	B. Sc. (Biochemistry) – Fifth Semester Examination (CBCS)  Tuesday, 7 <sup>th</sup> November 2017  10:00 a.m. to 1:00 p.m.	
	US05CBCH01: Molecular Biology - I	70
Note	Total Marks: 7  (1) Figures to the right indicate marks.  (2) Draw a neat and labeled diagram, wherever necessary.	• .
Q. 1	Choose the most appropriate answer from the four alternatives given:	[10]
i.		1.3
	(a) Oncogenes (b) Mobile genes (c) Pseudo genes (d) t - RNA genes	
ii.	Histones are smallproteins.	
	(a) Acidic (b) Basic (c) Neutral (d) None of the above	٠.
iii.	The shape of human mitochondrial DNA is	
4 N. F	(a) Single stranded linear (b) Supercoiled	V 173
<u> </u>	(c) Circular (d) Double stranded linear	7483
iv.	enzyme is responsible for $5' \rightarrow 3'$ exonuclease activity?	
ķ.	(a) DNA polymerase I (b) DNA polymerase II	
,A.V	(c) DNA polymerase III (d) DNA polymerase IV	ŧ
v.	DNA replicates during phase of cell cycle.	
2.5	(a) G0 (b) G1 (c) G2 (d) S	
· vi.	is the DNA sequence to which RNA polymerase binds to initiate transcription	1.
	(a) Enhancer (b) Operator (c) Promoter (d) Repressor	•
vii.	Which of the following introns are largest one?	
÷	(a) Group I (b) Group II (c) Group III (d) Both (a) and (b)	* 5
viii.	Wobble hypothesis state that	
* * * * * * * * * * * * * * * * * * *	(a) Single t-RNA can decode more than one codon for m-RNA	
; ,	(b) Single t-RNA can decode more than one codon for r-RNA	1
	(c) Single t-RNA can decode more than one codon for 28s r-RNA	
5	(d) Single m-RNA can decode more than one codon for t-RNA	
ix.	is not a nonsense codon.	* \$
	(a) AUG (b) UAG (c) UGA (d) UAA	1
52 ° <b>X.</b>	Which of the following is an example of ribozyme?	
	(a) RNA polymerase (b) Peptidyl transferase (c) Aminoacyl transferase (d) RNAse H	
: E		/ :

Q.2	Answer any <u>TEN</u> from the following:	[20]
i	. What are minichromosomes?	[20]
ii	Define the terms Gene and genome.	
iii	The state of the s	
iv		
, <b>, v</b>	What are the proteins needed to initiate replication at origin of E. coli?	
vi.		
vii.		
viii.	What are polycistronic and monocistronic m-RNA?	· "
ix.	Write differences between group-I and group-II introns splicing.	
х.	Justify that "ribosomes are complex molecular machine".	
xi,	"Transfer RNA acts as an adapter for protein synthesis" - Explain.	ŧ
xii.	Enlist characteristics of genetic codes.	
	the only transfer to the majority of a property and a	
Q.3 (a)	Explain the concept of gene.	[4]
(b)	Describe salient features of eukaryotic genomes.	[6]
	att Marings of <u>OR</u> and a second of the second of second	A .
Q.3 (a)	"Chromatin is chemically compound proteins"- Explain.	[4]
(b)	Justify that "Nucleosome is a organization unit of chromatin".	[6]
	The state of the s	~ -
Q.4 (a)	Write a note on Okazaki fragments.	[5]
(b)	Explain D - Loop mode of replication.	[5]
•	$\mathbf{OR}$ . The second of the second of the second of the second of $\mathbf{OR}$ . The second of $\mathbf{OR}$	. 1
Q.4 (a)	Describe Meselson-Stahl experiment.	[5]
(b)	Explain termination of DNA replication.	[5]
	e de la companya de La companya de la companya del companya del companya de la companya	414.3
Q.5 (a)	Explain the splicing of m-RNA.	[5]
(p)	Explain the mechanism of elongation of transcription in eukaryotes.	[5]
	$\overline{\mathbf{OR}}$ . The state of the	
Q.5 (a)	Describe reverse transcription.	[5]
(b)	Describe generation of 5' cap in eukaryotic m-RNA.	[5]
	多种。1500年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年,1940年	
Q.6	Explain the inhibitors of proteins synthesis in detail.	[10]
	and the state of t	- •
Q.6	Explain the process of initiation of proteins synthesis in detail.	[10]
	*****	
	[Page 2]	

# No. of SARDAR PATEL UNIVERSITY

## T.Y.BSc 5<sup>TH</sup> SEMESTER EXAMINATION NOVEMBER 2017

**BIOCHEMISTRY: USO5CBCH02** 

Title: ENZYMOLOGY

Q.1 Select proper option from following MCQ.  1. D- amino oxidase acts only on D- amino acid is the example of specificity.  a) Stereochemical b) Dual c) Linkage d) Group  2. Co-enzymes are required for following group of enzymes except  a) Decarboxylase b) Dehydrogenase c) Hydrolases d) Transaminase  3 of the following is not an activator for any enzyme. a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl  4. Enzyme bound to chromatin is a) Nucleosite triphosphatase b) Malate dehydrogenase c) Arginase d) Topoisomerase  5. Enzyme located in inner membrane of mitochondria is a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase  6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration  7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	Date:	9/11/17; Thursday	Time: 10:00 AM T	O 1:00 PM	TOTAL MARI	KS: 70
specificity.  a) Stereochemical b) Dual c) Linkage d) Group  2. Co-enzymes are required for following group of enzymes except  a) Decarboxylase b) Dehydrogenase c) Hydrolases d) Transaminase  3 of the following is not an activator for any enzyme. a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl  4. Enzyme bound to chromatin is a) Nucleosite triphosphatase b) Malate dehydrogenase c) Arginase d) Topoisomerase  5. Enzyme located in inner membrane of mitochondria is a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase  6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration  7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	Q.1 S	elect proper option	rom following MC	CQ.		[10]
specificity.  a) Stereochemical b) Dual c) Linkage d) Group  2. Co-enzymes are required for following group of enzymes except  a) Decarboxylase b) Dehydrogenase c) Hydrolases d) Transaminase  3 of the following is not an activator for any enzyme. a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl  4. Enzyme bound to chromatin is a) Nucleosite triphosphatase b) Malate dehydrogenase c) Arginase d) Topoisomerase  5. Enzyme located in inner membrane of mitochondria is a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase  6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration  7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	1.	D- amino oxidase ac	s only on D- amino	acid is the ex	cample of	
2. Co-enzymes are required for following group of enzymes except					•	· · · · · · · · · · · · · · · · · · ·
2. Co-enzymes are required for following group of enzymes except		a) Stereochemical	b) Dual	c) Linkage	d) Group	ł
a) Decarboxylase c) Hydrolases d) Transaminase  3 of the following is not an activator for any enzyme. a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl  4. Enzyme bound to chromatin is a) Nucleosite triphosphatase b) Malate dehydrogenase c) Arginase b) Malate dehydrogenase c) Arginase b) Hexokinase c) Protein kinase b) Hexokinase c) Protein kinase d) Collagenase 6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration 7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	2.			-		
3 of the following is not an activator for any enzyme.  a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl <sup>-</sup> 4. Enzyme bound to chromatin is a) Nucleosite triphosphatase b) Malate dehydrogenase c) Arginase d) Topoisomerase  5. Enzyme located in inner membrane of mitochondria is a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase  6. Which of the following method is based on change in solubility?  a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration  7. Which of the following is the feature of competitive inhibition?  a) Km is not changed b) Km is increased						
a) Mn <sup>++</sup> b) Mg <sup>++</sup> c) CN d) Cl  4. Enzyme bound to chromatin is		c) Hydrolases		d) Transami	nase	
<ul> <li>4. Enzyme bound to chromatin is</li></ul>	3.	of the	following is not an	activator for a	ny enzyme.	
<ul> <li>4. Enzyme bound to chromatin is</li></ul>		a) Mn <sup>++</sup> b) M	g <sup>++</sup> c) CN <sup>-</sup>	d) C	<b>1</b>	
a) Nucleosite triphosphatase c) Arginase d) Topoisomerase 5. Enzyme located in inner membrane of mitochondria is a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase 6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant b) Isoelectric focusing c) Iron exchange chromatography d) Ultra filtration 7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	4.					
c) Arginase  5. Enzyme located in inner membrane of mitochondria is		a) Nucleosite triphos				
5. Enzyme located in inner membrane of mitochondria is  a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase 6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant c) Iron exchange chromatography d) Ultra filtration 7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	33 .					
a) RNA polymerase b) Hexokinase c) Protein kinase d) Collagenase 6. Which of the following method is based on change in solubility? a) Decrease in dielectric constant c) Iron exchange chromatography d) Ultra filtration 7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	5.	Enzyme located in in	ner membrane of n	• •		
<ul> <li>6. Which of the following method is based on change in solubility?</li> <li>a) Decrease in dielectric constant</li> <li>b) Isoelectric focusing</li> <li>c) Iron exchange chromatography</li> <li>d) Ultra filtration</li> <li>7. Which of the following is the feature of competitive inhibition?</li> <li>a) Km is not changed</li> <li>b) Km is increased</li> </ul>			•			
<ul> <li>a) Decrease in dielectric constant</li> <li>b) Isoelectric focusing</li> <li>c) Iron exchange chromatography</li> <li>d) Ultra filtration</li> <li>7. Which of the following is the feature of competitive inhibition?</li> <li>a) Km is not changed</li> <li>b) Km is increased</li> </ul>	3.3	c) Protein kinase		d) Collagen	ase	
c) Iron exchange chromatography d) Ultra filtration 7. Which of the following is the feature of competitive inhibition? a) Km is not changed b) Km is increased	6.	Which of the followi	ng method is based	on change in	solubility?	
7. Which of the following is the feature of competitive inhibition?  a) Km is not changed  b) Km is increased		a) Decrease in dielec	tric constant	b) Isoelectr	ic focusing	
a) Km is not changed b) Km is increased	• 1			•	-	
	7.	Which of the following	ng is the feature of	competitive in	nhibition?	1 3
a) V m in decreased		a) Km is not changed		b) Km is in	creased	
c) Km is decreased d) V <sub>max</sub> is reduced		c) Km is decreased		d) V <sub>max</sub> is r	educed	:
8. What is feature of allosteric inhibition?	8.	What is feature of all	osteric inhibition?		1. S. J. F. 44 (1964)	
a) Km is not changed b) It is partially reversible	er i De	a) Km is not changed		b) It is parti	ally reversible	
c) It is not reversible d) Km is changed		c) It is not reversible		d) Km is ch	anged	
9enzyme is elevated in both obstructive jaundice and bone disease.	9.	enzyme is e	evated in both obs	tructive jaundi	ce and bone dise	ease.
a) SGOT b) SGPT c) Alkaline phosphatase d) Nucleotidase	24:					
10. All the enzymes are associated with myocardium except	10.	All the enzymes are	associated with my	ocardium exce	pt <u>* .:- * * * * * * * * * * * * * * * * * * *</u>	<u></u> •
a) SGOT b) LDH c) MB-CPK d) MM-CPK		a) SGOT b) LDI	e) MB-C	PK d	) MM-CPK	•
Q.2 Answer the following in short (Any ten) [20]	Q.2 A	Answer the following	in short (Any ten)	· )		[20]
1. Write two examples of co-enzymes involved in oxido-reductases.	1.	Write two examples	of co-enzymes invo	olved in oxido	-reductases	
2. List the factors affecting enzyme reaction?					-10440(4363,	
				/AA+		

3. Define activators and inhibitors		e e e	4.
4. List enzymes involved in lipid	metabolism.		
5. Enlist the enzymes involved in	nucleic acid syntl	nesis.	
6. How Change in pH is useful fo	r isolation of enzy	me.	
7. What are the features of non-co	ompetitive inhibiti	on?	
8. Define zero-order and first-order			
9. Write example for random mec	hanism.		
10. Write the role of saccharomyce	s species in wine	making and brewing.	
<ul><li>11. Write enzymatic method for the</li><li>12. Write enzymatic reaction which</li></ul>	e esumation of blo	ood glucose.	
	i is used for measi	urements of uric acid.	
Q.3 Write short note on:		g de la companya de	
a. Unit activity of enzyme.			
b. Enzyme specificity.		.01	
	OR	e september	•
a. Metaloenzymes.			
b. Co-factors.			
Q.4 Write short note on:		a Temporto de la M <sup>A</sup>	
a. Ultra centrifugation.	and the second s	e same e de la companya de la compa La companya de la co	
b. Ion exchange chromatography.		Harry M. C.	<u> </u>
or for exertainge enromatography.	OR	r og skiper for kalle i Med Skiper for kallet i Medisk	•
a. Electrophoresis.	OK	en e	
<b>b.</b> Gel filtration.	:		
O 5 Evoloin in detail about			[
Q.5 Explain in detail about uncompeti	tive inhibition.  OR		[
Q.5 Explain in detail about non-compe	etitive inhibition.		[
Q.6 Explain			١
a. Immobilization of enzymes.			[
b. Alanine transaminase.	_	and the second of the second	
Q.6 Write role of:	OR	and the second s	
a. Enzyme in treatment of cancer.			
b. Enzyme in treatment of genetic	deficiency disce-		
		the state of the s	[
BE:	ST OF LUCK		
	2		
AND THE STATE OF T		and the same of	

\ 1 T	۳.	٠	•	•	, ,	1.	*	,,,,
[3]/A9]		S	EAT :	No	******	- 11 may	, saț	
	100	1 - 1	11.5		1100	hari	Жİ	10
	B	San A.	<b>1</b>					

Q-1.

#### SARDAR PATEL UNIVERSITY

No. of Printed Pages ! 2

External examination—oct-2017

Biochemistry-V semester
paper code no. US05CBCH04
CELL BIOLOGY

Date:	Monday	Time: 10:0am -	L:OUPM	Total marks:70	
nsw	er the followings	ı <b>.</b>	#* v	en e	10
1.				le cells are ad actin D] α- and β-tubuli	
2.	proto fila	aments are bundled to	form an intermed	liate filament edita: 1944	•
	A]Ten b]	Eight c] one	d] two	era en	
3,	The end symbio	otic origin of mitocho	ondria is establish	ed based on	
	A] DNA sequ	ence data & other c	haracteristics.	b] molecular sequence dat	<b>a</b> *
4.	•	equence data and oth oplasmic protein con		d] DNA sequence data n the removal of denatured pol	
	A] Microbodi	es B] Proteasomes	C] Kinetosom	e D] Basal Granule	
5.	The position of	the centriole determi	nes the position of	fthe	
6.	A] Moving M	process forolecules "Uphili"	B] Pushing	Substances Out Of The Cell Substances Into The Cell	849821 25 Teach
7.				er nya Marin e Paran da yagi	
8.	A]intra water Proteosom are i	b] excess water nvolved in the remov	c] intra waste	d] intra protein	y (* 20
	A] Micro bodi	es B] denatured p	olypeptides C]	Kinetosome D] Basal Gran	ule
9.	A genetically-en	ngineered ONYX-01	5 lacking p53 is k	nown as	en e
	A] adenovirus	B] chaperone	C] securin	D]cohesion	¥*
10.	Separation of th	e sister chromatids re	equires	o digitari Maringan di managan di Managan di managan di	
	A] separase	B] chaperone	C] securin	D]cohesion	
		(I	) was a street of the		ر ه ۲

Q.2 Answer in short. (Attempt any ten)	<b>CH04</b> 20
1. Explain changes during metaphase of cell division.	
2. List different stages of meiosis	<b>V</b>
3. Give meaning for endo replication.	
4. How G-actin is recycled	
5. Write on difference between polymerization of intermediate	filaments and actin
6. Write Definition of eukaryotes and prokaryotes	diametrics diametria
7 Write on occurrence of Zonula occludes	:4998
8 Define Autophagy.	
9 Explain-how microtubules serve as the generating force for t	entre que entre
10 Write on forms of Endocytosis	
11 Enlist utilization of lysosomal enzymes	
12 Write importance and Examples of intermediate filaments.	elit Miller and Alberta
the section of the se	
LONG QUESTIONS 40 ma	and the second s
and a grapper of 200 de la latera	'KS Mary de la mente ( )
A Classify and explain role of cell junctions	algerga and the extra
	vel <b>ic</b> elo (1877)
B] Write on theory of Eukaryote Evolution OR	n desta New House discovery (List )
A] Differentiate between animal and plant cell	• •
Type of the control of the control of the control of the control of	ing cali day yangil — in idaN yangabb gi Nacasanandil da
A] Explain-Axonemal Microtubules Are Dynamic and Stable	ेत्रका प्रमुखीय हो।
Explain meaning and functions of centrioles.	our report Hoperny (1
$\mathcal{L}_{\mathrm{const}}(\mathbf{O},\mathbf{R}_{\mathrm{const}})$ , the property of $\mathcal{L}_{\mathrm{const}}(\mathbf{R}_{\mathrm{const}})$	Const. Const.
Write importance for F-Actin filaments	Tyrn hat hexyptonic 6
Write on Basal bodies platforms for building cilia	e light years flavo
] write on types of ER and explain structure and functions.	engera e ji ji jirma d
B] Write note on Golgi apparatus  OR	· ·
A] Write on – proteasomes	e all of passes as
B]write note on mitochondria.	j garaga (j
	·
Explain - The stages for mitosis with neat diagram	
OR	
rite detail account on cell cycle and explain events of each phases.	

, i

Q..3

Q..3

Q.4

Q.4

Q.5

Q.5

Q.6

Q.6

No. of a tagen

SARDAR PATEL UNIVERSITY V.V. Nagar-388 120

B.Sc. Biochemistry (V- SEMESTER)

