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(104) SARDAR PATEL UNIVERSITY

M.Sc. Semester-III: Analytical Chemistry Examination (CBCS) October-2018,

Tuesday, Date: 30.10.2018
Time: 2.00 p.m. to 5.00 p.m., Paper: PS03EANC21
Subject: Separation Methods

Q.1	off it	TT! -1-11-1 -1		is indicated at the end of the question y and indicate the same clearly.	th n
	11	Highlight the correct option	L ×n	aV sammer to only same to	1001
	1)	HEPT express by			[80]
		a) Lxn	b)		
	200	c) H/n	d)	A+B/H	
	ii)	clive the name of detect	tor	suitable for preparative	
		chromatography.			
				ECD of a diply sand	
	:::\	and the control of th	d)	TCD	
	iii)	The sould write to upon as	ad	etector in SFC over HPLC?	
8		w) FID	b)	TCD gishow malgx (d	
		c) ECD	d)	None a COT manufact	
	1V)	Give the expected retention vo	olun	ne for a solute which has	
		aroundation co-enticient 0.08 on	ı a	Collimn contains 50 ml of	
		bracionary phase and 0.3 mL, mo	bile	phase	
		a) 45.75 mL.	bl	4 57 mI	
		c) 4.57 Lit.	di	None 1	
	v)	which of the chromatography to	echr	idies does not have mabile	
		and stationery phase?	10	draged tooks not have mobile	
			b)	HPLC 032.0	
			d)		
	vi)	Capacity of resin depends on	-,	til Discusse the actuality to	
		a) porosity	bl	rigidity and the lax?	
7			d)	number of functional	
				mumber of functional	
	vii)	Which of the following preferr	har	groups and distributed a	
		alkaloid?	cu	as absorbent to separate	
		a) Cellulose			
			b)	Alumina alusmabus	
	viii)	Which of the following type of gold	d)	Silica gel	
	TOTAL	Which of the following type of gel a) Porous	18 111	led in GPC column?	
		e) Uand	b)	Spherical and amount	
2	a)	Attempt any SEVEN	d)	All of above	
	i)	Explain the the section 1			14]
	-)	Explain the theoretical equation of and its utility.	i 're	solution' in chromatography	
	ii)	data to delitey.			
	iii)	Discus the Kovats retention index	and	l its applications.	
		Explain bilely the SFE.			
	iv)	Why post trailing of peak is observ	ved i	n GSC? How to avoided it?	
	v)	Why HPLC is superior over GC?		, M	

ča.	vi vii viii ix	Discuss the Rf and Rx Values? Write a note on "Chromatotron".	
Q.3	a	Answer the following	[06]
	i: b)	Explain superiority of TLC over other chromatographic technique	[06]
	18.80	OR	
	(b)		
Q.4	a)	Discuss the characteristics of solvent for extraction.).
901	a,	theory'? Discuss Van- Deemter constant, equations and its	[06]
		usefulness to find H_{min} , and u_{opt} .	
	b)		[06]
	i)	principle.	[]
	ii)	A 4.20 meter column has a height equivalent to a theoretical	
		plate 0.70 mm. If the flow rate is 32.5 mL/min. calculate the base width in second of a peak for a solute having retention time. a) 39 Sec. b) 1 min. and 14 Sec. and c) 3 min. and 48 Sec. OR	
	b)	Explain working principle of TCD. Discuss the comparison between TCD and FID.	
2.5	a)	Explain the "gradient elution" in HPLC technique and discuss a detector suitable for it.	[06]
	b)	Discuss the resolution in Chromatography. Two solutes with distribution ratio 1.47 and 1.86 are to separate on a column whose volume ratio of stationary phase to mobile phase is 13.6, (a) How many theoretical plates are needed to ensure a resolution of 1.35. (b) What length of column is required for part (a), if H is	[06]
-3		0.250 cm.	
		jo (h OR are mortigo itselle len) (2	
	b)	Discuss the advantage of SFC over other chromatography. Explain the instrumentation and operating variables of SFC.	[06]
2.6	a)	Answer the following	[06]
	i ii	Explain the separation mechanism used in GPC.	[oo]
. 9		Discuss the principle and types of ion-exchangers in detail.	
	~,	Write a note on endosmosis and discuss on correction of endosmosis.	[06]
7	b)	Discuss various methods used for calibration of gel permeation chromatography.	[06]
[2:1]		e) Hard st Attempt any Savies 1) Explain me theoretical equation of resolution in characters.	6.0
		ii) Discus the Kongds r Tention lade: Explain briefly the Set. Why goes trailing of peak is closed.	
		The reverse same over the reverse strong and the reverse of the re	