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SARDAR PATEL UNIVERSITY

Vallabh Vidyanagar

Examination : B. Sc. Instrumentation (Vocational) US05CINV04 – Analytical Instrumentation

		Monday, 18 th November, 2019, 10:00 AM – 1:00 PM	
		Total Mari	ks: 70
		Note: The figures to the right indicate maximum marks.	
Q-1.	(1)	Multiple Choice Questions- The variation of is proportional quantity to pH. (a) current (b) resistance (c) potential (d) inductance	[10]
	(2)	pH of water is 7 because number of hydrogen ions arehydroxyl ions. (a) greater than (b) less than (c) equal to (d) none	
	(3)	pH measurement is dependent. (a) flow (b) pressure (c) level (d) temperature	
	(4)	The name of chromatography is defined from its (a) two phases (b) stationary phase (c) three phases (d) mobile phase	
	(5)	Flame Ionization Detection is a type technique (a) mixing (b) destructive (c) non destructive (d) equalizing	
	(6)	In GC Column is kept in (a) Cylinder (b) Detector (c) Injector (d) Oven	
	(7)	If mobile phase is liquid and stationary phase is solid it ischromatography. (a) Partition (b) adsorption (c) thin film (d) paper	
	(8)	Response time of any detector should be (a) very long (b) very short (c) long (d) not required	
	(9)	RI detector depends on	
	(10)	Refractive Index and UV absorbance detector can achieve universal detecting capability. (a) True (b) False (c) both (d) none	
Q-2.		Short answer type (attempt any ten)	[20]
	(1)	Draw neat labeled diagram of Glass electrode.	
	(2)	Explain in brief Hydrogen Electrode.	
	(3)	Enlist the property of carrier gas in GC.	
	(4)	Why helical tube columns are used?	

	(5)	What do understand by Paper Chromatography?	
	(6)	What are the criteria for selection of Liquid chromatography?	
	(7)	Draw diagram of sample injector in Chromatography.	
	(8)	Enlist drawbacks of large diameter column?	
	(9)	What are the requirements for detector in chromatography?	
	(10)	List different types of Laser detectors.	
	(11)	Explain briefly electrical conductivity detector principle.	
	(12)	Give the examples of Bulk property and Solute property detectors.	
Q-3.	(a)	Draw neat diagram of Reference electrode and explain it in detail.	[6]
	(b)	Write brief note about buffer. OR	[4]
Q-3.	(a)	Explain Zero corrected dc Amplifier type pH meter.	[6]
	(b)	Write Nernst equation and list design consideration of pH meter.	[4]
Q-4.	(a)	Draw block diagram of Gas Chromatography and discuss its components in brief.	[6]
	(b)	Briefly explain types of column. OR	[4]
Q-4.	(a)	Describe working principle of Flame Ionization Detector with its limitations.	[6]
	(b)	Explain working principle of Electron Capture Detector (ECD).	[4]
Q-5.	(a)	Draw block diagram of HPLC system. Describe its flow measurement and control.	[7]
	(b)	Explain Syringe type pump. OR	[3]
Q-5.	(a)	Discuss different types of Liquid Chromatography in brief.	[7]
	(b)	Write a note on modes of gradient elution.	[3]
Q-6.	(a)	Explain working of Refractive Index detector.	[5]
	(b)	With neat figure explain Fluorescence Detector. OR	[5]
Q-6.	(a)	Discuss working of UV absorbance detector.	[5]
	(b)	Describe working of Mass detector.	[5]