective longitudinal	(d) longitudinal				
(3)		Near the forbidden band the curvature of					
	(a)	negative	(c) zero(d) positive				
	(b)	constant	(u) positive				
(4)		The distance from the centre to boundary					
	(a)	$\frac{2\pi}{a}$	(c) π/a				
	(b)	$\frac{\pi}{2a}$	(d) 2π				
		/ Zu					
(5)		The classical Debyr-Huckel screening l					
	(a)	$(N_0 e^2 / T)^{-1/2}$	(c) $\left(N_0 e^2 / T\right)^{1/2}$				
	(b)	$\left(N_0 e^2/T\right)^2$	(d) $\left(N_0 e^2/T\right)$				
(6)		Change in 1/H through a single period					
	(a)	1/A _e 1/(A _e *hc)	(c) A_e (d) hc/A_e				
	(b)						
(7)							
	(a)	isotopic mass M as, $T \sim M^{-1}$	(c) $T_C \propto M^{-1/2}$				
		$T_C \propto M^{-1}$ $T_C \propto M^{-2}$	(d) $T_C \propto M^{1/2}$				
	(0)	$I_C \propto IVI$					
(8)	The width of energy gap in a superconductor at 0°K is nearly						
-	(a)		(c) $3T_{C}$ ·				
	(b)	$k_B T_C$	(d) $3.5k_BT_C$				

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Q:2 Answer any 7 of the following 9 questions briefly. [02 marks each]

- *1* Explain Umklapp scattering.
- 2 Explain electrostatic screening.
- 3 How is a reciprocal lattice obtained from direct lattice?
- 4 Using suitable schematic diagrams explain briefly the zone schemes.
- 5 Explain briefly the dHvA effect.
- 6 What anomalous skin effect ?
- 7 Explain the origin of energy gap in superconductors.
- 8 Explain with the help of a suitable diagram the term 'coherence length'.
- 9 Give in brief the experimental survey of superconductivity.
- Q:3 (a) State and prove Bloch theorem.

[6]

[14]

(b) Derive an expression for the band effective mass of an electron and [6] interpret the concept of hole.

OR

- (b) Write a note on Kronig Penny model. [6]
- Q:4 (a) Explain the formation of energy bands. Describe the empty lattice method. [6]
 - (b) What is orthogonalized plane wave? Obtain an expression for OPW form [6] factor.

OR

- (b) Describe the APW method of band structure calculation. [6]
- Q:5 (a) What is Fermi surface? Describe the Harrison's method of constructing [6] Fermi surface.
 - (b) Name various methods for the experimental mapping of Fermi surface and [6] discuss cyclotron resonance in detail.

OR

- (b) Obtain an expression for the Lindhard screening function. Also determine [6] its limiting values for $q \to 0$ and $q \to \infty$.
- Q:6 (a) Explain Josephson effects and derive expression for current density across [6] superconductor-insulator-superconductor junction.
 - (b) Explain the formation of Cooper pair. Enumerate the important features of [6] BCS theory of superconductivity.

OR

(b) Define Superconductivity. Explain in detail how superconductors are [6] classified into class-I and class-II superconductors.

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	L	M. Sc. (Physics) 4 ^t Wednesday,	^h Semeste 8 th Novem	r Examination ber 2017
		Time: 10:0	0 AM to 01	L:00 PM
		Paper Loo Course Title: Applied Cr	de: PS04EF	nhy and Bio-Physics
		course mue, applied of	y stanogr -	Max Marks: 7
NB:	i A	ll symbols have their usual meaning gure at the right side of the questic	g. on indicates	s full marks.
		orrect answer for each of the foll		50
1	The	e intensity of each peak in XRD patt	ern is diffe	rent due to
			b)	Applied current
	a)	Applied voltage	d)	None of the above
2	c) 	Specimen 1r circle diffractometer is used for	u)	
2		Powder specimen	b)	Single crystal
	a) ~)	For any crystalline material	d)	amorphous
2	C)	ne name of the technique where cry		-
3	11	he name of the technique, where ery		
	a)	Laue	b)	Rotation
	c)	Weissenberg	d)	precession
4	Fla	at Metal screen is used in techn	lique to rec	cord the XRD pattern.
	a)	Rotation	b)	Weissenberg
	a) c)	precession	d)	single crystal diffractometer
5	v i	ray diffraction from a f.c.c. structur		reflects the plane
5				(111)
	a)		b) d)	(311)
	c)	(220)		
6		When two cysteine molecules con	mome tog	ether, when type of contains is
		ormed hydrogen	b)	disulphide
	a)		d)	ionic
	c)	peptide n ESR, when external magnetic fiel		
	7 I	n ESR, when external magnetic her he internal field then Lande's facto	r is equal to	
			b)	8
	a		d)	2
	c			
1		Haemoglobin molecule possess hov	b)	3
	а	•	. *	
	C) 2	d)	1
Que2	Wı	ite answers of <u>any seven</u> questio	ons in brief	f.
-	1	Why the white radiation is must	for Laue m	ethod?
	2	What is the role of metal screen	in Weissen	berg method?
	3	Define integrated intensity and g	gives its un	it.
	4	• What is Wilson's plot?		
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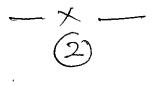
- 5 What is the role of gnomonic projection in Laue method?
- 6 Radiation produces damage to biological molecules -Explain with example.
- 7 State the factors affecting crystallization of biological macromolecule.
- 8 State and explain Pullman's criterion for carcinogenic activity.
- 9 Which three measurable parameters are commonly used for obtaining image of tissues in MRI?
- Que3 [A] What are different types of Laue techniques? Discuss the technique [06] highlighting the basic principle for recording diffraction pattern in such a set up.
- Que3 [B] Give the sketch of Debye-Scherrer powder method and explain in detail its [06] working. Why do you get doublet in such a photograph?
 - OR
- Que 3 [B] Derive the necessary formula for determining the particle size of [06] polycrystalline specimen under non-ideal condition. Also interpret the result.
- Que 4 [A] List the factors modifying the diffracted intensity. Discuss in detail multiplicity [06] factor and polarization factor.
- Que 4 [B] Discuss the graphical method to index a cubic XRD pattern from a [06] polycrystalline specimen.