Subject: HUMAN PHYSIOLOGY AND CLINICAL ENDOCRINOLOGY

Subject Code: US05CBCH05

DATE: 15/11/2017

Time: IDam to Loppen and distant

Q.1 Multiple Choice questions: (1 Mark each)  1. The following element exhibit down regulation of aging process  a. Sodium  b. Sulfur  d. Chlorine  d. Chlorine  d. Chlorine  2. Which of the following element use as emergency therapeutic in heart attack  a. Mn  b. Mg  d. K  3. Which of the following is consider as table salt  a. Na  b. NaCl  c. Cl  d. None of Above  4. How many peptides present in angiotensin II  a. 5  b. 8  c. 10  d. 2  5. PO2 in inspired air is  a. 20 mmHg  b. 70 mmHg  c. 30 mmHg  d. 158 mmHg  6. The following are the parts of respiratory tract except  a. Alveolarduct  b. Larynx  d. Trachea  7. Myosin are the example of  a. Structure protein  b. Regulatory protein  c. Motor protein  d. Sarcoplasmic protein  d. Sarcoplasmic protein  e. Sorbitol  d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  c. Adrenal gland  d. Pancreas	700	11. 13/11/2017	to 1.00 Pm TOTAL M	ARKS: 70
a. Sodium b. Sulfur c. Calcium c. Calcium d. Chlorine a. Mn c. Na b. Mg d. K  Which of the following element use as emergency therapeutic in heart attack a. Mn c. Na d. K  Which of the following is consider as table salt a. Na b. NaCl c. Cl d. None of Above  How many peptides present in angiotensin II a. 5 b. 8 c. 10 b. 8 c. 10 c. 10 d. 2  SommHg d. 158 mmHg d. 15	A 55.50	The Choice Illegione . /1 A/L	the and amountain and mucal arrests the	
b. Sulfur  c. Calcium d. Chlorine  d. Chlorine  a. Mn c. Na d. K  3. Which of the following is consider as table salt a. Na b. NaCl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein b. Regulatory protein c. Motor protein b. Cortisol c. Sorbitol d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as a. Pituitary gland b. Liver d. Answer in very short (Apy Tex)	1	The following element exhibit down as	rk each)	10
b. Sulfur  c. Calcium d. Chlorine  d. Chlorine  a. Mn c. Na d. K  3. Which of the following is consider as table salt a. Na b. NaCl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein b. Regulatory protein c. Motor protein b. Cortisol c. Sorbitol d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as a. Pituitary gland b. Liver d. Answer in very short (Apy Tex)		a. Sodium	fulation of aging process	
2. Which of the following element use as emergency therapeutic in heart attack a. Mn b. Mg c. Na d. K  3. Which of the following is consider as table salt a. Na b. NaCl c. Cl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 c. 10 d. 2  5. PO <sub>2</sub> in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein 8. Which of the following can cause neuropathy in hyperglycaemic condition a. Manitol b. Cortisol c. Sorbitol d. Barbitol 9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin d. Macula Densa a. Pituitary gland b. Liver d. Answer in very short (Any Tex)				
b. Mg d. K 3. Which of the following is consider as table salt a. Na b. NaCl c. Cl d. None of Above 4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 d. 2  5. PO2 in inspired air is a. 20 mmHg b. 70 mmHg d. 158 mmHg 6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein d. Sarcoplasmic protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein d. Sarcoplasmic protein c. Sorbitol d. Barbitol 9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin d. Macula Densa a. Pituitary gland b. Liver d. Pancreas	2	Which of the following of	d. Chlorine laying administrate (a	
b. Mg d. K 3. Which of the following is consider as table salt a. Na b. NaCl c. Cl d. None of Above 4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 d. 2  5. PO2 in inspired air is a. 20 mmHg b. 70 mmHg d. 158 mmHg 6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein d. Sarcoplasmic protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein d. Sarcoplasmic protein c. Sorbitol d. Barbitol 9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin d. Macula Densa a. Pituitary gland b. Liver d. Pancreas		a. Mn	mergency therapeutic in heart attack	
a. Na b. NaCl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 d. 2  5. PO2 in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein b. Regulatory protein d. Sarcoplasmic protein b. Regulatory protein c. Sorbitol d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Adrenal gland b. Liver d. Pancreas		h Ma	c. Na	
a. Na b. NaCl c. Cl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 5. PO <sub>2</sub> in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg 6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein b. Regulatory protein c. Sorbitol d. Barbitol 9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Adrenal gland b. Liver d. Pancreas	3	Which of the follows:	d. K	
b. NaCl b. NaCl c. Cl d. None of Above  4. How many peptides present in angiotensin II a. 5 b. 8 c. 10 d. 2  5. PO <sub>2</sub> in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein b. Regulatory protein c. Sorbitol d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa  10. LH is released from a. Pituitary gland b. Liver c. Cl d. None of Above d. None of Above d. 2  c. 30 mmHg d. 158 mmHg d. 158 mmHg d. 158 mmHg d. 158 mmHg d. Sarcoplasmic protein d. Sarcoplasmic protein d. Sarcoplasmic protein d. Sarcoplasmic protein d. Barbitol d. Barbitol d. Barbitol d. Barbitol d. Macula Densa c. Neurone d. Macula Densa d. Pancreas		of the following is consider as tak	ole salt	
a. 5 b. 8 c. 10 d. 2  5. PO <sub>2</sub> in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein b. Regulatory protein c. Sorbitol d. Barbitol g. Manitol b. Cortisol c. Sorbitol d. Barbitol g. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa c. Adrenal gland d. Pancreas		u. Iva	C1	
a. 5 b. 8 c. 10 d. 2  5. PO <sub>2</sub> in inspired air is a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg  6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein b. Regulatory protein c. Sorbitol d. Barbitol g. Manitol b. Cortisol c. Sorbitol d. Barbitol g. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa c. Adrenal gland d. Pancreas	4	How many and I	d. None of Above	10
b. 8  5. PO <sub>2</sub> in inspired air is  a. 20 mmHg  b. 70 mmHg  c. 30 mmHg  d. 158 mmHg  6. The following are the parts of respiratory tract except  a. Alveolarduct  b. Larynx  7. Myosin are the example of  a. Structure protein  b. Regulatory protein  c. Motor protein  d. Sarcoplasmic protein  e. Motor protein  d. Sarcoplasmic protein  e. Motor protein  d. Sarcoplasmic protein  e. Sorbitol  d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  c. Adrenal gland  d. Pancreas	121	namy peptides present in angiotensi	n II Chara on more at health to	
5. PO <sub>2</sub> in inspired air is  a. 20 mmHg  b. 70 mmHg  c. 30 mmHg  d. 158 mmHg  6. The following are the parts of respiratory tract except  a. Alveolarduct  b. Larynx  d. Trachea  7. Myosin are the example of  a. Structure protein  b. Regulatory protein  c. Motor protein  d. Sarcoplasmic protein  a. Manitol  c. Sorbitol  d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  d. 2  Answer in very short (Any Tar)				
a. 20 mmHg b. 70 mmHg c. 30 mmHg d. 158 mmHg 6. The following are the parts of respiratory tract except a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein e. Manitol b. Cortisol c. Sorbitol d. Barbitol 9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa  10. LH is released from a. Pituitary gland b. Liver c. Adrenal gland d. Pancreas	5		774-71-71-11-11-11-11-11-11-11-11-11-11-11-	
d. 158 mmHg  a. Alveolarduct b. Larynx c. Pharynx d. Trachea  a. Structure protein b. Regulatory protein d. Sarcoplasmic protein a. Manitol b. Cortisol c. Sorbitol d. Barbitol  In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa  c. Adrenal gland d. Pancreas		PO <sub>2</sub> in inspired air is		
d. 158 mmHg  a. Alveolarduct b. Larynx c. Pharynx d. Trachea  a. Structure protein b. Regulatory protein d. Sarcoplasmic protein a. Manitol b. Cortisol c. Sorbitol d. Barbitol  In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa  c. Adrenal gland d. Pancreas		a. 20 mmHg	c 30 mmHa	
a. Alveolarduct b. Larynx c. Pharynx d. Trachea  7. Myosin are the example of a. Structure protein b. Regulatory protein c. Motor protein d. Sarcoplasmic protein a. Manitol b. Cortisol c. Sorbitol d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa  10. LH is released from a. Pituitary gland b. Liver  2. Answer in very short (Any Tex)				
b. Larynx  7. Myosin are the example of  a. Structure protein  b. Regulatory protein  c. Motor protein  d. Sarcoplasmic protein  d. Sarcoplasmic protein  a. Manitol  b. Cortisol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  c. Pharynx  d. Trachea  c. Motor protein  d. Sarcoplasmic protein  d. Barbitol  d. Barbitol  d. Barbitol  d. Macula Densa  c. Neurone  d. Macula Densa  c. Adrenal gland  d. Pancreas	0.	The following are the parts of respirators	tract except	
b. Larynx  7. Myosin are the example of  a. Structure protein  b. Regulatory protein  c. Motor protein  d. Sarcoplasmic protein  a. Manitol  b. Cortisol  c. Sorbitol  d. Barbitol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  c. Adrenal gland  d. Pancreas		a. Mycolatuuci	c Phones	
b. Regulatory protein  d. Sarcoplasmic protein  c. Sorbitol  d. Barbitol  a. Nephron  b. Nebulin  c. Neurone  d. Macula Densa  c. Adrenal gland  d. Pancreas  d. Pancreas		b. Larynx		
b. Regulatory protein  d. Sarcoplasmic protein  c. Sorbitol  d. Barbitol  a. Nephron  b. Nebulin  c. Neurone  d. Macula Densa  c. Adrenal gland  d. Pancreas  d. Pancreas	7.	Myosin are the example of	d. Tracnea	
8. Which of the following can cause neuropathy in hyperglycaemic condition  a. Manitol  b. Cortisol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  c. Adrenal gland  d. Pancreas  c. Adrenal gland  d. Pancreas		a. Structure protein		
8. Which of the following can cause neuropathy in hyperglycaemic condition  a. Manitol  b. Cortisol  9. In human kidneys, functional unit of kidney is also classified as  a. Nephron  b. Nebulin  10. LH is released from  a. Pituitary gland  b. Liver  c. Adrenal gland d. Pancreas		b. Regulatory protein	c. Motor protein	0.0
9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa d. Macula Densa c. Adrenal gland d. Pancreas	8.	Which of the following can cause now	d. Sarcoplasmic protein	
9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa d. Macula Densa c. Adrenal gland d. Pancreas		a. Manitol	my in hyperglycaemic condition	
9. In human kidneys, functional unit of kidney is also classified as a. Nephron b. Nebulin c. Neurone d. Macula Densa a. Pituitary gland b. Liver c. Adrenal gland d. Pancreas		b. Cortisol	o. Borono	
b. Nebulin  10. LH is released from  a. Pituitary gland b. Liver  c. Neurone d. Macula Densa  c. Adrenal gland d. Pancreas	9.	In human kidneys functional wait 51:1	d. Barbitol	
b. Nebulin  10. LH is released from  a. Pituitary gland b. Liver  c. Neurone d. Macula Densa  c. Adrenal gland d. Pancreas		a. Nephron	y is also classified as	
10. LH is released from  a. Pituitary gland b. Liver  c. Adrenal gland d. Pancreas			c. Neurone	
a. Pituitary gland b. Liver c. Adrenal gland d. Pancreas  2.2 Answer in very short (Any Ton)	10.	LH is released from	d. Macula Densa	
b. Liver d. Pancreas  2.2 Answer in very short (Any Ton)		a Pituitary aland		
d. Pancreas  Answer in very short (Any Ton)		h Liver	c. Adrenal gland	8
		o. Hive	d. Pancreas	
	2.2	Answer in your shared		
20 Live of Old		Give the biological in		20
2. Write down the physiological role of L	2.	Write de l'ological importance of "carbon"	element in living system	20

- 1. 2.
- Write down the physiological role of Iron in human body. 3.
- List the components of renal and digestive system.
- Define cardiac cycle. Explain the two major events of cardiac cycle. 4. 5.
- Define Acidosis and Alkalosis.
- Write down the role of KHCO3 in RBC. 6.
- 7. Give a brief note: GFR

ο.	List out various structural protein with its function in muscles.	
9.	What is chemical synapse? Give the role of neurotransmitter in it.	
10.	Differentiate between endocrine and exocrine gland.	
11.	Explain the factors regulate insulin secretion.	
12.	Justify the statement "hGH is protein anabolic in nature"	
	· · · · · · · · · · · · · · · · · · ·	
Q.3	a) Explain the chemical, cellular and tissue level organization in living system.	[5]
:1	b) Write down the biological importance of major elements.	[5]
	the state of the s	
Q.3	a) Give accounts on: Integumentary and skletomuscular system.	[5]
	b) Explain the physiological role of hormone secreted by pituitary gland.	[5]
Q.4	a) Describe the sliding filament mechanism for muscle contraction in detail.	[5]
·	b) Define nephron. Explain the structure and function of kidney.  OR	[5]
Q.4	a) Give a brief accounts on : Chemical composition of muscles	[5]
	b) What is action potential? Explain the mechanism of nerve impulse transmission.	[5]
	Here the first the first state of the state	
Q.5	a) Define respiration. Explain the various phase of respiration in detail.	[5]
	b) Write down the role of haemoglobin as biological buffer.	[5]
	and the control of th	
Q.5	Write a note on: a) Chloride shift	[5]
	b) Carbon dioxide transport in blood	[5]
Q.6	Define Diabetes mellitus. Describe the effect of insulin on glucose metabolism.	[10]
<b>4</b>	Control of the Contro	
Q.6	Classify hormone based on solubility. Explain the mechanism of action of lipid	[10]
4	and water soluble hormone.	
	paratic formation of the second of the secon	
	and the state of t	

A CONTRACTOR OF THE CONTRACTOR

### SARDAR PATEL UNIVERSITY

V.V. Nagar-388 120

#### B.Sc. Biochemistry (V- SEMESTER)

Subject: BIOINSTRUMENTATION

US05CBCH06

DATE:	<b>17/11/2017</b>	TOTAL MARKS	S: 70
Q.1	Multiple Choice questions: (1 M	lark each)	10
1.		transmit wavelength less than specified	
	wavelength	The state of the s	
	a. Band pass filter	c. Long pass filter	
	b. Short pass filter	d. None of Above	
2.	Diodes are present in	and the more services and the Alberta Alberta.	
	a. Barrier cell	c. Voltaic tube	
	b. Multiplier tube	d. Photo emissive tube	3 (1)
3.	Infrared spectroscopy provides valual	ole information about	
	<ol> <li>a. Molecular weight</li> </ol>	c. Melting points	
	b. Conjugation	d. Functional group	4.7
4.	The following rotor is used in harvest		
		c. Vertical tube rotor	
	b. Swinging Bucket Rotor	d. Elutriator Rotor	
5.	Speed of centrifuge is denoted by		** 1
	a. G	c. RPM hading A separation with a sequence	
	b. RCM	d. RCF	
6.	Ion exchange chromatography is base	ed on the  c. ionic species	
	<ul> <li>a. electrostatic attraction</li> </ul>	c. ionic species	
	b. adsorption chromatography	d. partition chromatography	
7.	Salting out compound is required for		- ;
٠	a. HIC	c. HLC are reflected	
	b. HPLC	d. GLC	
8.		ed on specificity	*
	a. Gel permeation	c. Affinity	
	b. Ion Exchange	d. HPLC	
9.	In isoelectric focusing, proteins are se	•	, ·
	a. relative content of positively		
	b. relative content of negatively	charged residue only	
	c. size		
	d. relative content of positively	and negatively charged residue	
10.	Immuno-electrophoresis requires	<b>A</b>	
	a. SDS	c. Agarose	
	b. Polyacrylamide	d. Starch	
Q.2	Answer in very short (Any Ten)		20
l.	What is hyper chromatic shift?		
2.	Differentiate HPLC and GC.		
3.	Give disadvantages of filter photome	eter.	
3. 4.	-		
	_	ite down principle of IR spectroscopy.	٠,١
5.	Derive the equation of RCF.	C.P.T.	0.)

6.	Write down the application of ultracentrifugation.	
7.	What is gradient media?	
8.	Discuss the polymerization process occurs in acryamide gel.	
9.	Write down the role of: a. SDS b. β-mercaptoethanol	
10.	List types of monochromator and radiant energy sources.	
11.	Give a brief note on: Stationary and mobile phase used in chromatography	
-1.	Define partition co-efficient and effective distribution co-efficient.	
Q.3	a) Give an account on: U.V. spectrometer	[5]
	b) Explain with diagram: Photo emissive tube	[5]
	na n	
Q.3	a) Explain the principle and working process of colorimeter.	[5]
	b) Describe various types of monochromators.	[5]
,	mid to the second of the secon	
Q.4	a) Write a note on Isophynic Technique	[5]
	b) Explain various factors affecting speed of centrifuge.	[5]
	OR CHARLES THE CONTROL OR	
Q.4		[5]
	b) Give an account on- Rate zonal rotor	[5]
Q.5	a) What is cation and anion exchanger? Describe the principle and working	[5]
	procedure of Ion Exchange chromatography.	
	b) Write down the principle and procedure of Gel permeation	[5]
	chromatography.	
	types, the research of the control o	
Q.5	Write a note on: a) Principle and pumping system of HPLC	[5]
	b) Detector used in GLC	[5]
Q.6	Write down basic principal of electrophoresis. Describe principle and method	[10]
	of SDS-PAGE for protein separation.	
	the second of th	
Q.6	Write a short note on: Agarose gel electrophoresis	[10]
	The state of the s	
	. The second of	
	and a second control of the second control of the second control of the second control of the second control of	\ \{\psi}
	en de la companya de La companya de la co	

The contract of the second of

Common of Artist Commission of the March 1995

Salara Milliana (D.C.)

No. of Printed Pages : 2

### SARDAR PATEL UNIVERSITY

## B.Sc (Vth SEM) (CBCS) EXAMINATION

Tuesday, 7th NOVEMBER -2017

US05CBIT01: MOLECULAR BIOLOGY

TIME - 10.00 am TO 1.00 pm Total Marks - 70 Note: Figures to the right indicate full marks Multiple Choice Questions (Each question of one mark) Q-1 [10] Which of the following methods of DNA repair is most error-prone? 1) (a) Recombinational repair (c) Photoreactive repair (b) Excision repair (d) SOS repair DNA polymerase responsible for leading strand synthesis 2) (a) Pol ß (c) Pol & (b) Pol  $\alpha$ (d) Pol δ Regulation of eukaryotic DNA replication is carried out by 3) (a) RFC (c) PCNA (b) Cdks (d) NCR Mature eukaryotic mRNAs have a 5' cap that is residue of 4) (a) 7- methylguanosine (c) 3- methylation (b) 2-methylation (d) 4- methylguanosine Which of the following is not true of RNA processing in eukaryotes? 5) (a) Addition of 5'cap (c) Reverse transcription (b) Addition of a poly A tail (d) Splicing of RNA Movement of ribosome from one codon to another is done by -----6) (a) Translocation (c) Transition (b) Transformation (d) Transversion Proteins that facilitate the correct folding of protein synthesized is ------7) (a) Signal sequence (c) Ribozime (b) Chaperone (d) Ribosome The first step in the biosynthesis of polypeptide is catalyzed by 8) (a) Terminal transferase (c) Peptidyl transferase (b) Aminoacyl-t RNA synthase (d) Initiation protein Which of the following is not an example of transposon 9) (a) P element (c) Copia elelment (b) Tn element (d) Col element 10) Which of the following enzymes is essential for transposition events (a) Transposase (c) RNA polymerase (b) Transferase (d) Integrase

,			
0.2		Attempt any ton shout questions (F. 1)	(40)
Q-2	A	Attempt any ten short questions (Each question of 2 marks) Define Telomerase and give its functions	[20]
	B	Define DNA Polymerase? Give its significance	
	C	What is DNA Polymerase Shifting and give its significance	
	D	Give various functions of RNA polymerases	
	E	TTT to all the control of the contro	
	F	Write the various functions of Enhancer? What are promoters? Give its significance	
÷	G	What is SRP? Give its uses	Q. (4)
471	Н	What is the charging of t-RNA	
\$ 4	I	What is protein targeting?	1 - 48
	J	Describe the importance of target site in transposition	
	K	Define transposons and give its examples	
	L	What are autonomous elements?	
		management some kilder at the second of the	
Q-3		Company of Copyright Copyr	
* *	(a)	Discuss the initiation step of replication with its regulation.	[07]
	(b)	Enlist types of DNA damages and Explain any one	[03]
	` ,	OR	[05]
	(a)	Discuss the termination mechanism of eukaryotic replication	[06]
	(b)	Give an account on methyl directed mismatch repair system.	[04]
			F
Q-4	(a)	Describe the termination process of transcription.	[05]
Ţ.	(b)	Explain the processing of r-RNA and t-RNA	[05]
		OR THE AMERICAN	
	(a)	Discuss the initiation process of transcription	[05]
	(b)	Explain 5' capping and polyadenylation modifications	[05]
		en e	
Q-5			
	(a)	Explain in detail the initiation process of translation in eukaryotes	[10]
		OR	
	(a)	Discuss in detail the post translational modification process in eukaryotes	[10]
0.6			
<b>Q-6</b>	(-)	· 1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,1987年,	3-1
	(a)	Describe the virus like transposable elements with example.	[06]
	(b)	Give the application of transposons in r-DNA technology	[04]
	(a)	Describe the IC transferent a standard of the	
	(a) (b)	Describe the IS transposable element of prokaryotes.  Explain the mechanism of transposable element.	[06]
	(0)	Explain the internation transposable element.	[04]
			- 4.
		min (1)	



[41/A19]

SHAT No.\_\_\_\_

NO. OF PRINTED PAGES: 02

SARDAR PATEL UNIVERSITY
B.Sc. V SEMESTER EXAMINATION
THURSDAY, 9<sup>TH</sup> NOVEMBER 2017
10:00 A.M TO 1:00 P.M
BIOTECHNOLOGY: USO5CBIT02
MOLECULAR TECHNIQUES

a teaph of each act and the amortotal Marks: 70 for second of the second

Multiple Choice Question	ns.	[10]
The technique for separation of	f charged molecules was developed by	1
a) Tswett	c) Sveberg	
b) Tiselius	d) Sanger	
PCR provides t	the information about the activity of tumor cells and	
viruses.	,	
a) Inverse	c) Nested	1 112
b) Reverse transcriptase	d) Anchored	:
In site directed mutagenesis the	e nick is sealed with	
<ul> <li>a) DNA polymerase</li> </ul>	c) Taq polymerase	1 34.4
b) DNA ligase	d) Restriction endonuclease.	
The application of Southern blo	otting in genetic engineering includes	
a) DNA fingerprinting	c) RFLPs	
b) Molecular mapping	d) All of these, in the second and product the second	1 4 4 4 E
Northern blotting is commonly	used for identification of	i a
a) DNA-DNA interaction	c) DNA-RNA interaction	
b) RNA-RNA interaction	d) RNA protein interaction	y series
The range of micro satellite sec	quences in eukaryotes is	
a) 1-5bps b) 7-13 bps c)	) 14-20 bps d) 21-27 bps.	
RFLP stands for		
a) Random fragment lengt	th polymorphism	1000
b) Restriction fragment ler		
c) RNA fragment length p	polymorphism	
d) None of the above.	Augusta (Constants)	H 46 478
Which of the following enzyme	e commonly used for in vitro transcription?	1
a) Prokaryotic RNA polyi	merase	
b) Eukaryotic RNA polym		
	dymerase	1 10
d) None of the above.		
	oped by	
DNA lingerprinting was develo		
DNA fingerprinting was develor a) Francis Crick	c) Alec Jeffrey	
DNA fingerprinting was develor a) Francis Crick b) Khurana	c) Alec Jeffrey d) James Watts	r varia
<ul><li>a) Francis Crick</li><li>b) Khurana</li></ul>	d) James Watts	t e kity
<ul><li>a) Francis Crick</li></ul>	d) James Watts	t e kity

Q.2	Answer the following questions in short. (Attempt any 10)	[20]
i) ii) iii) iv) v) vi) vii) viii) ix)	Mention the role of SDS in SDS PAGE. How temperature affects the process of electrophoresis? Which properties are essential for designing primers for PCR? Give a note on Oligonuc eotide mutagenesis. Write about Dot blot hybridization. Define the terms: a) Hybridization b) Blotting. Write different applications of DNA fingerprinting. Define Satellite DNA. Write a note on FISH and McFISH.	
x) xi) xii)	Differentiate between cDNA library and genomic library. What do you mean by DNA sequencing? Write automated method for DNA sequencing Give the basic principle of DNA foot printing	g.
Q.3 a) b)	Explain in detail Agarose Gel Electrophoresis.  Give a detail note on RT-PCR.  OR	[05] [05]
Q.3 a) b)	Give an account on SDS-PAGE.  Explain basic methodology and applications of PCR.	[US] [0 <b>5</b> ]
Q.4 a) b)	Give an account on Southern Hybridization.  Explain Colony Hybridization.  COP  COP  COP  COP  COP  COP  COP  CO	[06] [04]
Q.4 a) b)	Write a short note Autoradiography.  Explain: In situ hybridization.	[04]
Q.5 a) b)	What is molecular marker? Explain RFLP in detail. Write a note on SNPs.  OR	[06] [04]
Q.5 a) b)	Explain in detail AFLP.  Give an account on construction of cDNA library.	[06] [04]
Q.6 a)	Explain in detail Chain termination method for DNA sequencing.	[10]
Q.6 a)	Describe the Maxam and Gilbert method for DNA sequencing.	[10]
	The state of the s	•



## [33/A27]

#### SARDAR PATEL UNIVERSITY EXAMINATION, 2017 TYBSc Fifth Semester, 11th November, 2017, Saturday and the same and t Subject Biotechnology Course: US05CBIT03 Paper---Plant Biotechnology

Time: 10.00am -1.00pm Total marks 70 MULTIPLE CHOICE QUESTIONS. Attempt all questions each carry one mark. **Q1.** [10]I Gene can be transferred to plant protoplast via endocytosis using-----A. Ribosomes C Lysosomes **B.** Liposomes **D.** Chromosomes II Function of vir E in Ti plasmid is to-----A. activate Vir G for transcription B. transfer T-DNA to plant cell C. recognize and nick T-DNA at border sequence. D. protect T-DNA from plant nucleases III. Root producing property of Agrobacterium rhizogenes is due to----A. Vir genes in its plasmid **B.** Genes for auxin in T-DNA of its plasmid C. Genes for cytokinin in T-DNA of its plasmid **D.** Opines in its plasmid IV. bar gene encodes for-----C phosphinothricin acetyl transferase **A.** Glutamine synthase **B.** pyrophosphatase D phosphoenolpyruvate V. CpTi gene in transgenic plants provide resistance to-----A. insects . C. viruses B. herbicide **D**. fungus VI. Temperature of liquid nitrogen is-----A  $-70^{\circ}$ C C  $-150^{\circ}$ C B  $-196^{\circ}$ C D  $-10^{\circ}$ C VII. Disarmed Ti plasmid is constructed by -----A. Adding Multiple Cloning Site
B. Removing tumor inducing genes
C. Adding selectable marker
D. Removing border sequences VIII. Hormone that controls closure of stomata in response to water stress is----C. GAD. IAA A. ABA **B.** BAP IX. What makes a plant to bend towards light as it grows? A. Because green plants need light to carry on photosynthesis B. Because green plants are phototropic C. Light stimulates plant cells on the illuminated side to grow faster D. Auxin accumulates on shaded side stimulating greater cell elongation X. Glycerol can be used as----A. Cryopreservant C Sterilant **B.** Cryoprotectant D A) and C) both

Q2.	Short questions. Attempt any TEN questions. Each carry Two marks	[20]
	a) Enlist different methods used to make transgenic plants.	
	b) Which genes were used to make golden rice, and why?	
	c) Explain the use of selectable markers in plants.	
2 1 1	d) Explain the microinjection method of gene transfer	
ŧ	e) Why cryoprotectants are required in germplasm preservation?	
	f) Give the full form of ABA ,EPSPS, CAT and DMSO.	
	g) Define somaclonal variations and micropropagation.	
	<ul><li>h) Define parthenocarpy and apical dominance.</li><li>i) Define secondary metabolites and enlist any four examples.</li></ul>	
	j) Explain the term ex situ and in situ cryopreservation.	
	k) What is triple response?	
	l) Explain the role of GA in seed germination.	
	2 - A - A - A - A - A - A - A - A - A -	
Q3a.	Explain the role of various genes involved in transferring T-DNA in plants.	[06]
Q3b	Write a note on electroporation.	[04]
01	OR	
Q3	Describe various markers used in screening transgenic plants.	[10]
Q4a.	Explain various strategies used to raise glyphosate resistant plant? Explain.	[07]
Q4b	Discuss the mode of action of cry proteins?	[03]
~	OR 19	[oo]
Q4a	Describe the biochemical changes during ripening of tomato.	[06]
Q4b	Define edible vaccine and explain its mode of action.	[04]
	randra de la composição d La composição de la compo	
Q5a.	Discuss the factors affecting cryopreservation.	[06]
Q5b	Explain the biosynthetic routes of auxins.	[04]
•		
Q5a	OR	£0.71
Q5h	Describe in detail the regulation of phototropism and geotropism in plants.  Define artificial seeds give their method of preparation.	[06]
QUU	Define artificial seeds give their method of preparation.	[04]
Q6a	Discuss in detail the molecular basis of somaclonal variations.	[07]
Q6b	Write a note on hairy root culture.	[03]
	OR	
Q6a.	Explain the methods used to isolate somaclonal variations in vitro	[06]
Q6b	What are bioactive compounds? Write a note on antitumor properties of plants.	[04]
	the control of the co	
	—X—	
	ho	

and the second s



DESI NO.

