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SEAT No. _____

No of printed pages: 04

SARDAR PATEL UNIVERSITY

M.Sc. (Chemistry) Semester-II Examination

Tuesday, 19th November 2019

Organic Chemistry-II (PS02CCHE22)

Time: 10.00 am to 1.00 pm

Total Marks: 70

Note: Right hand figures indicate marks

Q. 1 : Select the correct answer from each of the following.

[08]

- 1) β -amino carboxylic acid converted to β -lactum by using ----- .
a) DCC b) LDA c) R-Li d) DDQ
- 2) Which intermediate is formed in Bamford-Stevens reaction using protic solvent?
a) Carbocation b) Carbanion c) Carbene d) Nitrene
- 3) Which borane is formed by reaction of 2,3-dimethyl-2-butene and BH_3 ?
a) Catechol borane b) Thexyl borane c) Tripropyl borane d) Disiamyl borane
- 4) Selective alkylation of aldehydes and ketones can be carried out by using _____ reaction.
a) Wittig b) Stork-Enamine c) Robinson d) Mannich
- 5) Oxidation number of carbon-1 in acetic acid is _____.
a) 0 b) 2 c) +3 d) -3
- 6) Which of the following functional group is not reduced by NaBH_4 ?
a) Aldehyde b) Ketone c) Ester d) Acid chloride
- 7) Active methylene group can be selectively oxidized by _____.
a) SeO_2 b) DMSO/PTS c) MnO_2 d) $\text{Al}(\text{O-terbutyl})_3$
- 8) Better alternative for Michael addition is _____.
a) Wittig reaction b) Robinson annulations c) Peterson reaction d) Shapiro reaction

Q. 2: Answer the following short questions (Any 7)

[14]

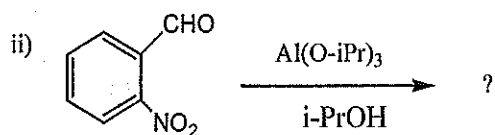
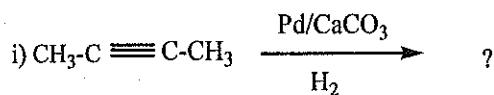
- 1) How does Bamford-Stevens reaction is differ from Shapiro reaction?
- 2) Why Wittig reaction should be performed under inert and dry condition?
- 3) Why does enamine preparation required only 2^o amine in Stork-Enamine reaction?

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4) Explain the oxidation of benzylic alcohol using MnO_2 with mechanism.

5) Write the product of the following reactions.

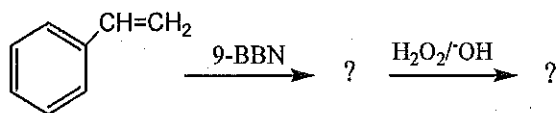


6) Give the synthesis of methyl vinyl ketone by using Mannich reaction.

7) What is the selectivity of $KMnO_4$ as reagent with varying conditions?

8) Discuss the mechanism of reduction of ethyl acetate by $LiAlH_4$.

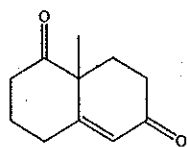
9) Complete the following reaction.



Q. 3. A) Answer the following questions.

[06]

i) Write the synthesis of following molecule by Robinson ring annulations.



ii) Explain the Schlosser modification.

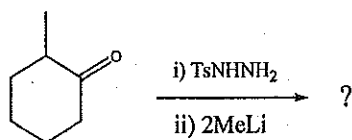
B) Do as directed.

[06]

i) Non-stabilized ylide preferentially lead to the formation of Z-alkene in Wittig reaction.

Explain.

ii) Complete the following transformation with detail mechanism.



OR

B) Answer the following questions.

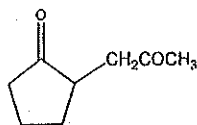
[06]

- i) Show that Peterson olefination follows anti elimination under acidic condition.
- ii) Nature of the solvent decides the pathway for Bamford-Stevens reaction.

Q.4. A) Answer the following questions.

[06]

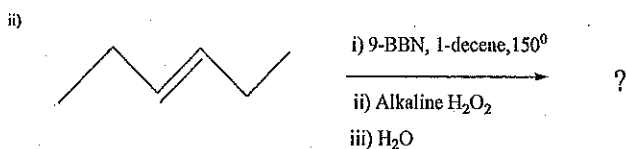
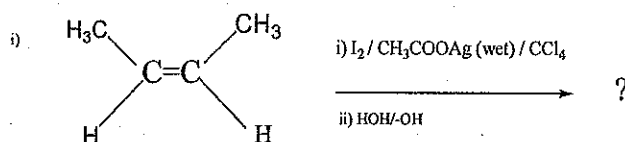
- i) Write the synthesis of following molecule by Stork enamine reaction.



- ii) Write a note on Bayer-Villiger oxidation.

B) Complete the following transformations.

[06]



OR

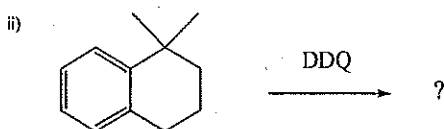
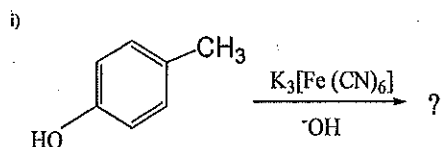
B) Answer the following questions.

[06]

- i) Give the detailed mechanism of Buchwald-Hartwig amination.
- ii) Discuss the importance of carbonylation of organoborane compounds in organic synthesis.

Q.5. A) Complete the following transformations.

[06]



(P.T.O)

B) Answer the following questions.

[06]

- i) Maleic acid on reaction with neutral KMnO_4 gives meso tartaric acid. Explain
- ii) Discuss the allylic bromination using NBS with detail mechanism.

OR

B) Answer the following questions.

[06]

- i) Write a note on Moffatt oxidation.
- ii) Discuss all synthetic applications of mercuric oxide as a reagent.

Q.6. A) Answer the following questions.

[06]

- i) Give the comparison between LiAlH_4 and NaBH_4 as reducing reagent.
- ii) The product formation in TBTH reduction of halo alkenes depends on concentration of TBTH. Explain

B) Write a note on:

[06]

- i) Clemmensen reduction
- ii) Wolf Kishner reduction

OR

B) Answer the following questions.

[06]

- i) Explain the Birch reduction with mechanism.
- ii) Describe the basic differences between reactivity of Gilman and Grignard reagents towards α,β -unsaturated aldehyde.

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