[108]

SARDAR PATEL UNIVERSITY 5th Semester B. Sc. (under CBCS) Examination-2020

Mon day, 25th December 2020 02.00 pm to 04:00 pm

Subject: PHYSICS (US05CPHY23)

(Thermodynamics And Statistical Mechanics)

Q.1	To answer the multiple choice questions choose the correct option					
	(1)	At absolute zero (a) increases	temperature, the (b) remains zero	entropy of a system (c) remains constant		[1
	(2)	The area under a	the curve TdS is e		(d) decreases	
	(3)	A reversible cyc	le has following p	rocesses.	atic (d) none of the above	
	(4)	Integral of dQ/T (a) S _i -S _f	of a reversible part (b) S _f -S _i	th is given by (c) S_i+S_f	(d) - S_{i} - S_{f}	
	(5)	Helmholtz funct (a) H = U + W	ion is given by (b) G = h – T S		(d) A = U - T S	
	(6)		cess occurs at Con (b) Pressure		(d) Temperature	
	(7)	Which of the foll canonical ensem	owing physical pa	rameters remain constant i		
			(b) [N, E, T]		(d) [Ε, V, μ]	
•	(8)	The Stirling form (a) $\ln \left(\frac{N}{e}\right)$	ula is ln N! = (b) N ln n		(d) N In N – N	
	(9)	Maxwell-Boltzmann law is for the				
		(c) Particles with half integral spin (d) Particles with integral spin				
	(10)	The Spin quantu	m number(s) of t	the is zero		
		(a) Photon	(b) Positron		(d) π - meson	
2	Fill in the Blanks and True-False State whether True or False					
	(1)	All Carnot engine	s operating betwee	n the same two reservoirs	have different efficiency	
	(2)	All Carnot engines operating between the same two reservoirs have different efficiency. During a reversible adiabatic process, the entropy of a system remains constant.				
	(3)	The triple point on a U-V-S surface is a plane triangle.				
	(4)			process the Gibbs function	n of the system not	

		in the blanks				
	. (5)	Phase space is a dimensional space.				
	(6)					
	(7)	In M-B System, the mean separation between the particles is than the thermal				
		length.				
	(8)	Maxwell-Boltzmann statistics cannot be applied to particle.				
Q.3	Answer briefly the following questions (Attempt any Ten)					
	(1) Define Refrigerator and Refrigerant.					
	(2)	Define Absolute zero.				
	(3)	Write two Ehrenfest's equations.				
	(4)	Show that for isotropic solid compressibility $k = 3\delta$.				
	(5)	Derive Energy equation $\left(\frac{\partial U}{\partial \nu}\right)_T = T \left(\frac{\partial P}{\partial T}\right)_V - P$.				
	(6) Give the difference between enthalpy and internal energy					
	(7)					
	(8)	Define: Phase Space and Phase point.				
	(9)	Define: Macroscopic States and Microscopic states				
	(10)	Obtain expression for entropy in the Maxwell – Boltzmann System				
	(11)	Define Degeneracy and Chemical Potential.				
	(12)	Explain Grand Canonical Ensemble.				
Q.4	Answer in detail of the following questions (Attempt any 4)					
	(1)	Obtaining Clausius' theorem derive the equation of second law of thermodynamics.	[32]			
	(2)	Derive Clausius Clapeyron's latent heat equation				
	(3)	Define entropy and obtain first and second T - dS equations				
	(4)	Explain Expansivity with proper diagram and figure.				
	(5)	State Liouville's theorem and prove that $\frac{d\rho}{dt} = 0$ using proper diagram				
	(6)	What is Gibbs Paradox in microcanonical ensemble? How it is removed				
	(7)	Define Bose- Einstein system Obtain the expression for Bose - Einstein				
		distribution of the particles among various states				
	(8)	Define Fermi - Dirac system obtain the expression for Fermi - Dirac distribution of				
-57		the particles among various states				