OR

- Que 4 [B]Sketch a four circle single crystal diffractometer and discuss its working.[06]Define primary and secondary extinctions.
- Que 5[A]Discuss primary, secondary and tertiary structure of protein.[06]Que 5[B]Discuss two different methods of protein crystallization at laboratory in detail.[06]OR0R08Oue 5[B]How Raman spectroscopy differs from IR spectroscopy? Explain the use of Raman[06]
- Que 5 [B] How Raman spectroscopy differs from IR spectroscopy? Explain the use of Raman [00] spectroscopy to study proteins and nucleic acids.
- Que 6 [A] State the basic principle of Infrared spectroscopy and discuss how proteins can be [06] studied with the help of this technique.
- Que 6 [B] Discuss delocalization in biomolecules specifically considering example of [06] benzene molecule and explain various parameters. What is the role of tight binding model to it.

OR

Que 6 [B] Write the full form of NMR. Discuss the basic principle and its application in [06] bio-physics.



Page 2 of 2

(70)

No. of Printed Pages: 3

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M.Sc. (Physics)(IVth Semester) Examination Date : 10 / 11 /2017, Day : Friday , Time : 10:00a.m to 1:00p.m. Subject : Crystal growth and imperfections in Solids, Paper No. PS04EPHY02 CBCS(choice based credit system) Important Note : Q.1 : Multiple choice questions (MCQ) carries one mark each. Q.2: Short questions carries two marks each (attempt any seven out of nine) Q.3 to Q.6 : Long questions carries 12 marks . Total Marks : 70 *Choose the appropriate options from the following in Q.1* Q.1 (8)i) Which rule tells whether or not a dislocation reaction will occur? (a) Frenkel (b) Schottky (c) Frank (d) Burger's technique is known as high temperature solution growth. ii) (a) Bridgman (b) Czochralski (c) Verneuil flame fusion (d) flux growth iii) An atom located at a position that is not a normal lattice site is known as (a) Frenkel defect (b) Schottky defect (c) dislocation (d) tilt defect ----- is a negative ion vacancy with one excess electron bound at the iv) vacancy (a)F-center (b) M-center (c) V-Center (d) K-center v) When structures are subjected to dynamic and fluctuating stresses it is termed as (b) Fatigue (c) Fracture (d) dislocation (a) Creep Self diffusion occurs by which mechanism vi) (a)vacancy (b) interstitial (c) ring (d) impurity Which is two dimensional imperfection? vii) (a) line (b) surface (c) point (d) void viii) The number of independent variable for the layer type phase equilibrium diagram above the melting point of Pb and Al is (a) 0 (b) 1 (c) 2 (d) 3

Page No. 1

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Q.2 *Answer any seven questions out of nine in 0.2*

- i) Obtain the expression for strain energy of an edge dislocation.
- ii) What is creep ? Draw a plot of creep strain as function of time.
- iii) Draw the diagrams showing like and unlike dislocations on the same plane, unlike dislocations separated by few atomic spacing, combination of dislocations.
- iv) Explain why in Brigmann technique grown seed crystal is not used?
- v) Define slip planes and slip directions.
- vi) Draw the cooling curve of pure and binary system and calculate number of degree of freedom for each case.
- vii) What is diffusion ? State Fick's first law of diffusion.
- viii) Determine the fraction of atoms in a given solid with the energy equal to 1.5 eV at room temperature (300K) and at 1500K where Boltzmann's constant $k = 8.614 \times 10^{-5} \text{ eV/K}$.
- ix) Explain Kirkendall effect.
- Q.3(a) Sketch and discuss in detail phase equilibrium diagram of two component system that are mutually soluble in liquid state and partially soluble in solid state.
- Q.3(b) Obtain the Gibb's Thomson equation for vapour and modified (6) Thomson's equation for melt.

