## SARDAR PATEL UNIVERSITY

M.Sc. (Analytical Chemistry) Examination, IVth Semester(CBCS)

Tuesday, Date: 28.04.2015

Session: Morning, Time: 10.30 a.m. to 1.30 p.m.

Paper: PS04ECHE05, Subject: Environmental Chemistry

Total Marks: 70

N.B.: i) The numbers of the marks carried by each question is indicated at the end of the question ii) Assume suitable data if considered necessary and indicate the same clearly.

Q.1	i)	A term TDS refers to	[08
		<ul> <li>a) Total dissolved solids</li> <li>b) Total dilution value</li> <li>c) Total density substituent's</li> <li>d) Both (a) and (c)</li> </ul>	
	ii)	Which is the first component of sampling train?	
		<ul><li>a) Collector</li><li>b) Vacuum source</li><li>c) Metering device</li><li>d) None</li></ul>	
	iii)	An air pollutant responsible for loss of metallic luster is	
		a) $SO_2$ b) $CO$	
		c) $CO_2$ d) $NO_X$	
	iv)	An ideal temperature of promoting pyrolysis is around	
		a) 1000 °C b) 550 °C	
		c) 700 °C d) 910 °C	
	v)	A gas transportation and disposal of solid wastes emit is	
		a) CO b) CO <sub>2</sub>	
		c) Both a) & b) d) SO <sub>2</sub>	
	vi)		
		into CO, H <sub>2</sub> and CO <sub>2</sub> carried out in a technique is called	
		a) Landfill b) Pyrolysis	
	***	c) Gasification d) MBT	
	vii)	Ozone layer exists at an altitude of	
		a) ~50 Kms b) ~11 Kms	
	:::\	c) ~23 Kms d) ~50 Kms	
	viii)	Plants absorb elemental nitrogen, via their roots, in the form of	
		a) Soluble nitrate salts b) Nitrogen c) Nitrogen diavide	
0.2		c) Nitrogen dioxide d) Proteins Attempt any Seven	[1/
Q.2	i)	State the terms 'TLV' and 'PHS'.	[14
	ii)		
	iii)	Discuss turbidity and hardness of water.	
	iv)	Define temporary and permanent hardness, stating ways of	
	11)	expressing them.	
	v)		
	vi)		
	vii)		
	viii)	What is composition of air? Average molar mass of atmosphere is	
	,	close to that of nitrogen gas! Explain [N=14,O=16,Ar=40,CO <sub>2</sub> =44].	
		[	

	ix)	What do you understand by hydrosphere? Name its important	
Q.3	a)	components with their significances. Illustrate sampling train for the air sample. A 27 L air sample when collected and absorbed in KI solution required 30.24 mL of 0.1018 M Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> solution to titrate librated I <sub>2</sub> equivalent to Cl <sub>2</sub> gas. Calculate the concentration of Cl <sub>2</sub> gas in the sample in ppm.	[06]
	<b>b)</b>	Answer the following Sulfur dioxide in a 10 L air sample was converted into SO <sub>4</sub> <sup>2-</sup> by	[06]
	ii)	appropriate treatment, and then $SO_4^{2-}$ into $BaSO_4(s)$ using 25.2 mL of 0.04 M $BaCl_2$ solution. $BaSO_4$ precipipates were filtered, and filtrate thus left required 36.0 mL of 0.02017 M EDTA solution to titrate excess $Ba^{2+}$ . Calculate concentration of $SO_2$ in ppm in the air sample [ $Ba=137.33$ , $S=32$ , $Cl=35.5$ ]. Write a note on ozone depletion.	
	b)	Outline following	
	i)	Analysis of NO–NO <sub>x</sub> and CO–CO <sub>x</sub> .	
	ii)	Effect of air pollutant on man and materials.	10.61
Q.4	a)	Enlist parameters which are employed for physical examination of water, and discuss them in detail.	[06]
	<b>b</b> )	Answer the following	[06]
	i)	Give a detail account of major components of water. Discuss	
	ii)	analysis of water and its significant effect.  Name various surface water pathogens, stating their importance.	
	,	OR	
	<b>b</b> )	State the term 'COD'. If a 1 L solution contains 425 mg of potassium hydrogen phthalate, KHP, calculate theoretical COD value in mg/mL	
Q.5	a)	Write a note on methods for disposal of wastes.	[06]
	<b>b</b> )	Answer the following Give the major causes of soil pollution.	[06]
	i) ii)	Describe the Biomagnifications process.  OR	
	b)	State municipal solid waste. Describe construction and demolition of wastes with suitable examples.	
Q.6	a)		[06]
	b)	What do you mean by weathering processes? Describe in brief processes of soil formation. State the common features of soil.	[06]
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