SARDAR PATEL UNIVERSITY

B.Sc. Vth Semester Industrial Chemistry (Voc) (CBCS) Examination

Industrial Chemistry (Vocational)

Course No.: US05CICV03

Subject: Technology of Petroleum and Petroleum Products

Time: 10:00 A.M. To 01:00 P.M.

15th November 2019 Friday

Total Marks: 70

N.B. (1) Marks allotted to the question are on its RHS

(2) Illustrate your answers wherever necessary with the help of neat sketches & chemical equations	
Q. 1	Answer the following MCQ'S:	(10)
1	Kerosene are distillate fractions of crude oil in the boiling range of°C	
	(a) 50-150 (b) 150-250 (c) 150-300 (d) 100-250	
2	Process is used in crude oil for removal of Sulphur.	
	(a) Selexol (b) Sulfinol (c) Gribotol (d) None of these	
3	Organic theory explaining the formation of is also known as Angler's theory	
	(a) Clay (b) Mineral (c) Soil (d) Petroleum	
4	The manufacturing of HCN by catalytic combination of, and	
	in presence of pt- Rhodium alloy catalyst.	
	(a) Ethane, Ammonia, Air (b) Methane, Ammonia, Air	-
	(c) Ethane, Ammonia, CO ₂ (d) Methane, Ammonia, CO ₂	
5	is used as scrubbing agent in the production of HCN.	
	(a) CH_4 (b) CS_2 (c) HCL (d) H_2SO_4	
6	is produced by the reaction of methane with sulphur at high temperature.	
	(a) CO (b) S (c) CS ₂ (d) CO ₂ .	
7	is manufactured by hydrolysis of ethylene oxide.	
	(a) Propylene glycol (b) Butylene glycol (c) Neopentyl glycol (d) Ethylene glycol	
8	are the raw materials for manufacturing vinyl acetate.	
	(a) Acetylene / Acetic acid (b) Benzene / Ethylene	
	(c) Acetylene / Ethylene (d) Nitro benzene/ Methane	
9	is the ideal structure for butadiene rubber production.	
	Н Н H ₃ C CH ₂ н сн ₃	
. :	H.C.C.C.C.H	
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ιV	is the ideal structure for isobutene.	
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	(a) (b) $\overset{H}{H} \overset{H}{H} \overset{H}$	

Q-2	Answer any ten of the following:	(20)			
1	Why inorganic theory fails to explain formation of crude oil?				
2	Distinguish between Octane number and Cetane Number				
3	Sketch Labelled diagram of bubble cap tray.				
4	Write properties and uses of Methanol				
5	Out line that how are the important petro-chemicals obtained from methane.				
6	Write properties & uses of HCN.				
7	Write the synthesis reaction of glycerine by Acrolein route.				
8	Give the outline of chemical obtained from ethane.				
9	Write properties and uses of acrylonitrile.				
10	Give the outline how the various chemical obtained from butane.				
11	What are the limitations of molecular sieve as catalyst?				
12	Give the outline how the various chemical obtained from butenes.				
Q-3 a	Explain in detail theories for the formation of petroleum.	(05)			
þ	Enlist the composition of petroleum and explain it in brief.	(05)			
•	OR				
Q-3 a	Write a note on Construction & working of bubble cap tray.	(05)			
b	Define petroleum and write a note on kerosene and diesel.	(05)			
Q-4 a	With the help of flow diagram explain the manufacturing of HCN.	(05)			
b	With the help of flow diagram explain the manufacturing of CS ₂ .	(05)			
OR					
Q-4 a	Out line that how are the important petro-chemicals obtained from Ethylene.	(05)			
b	Write a short note on properties & uses of CS₂ and Methanol.	(05)			
Q-5 a	Explain manufacture and use of ethylene oxide.	(05)			
b	Explain manufacture and use of styrene from benzene.	(05)			
	OR				
Q-5 a	Explain the manufacture of acrylonitrile.	(05)			
b	Explain the manufacture of ethyl chloride.	(05)			
Q-6	Explain the manufacture of butadiene.	(10)			
OR					
Q-6	Write a note on linear alkyl benzene sulphonate.	(10)			
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