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Sardar Patel University

B. Sc. (Semester – V) Examination

Date: 24-12-2020, Thursday

Time: 02:00pm to 04:00pm

Industrial Chemistry

COURSE NO: US05CICH01 (Organic Chemistry - II)

Notes: Figures to the right indicate full marks.

Total marks: 70

Q.1 Answer the following Multiple Choice Questions. (All are compulsory) (10)

- Pyridine undergoes nucleophilic substitution with NaNH_2 at 100°C to form
 - 2-Aminopyridine
 - 3-Aminopyridine
 - 4-Aminopyridine
 - None of these
- Which of the following reagents will react with pyrrole to form 2-formylpyrrole
 - HCOOH
 - CHCl_3/KOH
 - H_2O_2
 - $(\text{CH}_3\text{CO})_2\text{O}/\text{SnCl}_4$
- Furan reacts with ammonia in the presence of alumina at 400°C to give
 - Pyridine
 - Furfural
 - Pyrrole
 - Furoic acid
- Naphthalene undergoes nitration with $\text{HNO}_3/\text{H}_2\text{SO}_4$ at 60°C to give mainly
 - 1-Nitronaphthalene
 - 1,2-Dinitronaphthalene
 - 2-Nitronaphthalene
 - 1,5-Dinitronaphthalene
- All carbon atoms in Anthracene are...
 - sp hybridized
 - sp^2 hybridized
 - sp^3 hybridized
 - None of these
- Sodium borohydride is an important _____ reagent.
 - Bromination
 - Oxidizing
 - Reducing
 - Methylating
- Aldehyde having _____ undergo Aldol condensation
 - α -hydrogen
 - β -hydrogen
 - γ -hydrogen
 - δ -hydrogen
- _____ compounds doesn't undergo Aldol condensation reaction.
 - CH_3CHO
 - HCHO
 - $\text{CH}_3\text{CH}_2\text{CHO}$
 - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$
- Signal pattern of the CH_3 protons in the NMR spectra of the $\text{CH}_3\text{CH}_2\text{Br}_2$ and CH_3COOH is _____.
 - Triplet & Singlet
 - Doublet & Singlet
 - Triplet & Doublet
 - None of them
- Number of signals (ignoring the splitting patterns) would you see in the NMR spectra of the *p*-xylene and 2-Propanol is _____.
 - Two & Three
 - Three & Two
 - Three & Four
 - None of them

[1]

Q.2 Answer the following (08)

1. Thiophene is five membered rings which is most resonance stabilized among others five-member heterocyclic ring of similar class. True / False?
2. 2,5-dibromothiophene is the main product when thiophene reacts with Br₂ in benzene. True / False?
3. The Friedel–Crafts reaction of Naphthalene with Succinic anhydride using Nitrobenzene as a solvent give.....as a main product.
4. A reaction of Anthracene with Acetyl chloride using AlCl₃ and Nitrobenzene at lower temperature give.....as a main product.
5. Heterolytic cleavage of a carbon-carbon bond produces “Two carbonium ions” True / False?
6. “Methyl” carbocation has the least stability than “Tert-butyl”. True / False?
7. Infrared absorption requires a dipole change for the vibration, AND sufficient energy to promote the molecule to a new vibrational quantum state. True / False?
8. N₂ and O₂ molecules will show no absorption of IR radiations. True / False?

Q.3 Answer the following short questions (Any Ten) (20)

1. Give synthesis of Pyrrole.
2. Outline the rules for naming mono & di heterocyclic compound.
3. Write order of relative basicity of RCH₂NH₂ and RC≡N.
4. Write a synthesis of β-Naphthol.
5. Give resonating structures of Anthracene.
6. Write a reaction of Naphthalene with conc. sulphuric acid at 165°C?
7. Give synthesis of Aluminum isopropoxide.
8. Write preparation and properties of Osmium tetroxide.
9. Define a term rearrangement.
10. Write information obtained from H¹NMR-spectroscopy.
11. The NMR spectrum of compound C₂H₆O, shows one signal only, a singlet. Deduce the structure of it.
12. Giving a formula for calculation of DBE, calculated a value for MF C₇H₄N₂

Q.4. Long Answer Questions. (Attempt any 4 out of 8) (32)

1. Discuss properties of Pyridine and its constitution.
2. Electrophilic substitution in Thiophene.
3. Nucleophilic substitution in Pyridine.
4. Write notes on Electrophilic substitution reaction in Naphthalene.
5. Write preparation, properties and uses of Aluminum isopropoxide reagents.
6. Write preparation, properties and uses of SeO₂ reagents.
7. Write notes on Pinacol–Pinacolone and Benzilic Acid Rearrangement.
8. Write the principle of IR spectroscopy and discuss the applications of IR-Spectroscopy.