No. of Printed Pages: 2

### SARDAR PATEL UNIVERSITY

B.Sc Examination, 5<sup>th</sup> Semester

Monday, 13<sup>th</sup> November 2017

## US05CBIT04

(Immunology)

Time: 10:00 A.M -01.00 P.M.

Total: 70 Marks

Q-1	24	MULTIPLE CHOICE QUESTION	10
	<sub>/</sub> ,1.	What indicates a positive reaction in complement fixation test?	
	777	a) Ag-Ah precipitation	
		a) Ag-Ab precipitation c) Agglutination of sheep RBC b) No haemolysis of sheep RBC d) Haemolysis of sheep RBC	
	2.	Two or more cytokines that mediate similar function are attributes of	
		a) Pleiotrophy	,
	. 4	b) Synergy d) Redundancy	
	3.	A crossed precipitin line following double immune diffusion is due to	
		a) Shared epitopes between antigen c) Identify between antigen	
	+ 1	b) No common epitopes between d) Only few epitopes that are	
		antigen antigen	
	4.	antigen common between antigen Indicate the most appropriate test for detecting Typhoid bacterium in the serum	
		a) Agglutination c) ELISPOT	
	- 15	b) Sandwich ELISA d) RIA	
	5.	Originally was also called T-Cell growth factor.	
		a) IL-2 c) INF	
		b) TNF east, and d) chemokine	
	6.	The major antibody which is involved in classical pathway?	
	,43	a) IgM	•
	474	b) Ig A compared to the compar	
	7.	Hypersensitivity -type I reaction is also called as?	
		a) Reagine type c) Ab mediated lysis	
	144	b) Ag-Ab complex mediated d) Delayed type	
	8.	is potent activator of Naive T cell, memory T cell and effectors T cell.	:
		a) Dendritic cell c) Macrophages	
		b) B cell d) RBC	
	9.	Membrane attack complex of complement system is consist of	
		a) C1-C4 c) C5b-C9	
		b) C5 d) C3-C9	
	10.	Auto antibodies to intrinsic factor blocks vitamin B12 absorption,	
		a) Pernicious anaemia c) Drug induced anaemia	
		b) Haemolytic anaemia d) All of the above	
		(PT	ر ہ۔

Q-2		SHORT QUESTION (any ten)	**
	1.	What do you understand by coomb's test?	20
	2.	Give account on Zone Phenomenon.	
	3.	Explain Positive and negative Thymic selection in brief.	
	4.	Give components of complement system.	
	5.	Differentiate between T- dependent antigen and T- Independent antigen.	
	6.	What are the causes of primary immunodeficiency?	
	7.	What do you understand by Primary and Secondary mediators in Type-I	
	••	Hypersensitivity?	
	8	Enlist certain pathogenic factor for the initiation of Alternate pathway	
	9.	Give function of TNF.	
	10.	Explain in brief RID	
	11.		[H] :
	12.		
	141	Give a brief account on attenuated Vaccines with example.	
Q-3	A	Write a detailed note on RIA	
ζ-	B	Explain immuno-electrophoresis along with its variation	05
	1.	OR	05
	A	What is the purpose of using sheep and rabbit RBC in Complement fixation test and	
	••	discuss CFT in detail where the complement fixation test and	05
	В	Give a detail account on ELISA	
	L,	Core is detail, account on DIASA	05
Q-4	A		
V 1	В	Give an account on B cell Differentiation with diagram.  How do NK cells recognise foreign antigen?	06
	נו	appear to the cents recognise foreign antigen?	04
	A	Give a detail account on T call development at the control of the	
	А	Give a detail account on T cell development (a because of the attendance of the second particular of the second particula	10
Q-5	A	Give a détail account on SCID de la	
Q-5	B		06
	ָט	Describe alternate pathway of complement system with figure.	04
	A		
	.A.	Discuss the complement system pathway which is activated by formation of Immune Complex.	06
	$\dot{\mathbf{B}}$		
	D	Discuss HIV as Secondary Immunodeficiency disease.	04
0-6	A		
Q~0	B	Discuss in detail type-II hypersensitivity	06
		Write a note on Give an account on autoimmune disease mediated by stimulating and	04
		blocking auto antibodies	
	A	Discuss structure of MHC Class-I molecule	
	В	Discuss structure of MHC Class-I molecule	05
	D	Discuss the mechanism of graft is rejection.	05
			13
		ALL THE BEST 11 TO THE PROPERTY OF THE PROPERT	*



The second second

SEAT No.\_\_\_\_

NO. OF PRINTED PAGES: 02

### [27/A-11]

#### Sardar Patel University

B.Sc Biotechnology Fifth Semester Wednesday, 15<sup>th</sup> November 2017 10:00 a.m to 1:00 p.m

US05CBIT05 (Environmental Biotechnology)

Total Marks: 70 Note: Figures to the right indicates marks. Q.I Multiple Choice Questions [10] 1) is an indicator of organic water pollution. BOD a) b) Alkalinity c) Acidity TOC d) Soil pollution is/are caused by 2) Industrial activities b) Agricultural activities Mining activities d) All of these 3) Which of the following is not removed from water by tertiary treatment? Solids a) b) Phosphorus Nitrogen c) d) None of these 4) Thiobacillus group of bacteria involved in bioleaching are Phototrophs Chemolithotrophs b) c) Heterotrophs Autotrophs CuFeS<sub>2</sub> is an ore called 5) a) Chalcopyrite b) Arsenopyrite Covellite c) d) Hematite is not required for the process of biodegradation. a) Microorganisms Environmental conditions b) Adhesives c) d) Nutrients Bioremediation can be carried out by Fungi b) **Plants** a) c) Bacteria d) All of these biosensor is based on the movement of electron. a) Amperometric Potentiometric b) c) Piezoelectric d) Thermometric Which of the following is not a bioplastic? 9) PHA **PVC** a) b) c) PHB d) PLA Bioreceptor/s used in biosensor is/are 10) Microorganism Enzymes a) All of these c) Antibody d)

<ul> <li>a) Write the effects of land pollution.</li> <li>b) Enlist the control measures for air pollution</li> <li>c) Differentiate between BOD &amp; COD.</li> </ul>	[20]
c) Differentiate between BOD & COD.	
d) Define bioleaching.	
e) Enlist the microorganisms involved in the process of bioleaching.	
f) Give the significance of Bioleaching.	
g) Mention the factors affecting the process of biodegradation	
h) Give diagrammatic representation for construction of Superbug in bioremediation	n.
i) What is In situ bioremediation?	
j) Write about the components of Biosensor.	
k) Mention the name of natural sources used in the production of bioplastic.	
l) Give various applications of PHA.	
Q.III Explain any two biological treatment processes for sewage treatment based on	[10]
aerobic attached growth system.	
OR	
Q.III Write a detail note on the following:	
a) TOC	[05]
b) Activated sludge process of waste water treatment	[05]
Q.IV a) Give an account on the direct & indirect bioleaching.	[05]
b) Explain in detail types/methods of bioleaching.	[05]
OR	. ,
Q.IV a) Discuss in detail factors affecting the process of bioleaching.	[05]
b) Write in detail about Copper bioleaching.	[05]
Q.V a) Describe biodegradation of any one pesticide.	[05]
b) Write short note on Ex situ bioremediation.	[05]
OR	[]
Q.V a) Define biomagnifications. Explain it with the help of suitable example.	[06]
b) Write short note on Phytoremediation.	[04]
Q.VI a) Define biosensor. Explain its principle in detail.	[05]
b) Give an account on applications of biosensor.	[05]
OR	i 1
Q.VI Describe various approaches used for production of bioplastics and also write the properties of bioplastic.	e [10]



No. of Princes Pages 102

Sardar Patel University B.Sc Biotechnology Fifth Semester Friday, 17<sup>th</sup> November 2017 10:00 am to 1:00 pm US05CBIT06 (Cell Biology)

**Total Marks: 70** 

Note:	Figures t	o the right indicates marks.		n de la companya de La companya de la co	
I.Ç	Multin	le Choice Ouestions			[10]
1)	The flu	id mosaic model proposed in	by S	S. Jonthan, Singer & Gorth Nicolson.	
-,	a.	1975	c.	1670	
	b.	1980	d.	1972	
2)		is not a component of Plasma me	mbrane?		•
<del></del> ,	a.	Carbohydrate	c.	Protein	
	h	Linid	d.	None	
3)	If two	solutes are transport in the opposi	te directio	n across the plasma membrane the	
	nroces	s is called			
-	a.	Facilitate diffusion	c.	Symport	
	ь. b.	Uniport	d.	Antiport	
4)	0.	is also known as microfilame			
¬1	a.	Actin Filament	c.	Tubulin filament	
	b.	Myosine filament	d.	None of these	
<b>5</b> )		of the following cells lacks cytos	keleton?		
<i>3y</i>	a.	Eukaryotic plant cell	. С.	Both a & b	*1
	b.	Prokaryotic bacterial cell	d.	Prokaryotic & Eukaryotic animal	
4 to 3 to 3	0.		i i m	cell	e e
<b>6)</b>		is the ability of the system to re	eceive mul	tiple signals & produce unified	
•	respon		41.3		
40	a.	Cooperativity	c.	Desentization	V .
\$471	b.	Integration	d.	Amplification	- 1
7) ::	0.	is plasma membrane recep	tor that is	also an enzyme.	
93	a.	Tyrosine kinase	c.	PEP kinase	
	b.	Pyruvate kinase	d.	Hexokinase	
8) :	What	happen when insulin receptors bi	nds with in	nsulin?	
0)	a.	Phosphorylation of IRS-I	c.	Autophosphorylation of Grb2	
5 <sup>27</sup> 5	b.	Autophosphorylation of IRS-I			14
	. 0.	/tatophosphory latton of 1140 2		residue.	
0)	Whiel	h of the mostly associated with ap	optotic pa	thway?	
9)	a.	Myoglobin	C.	APAF I	
	а. b.	NADH	d.	α- Tubulin	
10\	• • •	mers are derived from			
10)		Endoderm	c.	Ectoderm	
	a. b.	Mesoderm	d,		
	D.	Mesodeliii	u,		рΤ

Q.II	Answer the following questions (attempt any TEN)	[20]
a)	Define Osmosis with example.	
b)	Draw a neat and labelled diagram of fluid mosaic model.	
c)	Give the difference between active and passive transport.	
d)	Draw and label Myosin I.	
· e)	What are MTOCs?	
f)	Give the functions of cytoskeleton.	
g)	Define Cell cigralling	
h)	Write basic characteristics of cell signalling.	
i)	Give the role of second messengers.	
j)	Define caspases with its role.	
k)	What do you mean by Metastasis?	
l)	Write the difference between benign and malignant tumors.	
Q.III	Explain various components of plasma membrane. Discuss membrane lipids in detail	[10]
O 777	OR CHARLES	F4.03
Q.III	Write note on following:-	[10]
n	a) Na <sup>+</sup> K <sup>+</sup> ATpase Pump b) Membrane fluidity	10.61
Q.IV a)	How assembly and disassembly of intermediates filaments occurs?	[06]
b)	What are MAPs?	[04]
	OR OR	50.63
Q.IV a)	Write a note on Myosin II and Kinesin.	[06]
b)	Describe the structure of Microtubules.	[04]
Q.V a)	Write a note on G- Protein coupled receptor.	[06]
b)	Write in brief about the role of cAMP as a second messenger with any one example.	[04]
Ť	OR	Ī
Q.V a)	Explain in detail about the Receptor Tyrosine kinase.	[06]
b)	Give an account on JAK-STAT pathway.	[04]
·		
Q.VI a)	Describe the mitochondrial pathway of apoptosis with neat and labelled diagram.	[06]
b)	Discuss the mechanism of cancer.	[04]
	OR The Medical Control of Cancer.	-
Q.VI a)	How the protooncogenes convert into oncogenes? Explain.	[06]
b)	Give the significance of apoptosis.	[04]
,		

200

And the state of t

1. 1. 1. 1. 1.

## SARDAR PATEL UNIVERSITY

B. Sc. (Bioinformatics) Examination, 5<sup>th</sup> Semester Tuesday, 07<sup>th</sup> November, 2017 US05CBNF01: Visual Programming

						•	
Time	e: 10:C	O AM	to 01:00 PM			Total Mari	ks: 70
	Note:	Answ shou	ver of all the qu ld be written in	estions (inclu the provided	ding ansv	Multiple Choice Questions) ver book only	
Q:1	Give [01]		ers of following	g Multiple Cl	noice	Questions	[10]
		(A) (C)	Compiler Technology		(B) (D)	Interpreter Assembly	
	[02]	Dim	is an Abbreviat	ion of			
		(A) (C)	Dimension Dimerization		(B) (D)	Dimer Diameters	
	[03]	The c	JIT is one type	of	.•		
		(C)	Hardware Measurement				
	[04]	In VI	3 .NET, inputbo	ox returns		type of data.	
		(A) (C)	String Integer		(D)	Double None of these	
	[05]	То со	ome out from a	ny loop		statement is used.	
		(A) (C)	Break Goto		(B) (D)	Exit None of these	
	[06]	Whic	ch statements a	re optional ir	an I	fThen statement?	
		(A) (C)	Else If		(B) (D)	Then All of the above	
	[07]	The '	Tick event is fo	und only in w	hich	object?	
		(A) (C)	Form Button		(B) (D)	Timer Textbox	
	[08]	Whi	ch control displ	ay hierarchy			
		(A) (C)	Treeview Timer		(D)	Label Scrollbars	
	[09]		is disco	nnected, in-n	nemo	ry representation of data.	
			DataReader DataAdapter		(B) (D)	DataSet DataCommand	
	[10]	Wha	at is the major o	component of	conn	ected data architecture?	
		(A) (C)	DataReader DataAdapter		(B) (D)	DataSet DataCommand	

Q:2	Ansv	Answer the following short questions (any Ten)		
	[01]	What is CLS?	·	
	[02]	Write a short note on MSIL?	·	
	[03]	List the parts of IDE.		
	[04]	Write down the syntax of If End If.		
	[05]	What is the use of 'By Val' & 'By Ref' keyword?		
	[06]	Write a difference between MDI and SDI.		
	[07]	Differentiate between radio button and checkbox.		
	[08]	Differentiate between label and linklabel.		
	[09]	Differentiate between listbox and combobox.		
	[10]	What are the features of ADO.Net?		
	[11]	Explain the use of server explorer in data access in .NET.		
	[12]	Mention different types of data providers available in ADO NET.	•	
Q:3	[A]	Explain CLR, CTS, and MSIL in detail.	[06]	
	[B]	Write a short note on Solution Explorer and property window	[04]	
		<u>OR</u>		
Q:3	[C]	Write a short note on IDE.	[06]	
	[D]	Write a short note on Boxing and Unboxing	[04]	
Q:4	[A]	Explain declaration of function in detail with example.	ine;	
Q.7			[06]	
	[B]	Explain For Next Structure in detail.	[04]	
0.4	ra:	<u>OR</u>		
Q:4	[C]	Explain Message box with example.	[05]	
	[D]	What is MDI form & explain how it differs from normal form?	[05]	
Q:5	[A]	Write a note on HScorllBar and VScrollBar.	[05]	
	[B]	Explain controls with its properties, methods & events: textbox and	[05]	
		button		
·		<u>OR</u>	٠	
Q:5	[C]	Explain try Catch final in detail with example.	[05]	
	[D]	Explain timer control with its properties, methods & events	[05]	
Q:6	[A]	Explain the connected architecture of ADO .NET in brief.	[04]	
•	[B]	Explain the steps to bind the application with the Database in ADO	[06]	
	r— J	.NET.	r1	
		<u>OR</u>	•	
Q:6	[C]	Explain the step, how can we retrieve data in DataSet?	[04]	
	[D]	Explain public methods of SqlCommand objects	[06]	
		-x-	-	

[42] SHAT NO.\_\_\_\_

#### SARDAR PATEL UNIVERSITY

#### T.Y.B.Sc : SEMESTER - V BIO INFORMATICS

#### **US05CBNF02**: Object Oriented Programming and Data Structure

	9/11/2017 Time: 10:00am to 01:00pm Max.Mar	ks:70
Q.1	ทุพธริสตรุ Multiple choice of Question:	10
Q.1	[1] is basic run time entity in object-oriented system.	10
	[A] Object [B] Class [C] Data [D] Function	
	[2] In data is hidden and cannot be accessed by external	
	functions.	
	[A] OOP [B] POP [C] SOP [D] None	
	[3] is a user define data type.	
	[A] Class [B] Variable [C] Operator [D] Function	
	[4] constructor contains an object as argument.	
	[A] Default [B] Parameterized	
ļ	[C] Copy [D] None of Above	
-	[5] For string manipulation header file is included.	
i	[A] string.h [B] ctype.h	
	[C] iostream.h [D] conio.h	
	[6] Which of the following is a valid function prototype in C++?	1
	[A] int sum(int a, int b, int $c = 0$ )	
	[B] int sum(int $a = 0$ , int b, int c)	
<u></u>	[C] int sum(int a, int $b = 0$ , int c)	
	[D] int sum(int $a = 0$ , int $b$ , int $c = 0$ )	
	[7] Defining a function with same name but different types & no. of	
	arguments is known as	
	[A] inheritance [B] function overriding	
	[C] function overloading [D] Inline function	
	[8] of the array represents the kind of data type.	
	[A] Size. [B] Base.	
	[A] Size. [B] Base. [C] Type. [D] Index.	
	[9] If the range of index varies from L U then size of the	
	array is	
	[A] $U - \overline{L + 1}$ . [B] $L - U + 1$ .	
	[C]U/L-1. $[D]L/U-1.$	
	[10] Two dimensional arrays are also called	
	[A] Row array [B] matrix array	
	[C] Column array [D] Index array	
Q.2	Answer the following in short (Any 10):	20
Austra (West	[1] Define class & object as concept of OOP.	,
	[2] Define encapsulation as concept of OOP.	
	[3] List out any four header file in C++.	***************************************
	[4] Explain cout in C++ with example.	
	[5] Explain cin in C++ with example.	
	[6] Write syntax for declare & initialize one dimensional array with	
	example.	
	[7] What is *this pointer?	
	[8] What is function? Explain briefly.	
	[9] What is inline function? Give an example of it.	
	[10] Write Algorithm for PUSH() operation on stack.	
	[11] Write Algorithm for POP() operation on stack.	
	[12] Define Primitive and Non-Primitive data structure.	

