56

(A-8) Seat NO:

Printed pages: 2

SARDAR PATEL UNIVERSITY T.Y. B.Sc. (CHEMISTRY), SEMESTER-5 (NC) INORGANIC CHEMISTRY (US05CCHE03) EXTERNAL THEORY EXAMINATION

Tim NB:	e: 3 Hrs. 10:30 aw Figures to the right	1-1-30pm t indicate marks Thus	sday, 12th May 2016	Total Marks : 70
Q:1. Choose an appropriate answer for each of the following. [10]				
	. The symbol σ_d use			
		(b) horizontal plane	(c) molecular plan	e (d) none of these
		ation reflection axis is		
	(a) C _n	X-2-11	(c) D _n	
	3. The energy gap be	etween t2 & e-sets of orbita		
	(a) $4/9 \Delta_0$		(c) both (a) & (b)	(d) none of this
	4. The electronic dis	tribution in $[Co(NO_2)_6]^{3}$ - co	omplex is?	2 2
	(a) t_{2g}^{6} e_{g}^{0}	(b) $t_{2g}^{6} e_{g}^{2}$	(c) \dot{e}_{g}^{4} t_{2g}^{2}	(d) $t_{2g}^{3} e_{g}^{3}$
	5. The d-orbital split	into degenerate or		
	(a) 2	(b) 3	(c) 5	(d) 10
		ntum number has originate		
	(a) θ equation	(b)	(c) R equation	(d) none of this
	7. What will be the degeneracy of a state characterized by $n_x^2 + n_y^2 + n_z^2 = 19$?			
	(a) 6	(b) 19		(d) 3
	8. Generally,c	o-ordinated complexes rea	ct faster than co-	ordinated complexes.
		(b) 4, 6		(d) none of these
		ction, ligand replaced by_		(D) 0.1
		(b) OH		(d) none of these
		and have strongest trans eff		/ IN 11 / O
	()		(c) NH ₃	(d) H ₂ O
Q:2. Answer the following [Any Ten] [20]				
	1) Define: Symmetry operation, Symmetry element			
	2) Give the multiplication table for point group C_2v .			
	3) Find the point group of H ₂ molecule.			
	4) Draw the diagram of splitting of d-orbital in Oh complex.			
	5) Arrange the ligands in increasing order in series based on their capacity to cause the			
	spiliting of d-orbitals: CN ⁻ , NH ₃ , NO ₂ ⁻ , F ⁻ , H ₂ O, Γ.			
	6) Find the term syn		aniaa .	
		postulate of quantum mech	anics.	
	8) What is a linear operator?			
	9) Write the Schrödinger equation in terms of polar co-ordinates.			
	10) What is the relation between β _n and K ₁ , K ₂ ,, K _n ?			
		s of attacking reagent.		
	12) What is Trans ef	rect?		

Q:3. (A) Explain the following by giving suitable illustrations: Rotation, Improper rotation. [64] (B) Identify the symmetry elements and detect the point groups of the following molecule [66] [Any two] (I) $[PtCl_4]^{2-}$ (II) (III) Cyclopropane OR Q:3. (A) Explain: Two & Three fold rotational axis present in pyridine and ammonia molecule respectively. (B) Give an account of C_nv point groups. Prove that C₃v point group is Non Abelian. [06] Q:4. (A) Differentiate: Paramagnetism & Diamagnetism. [04](B) Discuss about splitting of d-orbital in tetragonally distorted octahedral complex & square planner complexes. [06]OR Q:4. (A) Write short note on 'Jahn Teller effect'. [04](B) Calculate the LFSE of (I) Complex $[Co(H_2O)_6]^{+2}$ which is high spin complex and the value of P=22,500 cm⁻¹, $\Delta_0 = 9.300 \text{ cm}^{-1}$. (II) Complex $[Mn(CN)_6]^{-3}$ which is low spin complex and the value of P=28,000 cm⁻¹, $\Delta_0 = 38,500 \text{ cm}^{-1}$. Q:5. (A) Derive a three dimensional wave equation for a wave travelling in Y-direction and describe as Y(x,t)=f(x). $\Phi(t)$. [10]OR Q:5. (A) Give an account of particle in one dimensional box. [10]Q:6. (A) Discuss: Factor affecting the stability of complexes depends on nature of central metal ion. [04](B) Write about $S_N 1 \& S_N 2$ mechanism. [06]Q:6. (A) Write short note on 'Electrostatic polarization theory'. [04](B) Discuss about base hydrolysis reactions of six-coordinated Co(III) Ammine complexes. [06]