OR

- Q.3(b) Draw schematic diagram of the crystallization apparatus, and explain in (6) detail low temperature solution growth method of crystallization using slow cooling process.
- Q.4(a) Obtain the expression for concentration of Frenkel defects in equilibrium (6) condition in an ionic crystal.
- Q.4(b) What is dislocation? Describe edge and screw dislocation in a crystal (6) using appropriate diagram.

OR

Q.4(b) What are stacking faults? Explain how this fault are produced in f.c.c (6) crystal.

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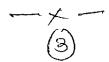
(6)

Q.5(a)	What is cross slip process ? Discuss how dislocation velocity can be measured with proper examples.	(6)
Q.5(b)	What to you understand by multiplication of dislocation ? Explain Frank- Read and multiple cross glide process. OR	(6)
Q.5(b)	State the methods used for observations of dislocations and explain any two of them in detail.	(6)
Q.6(a)	Discuss what is meant by vacancy diffusion and interstitial diffusion. Using equations explain non-steady state diffusion phenomena.	(6)
Q.6(b)	Explain carburizing and decarburizing process in steel. OR	(6)
Q.6(b)	What are cyclic stresses ? Discuss S-N curves with appropriate diagrams in detail.	(6)

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SARDAR PATEL UNIVERSITY M. Sc. (PHYSICS) IVth Semester Examination Friday, 10th November, 2017 10:00 am to 01:00 pm Course No.: PS04EPHY04. ADVANCED SOLID STATE ELECTRONIC DEVICES

1. 1. (13)

Note: All questions are compulsory.

SEAT No.

Total Marks:70

No. of Printed Pages: 02

Q.1 Multiple choice questions. (i) Band gap shrinkage is a BJT design limitation that occurs when (a) Base is made too thin (b) Base is made very lightly doped (c) Emitter is doped heavily (d) Collector is made narrow Cross doping is a serious problem in heterojunction devices. **(ii)** (a) Si - Ge (b) GaAs - InP (c) Si - GaP (d) CdSe - ZnSe For which of the following devices the carrier concentration in an (iii) active region is not controlled by doping of impurity? (a) BJT (b) MESFET (c) MODFET (d) CMOS (iv) In accumulation mode of MOS capacitor with p-type semiconductor, the hole density increases (a) within semiconductor (b) within oxide layer (c) within semiconductor away from interface (d) near semiconductor-oxide interface within semiconductor (v) At pinch-off, the channel current is (a) infinite (b) zero (c) I_{DSat} (d) none of this Which of the following is a most sensitive radiation detector? (vi) (a) P-N photodiode (b) Solar cell (d) Avalanche detector (c) PIN detector For making PIN diode, in the case of night vision applications, the (vii) material used is (a) GaP (b) GaAs (c) Hg_xCd_{1-x} (d) $Hg_{1-x}Cd_{1+x}$ (viii) The infrared wavelength range is microns. (a) 0.1-0.5 (b) 0.4-0.76 (c) 0.7-100 (d) 100-200. 0.2 Short answer questions (Attempt any seven) Differentiate between a MOSFET and MODFET. **(i)** (ii) Explain the preference of materials in Si based HBTs. Write down the important issues of short channel MOSFET. (iii) (iv) Draw the charge distribution in MOS capacitor in depletion, inversion and accumulation mode. **(v)** What is a CMOS circuit? Why it consumes less power than other devices? (vi) Explain the working of a phototransistor. (vii) Define various characterizing parameters of a solar cell.

- (viii) What is band gap tailoring? Explain.
- (ix) What is external quantum efficiency?

(14)

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Q.3(a) (b)	The second				
(b)	Sketch the multilayer structure of a MODFET and describe the role of each of the layers and discuss operation of MODFET in detail with the help of energy band diagram.	(6)			
Q.4(a)	Discuss the operation of a MOS capacitor. Draw and explain its C-V characteristics.	(6)			
(b)	Write down the key motivations used in MODFET and describe them.	(6)			
OR					
(b)	Explain different long channel issues in a MOSFET.	(6)			
Q.5(a)	What is CCD? Sketch the structure of CCD and explain the process of charge transfer in it.	(6)			
(b)	Discuss the absorption of light in semiconductors and explain how this property of semiconductors is used to construct radiation detectors.	(6)			
(b)	OR Obtain an expression for photocurrent in a P-N photo diode.	(6)			
(b)	Obtain an expression for photocartein in a ray photo aroue.	(-)			
Q.6(a)	Discuss various important issues in the PIN device design.	(6)			
(b)	Write a note on various LED materials.	(6)			
	OR				
(b)	Draw the structures of Surface Emitting and Edge Emitting LEDs. Explain their operation and describe their merits and limitations. XXXXXXXXXX	(6)			