Q.3	[A] Explain features (characteristics) of OOP in detail.	5
	[B] What is C++? Explain structure of C++.	5
	OR	
Q.3	[A] Explain different data type available in C++.	5
	[B] Explain basic concepts of Polymorphism and Inheritance	5
Q.4	[A] Explain nameless objects with proper example.	5
	[B] Explain destructors with an example.	5
	OR	
Q.4	[A] Explain constructor overloading with example.	5
	[B] Explain private and public area in class?	5
Q.5	[A] Explain the array of pointers with example.	5
	[B] Explain the pointer to function with example.	5
	OR	
Q.5	[A] What is function overloading? Explain it with example.	5
	[B] Explain the concept of friend function with suitable example.	5
Q.6	What is Queue? Write an algorithm for Insert and Delete operations on Circular Queue.	10
	OR	
Q.6	Write detail note on Array.	10



#### SARDAR PATEL UNIVERSITY

Bachelor of Science (B.Sc.)
Fifth Semester Examination November – 2017
Saturday, 11<sup>th</sup> November, 2017
10 A.M to 1:00 P.M
Subject: Bioinformatics
COURSE: US05CBNF03

(Basics of Immunology)
Total Marks: 70

Note	: (1) Figures to the right indicate marks. (2) Draw a neat and labeled diagram, wherever necessary.	Total Marks, 70		
Q. 1	Choose the most appropriate answer from the four alternat			
(1)	Which of these is called Altered self cell: A) Tumor cell B) Virus infected cell C) both A	& B D) B lymphocyte		
(2)	Which of these cells is phagocytic?  A) B-cell B) TH cell C) Dendritic cell  D) Both A and C			
(3)	The substances that posses antigenicity but lacks immunoge A) Adjuvants B) Haptens C) Superantigens	<del>-</del>		
(4)	Which of the antibody molecule can cross the placenta?  A) IgA B) IgE C) IgG D) IgM			
(5)	How many classes of MHC molecules are there: A) 2 B) 3 C) 4	D) 5		
(6)	Principal of RID states that concentration of antigen is dire A) Concentration of Antibody B) Diamete C) Time of incubation D) Size of the plat	er of the precipitin ring		
(7)	Major histocompatibility locus in human beings is known a A) H-2 complex B) HLA complex C) Ig complex			
(8)	Antigen Antibody interactions are important because of:  A) Specificity  B) Sensitivity  C) Non-ambiguity	D) All of these		
(9)	Phagocytic pathway would process:  A) Endogenous Antigen  B) Exogenous Antigens  C) antigens	Haptens D) Super		
(10)	The Hybridoma technology was invented by:  A) Kohler & Milstein B) Porter C) Edelman	D) Watson & Crick		
		DTA		

Q.Z	Answer any <u>TEN</u> from the following:	[20]
(1)	Define Innate immunity and mention its components.	, -,
(2)	Briefy explain how haptens are different from antigens?	
(3)	Transfer its reaction.	
(4) (5)	1 8 5 1 1 1 1 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
(6)	Define Secondary lymphoid organ. Mention two examples. What are the functions of antibody molecule.	
(7)	Enumerate various features of Antigen antibody reactions.	
(8)	What are the main differences between agglutination and precipitation reaction.	
(9)	Define Radio Immuno Assay. What is its limitation.	
(10)	Write various applications of monoclonal antibodies.	
(11) (12)	What are the various functions of MHC molecules?	
(12)	What are the main differences between endogenous and exogenous antigen?	
Q.3 (a)	Define Innate immunity. Briefly explain various components of innate immunity	[5]
(b)	Give a Comparative account of Primary and secondary Immune response	[5]
	<u>OR</u>	1 1
Q.3 (a)	Briefly explain second line of defence, its features and components.	[5]
(b)	Give a Comparative account of Active and Passive immunity	[5]
Q.4 (a)	Explain the structure of antibody molecule with Diagram.	[5]
(b)	Write a short note on thymus as an lymphoid organ	[5]
	OR	
Q.4 (a)	Briefly explain various immune cells and their functions in brief. Write a Give a	[5]
(b)	comparative account of IgG and IgM Molecule.	[5]
Q.5 (a)	Write an comparative account of RIA & ELISA	[5]
(b)	Briefly explain precipitation reaction and its types.	
	OR	
Q.5 (a)	Explain CFT with neat and labelled diagram.	[5]
(p)	Explain Sandwich ELISA with a diagram.	[5]
Q.6(a)	Explain Hybridoma technology with proper flow chart.	[5]
(b)	Explain structure of MHC Class-I Molecule with diagram.	[5]
	OR	
Q.6	With the help of labelled diagram explain the processing of endogenous antigen in detail.	[10]
	XX	



#### No. of Printed Pages 1 2

#### Sardar Patel University Semester examination-2017

B.Sc Vth Semester,

Subject - Bioinformatics

Course no. US05CBNF04,

Date - 13.11. 2017, Monday

Genetic Engineering-I

Time-3hrs (10.000m to 1.000m)

Marks-70

NOTE- Figure in the right indicates marks

All questions are compulsory. Make necessary diagram wherever needed.

O.1. Multiple Choice Question (MCQ). Select correct answer from given MCQ.

- 1.a. Type II Restriction enzymes are always recognizes target site that are
  - (A) Palindromic DNA sequences
- (B) AT rich sequences
- (C) GC rich sequences
- (D) Repetitive DNA

1.b. Which of the following bond are join by DNA ligase

- (A) Phosphodiester bond
- (B) Hydrogen bond
- (C) Glycosidic bond
- (D) Covalent bond

1.c. Which of the following vector considered as work horse of a gene cloning laboratory

(A) YAC

(B) BAC

(C) PBR 322

(D) PUC 8

1.d. Ti plasmid are present in

(A) E. Coli

- (C) Agrobacterium Tumifacience

(D) Pseudomonas Putida

(B) Bacillus subtilis

1.e. Select the vector used for gene transfer in mammalian cell

(A) YAC

(B) Retrovirus

(C) PBR 322

(D) YCP

1.f. Select the methods that used for identification of Recombinant clones

- (A) Split gene expression
- (B) Insertional inactivation
- (C) Base pairing
- (D) Insertional activation

1.g. Which of the following bacterial cell are used for transformation of rDNA

- (A) Competent cell

(B) Wild type cell (C) Protoplast (D) Spheroplast

1.h. Agarose gel electrophoresis can be used for separation of DNA fragments

(A) 0.5 to 60 kb

(B) 50-200 kb

(C) 150-450 kb

(D) 450-600kb

1.i. 2,3 dideoxynucleotides can terminate the synthesis of DNA through

- (A) Block the formation of phosphodiester bond with dNTPs
- (B) Block the formation of Hydrogen bond between bases
- (C) Block the formation of N glycosidic bond between sugar and base
- (D) Block the formation of covalent bond

1.j. Which of the following are template for Reverse Trscription (RT) PCR

(A) cDNA

- (B) Genomic DNA
- (C) Mitochondrial DNA
- (D) All of the above

P.T.O



Q.2. Short questions (2 marks each) attempt any ten	(2x10=20marks)	
[1] What is r DNA?		
[2] Write notes on alkaline phosphatase.		
[3] Define cosmid vector.		
[4] What is T DNA?		
[5] What is retrovirus vector?		
[6] Enlist essential features of cloning vectors.		
[7] What is transformation?		
[8] Write short notes on application of genomic DNA library.		
[9] How cDNA can prepare in laboratory?		
[10] Enlist various reagents for agarose gel electrophoresis.		
[11] What is chromatography?		
[12] Write brief notes of application of DNA sequencing.		
Q3.a Explain the features and properties of PUC 8 vector.	[05]	
Q3.b.Discuss the important properties and function of restriction enzymes.  OR	[05]	
Q.3.a. Explain the features and properties of pBR322 plasmid	[05]	
Q.3.b. Write short notes on properties and application of DNA polymerase.	[05]	
Q.4.a. What is YAC? Explain.	[05]	
Q.4.b. Write notes on Binary vector.	[05]	
OR	. ,	
Q.4.a. Explain co integrate with neat diagram.	[05]	
Q.4.b. Differentiate between YEP and YIP.	[05]	
Q.5.a What is insertional inactivation? Explain with suitable examples.	[05]	
Q.5.b. How will you construct genomic DNA library? Explain	[05]	
OR		
Q.5.a. How will you construct cDNA library? Explain with suitable steps.	[05]	
Q.5.b. Write s notes on application of recombinant selection and screening.	[05]	
Q.6. a.Explain the process of classical PCR in detail.	[05]	
Q.6.b. How a DNA fragments sequenced by Maxam and Gilbert Method? Explain.		
Q.o.b. How a DNA fragments sequenced by Maxani and Gibert Method: Explain.	[00]	
OR		
Q.6.a. Explain the various factors required for polyacrylamide gel electropho		
Q.6.b. Write notes on Sanger method of DNA sequencing.	[05]	
XX		



*	SEAT No	
Ĺ	No of Prince of the	
	No. of Printed Pages: 02	
	EXTERNAL EXAMINATION	
	DATE - 15/11/17 DAY- WEDNESDAY TIME 10:00 TO 1:00 pm	
	Course- US05CBNF05 SUBJECT: BIOINFORMATICS	*. *
	CLASS- T.Y.B.Sc V Sem TITLE BIOINFORMATICS APPLICATION-I	
	TOTAL MARKS: 70	
Q1- Se	elect the correct from the following Multiple Choice: [1 X 10]	
(i)	Which of the following organisms is not considered to be a model genetic system?  a) Mice b)Fruit flies c) Humans d) Yeast	
(ii)	Gene duplication has been found to be one of the major reasons for genome expansion	a in
	eukaryotes. In general, what would be the selective advantage of gene duplication?	
	<ul> <li>a) If one gene copy is nonfunctional, a backup is available.</li> </ul>	
	b) Larger genomes are more resistant to spontaneous mutations	
	c) Duplicated genes will make more of the protein product.	
(:::N	d) Gene duplication will lead to new species evolution.	
(iii)	What is the most challenging issue facing genome sequencing?	
	a) the inability to develop fast and accurate sequencing techniques  b) the ethics of using information from someones at the individual leading to the individual leads to the	
	b) the ethics of using information from genomes at the individual level c) the availability and stability of DNA	
	d) all of the above	
(iv)	Funding for the Human Genome Project comes from the	
` /	a) NIH b) DOE c) NIH and DOE. d)NIH, DOE and ELSI	
(v)	Chromosome walking	
	a) allows one to move from one chromosome to another	
٠	b) requires overlapping cloned sequences	
	c) requires FISH	•
	d) requires pedigree analysis	
(vi)	Each ds DNA has frame in ORF.	
(viii)	a) 10 b)3 c) 6 d) 2	
(vii)	Block is a) Sequence with indel b) sequence with mismatch only	
	a) Sequence with indel b) sequence with mismatch only c) Sequence with match only d) sequence with gaps only	
(viii)	Well-conserved regions in multiple sequence alignments	
()	a)reflect areas of structural importance. b) reflect areas of functional importance	
	c) reflect areas of both functional and structural importance	
	d) reflect areas likely to be of functional and/or structural importance.	
(ix)	Why are colour schemes important in creating and analysing sequence alignments?	
	a)They look pretty.	
	b) To make clearer printouts and presentations.	
-	c) To allow you to distinguish conserved residues and residue groups more easily	
( )	d) To allow you to detect active sites of proteins	
(x)	The regions of DNA in a eukaryotic gene that encode a polypeptide product are called: a) hnRNAs b) exons c) enhancers d) leader sequences	
Q2	Answer the following in brief (any ten) [2X10]	
(i)	Why comparative genomics is important?	
(ii)	Differentiate conservative and replicative mode of transposons.	
(iii)	Explain chromosome walking.	

(iv)	Differentiate prokaryotic genome and eukaryotic genome				
(v)	Name few model organisms used for human genome project and why they were used? Give the importance of sequence logo.				
(vi)				in outcomustos	
(vii)		atically explain str omoter and its utili			
(viii)	<u> </u>	EST]-{DW}-2(X)-		nession	
(ix) (x)		out the history of I			
(x) (xi)		atellites differ from		tec?	
(xii)		ne sequencing is i		103;	
(XII)	willy genor	ne sequeneing is i	пропине.		
	LONG QU	JESTIONS			
Q3	Write a sł	ort note on			(10)
	i)	Transposons		ii) Alternative splicing	
	iii)	Pseudo genes		iv) Repeat regions	
				O.D.	
0.0	· · ·	1 . 11 1 1		OR	(10)
Q3	Explain ii	n detail about euka	ryotic genome	e organization	(10)
Q4	Discuss tl	ne method and sign	nificance of ge	enome sequencing.	(10)
		Z.	Ü	OR	, ,
Q4	Elaborate	the aims, objectiv	es and applica	ition of HGP.	(10)
05(a)	What is C	ND E2 Thioniss tha 1	ogia plaorithm	n for gene identification.	(05)
Q5(a) Q5(b)		nort note on ANN.		Tot gene identification.	(05)
Q5(0)	WIIIE a SI	fort flote on Aiviv.		OR	(03)
Q5	Explain t	he structure of pro	karvotic gene	structure and different methods for its	
42	prediction		itary out going	01. 40.47 0 41.44 41.14. 01.1 17.007.000 707 110	
	•				
Q6	Write a sl	hort note on the fo	llowing: (any i	2)	
	i) Profile	j	i) Block	iii) pattern	(10)
				OR	
Q6(a)		equence logo? Ex			(05)
Q6(b)	How HM	M model can be u	sed for multip	le sequence alignment.	(05)
*****	*****	*******	** —X —	· X ******************	****
			1 -		

9. Explain protein – protein interaction.

10. Explain principle of protein folding.

11. What are prions.

Q.1

Q.2

12. Explain about Cystic fibrosis disease.

Q.3	a) b)	What is Normalization? Explain 1st and 2st Normal Form.  List various basic data types used in oracle. Explain in detail.	5]
Q.3	a) b)	OR Explain E.F.Codd Rules. What is E-R diageam?Draw the different symbol used in it.	[5] [5]
Q.4	a)	Explain various ways to change structure of a table using alter statement	nt [5]
	b)	Explain various ways to insert records in a table.	[5]
Q.4 Q.5 Q.5	<ul><li>a)</li><li>b)</li><li>a)</li><li>b)</li></ul>	OR Explain PrimaryKey Constraint in detail with Example. List Different operator used in sql?Explain any one of them.  What are prion? How is cause infection and its consequence. Discuss any two disease in detail for protein misfolding.  OR How the misfolding of protein occur and its consequence. Write a note on mad cow disease.	[5] [5] [5] [5] [5]
Q.6	a) b)	What are protein? Discuss their types. Write a note on yeast to hybrid method.	[5] [5]
Q.6	a) b)	OR What are protein motifs? Explain them. Discuss the primary and secondary structure of protein.	[5] [5]

(iii) Iodine is obtained from:

#### No. of Printed Pages: 2

#### SARDAR PATEL UNIVERSITY

B. 52. 5th Semester Examination
BOTANY-US05CBOT01

(Algae, Bryophytes, Pteridophytes and Gymnosperms)

Date7/11/17 Day- Tuesday

Time: 10.00 a.m.-1.00p.m.

Q-1 Choose the correct answer
(i)Filaments are unbranched in:
(a)Zygnema (b)Spirogyra (c)Oedogonium (d)all the above

(ii) Fusion between motile gametes of unequal sizes is:
(a)isogamy (b)anisogamy (c)dichogamy (d)hologamy

(d)nologani

(a)Sea weeds (b)bryophytes (c)gymnosperms (d)fungi

(iv) Elaters are not found in the capsule of:
(a) Marchantia (b) Riccia (c) Pellia (d) Anthoceros

(v)The sporophyte of which of the following bryophytes is considered to be the most advanced?

(a)Riccia (b)Pellia (c)Sphgnum (d)Funaria

(vi) Heterospory is found in:
(a) Pteris (b) Equisetum (c) Isoetes (d) Lycopodium

(vii)Which light is most effective in inducing apogamous development of sporophyte?
(a)far-red (b)green (c)red (d)white

(viii)The elementary process of the telome concept is:

(a)curvation (b)reduction (c)overtopping (d)all the above

(ix)How many cells are present in the male gametophyte of *Ephedra* at the time of pollination? (a)3 (b)4 (c) 5 (d)7

(x)Endosperm in gymnosperms is:
(a)haploid (b)diploid (c)triploid (d)tetraplo

(c)triploid (d)tetraploid

Q-2 Attempt any ten.	(20)				
(i) Differentiate between zoospore and aplanospore.	` /				
(ii) Describe: Haplontic life cycle in algae.					
(iii) In what ways can algae be of use to space travelers? (iv) Arrange the following cells in the archegonium of bryophytes from base to apex-					
(v) Name the types of vegetative reproductions are found in bryophytes.					
(vi) Write about medicinal uses of bryophytes.					
(vii) Name the elementary processes responsible for the development of higher vascular plants.					
(viii) Draw the <i>Ophioglossum</i> plant and label it.					
(ix) Define:Heterospory.					
(x) Differentiate monoxylic and pycnoxylic wood.					
(xi) Draw a labeled diagram of gymnosperm ovule.					
(xii) Describe fertilization in <i>Cycas</i> .					
Q-3 Write an illustrated account of the habitat of the algae.	(10)				
OR	(10)				
Q-3 Write an account of economic importance of algae.	(10)				
Q-4Give a comparative account of the sporophytes of Marchantia, Anthoceros and Funaria	ı.(10)				
OR	(20)				
Q-4 (a)Write a note on classification of Bryophytes.	(06)				
(b) What is the difference between elaters and elaterophore?	(04)				
(o) 11 22 22 22 22 22 22 22 22 22 22 22 22	(* ')				
Q-5 What is apospory? Give its special features and describe various factors that affect this					
phenomenon in pteridophytes.	(10)				
OR	(10)				
Q-5 Write about the external characters and reproduction in <i>Dryopteris</i> .	(10)				
Q 5 White about the external characters and reproduction in 20 yopics is.	(10)				
Q-6 Describe: Pollination and fertilization in gymnosperms.	(10)				
OR	(10)				
Q-6 Write about the economic importance of gymnosperms.	(10)				



Printed pages: 02

#### SARDAR PATEL UNIVERSITY

#### B.Sc. (Fifth Semester) Examination, November-2017 Subject: BOTANY

(Paper: US05CBOT02- Structural, Adaptive and Reproductive Biology of Angiosperms)

	ate: 9/11/17 nursday		Time:10.00 a.m 1.00 p.m. Marks-70	
).	-1 Multiple Choice Questions		(10)	
	(i) Which of the following has co (a) Gujarat University (c) Saurashtra University	ontributed much to plant anatom (b) Sardar Patel University (d) M.S. University	y research?	
•	<ul><li>(ii) Which of the following is/are</li><li>(a) xylem</li><li>(b) (c)both (a) and (b)</li></ul>	e involved in transport of food in (b)phloem (d) neither (a) nor (b)	plants?	
	<ul><li>(iii) Velamen tissue is found in?</li><li>(a) Tubers of potato</li><li>(c) roots of Vanda</li></ul>	(b) rhizomes of Osmunda (d) fruits of coconut		
	(iv) Avicennia is known for Res (a) prop roots (c) respiratory roots	piratory roots are produced by: (b) photosynthetic roots (d) clinging roots		
	(v) Salt glands are present in: (a) Tamarix (c) Turmeric	(b) Tamarind (d) All the three		
	<ul><li>(vi) Sunken stomata is a common</li><li>(a) epiphytes</li><li>(c) hydrophytes</li></ul>	n feature of: (b) halophytes (d) xerophytes		
	<ul><li>(vii) Vivipary is often found amor</li><li>(a) halophytes</li><li>(c) mesophytes</li></ul>	ng: (b) hydrophytes (d)xerophytes		
	(viii) A fruit wherein fruit wall an (a) carcerule (c) cypsela	nd seed coat are fused in called: c (b) caryopsis (d) cremocarp	aryopsis is a fruit:	
	(ix) Pappus is a modification of c (a) hypogynous ovary (c)both (a) and (b)	alyx which is associated with: (b) epigynous ovary (d) neither (a) nor (b)		
	(x) In a culture medium which o (a) thiamin (c) solid carbon	ne of the following is a source of (b)sucrose (d) all the three	carbon?	

Q-2 Answer in brief to any ten of the following:

(20)

(a)	How plant anatomy can be employed in forensic science?	
(b)	Name any two plant embryologists with their affiliation and one major contribution.	
(c)	With the help of a simple sketch, explain apical cell theory.	
(d)	What are histogens? Where are they located? What is their significance?	
(e)	What are Raphides? Where are they found? How are they different from starch grains?	
(f)	What is vascular cambium? What is its major function?	
(g)	Why xylem and Phloem are known as complex tissues? How do primary xylem and primphloem differ each other?	ary
(h)	What is pollen viability? What are the factors that influence this property?	
(i)	What is parthenocary? What are its advantages?	
(j)	What are the Photosynthetic roots? Where are they found?	
(k)	List various changes occurred during the development of fruit from ovary.	
(1)	With the help of suitable examples, list the adaptations of seeds and fruits for wind dispersal.	
Q-3 V	Vith the help of neatly labelled diagrams, describe secondary growth in dicot roots.  OR	(10)
Q-3 V	Vith the help of labelled diagrams, describe the structure and function of endodermis.	(10)
Q-4 V	Vrite a comparative account of anatomical adaptations of hydrophytes and xerophytes.  OR	(10)
•	Sive notes on:	/a=1
	a) Salient features of Phloem in deciduous trees. b) Structure and function of cuticle.	(05) (05)
,	b) Structure and randion of Gallors.	(,
Q-5 ł	lighlighting their applications, explain in detail about embryo culture and parthenocarpy  OR	(10)
Q-5 \	Write an account on the major contributions of P.Maheshwari in Plant embryology	(10)
Q-6 [	Describe the important features of pollen-pistil interaction.  OR	(10)
•	Write notes on: (a) Pollen allergy (b) Dehiscent fruits	(05) (05)

## No. of Printed Pages: 2

# SARDAR PATEL UNIVERSITY B. Sc. - T \sim, BOTANY-US05CBOT03 (Microbiology, Microtechnique and Forestry) 11/11/2017, Saturday Time-10.00 a.m

11/11/2017, Saturday	Time-10.00 a.m1.00 p.m.
	Marks-70
Q.1 Choose the correct answer	(10)
(i) Bacteria were first discovered by (a)A.V.Leewenhoek (b)N.D.Zinder	(c)Robert Koch (d)Robert Hooke
(ii) Gram stain is used in the study of  (a) Bacteria (b) Fungi (c) Bryoph	hytes (d) None
	ormed on which bacteria? monella eurella pestis
(iv)Formaldehyde is an example of: (a)Fixative (b)Dye (c)Stain	(d)none of these
(v)The knife of sliding microtome is made up (a)Glas (b)Diamond (c)Steel (d)	of )none of the above
(vi)Periodic Schiff's staining is used to mark_ (a)Collagen (b)Cell nuclei (c	c)Carbohydrates (d)Cytoplasm
(vii)An example of extremely durable wood is (a)Bombax ceiba (b)Albizzia lebbeck (c	s c)Sterculia urence (d)Tectona grandis
	a monosperma miphora wightii
(ix)Resin is obtained from  (a)Pinus (b)Cycas (c)Sabei grass	(d)Acacia nilotica
(x)Cellulose fibers mechanically separated fi (a)charcoal (b)drugs (c)paper	from wood and fiber crops are used to make er (d)none of these
	CPT

		,
Q-2 Short answer questions(any ten).	(20)	
(a)What are gram positive bacteria?		
(b) Explain: Viruses are as living or non-living.		
(c) Give economic importance of Actinomycetes.		
(d)What is meant by vital staining?		
(e)What are killing and fixative?		
(f)Write the uses of acetocarmine stain.		
(h)Define:Maceration.		
(i) Write differences between wood and timber.		
(i) What is the characteristic of ply wood?		
(k) Write the sources of Lac shellac.		
(l)Write the economic importance of isabgol.		
	(0.6)	
O-3 Write notes on: (a)Actinomycetes.	, ,	•
(b)cell-wall structure of gram positive and gram negetaive bacteria.	(04)	
OR	(04)	
Q-3Describe (a)Transmission of viruses.	` '	
(b)Economic importance of bacteria.	(00)	
O 4(a)What is the significance of dehydration and infiltration?	(05)	
(h)Write on: Maceration	(05)	
OR		
	(04)	
(b)Carnov's fluid.	• •	
(c) Whole mount preparation.	(03)	
O_5Write a brief note on sources .uses and properties of dyes.	(10)	
OR		
•	• •	
(b)Tannins.	(05)	
O-6 Briefly explain regarding natural rubbers.	(10)	
OR	/4.63	
Q-6 Write a note on composite wood and improved wood.	(10)	
<b>V</b>		
	(a)What are gram positive bacteria? (b)Explain: Viruses are as living or non-living. (c)Give economic importance of Actinomycetes. (d)What is meant by vital staining? (e)What are killing and fixative? (f)Write the uses of acetocarmine stain. (g)Explain:Seasoning. (h)Define:Maceration. (i)Write differences between wood and timber. (j)What is the characteristic of ply wood? (k)Write the sources of Lac shellac. (l)Write the economic importance of isabgol.  Q-3 Write notes on: (a)Actinomycetes. (b)cell-wall structure of gram positive and gram negetaive bacteria.  OR Q-3Describe (a)Transmission of viruses. (b)Economic importance of bacteria.  Q-4(a)What is the significance of dehydration and infiltration? (b)Write on: Maceration.  OR Q-4Write notes on: (a)Process of embedding. (b)Carnoy's fluid. (c)Whole mount preparation.  Q-5Write a brief note on sources, uses and properties of dyes.  OR Q-5Explain: (a)Fibers. (b)Tannins.  Q-6 Briefly explain regarding natural rubbers.  OR	(a) What are gram positive bacteria? (b) Explain: Viruses are as living or non-living. (c) Give economic importance of Actinomycetes. (d) What is meant by vital staining? (e) What are killing and fixative? (f) Write the uses of acetocarmine stain. (g) Explain: Seasoning. (h) Define: Maceration. (i) Write differences between wood and timber. (j) What is the characteristic of ply wood? (k) Write the sources of Lac shellac. (l) Write the economic importance of isabgol.  Q-3 Write notes on: (a) Actinomycetes. (b) Economic importance of bacteria.  Q-4 (a) What is the significance of dehydration and infiltration? (b) Write on: Maceration.  Q-4(a) What is the significance of dehydration and infiltration? (b) Write on: Maceration.  Q-4 Write notes on: (a) Process of embedding. (b) Carnoy's fluid. (c) Whole mount preparation.  Q-5 Write a brief note on sources, uses and properties of dyes. (b) Tannins.  Q-6 Briefly explain regarding natural rubbers.  Q-6 Briefly explain regarding natural rubbers.  QR

[34] SEAT NO.\_\_\_\_

No. of Printed Pages: 2

#### SARDAR PATEL UNIVERSITY

Bsc. (5th SEMESTER) Botany

Subject code - US05CBOT04 - ECOLOGY OF PLANTS

Date: 13/11/2017

Day:Monday

Time: 10:00a.m. To 1:00 p.m.

