

C109)

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SEAT NO: _____

No. of Printed Pages: 03

SARDAR PATEL UNIVERSITY

M.Sc. Semester-III: (Physical Chemistry) Examination (CBCS)-2018

Tuesday, 30th October, 2018

02:00 p.m. to 05:00 p.m.

PS03EPHC22, Advanced Characterization Techniques

- N.B.:*
- i. Attempt all questions.*
 - ii. Figures to right indicate full marks.*
 - iii. Unless otherwise mentioned, symbols and notations have their usual standard meanings.*
 - iv. Neat sketches are to be drawn to illustrate answers, wherever required.*
 - v. Assume suitable data, if necessary and indicate the same clearly.*

- Q.1** The question i) to viii) contain a Multiple Choice Questions (MCQs). Each question has four alternative responses marked (a), (b), (c) and (d) out of which *only one* is the correct response. Please mark correct response i e (a) / (b) / (c) / (d). **[08]**
- i)** The deformation per unit length is called
(a) Strain (b) Stress
(c) Elasticity (d) None of these
 - ii)** Two parallel, equal and opposite forces acting tangentially to the surface of the body is called as _____
(a) Complementary stress (b) Compressive stress
(c) Shear stress (d) Tensile stress
 - iii)** Viscoelasticity exhibits which of the following properties?
(a) Time-dependent (b) Temperature-dependent
(c) Temperature and time dependent (d) Temperature and time independent
 - iv)** Shape of true stress-strain curve for a material depends on
(a) Strain (b) Strain rate (c) Temperature (d) All
 - v)** The presence of solvent in a crystalline sample of a compound could be detected and quantified using:
(a) UV-VIS (b) AAS (c) TGA (d) ESI-MS
 - vi)** In DTA, sharp endothermic peaks give ideas of changes in _____, whereas broad endotherms signify _____ reactions.
(a) amorphous, hydration (b) crystallinity, dehydration
(c) amorphous, dehydration (d) crystallinity, hydration
 - vii)** **Statement 1:** Brookfield viscometer is very robust, is supported by software and even shows values of K and n, but, sometimes particles suspended in the fluid cause problems.
Statement 2: Drawbacks of plate and cone viscometer: edge-effects, turbulence and temperature effects.
(a) True, False (b) True, True (c) False, False (d) False, True



(P.T.O.)

- viii) What is the correct definition of a pseudoplastic liquid?
- A liquid which becomes less viscous as the rate of shear increases
 - A liquid which becomes more viscous as the rate of shear increases
 - A liquid which becomes less viscous over time when a constant shear stress is applied
 - A liquid which becomes more viscous over time when a constant shear stress is applied

- Q.2** Answer the following questions as directed (**ANY SEVEN**). [14]
- Explain how liquid crystalline polymers at the molecular and supermolecular level are useful to understand mechanical properties of polymer.
 - Define: Poisson's ratio (ν), Shear modulus (G)
 - Discuss *viz-a-viz* : Elastic solid and Viscous liquid.
 - Define: Storage modulus G_1 and Loss Modulus G_2
 - How instrument for Thermogravimetry records the results?
 - List one application each of Differential Thermal Analysis (DTA) in Physical chemistry and Analytical chemistry.
 - Explain Basic continuum and multi-scale models as a foundation rheology.
 - Describe succinctly: Network of Entanglements.
 - Mention one application of rheology in Food and processing industries.

- Q.3**
- Elucidate how blends, grafts and copolymers at the molecular and supermolecular level are valuable to recognize mechanical properties of polymer. [06]
 - Discuss Engineering components of strain. Present component of strain. [06]
- OR**
- (i) Define: Deformation ratio. Give details of finite strain deformation of a cube, of unit dimensions in the undeformed state to the rectangular parallelepiped. [03]
(ii) Define: Young Modulus (E) and Bulk Modulus (K) [03]

- Q.4**
- Derive the Strain energy function $U = \Delta A = \frac{1}{2} NkT (\lambda_1^2 + \lambda_2^2 + \lambda_3^2 - 3)$ [06]
 - Describe Kelvin model and derive the expression: [06]
 $e = J\sigma [1 - \exp(-\frac{t}{\tau})]$ where J is the Spring compliance.

OR

- (i) Define: Creep and Creep compliance $J(t)$; Retardation time (τ'); Relaxation time (τ) [03]
(ii) What is Stress Relaxation? Show that for amorphous linear polymers at high temperature the stress may eventually decay to zero. [03]



- Q.5**
- a) Write working principle of differential thermal analysis (DTA). How Environmental factors and sample factors affect the DTA curve? [06]
 - b)
 - i) Compare: Derivative Thermogravimetry (DTG) and Thermogravimetry analysis (TGA). [03]
 - ii) Write Borchardt and Daniels theoretical basis to find a relationship between sample parameters and the area under the peak of differential thermal analysis (DTA). [03]

OR

- b) Explain working principal, instrumentation and applications of differential scanning calorimetry (DSC). [06]
- Q.6**
- a) Expound following non-Newtonian behaviour with suitable illustrations: [06]
 - (i) Shear Thinning Fluids (ii) Dilatant Fluids (iii) Bingham Plastic
 - b) Elaborate viscoelastic effects of following common non-Newtonian behavior [06] with appropriate examples:
 - (i) Weissenberg effect (ii) Fluid memory and (iii) Die Swell

OR

- b)
 - (i) Define: Relaxation modulus. Explain $\log G$ vs $\log T$ graph for stress relaxation. [03]
 - (ii) Define: Deborah number. How D_e is valuable to understand solid and liquid behavior? [03]



