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SARDAŔ PATEL UNIVERSITY

M. Sc Pharmaceutical Chemistry (Semester-I) Examination Thursday, 27/10/2016; Time-10:00 AM to 01:00 PM SUBJECT CODE: PS01CPCH03

SUBJECT TITLE: Basic Chemistry for Pharmaceuticals

Maximum Marks: 70 Note: (1) All questions are compulsory. (2) Figure to right indicates total marks of question. Choose the correct option for the following: Q-1 1×8 When the carbon atom is sp2 hybridized in a compound, it is bonded to a. 5 other atoms b. 4 other atoms c. 2 other atoms d. 3 other atoms Which of the following hydrocarbons has the shortest C-C bond a. $CH_2=CH_2$ b. CH₃-CH₃ c. CH≡CH d. None Which of the following is not a nucleophile. a. NH₃ b. HSO₃c. AlCl₃ d. OH-Which of the following is an electrophile a. CH₃O b. CH₃CH₂⁺ c. NH₃ d. CH₃CH₂⁻. The carbon atoms in a benzene rings are a. sp hybridized b. sp2 hybridized c. sp3 hybridized d. None Which of the following is not aromatic a. pyridine b. Pyrrole c. Furan d. Ethylene Tap water contains impurities of a. Ca²⁺, Mg²⁺, Na⁺ ions b. Cl⁻, CO₃²⁻, SO₄²⁻ d. None of above In the limit test for Arsenic, arsenic converted into a. Arsine (AsH₃) b. As_2S_3 c. AsCl₃ d. None O-2 Answer the following (Any Seven). 2×7 Indicate the type of hybridization of each atom in the following structure: 1. a. CH₃CH₃ b. CH₃C≡CCH₃ Define hemolytic fission with suitable example? 2 Explain why phenol is nitrated more readily than benzene. 3. Elaborate e1 mechanism with suitable example. 4.

	5.	Define Huckel's rule?	
	6.	Describe that antracene is aromatic compound.	
	7.	Describe the limit test for lead?	
	8.	Define assay of tolerances.	
	9.	What is the Heavy metal impurities?	
Q-3	A.	Write a note on a. sp3 hybridization b. sp2 hybridization	(
	B.	Describe Carbenes, Nitrenes and benzyne with suitable example. OR	(
Q-4	В.	Write a note on factors which influence a reaction.	(
	A.	Explain why amino group (-NH ₂) acts an ortho, para director while nitro (-NO ₂) act as a meta-director.	(
	В.	Describe the nitration and Friedel Craft's alkylation of benzene.	(
		OR	
	В.	Write a note on Phase transfer catalyst and Crown ethers.	ć
Q-5	A.	Describe the structure of benzene.	6
	В.	Explain why: a. Pyridine is more basic than pyrrole b. Pyridine undergoes nucleophilic substitution at 2- position. OR	6
÷	В.	How is furan synthesized? Describe its important reactions.	
Q-6	A.	Write note on Major sources of impurities.	6
	В.	Describe the pharmacopeial methods for the control of impurities.	6
		OR	
	В	Describe the major sources of errors	6