Total Marks: 70

Note: \* All questions are to be attempted. \*Figures to the right indicate marks.

Q. 1 10 Choose the most appropriate answer. 1. The collective and continuous growth of plants in space is called... 1. Flora 2. Vegetation 3. Species 4. Factor 2. The study of the ecology of an individual or a particular species of organism called.... 1. Synecology 2. Population ecology 3. Autecology 4. Habitat ecology 3. The % distribution of species among the various live form of a flora is called. 1. Biological clock 2. Biological spectrum 3. Life form 4. Ecotone 4. Organism that occupy the same ecological niche in different geographical regions is known as 1. Ecological equivalents 2. Standing crops 3. Standing state 4. Ecotone **5.** Flowering in plants is affected by temperature through..... 1. Thermoperiodism 2.Precipitation 3. Photoperiodism 4. Humidity **6.** Ozone layer is present in 1. Stratosphere 2. Atmosphere 3. Lithosphere 4. None of the above 7. Soil erosion can be prevented by 1.Deforestation 2. Raising forests 3.Excessive use of fertilizer 4.Overgrazing 8. Major source of formation of soil is.... 1. Snow covered mountain 2. Riverbeds 3. Rocks 4. Volcanoes 9. The occurrence of relatively definite sequence of communities over a period of time in the same area is known as ..... 1. Competition 2. Colonization 3. Composition 4. Ecological sccession 10. In a barren there reach the seeds and propagules of the species, this is known 1. Migration 2. Ecesis 3. Dominance 4. Composition as....

Q. 2	Give brief answer the following questions (Any ten).	20			
	1. Define: Autecology.				
	2. What is role of producers in ecosystem?				
	3. What is energy flow? Explain it.				
·	4.Explain: Dessication.				
	5. What is humidity in the air?				
	6.Explain: Seed dormancy.				
	7. Why water is essential for life?				
	8. Write full form of CPCB and COD				
	9.Explain- Climate.				
-	10.Define: Dominance.				
-	11.Define: Migration				
	12.Define: Species diversity.				
Q. 3	Explain the following terms.				
	Conservation, Standing state, Adaptation, Succession, Pollution	10			
	OR	10			
0.3	Explain the following terms.				
Q. 3	Ecades, Population ecology, Ecological pyramid, biome, Biomass,	10			
Q. 4	Write short note on:				
	1. The effects of light on plants	06			
	2. Liebig's law of Minimum	04			
	OR				
Q-4.	Write short note on:				
	1. Biological clocks	06			
	2. Ecological indicators	04			
	av av sagarat materiots				
Q. 5	What is natural resources? Give an account of forest resources.	10			
	OR				
O #	Explain it:				
Q. 5	1. Soil profile.				
	2. Biological method of soil conservation.	05			
0.6		05			
Q. 6	Explain the following:				
	1. Quadrate method.	05			
	2. Fresh water ecology.	05			
	OR	05			
Q. 6	Explain the following:				
	1. Marine ecology.	05			
	2. Transect method.	05			

SZ.

No. of Printed Pages: 02 T297 SARDAR PATEL UNIVERSITY 5<sup>th</sup> Semester **BOTANY-US05CBOT05** (Cytology and Genetics) Date-15/11/17 Time:10.00 a.m.- 1.00 p.m. Day-Wednesday Marks-70 Q-1 Multiple Chioce Questions. (10)(i)Non-histone protein contains aromatic aminoacids: (a)Lysine (b)Leucine (c)Arginine (d)Tryptophan (ii) Nucleolus was first discovered by: (a)Watson (b)Robert Brown (c)Waldyer (d)Fontana (iii) What corresponds to a single genetic locus? (a)centromere (b)chromonema (c)chromatin (d)chromomere (iv)Chlorophyll is present in: (a)membrane (b)ribosome (c)stroma (d)grana (v)Which of the following organelles contains its own DNA? (a)chloroplast (b)mitochondria (c)nucleus (d)all of the above (vi)Which of the following is not a membrane bound cell-organelle? (a)Lysosome (b)Chloroplast (c)Mitochondria (d)Ribosome (vii)The gene for red green colour blindness is located on: (a)Y-chromosome (b)X-chromosome (c)Autosome (d)none of these (viii) Variation in chromosome number in organisms is known as: (a)ploidy (b)polyploidy (c)euploidy (d)nullisomy (ix) Bar-eye in Drosophila is an example of: (a)inversion (b)deletion (c)frame-shift mutation (d)duplication (x) Kappa-particles in Paramecium are an example of: (a)cytoplasmic inheritance (b)nuclear inheritance (c)particulate inheritance (d)sex-linked inheritance Q-2 Answer the following (any ten). (20)(a)Define-endomitosis. (b)Write the functions of nucleus. (c)Mention the names of single membrane cellular organelles. (d)Briefly describe the structure of primary cell-wall. (e)Name the enzymes of peroxisome. (f)Explain:Rough ER, (g)Differentiate between 70s and 80s ribosomes. (h)Write the silent features of sex-linked inheritance. (i) What is Kappa particle? What is its role?

(i)Define-Mutation.

(I)Define-deletion.

(k) Differentiate between transition and transversion.

Q-3Define:Chromosome. Write an account of special types of chromosomes.	(10)
OR	
Q-3Describe the ultra structure and functions of nucleus.	(10)
Q-4Write the ultra structure of: (a)Peroxisome	(05)
(b)Ribosome	(05)
OR	
Q-4Describe the ultra structure and functions of mitochondria.	(10)
Q-5Describe: Shell coiling in Snail.	(10)
OR	
Q-5Describe following: (a)Cytoplasmic influence in snail	(05)
(b)Plastid inheritance in 4 o'clock plant	(05)
Q-6Describe the process of transcription.	(10)
OR .	
Q-6Describe the structure and semi conservative method of DNA-replication.	(10)

\_\_\_ X \_\_\_

### No. of Printed Pages :02

[22]

Sardar Patel University

B. Sc. (Semester - 5) External Examination US05CBOT06 - PLANT PHYSIOLOGY

Friday, 17 <sup>th</sup> Novem	nber 2017; Time: 10.00 a.m. to 1.00 p.m.
N.B.: Figures in the right indicate n	narks. Max Marks: 70
Q1. Select the appropriate answer f	or the following multiple choice questions: (10)
(i) Plant hormones are also ki	nown as:
(a) Growth factors	(b) Growth regulators
(c) Phytohormones	(d) All of these
(ii) have been demons:	two tool to be affect to be a second
(a) Gas	trated to be effective in breaking dormancy of seeds (b) Auxins
(c) Cytokinins	(d) None of these
( ) = 3 = 0 = 1	(d) None of these
(iii)The term	was suggested to designate the response of
S to totalite telisi	th of the day and night
(a) photoperiod	(b) photoperiodism
© photonasty	(d) de-etiolation
(iv) The PR and PFR are two	different forms of
(a)Phytochrome	(b) Cytochrome
(c) Chlorophyll	(d) None of these
(a) Meristem (c) Cambium  (vi) Stress caused by flood is or	n dry weight is the definition of  (b) Growth  (d) Parthenocarpy  ne type of:
(a)Biotic stress	(b) Abiotic stress
(c) Osmotic pressure	(d) None of these
or senescence such as adscis	of ageing and senescence and some specialized aspects ssion is known as  (b) Phytochemistry
(c) Phycology (d	i) None of these
(VIII) Which one of the following	g is known to retard senescence?
(a) Ethylene	(b) GA
(c) Auxin	(d)All of these
(ix)Which one of the following is	s known for promoting senescence?
(a) Kinetin	(b) Ethylene
(c) GA	(d)All of these
(x) Foolish seedling is correlated	with:
(a)Auxin	(b) GA
(c) ABA	(d) Kinetin

## Q2. Answer any TEN of the following questions in brief:

 $(10 \times 2 = 20)$ 

- (i) Naming the principal naturally occurring auxin in all the higher plants, list out the names of other naturally occurring auxins
- (ii) Comment upon the statement, "Phytohormone has been found to control a number of physiological responses of the plants'.
- (iii) Write a brief note on long day plants.
- (iv) Give four examples of DNP.
- (v) Write a brief not on flowering hormone
- (vii) What is senescence?
- (viii) Explain the term 'stress' and present the list of various environmental stresses.
- (ix) Differentiate between capacity adaptation and resistance adaptation
- (x) Briefly discuss the importance of Post Harvest Physiology
- (xi) Define 'Vernalization'.
- (xii) What is the role of Plant Physiology in agriculture/2

Q3 What are plant hormones? Discuss the physiological roles of (a) Ethylene and (b) GA  OR	(10)
Q3 (a) Write an explanatory note on factors affecting growth (b) Briefly explain the mode of action of auxin	(06) (04)
Q4 Write an essay on Photoperiodism.  Q4 Write an essay on phytochrome.	(10)
Q5 (a) Discuss the types of senescence and add a note on the biological significant senescence?  (b) Write a note on sequential leaf senescence  OR	(10) nce of (06) (04)
Q5 (a) Write in detail about physiological and biochemical changes during senescence.  (b) Present a concise note on the flower senescence.	(07) (03)
Q6 (a) Discuss in detail the types of Abiotic environmental stresses.  (b) Write a brief note on mechanism of stress injury  OR	(07) (03)
Q6 (a) How plant develops resistance against stress?  (b) How plants respond to the stress due to temperature?	(05) (05)

10

No of printed pages: 02 +06

#### SARDAR PATEL UNIVERSITY

B.Sc. (Semester-V) Examination (CBCS)(Regular & NC) Tuesday, 7th November-2017 10:00 a.m. to 01:00 p.m. US05CCHE01:- Organic Chemistry

Total Marks:70

Note: 1. Figures on the right side indicate the marks. 2. All questions are to be attempted. Choose the correct option for the following: Q.1

Which of the compound have the properties of 20 aliphatic amine? (b) Thiophene (c) Pyrolle (d) Pyridine

(a) Pyrrolidine

2 ... The Electrophilic substitution reaction in pyridine takes place at..... (b) Position-2 (c) Position-4 (d) None of these.

(a) Position-3

materials of the School of Carrier 3 Which of the following is not a six member heterocyclic?

(a) Picoline (b) Piperidine (c) Furan (d) Pyridine

र क्रिके क्रीनार नाम १९०० व्यक्त हुन्युना रहा

How many NMR signals would you expect from m-Xylene?

(c) 3 1. 1 1<sub>3</sub> 2 threshed (d) 5 (24 (3) 24 (3)

5 How many CMR signals would you expect from n-hexane?

(a) 3

(b) 2

(c) 4

How many isomers are possible for  $C_2H_4Cl_2?_{\text{started}}$  and the started possible for  $C_2H_4Cl_2?_{\text{started}}$  and the started possible for  $C_2H_4Cl_2?_{\text{started}}$ 

(a) 1 Allowing with (b) 3 Applyone than (c) 2, Applyon (c) 4. Applyone (c) 2.

Committee was seed and congress and the description of the specific was that have given you was and

Which of the following diene is more stable?

(a) 1,3- Butadiene (b) 1,2- Butadiene (c)1,2- Pentadiene (d) None of these Control of the Communication of the Control of the

Which of the following is the example of Co-polymer? 8

(a) PVC

(b) SBR (c) Orion

(d) Plexiglas

is the detergent of ampholytic class?

(a) Igepon-T

(b) SLR

(c) Miranol C<sub>2</sub>M (d) None of these.

Consider the transfer and analysis of the state of the st Which of the following insecticide is the derivative of carbamic acid?

(a) Baygon

(b) Heptachlor

(c)Malathio

(d) Ferbum

Answer the following: (Attempt any Ten). it is the set of the state of the state of the set of the state of the set of th 0.2

- Discuss the Chichibabin reaction.
- Describe the structure of furan.
- Give the synthesis of 1-Methyl isoquinoline from benzene by using Bischler-Napieralski 3 designately the waters of the sequential light and state of the synthesis.
- Give the various aspects of CMR spectroscopy.
- 5 Differentiate between Diastereotopic proton and Enantiotopic proton.
- Explain the shielded and deshielded protons.
- 7 Explain the term Hyperconjugation in Propylene.
- What is Vulcanization? Why rubber is vulcanized?
- Why propylene is 2.7 Kcal more stable than ethylene?
- Write the application of stomach insecticide. 10
- What is fixative? Describe the function of fixative. 11
- Give the application of detergent used as scouring agent.

				. :	
	1 4,1	aga r	Give the detail step synthesis of 5, 6-benzoquinoline from 2-amino naphthalene by	4	
Q.	.3	A	Skraup synthesis.		
		В	Why electophilic substitution reaction of pyrrole takes place at position-2 but not at the	3	
		ט	nocition-3?	_	
		С	Give the synthesis of 1-phenylisoquinoline from the toluene.	3	
			OR		
			A linhatic amine and pyridine Give	4	
Q	.3	Α	Arrange the increasing basicity order for the pyrrole, Aliphatic amine and pyridine. Give		
		D	detail explanation of your answer.  Why nucleophilic substitution reaction in pyridine is preferred at the position-2 and	3	
		В	modition 4 but not at position-39 Explain		
		С	Give the synthesis of 3,5-Dicarbethoxy-2,4,-dimethyl pyrrole from α-amnio-p- ketoester	3	
		Ŭ	and ethyl acetoacetate by Knorr- Pyrrole synthetic route.		
			Deduced the etructure of	10	
Q	).4		What is NMR? Give the various aspects of NMR spectroscopy. Deduced the structure of	10	
			compound having following spectral data. Label all kinds of a protons / carbons and gives		
			appropriate explanation for the structure.  (I) Molecular formula: C <sub>10</sub> H <sub>12</sub> O		
			IP (om <sup>-1</sup> ): 3100 3000 2950 2900 1670 1620 1370 1385 750		
			NMR(8, ppm): (a) Doublet, 6H,1.25 (b) Multiplet, 1H, 3.4 (c) Singlet, 5H, 7.67		
		•	and of the control of the control of the page that the come and the		
			(II) Molecular formula: C <sub>6</sub> H <sub>13</sub> N		
			CMR(δ, ppm): (a) 22.7, Quartet (b) 31.5, Doublet		
			(c) 35.8, Triplet (d) 46.9, Triplet		
			OR		
,	Q.4		Discuss splitting of NMR signal by spin-spin coupling. How will you differentiate	10	
`	Į. <del>4</del>		geometric isomers by using CMR spectroscopy? Draw all possible isomers for the		
			formula C.H., and tell how many CMR signals do you expect from each of the isomers?		
			Deduce the structure of compound having following spectral data, label all kinds of		•
			carbon/protons and give appropriate explanation for the structure.		
			Molecular formula: C <sub>10</sub> H <sub>12</sub> O <sub>3</sub>		
			NMR (δ, ppm): (a) 1.35,3H, Triplet (b) 4.35,2H, Quartet (c) 3.8,3H, Singlet (d) 7.5,4H, Quartet		
			(c) 3.6,311, Singlet (d) 7.5, 123, Quarter		
1	Q.5	Α	Give the mechanism for polymerization of styrene in presence of sodium metal and	4	
•	Q.D		nonhthalene		
		В	Give the distinguishing features of addition and condensation of polymerisation.	3	
		C	What are plastics? Give there classification and discuss its properties.	3	
			The state of the s		
js er			OR Write about the role of Ziegler-Natta catalyst in the synthesis of polymers.	. 4	
	Q.5	A	Discuss the addition of HCl to 2, 4-Hexadiene.	3	
		B C	Give the advantages of Co-ordination polymerization over free radical polymerization.	3	
			,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的。		
	Q.6	A	Give the synthesis of DDT with its advantages.	4	
	•	В	Write note on: the principle of cleansing action of detergents.	3	
		С	Give the synthesis of coumarin from the cheapest raw materials.	3	
			ge verg des burers of the control of had properly a separate described as the control of the con		
			of the state of the state of the state and discuss in detail any one	4	
	Q.6	A	of them.	-	
		В	Of them.	<b>6</b> ,	
		D	(i) Compound which gives floral soap and detergent.		
			(ii) Compound containing triazole moiety which used as whitening agent.		

#### CHARACTERISTIC PROTON CHEMICAL SSUPE

Type of pr	otos	Chemical shift, ppu	3	in the	e erre e j
<u>, , , , , , , , , , , , , , , , , , , </u>		3	Andrew Brings.	Matter Color of the South	100
Cyclopropene		0.2		to the second of the second	for Media
Primary	RCH,	<b>6.9</b>			e for the parameter
Secondary	R,CH,	4.3.	•	The second of the second second	
Tertiary	R,CH	1.5			网络人名 人名英格兰
Vinylic	O-C-H	4.6-5.9		· ·	installed a subject of
Acctylenic	CMC—H	2-3		<i>,</i>	4.4 (*)
Aromatic	<b>∧ı—¥</b> [	6- <b>3.</b> 5		** * .	
Benzylic	Ar-C-H	2.2-3		\$ - · · ·	
Allylic	C-C-CH,	1.7			
Fluorides	HCF	4-4.5		w. 5.7.	
Chlorides	HC-CI	3-4		•	•
Bromides	HC-Br	2.5-4			
Todides	HC-I	2-4			
Alcohols	нс-он	3.4-4			
Ethers	HC-OR	3.34		-	
Esters	RCOO-CH	3.7-4.1		•	
Esters	HC-COOR	2-2.2			
Acids	HC-COOH	2-2.6	•		
Carbonyl compounds	BC-C-O	2-2.7	•	•	
Aldebydic	R- CHO	910		a	
Hydroxylic	R - OH	1-5.5		on the state of the party	
Phenolic	Ar - OH	4-12		Secretary of the	in language
Enelic	C-C-OH	15-17			
Carboxylic	RCOOH	10.5-12	4	•	
Amino	RNH,	1-5		e e e e e e e e e e e e e e e e e e e	الإسراء الأس

#### Characteristic Infrared Absorption Frequencies

Bond	Compound type	Frequency range, can
C-H	Alkanes	2850-2960
		1350-1470
C-H	Alkenes	3020-3080 (m)
		675-1000
C-H	Aromatic rings	3000-3100 (m)
		675-870
С-Н	Alkynes	3300
C=C	Alkenes	1640-1680 (v)
C=C	Alkynes	2100-2260 (v)
C≂C	Aromatic rings	1500, 1600 (v)
C-O.	Alcohols, ethers, carboxylic acids, esters	1080-1300
C=O	Aldchydes, ketones, carboxylic acids,	1690-1760
	esters	
O-H	Monomeric alcohols ,phenols	3610-3640 (v)
÷	Hydrogen bonded alcohols, phenols	3200-3600 (broad)
	Carboxylic acids	2500-3000 (broad)
N-H	Amines	3300-3500 (m)
C-N	Amines	1180-1360
C≌N	Nitriles	2210-2260 (v)
$NO_2$	Nitro compounds	1515-1560
314		1345-1385



#### Characteristic absorption for dienes

1,5
214
253
+30
+5 -
45
+0
<b>.</b> +€
+ 30
<b>-</b> 1- \$
4-60
+0
$\lambda_{cate} = Total$

## Characteristic absorption for substituted benzene derivatives

	Arcor/archo/arco,h/ai	CO <sub>2</sub> R	λ <sub>max</sub> (nm)
Parent	chromophore: Ar = C6H5		•
G ≂	Alkyl or ring residue, (e.g., A	irCOR)	246
	H, (ArCHO)		250
G ∓	OH, OAlk, (ArCO,H and Arc	CO₂R) .	230
increm	ent for each substituent on Ar kyl or ring residue		****
	ryr di ing residue		-
O14	-OCH <sub>2</sub> , -OAlk	p-	+10
V.,	COIII, WAIK	法国际 医二氯基苯酚 网络斯坦	+7
0~ (	oxyanion)	p.	1 25
~ (	on i moon	0	4.11
		4 4	+ 20
-C!		p-	+ 78*
-C,		0-, m-	
$-\mathbf{B}r$		•	4 10
124		0-, m-	4.2
<b>NO</b> 1			+ 15
NH,		, o-, m-	4 13
3375-	.0.011	<b>p</b> -	4.58
MHC	OCH,	o-, m-	+ 20
		р-	+ 45
- NHC		p-	± <b>7</b> 3
N(C∤	₹ <sub>3</sub> ),	o-, m	+ 20
	•	p-	$\pm 85$

