[106|A-9]

SARDAR PATEL UNIVERSITY

S.Y.B.sc. 3rd SEMESTER EXAMINATION JAMUARY, 2021

BIOCHEMISTRY: USO3CBCH22

TITLE: BIOPHYSICAL BIOCBEMISTRY	
Date: 01/01/2021; Friday Time: 02:00 PM TO 04:00 PM TOTAL MARKS: 70	
Q.1 Select proper option from following MCQ. [10]	
1) Which of the following fluid is minimum in our body?	
a) ECF b) ICF c) ITF d) CSF	
2) Water intoxication is due to	
a) Pure water dehydration b) pure salt dehydration	
c) poor secretion of ADH d) renal failure	
3) Minimum excretory volume is about ml / 24 hours	
a) 100-200 b) 200-400 c) 500-600 d) 700-900	
4) Unequal distribution of diffusible ion of RBC is due to	
a) surface tension b) diffusion	
c) donnan membrane equilibrium d) viscosity	
5) Emulsoids arecolloids.	
a) liophilic b) liophobic c) protective d) association	
6) In various diseases of blood the viscosity of blood	
a) increases b) decreases c) remain constant d) none	
7) PH meter is composed of	
a) a reference electrode b) a glass electrode	
c) an electrometer d) all of these	
8) Most important buffer system regulating blood pH isbuffer	
a) Hb b) bicarbonate c) phosphate d) protein	
9)radioisotopes are frequently used for investigation of metabolic pathway	
a) 14c b) 14N c) 14P d)14ca	
10) is based on excitation of solid or solution	
a) scientilation counting b) GM counter c) autoradiography d) all of these	

Q2. Fill in the blanks and true false	[8]
 The bond angle between H and O atom in structure of water is Suspensoid are example of colloid. Formula of calomel is SI unit of radioactivity is 	3
True or false	
 5. The most important buffer system regulating blood PH is Hb b 6. In plants opening and closing of stomata is regulated by diffus 7. Volume receptor is sensitive for intravascular pressure. 8. Obligatory losses is approximatly200ml. 	ouffer. ion.
Q3. Answer in short. (Any ten)	[20]
 Define buffer with examples Define hydrogen bond Define diffusion. Define viscosity What is Tyndall effect? What is role of ADH? Define radioisotopes. Give reason structure of ware is V shape. Make flow chart for distribution of body water. What is function phosphate buffer? Define atomic and mass number. Define pH. 	
Q4.: Long answer questions. (any four) (8 marks each)	[32]
 Give an account on normal water balance Explain physiological buffer system. Explain classification of colloids. Explain biological significance of osmosis. Pure water dehydration. Explain Acid Base and derived H-H equation. Application of radioisotopes Explain GM counter 	
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