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SEAT No. _____

No. of Printed Pages : 2

Sardar Patel University

M. Sc THIRD SEMESTER (Physical Chemistry) Examination – 2019

Friday, 22nd March 2019

2.00 p.m. to 5.00 p.m.

PS03CPHC 22, Nuclear and Radiation Chemistry

N. B: (i) Attempt all questions

(ii) Figures to the right indicate full marks

(iii) $u = 1.6605 \times 10^{-27}$ kg, $h = 6.6262 \times 10^{-34}$ J.s, $1\text{eV} = 1.6022 \times 10^{-19}$ J
 $m^1\text{H} = 1.0078$ u, $m^2\text{D} = 2.0141$ u, $m^3\text{H} = 3.0160$ u

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1. Multiple Choice Questions

(i) Which of the following has highest cross section ?

(a) (d,t) (b) (t,n) (c) (d,p) (d) (t,d)

(ii) In n Tc, T_c is defined as:

(a) Fusion temp. (b) break even point (c) critical temp. (d) combustion temp.

(iii) $e^- + ? \rightarrow \text{H}^+$ (a) H^+ (b) H (c) H_2O^+ (d) OH^-

(iv) The zig zag path indicates:

(a) alpha particles (b) aqueous electrons (c) beta particles (d) gamma particles

(v) $d + p \rightarrow ? + \gamma$ (a) ${}^2_3\text{He}$ (a) ${}^2_3\text{He}$ (c) ${}^3_2\text{He}$ (d) ${}^4_2\text{He}$ (vi) $I.C_i$ is equal to :(a) 3.70×10^{10} dps (b) 2.40×10^{10} dps (c) 3.70×10^{10} dps (d) 2.40×10^{10} dps

(vii) Following radiation does not produce ion-pairs:

(a) gamma rays (b) alpha rays (c) atomic electrons (d) beta rays

(viii) Barn is the unit of:

(a) flux (b) cross section (c) yield (d) both b and c

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2. Attempt ANY SEVEN:

(i) How one can express the force of repulsion in nuclear reaction ?

(ii) ${}^{63}\text{Cu} ({}^2\text{H}, {}^1\text{H}) {}^{64}\text{Cu}$: is this reaction an example of stripping or pick up reaction ?(iii) Consider the reactions: (γ, n) and (γ, p) and classify them by their type

(iv) Give two basic characters of a tracer nuclei.

(v) Prove that the hydrogen of the water goes to the alcohol in the ester hydrolysis reaction.

(vi) How one can determine the antibiotic produced in a bacterial broth ?

(vii) Derive $p = h / \lambda$

(viii) Give two examples of reactions that involve charged particles.

(ix) What are scintillation counters ?

(1)

(P.T.O.)

3. (a) Explain in detail different types of particle – particle reactions and give the unique features of (d,p) reactions 6
- (b) (i) A deuteron particle is accelerated under 1 MeV energy and estimate its velocity. 3
- (ii) Describe the elements of Bethe's notation and explain with two examples. 3

OR

- (b) (i) Define and discuss nuclear instability. 3
- (ii) Which one is correct $E_{th} > Q$ or $Q > E_{th}$? Prove it with proper explanation. 3
4. (a) Discuss the different features and conditions used under magnetic and inertial confinement to control the nuclear fusion reactions. 6
- (b) (i) Give at least three reactions of importance for fusion process. 3
- (ii) What is a breeding cycle? Why it is required. 3

OR

- (b) Explain the inverse law. 6
5. (a) Are all the P-Cl bonds in PCl_5 equivalent or not? Justify your answer taking the help of the tracer technique. 6
- (b) What is DID analysis method? Describe how it can be used to estimate the volume of blood in humans 6

OR

- (b) Give the features of neutron activation analysis and how it is useful in tracing the arsenic poisoning. 6
6. (a) What is meant by BREMSSTRHALUNG? Discuss features of interaction of Beta Particles with matter. 6
- (b) Describe and discuss various physical effects and photochemical reactions during the slowing down of ionizing radiation. 6

OR

- (b) Write a note on Solid State Ionization detectors. 6

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