# Characteristic absorption for α, β-unsaturated carbonyl compounds

Enone Dicnon  Base values  Acyclic α,β-unsaturated ketones  Six-membered cyclic α,β-unsaturated ketones  Five-membered cyclic α,β-unsaturated ketones  α,β-Unsaturated aldehydes  α,β-Unsaturated carboxylic acids and esters  Increments for  Double bond extending conjugation	= C — C := (nm) (nm) 215 215 202 210 195
Base values Acyclic α,β-unsaturated ketones Six-membered cyclic α,β-unsaturated ketones Five-membered cyclic α,β-unsaturated ketones α,β-Unsaturated aldehydes α,β-Unsaturated carboxylic acids and esters Increments for	(nm) 215 215 202 210
Acyclic α,β-unsaturated ketones Six-membered cyclic α,β-unsaturated ketones Five-membered cyclic α,β-unsaturated ketones α,β-Unsaturated aldehydes α,β-Unsaturated carboxylic acids and esters Increments for	215 202 210 195
Six-membered cyclic α,β-unsaturated ketones  Five-membered cyclic α,β-unsaturated ketones α,β-Unsaturated aldehydes α,β-Unsaturated carboxylic acids and esters  Increments for	215 202 210 195
Five-membered cyclic α,β-unsaturated ketones α,β-Unsaturated aldehydes α,β-Unsaturated carboxylic acids and esters Increments for	2)0 195
α,β-Unsaturated addenydes  a,β-Unsaturated carboxylic acids and esters  Increments for	2)0 195
α,β-Unsaturated aidehydes  a,β-Unsaturated carboxylic acids and esters  Increments for	2)0 195
a,β-Unsaturated carboxylic acids and esters  Increments for	2)0 195
a,β-Unsaturated carboxylic acids and esters  Increments for	195 sedi 195 sedi 196
The State of the S	The state of the s
The State of the S	The state of the s
Increments for Double bond extending conjugation	+30
Double hand extending conjugation	+30
	4.30
Alkyl group, ring residue a	. 10
B	4.10 4.10
y and higher	2.12
Polar groupings: -OH o	435
ß	+30
$\boldsymbol{\delta}$	₹ 50
$OAc  \alpha, \beta, \delta$	+ 6
OMe a	+35
β	+30
γ	+17
8	` +-31
SAlk β	4.85
er e	<b>#15</b>
$oldsymbol{eta}$	412.
—Br a	+25
and the party of $oldsymbol{eta}$	+30
-NR, A	+ 9.5
Exocyclic double bond	4.5
Homodiene component	139

4

<sup>13</sup>C shifts for terminal and internal systems

*****	Y Y
p and	BB
Torres	Leformel

o ex manual as a		ACTIVITY TOUG			
· ·			β		
Tennénsi	Internal	Temples	internal		
+ 9	+ 6	+10	+ 8	-2	
+20	•	+ 6	-	-0:	
+ 45		+ 5,5		-3	
+21	+16	<b>+ 3</b>	+ 2	-2	
+25			_	-2	
+20		•	-	~2	
			_	-	
+22		+ 25	` **	~O.	
+30	+ 24		4 3	-2	
+31			, .	2	
+23	+17	-	+ 7	-2	
448			•	-5	
4 5A	+51	+ 8	-	-4	
+51	+45	+ 6	•	<b>3</b>	
+29		_	•	-5	
			• •	-5	
+37	+ 31	+ 8	•	· d	
+42		+ 6	•	<b>.</b> ~3	
+31		+ 5		-7	
+63	+ 57	+ 6	+ 4	√ .	
+ 4	+ 1	+ 3	-	3	
+11	+11	+12		~4	
+20		+ 7		~. j	
+ 68	+ 63	+ 9	+ 6	4	
+31	+32	+11	-	-4	
+20	+25			-3	
- 6	+ 4	+11	+12	~1	
	+ 9 + 20 + 45 + 21 + 25 + 20 + 33 + 22 + 30 + 31 + 23 + 48 + 59 + 51 + 29 + 26 + 37 + 41 + 53 + 41 + 53 + 41 + 53 + 51 + 51 + 20 + 31 + 25 + 26 + 37 + 31 + 26 + 37 + 37 + 37 + 37 + 37 + 37 + 37 + 37	Tennenal Instanced  + 9 + 6 +20 + 45 +21 + 16 +25 + 20 +20 + 17 +33 + 28 +22 +30 + 24 +31 +23 + 17 +48 + 41 +58 +51 +51 +45 +29 +24 +26 +24 +37 +31 +42 +31 +63 +57 +4 + 1 +11 +20 +68 +63 +31 +32 +20 +25	Terminal Internal Terminal  + 9 + 6 + 10 +20 + 6 + 4.5 + 5.5 +21 + 16 + 3 +25 + 20 + 5 +20 + 17 + 3 +33 + 28 +22 + 2.5 +30 + 24 + 1 +31 0 +23 + 17 + 9 +48 + 41 + 10 +58 +51 + 8 +51 + 45 + 6 +29 + 24 + 11 +26 + 24 + 8 +37 + 31 + 8 +42 + 6 +31 + 5 +63 + 57 + 4 +4 + 1 + 3 +11 + 11 + 12 +20 + 7 +68 + 63 + 9 +31 + 32 + 11 +20 + 25 + 11	Terminal Internal Terminal Internal  + 9 + 6 + 10 + 8 + 20 + 6 + 45 + 5.5 + 21 + 16 + 3 + 2 + 25 + 20 + 5 + 3 + 20 + 17 + 3 + 2 + 33 + 28 + 2 + 30 + 24 + 1 + 1 + 31 0 + 23 + 17 + 9 + 7 + 48 + 41 + 10 + 8 + 58 + 51 + 8 + 5 + 51 + 45 + 6 + 5 + 29 + 24 + 11 + 10 + 26 + 24 + 8 + 6 + 37 + 31 + 8 + 6 + 37 + 31 + 8 + 6 + 37 + 31 + 8 + 6 + 31 + 3 + 3 + 11 + 11 + 12 + 11 + 20 + 68 + 63 + 9 + 6 + 31 + 32 + 11 + 10 + 20 + 25 + 11 + 10	

<sup>13</sup>C Shifts for some linear and branched chain alkanes

	₩ 4. Siidhahaanaan		-		
Compound	C1	C-2	C3	C-4	C-\$
Methano	-23	,	Andrew Colonies Colonies	<del>Management Asset 4</del>	*************
Ethane	5.7		,	vila.	
Propane	15.8	16.3	158	1975	Asi.
Buteno	13.4	25,2	25.2		34%
Pentane	13.9	22.8	34.7	33.0	
Hexage	14.1	23.1	32.2	32.8 32.2	13.9
Heptane	14.1	23.2	32.6		23.1
Octane	14.2	23.2	32.6	29.7	32.6
Nonane	14.2	23.3	32.6	29.9	29.9
Decene	14.2	23.2	32.6	30.0	30.3
Isoboteno	24.5	25.4	32.8	LIE	30.5
leopontane	22.2	31.1	** *		N. s
Isohoxane	22.7	28.0	32.0	11.7	
Veopentage	31.7	28.1	42.0	20.9	14,3
2.2 Dimethymatane	29.1	30.6	97.0		
Mothylpontene	113	29.5	36.9	8.9	
and discount	4115	29.3	36.9	(18.8,	_
-3-Dimethylbutane	19.5	24.2		3-CH	.)
2,3-Trimothylbutane		34.3	40.0		
3-Dimethylpeniane	27 A	33.1	38.3	16.1	:
Servin themes	7.0	25.3	36.3	(14.6,	
				3436	<b>)</b> ;

<sup>13</sup>C shifts for substituted benzenes
Base value for benzene is 128.5 ppm

	C-I (Andrea)	C.3	C-3	C-4	C of Substituted (ppm from TMS)
	the state of the s	neturnen de des de partir de la companya de la comp		0.0	
L	0.0	0.0	0,0	~29	21.3
	+9,3	+07	0.1	-2.6	292 (८५), १५४ (८५)
HCH.	+156	-05	0.0		34.4 (CH), 24.1 (CH)
ALCI).	+20.1	20	0.0	~2.5	34.5 (C), 31.4 (CH <sub>2</sub> )
	+22.7	-3.6	-0.4	-3.1	137.1 (CH), 113.1 (CH)
XCHP?	+9.1	-2A	+0.2	-0.5	5/1 (CD), 1400 (Cd)
H=CH <sub>2</sub>	-5.8	+6.9	+0.1	+04	84.0 (C), 77.8 (CTI)
CH	4 12.1	-1.8	-0.1	-1.6	
C.H.	+13.3	-0.5	-0.6	-0.4	64.5
CHPOH.	+7.7	~0.0	~0.0	~0.0	20.7 (CH.), 66.1 (CH.).
CHOCCH!	* I.I	54,54			170.5 (CO)
Ô		1.00		** *	dia di participa d
-	+26.6	-12.7	+1.6	-73	Left etg.
OH	+31.6	-144	+1.0	<b>-7.7</b>	54.1
OCH <sub>2</sub>		-9.4	41.6	~ 5.3	
oc <sub>t</sub> r,	+ 29.0				
Ŷ	,				01/1/201 \ 1607 (Cmm())
ľ	+ 22.6	-7.1	-0.4	~3.2	23.9 (CH <sub>3</sub> ), 169.7 (C∞O)
OCCH,	4 8200	and the second			
Q	**				192.0
Cii O	+8.2	+1.2	+0.6	+ 5.8	1350
		al aged data setting for the			:
Q ·	and the second second			+ 2.8	24.6 (CH <sub>2</sub> ), 195.7 (C==O)
CCH,	+7.8	-0.4	-04	7 40	Stra Country and Co.
Q			0.0	4-3.8	48/ 4 (85-m45)
CC.H.	÷9.1	+15	-0.2	7-340	IMA (Caro)
Ŷ.		4 17	+0.7	+6.7	•
ĊCF,	-5.6	+1.8	TWI		• .
9	+2.9	+13	+0.4	÷4.3	168D
COH	44.9	V 3.55			•
Q				_	CL D (COLL ) 166 8 (Comp)
1	+2.0	+12	-0.1	+ 4.B	51.0 (CH <sub>3</sub> ), 166.8 (C==O
ÇOCH,					168.5
o ca				. 70	•
ks.	+4.6	+29	40.6	+7.0	
Ç.,					
Ç				. 2 /	
CNH <sub>2</sub>	4 5.0	- 1.2	0.0	+3.4	119.5
CMEN	~16.0	+ 3.6	4 Q.6	+4.3	€ E ₹ ayî
	+19.2	-124	+1.3	-9.5	40.3
ин,	+ 22.4	15.7	<b>₹</b> 5,6	4114	40,3
in(CH <sup>γ</sup> ) <sup>γ</sup>	7 ====				
· · · · · · · ·					
NHCCH,	+11.1	~9. <del>9</del>	+0.2	-5.6	
NO <sub>t</sub>	+19.6	~53	+0.9	+6.0	129.5
Mark Care () Lend	+5.7	-3.6	+1.2	-2.8	1474
	+ 35.1	-143	+ 0.9	~4.5	
F	+6.4	+0.2	+10	-2.0	
a		+3.4	+22	-1.0	
Br	~5.4	+9.9	+2.6	-7.3	
Į.	-32.2		+ ().4	+3.4	
CP <sub>3</sub>	+2.6	-3.1	+0.2	-3.3	
SH	+2.3	+0.6		-3.6	15.9
SCH,	+10.2	-18	+0.4	+33	
SO <sub>1</sub> NH <sub>1</sub>	+15.3	-29	+0.4		
St(CH)	+13.4	+ 4.4	~1.1	- 1.1	1

Influence of functional group X on the chemical shift position  $(\delta)$  of nearby carbons in alkane chains

	المرتبون والمرتبون والمرتب	o-shift	β-shift	γ-shift	
	ХСН	хсн-     R	- x - C - R	, , ,	. ,
	, (1)	or 2°	or 3°	a galagian dari ang panggarang sang panggarang da	description of the state of the
-CH <sub>3</sub>	9	6	3	9	·». 3
R: see table 3.11					
axial—CH3	ţ	/-X1	ad (), i	5	mb.
equatorial—CH3	6	Sec.	198	9	0
(în cyclohexanes)					- : **
-CH = CH <sub>2</sub>	22	16	12	7	<b>-2</b>
-c=cH	4		***	3	3.
$-C_6H_5$ , $-Ar$	23	17	11	10	-∗43
-F	70	y-84	شن	8	7
Cl	31	35	42	10	~-5
—Br	19	28	37		~. <b>4</b>
unger 1	~ 7 to 2	0 -	<b></b> ,	11	
-NH2NHRNR2	29	24	18	11	4
-NO <sub>2</sub>	62	<b>4</b> .9	e e	3	#re. <b>5</b>
-NHCOR, -NRCOR	10	4.	. 25	0	0
-NH3*	25	715	Ψ,""	7	m=3
-(N	3	4	1416	<u>"</u>	<b>~3</b>
-SH	2	-f-me	.40	. 19 j.	10.
OH	50	45	40	9	3
-OR	50	24	17	10	-6
-OCOR	52	50	45	7	rs< <b>6</b>
-coohcoorcon	20	16	(3	2	~-3
-COR,-CHO	30	24	17	2	£
COR CHO SO <sub>3</sub> H, SO <sub>2</sub> N	50		<u></u>	3	0

Influence of functional group X on the chemical shift positions ( $\delta$ ) of nearby carbons in alkene groups and benzene rings

Base values:	ethylene (	δ 123)	and	benzene	(8 128)	
Ammont de l'annière reserve de l'annique de l'annique plus magnin des que de l'annique de l'anni	x,	C-2		Ž	ortho meta	
•	LAll	cones	L	Benz	enes	A STATE OF THE PARTY OF THE PAR
	( 1	C1	C-1 (tpso)	· ortho	meta	Nata
-CH;	10	nu.\$	ģ	0	Q	mananananan di mandisan masaya. Mananananan
8.	16	~8	15	0	0	~·2
$R, \longleftrightarrow$	23	wy. <b>8</b>	21	Q	0	2
-CH = CH,	15	~6	9	0	0	· · · · · · · · · · · · · · · · · · · ·
-CHECH	a vide	ale:	~6 ·	4	0	0
-Colley-Ar	13	-11	13	·	1	11 to \$1
mage.	25	34	35	-14	1	<b>\$</b>
CI	3	~6	Ġ	0	<u> </u>	-2
-Br	~8	enar I	<b>5</b> .	3	7	m 2
ine	-38	7	-32	10	Ng Ng	ner }
···NH <sub>2</sub>	A*** .	dav.€	18	-13	j	10
-NHR	¢+	n de q	20	- 14	1	~10·
mNK <sub>2</sub>	/**	24.0	22	1 <i>6</i>	1	10
-NO;	22	··· ]	20	<b>~</b> \$	1	6
-NHCOR, -NRCOR	ece:	4.	10	-7	1.	5.11 <b>4</b>
-CN	-15	. 15	~16	4	1	6
-SH	, min	-27/74	4	ı	1	<i>-</i> -3
-ОН	20	***	27	-13	1	_7
-OR	29	- 39	30	-15	1	<del>8</del>
-OCOR	18	~ 27	23	~6		~1 <b>2</b>
-COOH, -COOR, -CON	4	9.	2	2	0	5
-COR, -CHO	14	13	9	į	f ·	6
-so, H, -so, N	-ui-	, wax	16	0	0	4.
-PMc,		rya trya	14	1.6	0	
-PAI:	-oze	994	9	\$	0	-1 0

[44/A20]

SEAT No.\_\_\_\_

No. of printed pages: 03

#### SARDAR PATEL UNIVERSITY

B. Sc. **V**-Semester Examination Thursday, 9<sup>th</sup> November, 2017

10.00 a.m. to 1.00 p.m. US05CCHE02 - ORGANIC CHEMISTRY

Total Marks: 70

Note	te:(i) All questions are to be attempted. (	ii) Figures to the right indicate marks.
a f	and the second s	Annual Carlos de Car
Q.1	Choose the correct option for the followi	ng: 10]
(i)	The intermediate formed in Sommelet rearra	angement is known as
( )	(a) ylide (b) cation (	c) radical (d) betaine
(ii)	Using reaction carboxylic acid can b	e converted into primary amine.
	(a) Benzilic acid (b) Curtius-Schmidt (	c) Wittig (d) Beckmann
(iii)	reaction fails to proceed	in the dark.
. ,	(a) Crigee – Kasper (b) Hoffmann–Loeffler (	(c) Baeyer – Villager (d) Benzoin
(iv)	is used to kill / remove the worn	<b>18.</b> gr <sub>owski</sub> group og grover og gledding skiller
	(a) anthelmintics (b) antihistamine (	c) antipyrine (d) antimalerial
(v)	Mily Mily Mily And Mily Mily Mily Mily Mily Mily Mily Mily	
\$13 ×	(a) C (b) E	c) B (d) none of these
(vi)	(a) C (b) E ( Acedapsone belongs to	of drugs.
	(a) antibacterials (b) antileprotic	(c) antibiotics (d) none of these
(vii)	General molecular formula of terpenoid is .	(4) (011)
3	(a) $C_5H_8$ (b) $-(C_5H_8)_n$	(c) $-C_5H_8-$ (d) $-(C_5H_{10})_0-$
(viii)		ne separation of Nerol and Geraniol
	from essential oils?	(1) 0.01
9377	(a) CaCl <sub>2</sub> (b) CdCl <sub>2</sub>	(c) annydrous CaCl <sub>2</sub> (d) CuCl <sub>2</sub>
(ix)	Oestrone was isolated by	
	(-,	(c) E. Laqueur (d) both 'a' & 'b'.
(x)	Androgens are hormones.	(-) to the (-) 0 (b) (d) many of those
	(a) Female (b) Male	(c) both 'a' & 'b' (d) none of these
 	Answer the following (Attempt any ten):	
Q.2	Explain Birch reduction of toluene.	and the second of the second o
(i) (ii)	State Benzoin condensation reaction. Why	
(11)	Benzoin condensation reaction?	•
(iii)	Explain Perkin condensation reaction using	g suitable illustration.
∍(iv)	What is Drug ? What are the requirements	of an ideal drug?
(v)	Write the structure and uses of (a) Viot	orm & (b) Dimennyarinate.
(vi)		ICINAL CREMISTRY.
(vii)	) What is gem dialkyl group and its rule? G	ive its utility in the structure

(viii)	Complete and re-write the given reaction :	
	$\frac{P_2O_5}{\text{heat}}  \text{Camphor}  \frac{\text{conc. HNO}_3}{\text{conc. HNO}_3} = \frac{?}{?}$	
(ix) (x) (xi)	heat Enlist the methods used for isolation and separation of terpenoids from the plant Define : Hormones and Vitamins. Explain : $\alpha$ , $\beta$ - unsaturated carbonyl compounds are more reactive than simple alkenes.	s.
(xii)	Give classification of Hormones.	
Q.3	t market in the first transfer of the second	
[a] [b]	- CH₂ - group in Mannich base one is from formalhedyde.	4] 3]
[c]	State Benzilic acid rearrangement. Also explain that in an unsymmetrical benzil, the aryl group with electron donating character migrates faster than simple aryl group.	3]
Q.3		
[a]	Write reaction mechanism for Beckmann rearrangement and show that it is highly stereospecific reaction.	4]
[b]	Discuss Equandii ea a mana a	3]
[c]	Complete the given westign and	3]
A state	Cyclohexanone + $(Ph)_3P = CH_2$	
Q.4	Give the broad classification of drugs. Discuss the mode of action of [10] antipyretic drugs. Write synthesis and uses for Novalgin and Chloroquine.  OR	0]
	Which drug is used to cure the infections caused by cocci bacteria? Discuss the mode of action of it. Write synthesis and uses for Phenobarbitone, Warfarin and Miracil-D.	0]
Q.5	en de la companya de La companya de la co	
[a]	Discuss Wallach's oxidative degradation for elucidation of position of double bond and <i>tert</i> . alcoholic group in the structure of α-terpineol.	4]
[b] [c]	Write synthesis of Citral using Arens – Van Dorp's synthesis.  Show that Nerol is a geometrical isomer of Geraniol.  OR	
Q.5	OR in the following statement of the second statement	
[a]	Define isoprene rule and special isoprene rule. Show that special isoprene	1]
[b]	rule is a guiding principle and not a fixed rule.  Discuss the uses of following reagents for the structure elucidation of [3]	_



[c]	terpenoids: (i) Tilden's reagents (ii) alkaline KMnO₄/CrO₃ (iii) (CH₃CO)₂O Write synthesis of: (i) Terebic acid using ethyl acetoacetate. (ii) Linalool via Ruzicka synthesis.	[3]
Q.6 [a]	How will you determine the presence of steroid nucleus, three double bonds, benzenoid ring and position of phenolic —OH group in the structure of oestrone?	[4] [3]
[b]	Write synthesis of testosterone from cholesterol. Discuss electrophilic addition and nucleophilic addition reaction mechanism for $\alpha$ , $\beta$ -unsaturated carbonyl compounds using suitable illustrations. OR	[3]
Q.6		[4]
[a]	Discuss the constitution of testosterone.	[3]
[b]	Write synthesis of osterone.	[3]
[c]	Explain mechanism of Michael addition reaction using suitable illustration.	f_1
	<u> : : : : : : : : : : : : : : : : :</u>	

•

the state of the s

[36/A22]

SEAT No.

. . . . . . . . . . . . . . . . 2

#### SARDAR PATEL UNIVERSITY

**B.Sc.** (Semester-v) EXAMINATION

Saturday, 11th Nov 2017, 10:00 am To 01:00 pm Subject Code: Inorganic Chemistry (US05CCHE03)

	То	tal marks	s: 70
Q:1	Choose the correct option from the following.		10
1	The point Group of H <sub>2</sub> O molecule is		
-	(a) C <sub>2</sub> v (b) C <sub>3</sub> v (c) D <sub>3</sub> h (d) Oh		
2	Which of the following is the principle axis of rotation in [PtCl <sub>4</sub> ] <sup>2</sup> molecule?		
	(a) $C_1$ (b) $C_2$ (c) $C_3$ (d) $C_4$	•	
3	Which of the following geometries contain infinite fold axis of symmetry?		
	(a) Linear (b) Tetrahedral (c) Octahedral (d) Square Plannar		
4	The paramagnetic property of transition elements is due to	78 e 1	
	(a) fully filled configuration (b) partially filled d- orbitals		
	(c) partially filled f-orbitals (d) both 'b' and 'c'		
5	What is the value of "L" for <sup>1</sup> D?		
	(a) 4 (b) 1 (c) 5 (d) 2 Fourth postulate may be stated as follows: the time development of a wave function is given	hv .	`
6		ω <b>γ</b>	
	ih/2π • dΨ/dt = (a) $\nabla^2 \Psi$ (b) $V_{tt} \Psi$ (c) $H\Psi$ (d) $\Psi_{(t)}$		
~	The electronic degeneracy is reduced on slight of a system.		
7	(a) Vibration (b) distortion (c) improve (d) deviation		
8	is more stable chelate for the chelate ring containing single bond.		
8	(a) 4 – member in ring (b) 5- member in ring		
	(c) 6-member in ring (d) 'b' and 'c' both		
9	SN <sup>2</sup> mechanism is known as		
	(a ) substitution mechanism (b) association mechanism		
	(c) Dissociation mechanism (d) Formation mechanism		
10	In aqueous solution the concentration of water is  (a) [ 5.55 M] (b) [0.55M] (c) [555M] (d) [55.5M]	48.0	1
	(a) [5.55 M] (b) [0.55M] (c) [555M] (d) [55.5M]	111	
Q:2	Answer the following. (Any Ten )	1 1	20
1	and the second s		
2			
3	10 CC (11) CO		
4	and the second s		
	- 1		
5			
6	,		
7	and the second s		
8			
9			
10			
11	(ii) substrate		
12	1		
14	Commission sections of the second of the sec		

Q:3	Prove with Proper Example :	10
	(i) $S_n^n = E$ for $n = even number$ (ii) $S_n^{2n} = E$ for $n = odd number$ OR	
O:3	Give an account on : (i) Improper rotation	
•	(ii) Reflection	10
	(iii) Cubic Point Group	
Q:4	(a ) Discuss the splitting of d-orbitals in an octahedral field.	5
	(b) Distinguish between diamagnetism and paramagnetism.	5
	OR	_
Q:4	(a) Explain " $[V(H_20)_6]^{+3}$ is green in color".	5
	(b) For $[Fe(H_2O)_6]^{24}$ ion the mean pairing energy is 17,600 cm <sup>-1</sup> , the magnitude of $\Delta_o$ is 10,400 cm <sup>-1</sup> calculate LFSE for complex in corresponding both states, which state is more stable? why?	5
Λ.Ε	(a) Discuss the electric College (control of the control of the co	
Q:5	(a) Discuss the electron in three dimensional box.	5
	(b) calculate the energies of the lowest two quantum states for an electron confined to one dimensional box of 5 A° length. Calculate the wavelength of a radiation emitted in transition of electron from the excited state to the ground state. Given: $h = 6.625 \times 10^{-27} \text{ erg.sec}$ $m_e = 9.108 \times 10^{-28} \text{ gm}$ OR	5
Q:5	(a ) Give an account on : (i) Hermition operator (ii ) unitary operator	. 5
	(b) An electron moves in three dimensional box which has edge 1.00 cm.calculate the energy required in ergs to raise the electron from the ground state to the state where $n_x=2$ , $n_y=2$ and $n_z=1$ Given; $h=6.625 \times 10^{-27}$ erg. sec $m_e=9.108 \times 10^{-28}$ gm	5
Q:6	(a )Discuss the SN <sub>1</sub> mechanism in ligand substitution reaction in octahedral complex.	5
	(b) Discuss the continuous variation (Job's) method for the determination of composition of the complex.	5
	OR	
Q:6	(a ) Define: Trans effect. Discuss the electrolytic polarization theory of trans effect.	5
	(b) Discuss the stepwise formation of complexes, stepwise formation constant and overall formation constant. Give the relationship between $\beta_n$ and $k_1, k_2, k_3, \ldots, k_n$	5

	No	No. of Printed Pages:	02
35	SARDAR PATEL	UNIVESITY	
	B. Sc. Examination (Fift	•	
	Monday, 13 <sup>th</sup> Nove		* * * *
	-		
	Session:- Morning, Time:- 10	the control of the co	
	Subject:- Chemistry, Course	the control of the co	-
	Course Title:- Inorga	anic Chemistry-	
	(1) All questions are compulsory	Total Mark	s: 70
. (1994)	(2) Figures to right indicate full marks.	grand in the many of the property of the	F 1
Section 1			****
Q:16	Give the most correct choice to the following	multiple choice questions.	(10)
i.	Which one of the following is not Lewis acid?	?	
13.5	(a) $SO_3^{2-}$ (b) $Cu_{-}^{2+}$ (c) AIC		a din a a
ii.	According to Pearson's SHAB principle which	h metal ions occur in nature both as	1
į.	(a) $Cu^{2+}$ , $Ni^{2+}$ (b) $Ca^{+}$ , $Pb^{2+}$ (c) $Mg^{2}$	$_{5}^{2+}$ , $Ca^{3+}$ (d) $Mg^{2+}$ , $Ag^{+}$	
iii.			
Poli	(a) Liq. NH <sub>3</sub> (b) Liq. HF (c) Liq.	. SO <sub>2</sub> (d) Hydrazine	1
iv.	The strength of H <sub>2</sub> O <sub>2</sub> solution labeled as '16 ve	/olume' is	•
	(a) 10/7 N (b) 7/20 N (c) 10/1	14 N (d) 20/7 N	
V.	The number and types of bonds between two (	C – atoms of acetylide ion are	- 1
	(a) one bond $\sigma$ and one bond - $\pi$ (b) two		
	(c) one bond $\sigma$ and two bond $-\pi$ (d) one	e bond $\sigma$ and half bond $-\pi$	à .
vi.	Fused silica is also known as		
		osilicate glass (d) flint glass	
VII.	is mechanical process for making large		
	(a) Pressing (b) Jolling (c) Kno		
VIII.	The long chain vitreous sodium phosphate pol		
iv	(a) electric (b) ceramic (c) rubl		
ıx,	A is a mixture of equal proportions of l (a) silicone resins (b) silicone rubber (c) sili		
w.	Inorganic polymers are	(d) stricone greases	
46.	(a) stronger, harder & brittle than org. polymer	ers (b) stiffer than are naturals	
	(c) soluble in polar solvents	(d) all of above	
	( ) · · · · · · · · · · · · · · · · · ·	(4) 411 01 40070	
Q:2	Attempt any ten short answer type questions	s of the following.	(20)
	Liquid ammonia is best non aqueous solvent. I		(=0)
	Give the limitations Lewis concept.	•	
	Explain the term "amphiprotonic substances".		
	List uses of carborundum.		
	Page no.:	-01	. ^
		CPT	<b>'</b> ره

$\mathbf{v}$	Calculate degree of hardness of water containing 6 mg of MgSO <sub>4</sub> (M-120) per kg of wat	er.
vi.	How will you remove both type of hardness of water by washing soda?	
vii.	List the basic raw materials used for the manufacture of ceramics.	•
,viii.	Explain the term 'annealing': 1987 18 18 18 18 18 18 18 18 18 18 18 18 18	
ix.	What are pyrex glasses? The Constraint of the American Administration of the American Adminis	
x.	Give all the possible structures of imides of sulphur.	
хi.	Write the general properties of inorganic polymers.	
xii.	What are borophosphate glasses?	
0:3[	A] Discuss the periodic variations of acidic properties of oxyacids.	(5)
	B] Describe the general chemical reactions that occur in ionizing solvent.  OR	(5)
0.31	[A] Give a brief account on Usanovich concept of acids and bases.	(5)
	[B] Classify solvents in a different ways.	(5) (5)
	[D] Classify solvents in a different ways.	(5)
<b>Ω • 4</b> [	[A] Give preparation, properties, structure and uses of aluminium carbide.	(5)
	B] 20 mL of a '10 volumes' labelled solution of H <sub>2</sub> O <sub>2</sub> after acidification with dilute	.: .:
•	$H_2SO_4$ required 30 mL of N/12 KMnO <sub>4</sub> for complete oxidation. Calculate the percentage of $H_2O_2$ in the solution.	(E)
	percentage of $\mathbf{n}_2 \mathbf{O}_2$ in the solution. The solution $\mathbf{O}_2$ is the solution of $\mathbf{O}_2$ and $\mathbf{O}_3$ are shown in the solution of $\mathbf{O}_3$ .	(5)
$0 \cdot 41$	[A] Give the general methods of preparation of carbides.	(5)
	B] Discuss the large scale methods of preparation of hydrogen peroxide.	(5)
		` '
Q:5	What are ceramics? Classify ceramics in a different ways and explain in detail ;-	
	recuperative pot furnace.  OR	(10)
Q:5	What are porcelain and china? Discuss the raw material used in production of porcelain and china and describe regenerative tank furnace.	(10)
0.7	gan sanga sa nga katalah sa na sa	.(E)
	[A] Discuss the structure of cyclic triphosphonitrilic chloride.  [B] Compare the chemical properties of inorganic benzene with that of organic benzene.	(5) (5)
0.6	OR A What are silicones? Describe high thermal silicone and silicone oils.	<i>(</i> 5)
		(5) (5)
		(3)
	range stational station and the station of the stat	7
	and the are well with prop) is every for and and direct the edical expension (2)	
	sught to profit the state of th	
	gat gabat sakka <mark>O</mark> ronij, og å se samt i sakkaj kar hyprag	
	s federal to the contract of t	
	appropriate the first factor of the first fact	4
	age a gallocal extra adding et are reading by a co	:11

No. of Printed Pages: 03 30/A-127 SARDAR PATEL UNIVERSITY B.Sc. (Semester - V) Examination Physical Chemistry USO5CCHEO5 Date: - 15/11/2017 Time: 10:00 am to 1:00 pm. Total Marks: 70 Day: - Wednesday Note: - 1. Figure to the right indicate full marks. 2. All questions are to be attempt. Choose the correct option and rewrite the sentence. [10] Q.1. 1. \_\_\_\_\_ is the best source of UV radiation. (a) Mercury Lamp (b) Candle (d) Tungston Lamp (c) Sodium Lamp 2. Radiation is the type of \_ (a) Monochromator (b) Detector (d) Source of light (c) Filter 3. Crystal can be classified into \_\_\_\_ crystal system. (b) 5 Page 10 (c) 7 Page 14 (d) 4 Page 15 4. In the powder diffraction method, the diffracted X-rays patterns are collected on \_\_\_ (b) Camera (a) Blank Paper (d) Photographic Plates (c) Screen 5. Natural rubber is basically a polymer of \_\_\_\_\_. (b) Ethylene (a) Isoprene (d) Propylene (c) Propane 6. A plastic which can be soften on heating and hardened on cooling is called (b) Thermoplastic (a) Thermoelastic (c) Thermosettimg (d) Thermite 7. Polymer are made up of two different types of monomeric units in their chain is called\_\_ (b) Graft copolymer (a) Block copolymer (d) Random copolymer (c) Co-polymer 8. Mark- Houwink Sakurade equation is given by \_\_\_\_\_ (b)  $\eta_{red} = \eta_{sp}/c$ (a)  $\eta_{sp} = \eta_{rel-1}$ (d)  $[\eta] = km^{\alpha}$ (c)  $\eta_{rel} = \eta / \eta_0$ 9. Which of the following technique yield a weight average molecular weight? (a) Light Scattering (b) Vapour Pressure Osmometry (d) Osmometry (c) Viscometry

(b) Soluble in water

(d) Soluble on monomer

10. In Emulsion polymerization, the initiator is \_\_\_\_\_

(a) Insoluble in both

(c) soluble in both

Q.2		Answer the following. [Any Ten]	[20]
	1.	Define: (a) Incandescence (b) Luminescence	
	2.	Differentiate between thermal and photochemical reactions.	
wang s	3.	State Lambert's law. Give its mathematical expression.	s test
	4.	Write the procedure for determing the Miller Indices for a plane.	Transfe
(A)	5.	Define : (a) Ionic Radii (b) Axis of Symmetry	
	6.	Define: (a) Constructive Interference (b) Destructive Interference	
	7.	Distinguish between HDPE and LDPE.	
	8.	Show that bifunctionally in a monomer is a must for the growth of a polymer chain.	÷
	9.	Write the salient features of anionic polymerization.	•
,	10.	Explain the term "Auto Acceleration".	
	11.	Define: (a) Relative Viscosity (b) Tyndall Effect.	
	12.	Define: (a) Specific Viscosity (b) Polydispersity	# # * .
ę.3	(a)	What is Quantum yield? Explain the reason of high and low quantum yield as well as explain the factors affecting quantum yield.	[05]
	(b)	For the photochemical reaction, $B \rightarrow C$ , $1.0 \times 10^{-5}$ mole of B was formed on absorption of $6.62 \times 10^{7}$ ergs at $3600A^{0}$ . Calculate quantum yield.	[05]
	•	[ Given : $h = 6.62 \times 10^{-27}$ erg.sec , $c = 3 \times 10^{10}$ cm / sec $\frac{1}{2} \times 10^{10}$ molecules / mole ]	ŧ
Q.3	(a)	State Beer's law. Give the factors responsible for the deviation from Beer's law.	[05]
45	(b)	Calculate the energy in calories per mole or per Einstein for radiations of wavelength 1000A <sup>0</sup> .	[05]
		[ Given: $h = 6.62 \times 10^{-27} \text{ erg.sec}$ , $c = 3 \times 10^{10} \text{ cm} / \text{sec}$	
		$N = 6.02 \times 10^{23} \text{ molecules / mole}$	
		a programme de la completa de la co Completa de la completa del completa de la completa de la completa del completa de la completa del la completa de  la completa de la completa del la completa de la completa de la completa de la completa de la c	
		· · · · · · · · · · · · · · · · · · ·	

Derive Bragg's equation with suitable diagram. Discuss its [05] Q.4. a limitations. What are the miller indices for planes with the following [05] (b) intercepts each expressed in terms of the unit cell dimensions? (2) [ ¼, ½, ½ ] (5) [ 2a, 3b, c] (1) [ 1, ½,½ ] (4) [ $\infty$ , 1,  $^2/_3$ ]  $(3) [1, \infty, \frac{1}{2}]$ [0R] Discuss the powder method of X-ray crystallography to [05] 0.4. (a) determine the structure of a crystal. Tugston has a BCC lattice and its density and molecular weight [05] (b) are 19.30 gm/cm<sup>3</sup> and 183.25 gm/mole respectively. Calculate the volume of the Tugston atom and the distance between d200,  $d_{110}$  and  $d_{222}$  planes. Explain mechanism of free-radical chain polymerization. [10] Q.5 Describe kinetics of chain polymerization. Also give its limitation and characteristics. [OR] What are polycondensation reaction. Give its suitable example. [10] Q.5 Explain kinetics of catalysed and non-catalysed polycondensation. Write the principal, draw the sketch and describe the membrane [05] Q.6 (a) osmometric method for the determination of molecular weight of polymer. The intrinsic viscosity of solution polyisobutylene at 20°C is [05] (b) 180 cm<sup>3</sup>/gm. If [n] is related to the viscosity average molecular weight. Calculate the molecular weight Mv of polymer. Given:  $\alpha = 0.64$  $K = 3.60 \times 10^{-4}$ [OR] List out the types of polymerization technique. Describe any [05] **Q.6** one polymerization technique. Mention the advantages, disadvantages and its application. Polymer molecule weight 2 X103, 1 X 104 and 3 X 104 are mixed [05] (b) talking 10%, 80% and 10% respectively. Calculate Mn and Mw assuming percentage by weight and percentage by number.

A Recognition of appearing the security Mark Marketting and Control of the Control of th ٠... and the state of the property of the state o Ţ. the first the second of the se was a second of the second of Beer Care The second of the 经自治的 and the second of the second o

[23 & A-11]

# SARDAR PATEL UNIVERSITY S.Sc. EXAMINATION (SEMESTER-V) NOVEMBER-2017

	MOAEMREK-501
DATE .	PHYSICAL CHEMISTRY (US05CCHE06) 17/11/2017(FRIDAY) TIME:10,00 to 1,00 pm
DAIL .	17/11/2017(FRIDAY) TIME:10.00 to 1.00 pm
QUE-1	Choose the most appropriate option for the following to the first [10]
, <b>1</b>	How many degrees of freedom are there in a system of liquid water and
	water vapor at equilibrium at pressure of 1 atmosphere?
	(a)0 (b) 1 (c) 2 (d) 3 (d) 3 (d) (d) (d) (d)
2	A compound when heated giving a new solid phase and a solution with a
	composition different from that of solid phase is said to possess
	(a) Incongruent melting point (b) Congruent melting point
	(c) Peritectic temperature (d) Eutectic temperature
3	How many phases are present in $CaCO_{3(s)} \Rightarrow CaO_{(s)} + CO_{2(g)}$
	(a)0 (b) 1 (c) 2 (d) 3
4	Which one of the following is an incorrect statement for physisorption?
region fr	(a)It is reversible process (b) It requires less heat of adsorption
40 A # PE 25 A	(c) It requires activation energy (d) It takes place at low temperature
5	Rate of physisorption increase with
	(a) Decreases in temperature (b) Increase in temperature
<b>表寫</b> []	(c) Decrease in pressure (d) Decrease in surface in area
6 	In GLC quantitative determination is done by
	(a)Area of peak (b) Position of peak
	(c) Number of peaks (d) Resolution of peak selection of peak
7	The quality and quantity of heavy metals present in organometallic
	compounds can be identified by
v t	(a) Thermal conductivity detector (b) Flame ionization detector
W 1	(c) Flame photometric detector (d) Electron capture detector
8	Increase in number of theoretical plates increase with the state of th
· Wasa	(a)decrease in column length (b) increase in column length
	(c)decrease in column diameter (d) both b AND c
9	The current due to the concentration gradient between two part of cell
	solution is known as
	(a) Diffusion current (b) Capacitive current
10	(c) Migration current (d) Charging current
a O	In polarography, quantitative determination can be done by measuring
	(a) Half wave potential (b) Limiting diffusion current (c) Applied voltage (d) Residual current
QUE-2	(d) Nondation (d) Nondation
1	Answer the following in very short (Any Ten)  Define: Degree of freedom and polymorphism.
	Calculate number of degree of freedom in aqueous solution of
Magazina (	(i) sodium chloride (ii) Acetic acid
3	Why it is not possible to have quadruple point in a phase diagram for a
-	one component system?
4	Differentiate between adsorption and absorption.
5	Define: (1) Adsorption isotherm (2) Sorption.
6	Under which condition Langmuir adsorption isotherm becomes identical
	with Freunlich adsorption isotherm?
7	Enlist important requirements of carrier gas.
8	Explain the term: Retardation factor.
_	_ 1.

Enlist important requirements of stationary liquid phase in GC.

9

10	Give merits and demerits of DME.	
11	What are the advantages of polarography?	
12	Describe polarographic maxima, and the second polar po	
QUE-3	Attempt the following	70F3
[A]	Draw and discuss the phase diagram of sulphur system.	[05]
[B]	Rhombic sulphur changes to monoclinic form at a temperature of 95.6 °C	[05]
ři Z í	and 1 atm pressure. What would be the change in the transition the temperature per atm change of pressure? Given: Heat absorbed in the	
r	change = 597.5 cal/mole, density of rhombic sulphur is 2.05g/c.c. and	
	density of monoclinic sulphur is 1.95g/c.c.	
	vidina madata – i ta mining basa ang <b>08</b> ya king batasat matan kabana ang at	
QUE-3	Attempt the following paragraph of the control of the paragraph that will be a proven on the control of the con	
[A]	Draw and discuss the phase diagram of simple eutectic system.	[05]
[B]	The vapor pressure of water at 95 °C is found to be 634 mm of Hg. What	[05]
	would be the vapor pressure at a temperature of 100 °C? The molar heat	
	of vaporization in this range of temperature may be taken as	
QUE-4	40593 J/mole.	
[A]	Attempt the following Starting with assumption, derive Langmuir adsorption isotherm.	[05]
[8]	Describe various types of experimental physical adsorption isotherms.	[05]
	OR	
QUE-4	Attempt the following	
[A]	Discuss Frendluich adsorption isotherm. Give its limitations.	[05]
[8]	Describe the forces responsible for physical adsorption.	[05]
QUE-5	Describe the general requirement of gas chromatography detectors.	[10]
	Describe thermal conductivity detector and Flame ionization detector in	•
	detail.	
QUE-5	Skotch the block diagram of assignment or able to be back the back.	[10]
QUL-3	Sketch the block diagram of gas chromatography. Describe the basic principle and working of it. The particular to the basic principle and working of it.	[TO]
QUE-6	Attempt the following was a fit to the form of the problem of the problem.	
[A]	Write a note on: (i) Catalytic current (ii) Kinetic current	[05]
[B]	A DME has following characteristics. (i) Weight of mercury collected	[05]
	through capillary for 100 seconds is 0.196 gms. (ii) 5 drops of mercury	
	collected in 21.6 seconds. When above DME was used then it gives	
	diffusion current of 8.76 $\mu A$ for the solution of 0.001M Zn <sup>+2</sup> , After this	
	experiment, A new DME was used having drop time 6.13 second and	
	flow rate of mercury of 3.85 mg/sec. Calculate the concentration of Zn <sup>+2</sup>	
AL LANCE	which gave diffusion current of 16.3 μA with new electrode.	naj.
QUE-6	まれば 東西山 OR to provide provide passes and common the following におきない ロート・マート (大変などのできる) こうできる (100 page) によっては	
[A]	Write a note on: (1) Direct comparison method (2) Pilot ion method	[05]
[B]	Diffusion current constant of Cd <sup>+2</sup> in 0.1M KCl is 3.42. Calculate the	[05]
	diffusion current in 0.001M Cd <sup>+2</sup> using a capillary with a drop time of 3	
	seconds. The weight of 20 drops of mercury is 100 mg.	
	ent <u>en er refer i X. de lite —</u> for nombre excitação dista Lagrando — — — — — — — — — — — — — — — — — — —	· ,
	· Topics ( ) management ( ) within	. "
	and alterated process or over connections or statement and any many and another the extension of the extension of	1)
	en informe inapparejon to the major stiffer	

The state of the s

#### SARDAR PATEL UNIVERSITY

T.Y. B.Sc. EXAMINATION, SEM-V

	CCSC01 : OBJECT ORIEN	TED PROGRAMMI	ING USING C++
		am To 01:00pm	Max. Marks: 70
Q1 Multi	iple Choice Question		[10]
i)	C++ allows declaration of varia	bles at?	
	A, anywhere in the scope		executable statement
	C. Both (A) and (B)	D. None	
, ii)	refers to fixed value t	hat do not change durin	g the execution of a
	program.		18
	A. Constant	B. Variable	
+11	C. Both (A) and (B)	D. None	
iii)	The header file should	be included that use inp	ut/output statements.
	A. iostream	B. stdio	
	C. conio	D. io	
iv)	Object without name is known	as	
	A. nameless object	B. live object	
	C. constant object	D. default object	
v)	For string manipulation	header file is inclu	ded.
•	A. iostream.h	B. ctype.h	
	C. string.b	D. conio.h	
vi)	Defining a function with same p	prototype in base class &	derived class is 👉 💎 💮
,	known as		
	A inheritance	B. function overriding	ng - Salara - Salara Salar
	C. function overloading	D. Inline function.	e de la companya de
vii)	One base class & multiple deri	ved classes means	_ inheritance.
,	A. multiple	B. hierarchical 🦠	
		D. hybrid	÷
viii)		to initialize string data-r	nember inside the
,	parameterized constructor?		
	A. strstr() B. strlen()	C. strepy()	D. None of these
ix)	A pointer is	4	
	A. A keyword used to create va		
	B. A variable that stores addre		
	C. A variable that stores addre		
	D. All of Above.		
x)	Which of the following declara	itions are illegal?	
~)	A. void *ptr	B. char *strl =	z "XYZ"
	C. char str2 = "abc"	D, const *int p	
Q-2 Ans	swer the following in short. (Any	•	[20
i)	Define encapsulation as conce		·
ii)	Define class & object as conce		
iii)		on and data encapsulatio	on.
iv)		declare string in C±+.	
v)			
vi)			
			line.
vii)			
viii •	,		=-
ix)		mand one the Quar	
x)		s of the operators << & :	>> in ('++.
Xi) Vii	•		

Q 3	a)	What is C++? Explain structure of C++.	5
	b)	Explain different operators available in C++,	15
		OR	•
Q-3	a)	Differentiate: OOP & POP.	5
٠.	b)	Describe with examples the uses of enumeration data types.	5
Q – 4	a) :-	Can we give the objects as arguments in function? If yes then explain it with	[5
		an example, and if no then give the reason.	
	b)	Explain destructors with an example.	[5
0 1		OR	
Q – 4	a) 	How many types of constructors we can create in C++ programs? Also write the characteristics of constructor.	[5]
	b)	Explain all operations on Arrays.	[5]
Q – 5	a) -	Discuss inheritance in protected mode & its effects on accessibility of base- class members.	[5]
	b)	Explain the concept of friend function with suitable example.	[5]
0 6	`	OR	
Q – 5	a)	Discuss inheritance in public mode & its effects on accessibility of base-class members.	5
	b)	Explain multiple & multilevel inheritance with suitable example	[5]
Q – 6	a)	Explain the restriction and limitations in overloading operator.	[5]
	b)	What is file mode? Describe the various file mode operations available.  OR	[5]
Q - 6	a)	Explain seek() and write() function with example.	[5]
	b) -	Explain binary operator overloading with example.	5



## SARDAR PATEL UNIVERSITY

T.Y. B.Sc. EXAMINATION, SEM - V

US05CCSC02: Business Data Processing Through DBMS. /2017 Time:10:00am To 01:00pm Max.

	0044	2017 Time:10:00am To 01:00	m Max. Marks	: 70
Date:	(19/11		1	
	Thus	aay		ftůi
Q - 1	Multij	ole Choice Question	ands if to Young, then Lee is	
:	i)	When Lee makes out a cheque for Rs.50 and s	Chus a to Toolg, we	
		known as: a) The Banker b) The C	Proditor	
		a) the mine		
	1	e) The Payee d) The l		<u> </u>
• •	ii)	A debit balance of Rs. 100 in a cash account sh	iows max.	
		a) Rs. 100 was the total of cash paid out	,	:
		b) There was Rs.100 eash in hand		
		e) Cash has been overspent by Rs.100		!
		d) The total of eash received was less than Rs.	.100	
	iii)	Which of the following is a liability?	•	:
		a) Creditors for goods c) Motor Vehicles d) Cash at	ery	
	į '	c) Motor Vehicles d) Cash at	Bank	. •
	iv)	is used to print results on printer.	Contract the second	İ
	1 "	(a) ? (b) ?? (c) ???	lone of these	
	v)	specifies a two-digit year format.		1
	• • •	a) set century off b)set cent	ury on	
1		e) set year off d) set year	on Programme	
		The default format of date data type is		į ;
	vi)	b) mm/dd/	${f yy}$ . The second of the	
		a) mm dd yy: (1) yy/mm/(	dd - Commercial Section 1997	
		The default extension of program file is	.prg d)None of these	
	V11)	a) .idx b) .frx e)	prg d)None of these	:
	!	a) .idx b) .frx c) is the extension of simple index file		
	viii)		of d) ,ind	1
}		a) .idx b) .srt c) .dr A DO WIILE programming structure ends	with	i ,
	+ ix)	A DO WITTE programming structure costs	LE	:
	į	(4) (3) (3) (4) (4)		
į		c) ENDDO d) None To validate the data entered with @ GET us	clause with (a) GF	$T^{\frac{1}{2}}$ . $\Box$
	x)	To validate the data entered with a COLD to	CK	1
	1		(a) and (b)	;
	1	e) CHECK d) Both	(4) 4110 (2)	
į.	i			1 (20)
Q	2   Ans	wer the following in short (Any 10).		
	i)	Define the term Assets.		1
	ii)	Define the term Income.	•	
1-	iii)	Define the term Debts.		•
į į	iv)	the state of the s		1
	v)	Differentiate: ? and ??? .		
t •	vi	1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2		į
-	vii			
!	vii	A CONTROL OF THE PARTY		, 1
1	ì	Explain KOEO and BOFO functions.		
	ix	The contract of the second of the contract of	Command file	
	X			
	xi	the second second second the CAN	l' command.	
•	xi	) + write down the syntax and use or a cover-		
		A second	· · · · · · · · · · · · · · · · · · ·	

Q-3	a)	Describe advantages and limitations of accountancy in brief.	[5]
	h)	Explain the classification of accounts in detail.  OR	[5]
Q-3	a)	Explain advantages and limitations of journal in book-keeping.	[5]
	b)	Explain basic rules of accounting.	[5]
Q 4		Explain the following commands with example.	[10]
	A Partie	1. DELETE 2. LOCATE 3. REPLACE 4. INPUT OR	
Q - 4	-	Explain the following functions with example.  1. SUBSTR() 2. REPLICATE() 3. CMONTH() 4. INKEY()	110
Q - 5	a)	Explain the concept of multiple databases with Select & Set relation	(5)
V O	"'	command.	[5] -
	b)	How corrupted index file is reorganized? Explain it with suitable example.  OR	151
$Q=\underline{5}$	a)	What do you mean by indexing? Explain the different types of index file.	[5]
	b)	What is sorting? Explain SORT command with example.	
Q-6	a)	What is procedure? How procedures are created & how parameters are	151
		passed to them? Also give example.	
	b)	Explain in detail 'JOIN' command with appropriate example OR	[5]
Q - 6	a)	Explain in detail 'APPEND FROM' command with example:	151
	b)	Explain the following commands with example	151
!		1. Do Case Endease 2. Sean Endscan	



1377	PAT No.	
()	فقعه صفيا المجهاد التعالم موالدي الراه الواراني	ė

Total No. of Printed Pages: 02

#### SARDAR PATEL UNIVERSITY

B.Sc (Semester - V) (CBCS) (Reg. & NC) Examination 2017

**US05CCSC03:** Visual Programming through VB.NET

Saturday, 11<sup>th</sup> November' 2017

x. Marks: 🛚	/U
	x. marks:

_	elect an appropriate a	answer for the following.		[10]
1.	Through Visual Basic p	projects you can create	<b>-</b>	
	[A] Circuit	[B] Windows Application	[C] Both a & b	[D] None of these
2.	VB.Net is Called as	Programming Langu	ıage.	The second second second
,	[A] Assembly	[B] Procedural	[ <b>C</b> ] OOP	[D] Functional
·3.	The JIT is one type of		7. ·	
	[A] Hardware	[B] Measurement Unit	[C] Time	[D] Compiler
4.	SDI stands for		_	
	[A] Standard Docu	ument Interface	[B] Single Docun	
1	[C] Standard Docu	ıment Integrated	[D] None of thes	<b>6</b> 
5.	is default	event for form.		
, <del>.</del> .	[A] Click		[C] Load	[ <b>D</b> ] Move
6	loon exec	ute at least once.	ing and the second of the seco	and the State of t
ν.		[B] dowhile	[C] for	[D] if
. <b>8.</b>	Which control display h	nierarchy of nodes.  [B] Timer	[C] Scrollbar	[D] TreeView
a	metl	hod is used to populate Data	aSet.	
:	[A] Populate		[C] Open	[D] Store
1(		nd delete SQL commands, $\_$	met	hod is used.
	[A] ExecuteDataR [C] ExecuteNonQu		[B] ExecuteScala	ar Ier — 1990 (1818) sa
		Jel V	[D] Executereat	ICI A A A A A A A A A A A A A A A A A A
	[C] ExecutemonQt	,		4 V
771.	[C] ExecuteMonGr			+ - N
7-11-x		questions in brief: (Atten		
2 A				e v
2 A	Answer the following  What is CLS?		npt any Ten)	e v
2 A 1. 2.	Answer the following  What is CLS?	questions in brief: (Atten	npt any Ten)	e v
2 A 1. 2. 3.	Answer the following  What is CLS?  List types of project.	questions in brief: (Atten	npt any Ten)	e v
2 A 1. 2. 3.	Answer the following  What is CLS?  List types of project.  List the Data Types su  Differentiate: MDI V	questions in brief: (Attenuations) upported by VB.NET. V/S SDI.	npt any Ten)	e v

	8. Explain use of Timer.					
	9. Compare RadioButton and Checkbox.		- 1 -			
1	10. What are the applications of ADO.net?			4 (4)	· 通用 医二种 ( ) ( ) ( ) ( ) ( ) ( ) ( ) ( ) ( ) ( )	, ander
	11. Explain DataGrid control.				e de la companya de l	aye Alberta
	12. Explain System.Data.SqlClient namesp	ace.				2,7
				. 111	A CONTRACTOR	
Que:	3 Explain .NET Architecture in detail. Discuss	VB.NE	T with its	features.	Professional Profession	[10]
	Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Marian Maria Marian Marian	<u>OR</u>	7.5		· · · · · · · · · · · · · · · · · · ·	• -
Que:	3 Write a short note on IDE and Solution Exp	olorer.	; 	# M	14 (1) (1) (1) (1) (1) (1) (1) (1)	[10]
Que:	4 [A] Explain Form life cycle in detail.				e to de la final	[05]
	[B] Explain Modules in detail.	•			11 A 1	[05]
		<u>OR</u>			in the early of the	F1
Que: 4	<b>4 [A]</b> Explain Message box with example.		:	i se i king ngak Kiningan	. Straight fair	[05]
	[B] Explain the ways of passing arguments	s with e	xample. Ex	kplain use d	of 'Bv Val' & 'Bv R	
	keyword.		31. 13. 4p. 5		,	[05]
Que: !	<b>5 [A]</b> Explain different type of error handling	ı. Expla	in anv one	in detail.	· · · · · · · · · · · · · · · · · · ·	[05]
	[B] Explain Listbox, Combobox and Checkl				自然的 (1.8.基) 1	[05]
	and the second s	OR	a 1 40 m	er e .	a New Art Allen	[02]
Que: 5	[A] Explain try catch final in detail wit	h exam	ple.	5 °	(1) (1) (1) (1) (1) (2) (2) (2) (2) (2) (2) (2) (2) (2) (2	[05]
	[B] Explain Button control with its properti		-	events.	14 - Hangy 11 477	[05]
	graphic of the second of the s			1 W. H		[00]
Que: 6	[A] Explain the connected architecture of A	NO NE	 T in hrief	1		Fow1
-	[B] Explain major ADO .net objects.	100.11	, in bilei,	Harris Communication		[05]
		<u>OR</u>	18	Market S. S. Services	olothepolotika († 19 16. sp. – Francis	[05]
Que: 6	[A] Explain use of "ExecuteScalar", "Execution detail.	uteNon(	Query" and	"ExecuteR	leader" methods	[05]
	[B] Explain the sten, how can we retrieve	data in	DataC-ta			
: "	[B] Explain the step, how can we retrieve of	iala III	valaset(	ি শিক্ষি ক্রি	ing site of a second	[05]
					and the second s	F 2

6. How functions are declared in VB .NET?7. Explain any four properties of RichTextBox.

### ### Best @f Luck ###



# SARDAR PATEL UNIVERSITY T.Y. B.Sc. (SEMESTER – V) EXAMINATION COMPUTER SCIENCE

US05CCSC04: COMPUTER NETWORKS

Date:	13/11	1/2017, Morelay Time: 10:00am To 01:00pm M	ax. Marks	s:70
		iple Choice Question		[10]
(3)	i)			(b) - b -
11	1)	a) Sending message to some computer	1 lisenss 1	(1)
121		TANK TO THE CONTRACTOR OF THE	w	
		C 11 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	al surviv	(B + -
[5]		d) Receiving message from some computer in the network	with the same	
101	ii)	A computer network permits sharing of		
(47.1	11)	a) Resources b) Information	nisiqxii	
101		c) Both a & b d) None of these		
	iii)	If a computer on the network shares resources for others to use,	it is called	
	)			
		a) Router b) Client c) Mainframe d) Server		
121		c) Mainframe d) Server	mulqxa	
	iv)	c) Mainframe d) Server Which of the following modulation requires the lowest bandwid	th?	
	~,	a) ASK b) PSK c) FSK d) QPSK		
1 1 1	v)	For long distance data transmission, what is the preferal	ole mode	of
[6]	.,	communication?		
		a) Simplex b) Half duplex	y.	
		c) Serial transmission d) Parallel transmission	n	
	vi)	Transmission media are usually categorized as	i" 39	
		a) Fixed or unfixed b) Metallic or Non-meta	ıllic	
		c) Determinate or indeterminate d) Guided or unguided	d	
	vii)	Which of the following is a method of Wireless Transmission?		
	,	a) Electromagnetic Spectrum b) Infrared and millimet	er waves	
		c) Light wave transmission d) All of the above		
	viii)	A Bluetooth network can have master(s).		N
		a) Eight b) Two c)Three d) One		
	ix)	A repeater takes a weakened or corrupted signal andit		
		a) Regenerates b) Reroutes c) Amplifies d) Resa	ımple	
	x)	Which topology requires a central controller or hub?		
		a) Ring b) Mesh c) Bus d) Star		
				1001
Q-2		wer the following in Short.(Any 10)		[20]
	i)	Define Host.		
	ii)	Define Network Topology.		
	iii)	Define Spectrum.		
	iv)	Explain serial transmission in brief.		
	v)	Explain phase modulation in brief with example.		
	vi)	Explain amplitude modulation in brief with example.		7 5 6 1 1 1
	vii)	Write Full Form of: POP & HTTP	A B C M To 1	
	viii)	Last seven my ers or other	n hens Fari Da	
	ix)	Politic Politics		
	x)	Explain Router with example in brief.	(E. )	- II 3
	xi)	Explain switches with example in brief.		
	xii)	Explain Gateway with example in brief.		
		(7)	C	COTA
		( <del>L</del> )		

		v i
-3 a)	Name the two popular LANs. Also explain the two division of broadcast	[5]
<b>b</b> )	network in brief. Discuss the importance of computer networking.	[5]
-3 a)	Write short note on MAN and WAN. The agree of the state o	[10]
4 a) b)	What is switching? Explain packet switching in detail.  Discuss FDM in detail.  OR	[5] [5]
4 a) b)	Write the difference between circuit switching and packet switching. Write a short note on "Fiber Optic".	[5] [5]
- 5	Explain any four layers of OSI model.	[10]
5	OR List and explain design issues for layers in detail.	[10]
6 a) b)		[5] [5]
6 a)	Write a short note on "Mesh Networks".	[5] [5]
ĺ		
	e degree de la Light de la constant	
	and the second of the control of the	44
	the state of the s	
	and the analysis of the fooders of a gains at the decision of the self-self-self-self-self-self-self-self-	: 18 ±
	· Profession (1995) Charles transported page (1995) Profession (	
	TOTAL TO THE CONTROL OF THE CONTRO	y 11.
	en de la composition br>La composition de la	5 % *
		1 ark
	the management of the second o	
	表现的一种技术的企业。	i
	en la proposition de la company de la company de la company de la company de la company de la company de la co La company de la company d	
	en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	
	and the second of the second o	
	in the state of th	
	i depute e la la la filia de la travalla de la companya de la companya de la companya de la companya de la comp	1.5
	detail to the description of the second state of the second secon	111
	(1) 大块 人工 無 資本 (2) 维亚克克克·特克	\$ \$ 1 × 2
	· · · · · · · · · · · · · · · · · · ·	7120 i
	was a second of the second of	(34)
	र्याच्यां सम्बद्धाः स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स्वत्यां स	1
	The first and the first of the second of the	19
	and the second of the second o	

٠,,

SEAT	No:	
		The same of the sa

Total no of Pages: 02

## [31]

### SARDAR PATEL UNIVERSITY

## B.Sc Sem-V (Computer Science) November - 2017

## US05CCSC05 || System Analysis and Design

Date:	15-11-2017	Time: 10:00	O A.M. To 01:00 P.N	Marks: 70
Q-1	M.C.Q			
1	M.C.Q	System is an avamn	do of Notural greaters	1 more more 1
	a) solar	b) information	le of Natural system. c) business	1) 1 6
2	u, 00141		le of man-machine system.	d) defense
	a) solar	h) information	c) business	1) 1 0
3	The third step of	f SDLC is	c) business	d) defense
	a) Problem ident	tification	b) System avalue	
		ement analysis	b) System evalua	uon . mand entile vide
4	SSADM does no	ot includes	d) System study	reifa di nicionali I
	a) System Surve	v b) Structured Des	ign c) Maintenance d) S	
5	is	considered as an inni	it of the Configuring Hardw	System Analysis
	a) Physical Requ	irements b) Hard	ware c) Packaged Design	are Study.
6	The processed da	ata with specific purr	ose are called	d) Test Plan.
	a) Data	b) Information	c) Picture d) Sys	
7			c) Picture d) Systect error.	stem
	a)Data Capture	b) Computer In	put c) Data Validation	ADA
	preparation	o) computer in		oden meter
8		not the fact finding te		
1 712	a) Interviewing	b) Questionnaire	es c) Record inspection	na David Isid B
9	DFD stands for	-) (woodonnan	c) Record hispection	d) Data input
	a) Data Flow Dia	ıgram	b)Database Flow Diagram	,
	c) Data Flow Des		d) None of above	1
10	The DFDs that sl	nows "What is going	on" is known as	DFDs.
	a) Physical	b) Logical	c) Physiological	d)None
		-)84.	o) i nysiological	ujivone
Q-2	<b>Short Question</b>	(Any Ten)		20
1	Define system.			20
2	List steps of SDL	.C.		
3	Differentiate bety	ween open system and	d closed system.	
4	Explain Documer	ntation in short.		
5	List any two Tan	gible Benefits.		
6	Write any four ste	eps of SSADM.		
7	List any five poin	nts to remember while	e Interviewing.	
	Describe any two			
9	List any four desi	gn principles of outp	mt	
	Draw the context	level diagram for rol	ilway reservation system.	
11	Explain Code Ge	nerator	iway iescivation system.	
12	Explain Any two	advantages of CASE	Tools	
		unugos of CASE	7 1 0015.	/D ==:
			* 5	(P.T.O)

List the types of systems. Give some examples of e	ach system.
OR	er en en en en en en en en en en en en en
Explain the Problem Identification step of SDLC.	
Explain evaluation of the system.	
LANGUA TANÀN TANÀN	
Explain Structured Analysis in brief.	
Explain structured design in brief.	
OR	
OK .	
What are the advantages of SSADM?	
List different stages of SSADM. Explain any three.	, as S
Dist different stages of BBADIVI. Explain any tiree.	n de la companya de l
	en en en en en en en en en en en en en e
Explain basic steps in Data capture.	en en en en en en en en en en en en en e
Explain Different types of output.	and the second second second in the second s
or one of the control	
and the second of the second o	
Evaloin Donald in constitution	en en en en en en en en en en en en en e
Explain Record inspection.	
Describe Questionnaires in detail.	
What is DED? Draw Symbols of DED. White Dules	CDPD D
What is DFD? Draw Symbols of DFD. Write Rules	of DFD. Prepare context
level diagram and zero level data flow diagram for t system in nationalized bank.	the saving and withdrawal
OR	
List Case Components. Explain and limitation of Ca	see Components

5.4

(36) 54

SERI IV

No. of Printed Pages: 2

#### SARDAR PATEL UNIVERSITY

B.Sc. (SEM V) (Electronics & Communication)

Subject: Analog Circuit Design & it's Applications

Date & Day: 07-11-2017, Tuesday

Note: The figure right indicates full marks

Time: 10-00 Am. TO 1-60 PM.

Total No. of Printed Pages: 02

USO5CELCO1

Total mark: 70

#### Q-I Choose the correct answers (10)Class AB operation is often used in power amplifier in order to 1. (c) (d) Reducing collector (a) Overcome cross **(b)** get maximum Remove even dissipation over distrotion efficiency harmonics In a amplifier, the current flows only during positive half cycle. 2. (a) Class A (b) Class B (c) Class AB (d) Class C Negative feedback in an amplifier 3. (a) Reduces (b) Increases noise (c) Reduces gain (d) Increases frequency bandwidth 4. Voltage series feedback results in Decreases both (d) Decreases o/p & (a) Increases both (b) (c) Increases i/p & i/p & o/p decreases o/p increases i/p i/p & o/p impedance. impedance. impedance. impedance. 5. oscillator employs two capacitor in series & inductor in parallel. Hartley Phase shift (d) Crystal (a) Collpitt's (b) (c) is a very popular circuit & is commonly used in a local oscillator in the radio receiver. 6. Collpitt's (a) Phase shift Hartley Crystal (d) (b) (c) oscillator oscillator oscillator oscillator phase shift is obtained. In a single RC network 7. (b) $=90^{\circ}$ <900 (d) >800 8. The 555 timer Ic can be operated at supply voltage (a) 5-15V (b) 1-18V (c) 2-20V (d) 5-18V 9. The 555 timer Ic is used to provide Rectification Oscillation (a) Time delay (b) (c) Amplification (d) 10. The efficiency of a Class A amplifier is \_\_\_\_\_ 68.5 70 (c) (d) (a) 78.5 (b) 50. (20)Attempt any ten of the following Q-2 Sketch out the circuit diagram of PPM and explain it. 1. 2. Differentiate between Class A and Class B amplifier. Define: Oscillator 3. 4. Discuss the miller's effect.

Why feedback is necessary in amplifier?

List out the salient features of 555 timer Ic.

What do you mean by cross over distortion?

Note down the frequency of Hartley & Collpitt's oscillator.

Give the comparison between the negative and positive feedback.

Explain the Bi-stable multivibrator as a RS flipflop.

List out the different types of feedback in amplifier.

5.

6.

7.

8.

9. 10.

11.

Q-3 (a)	What do you mean by power amplifier? Explain class A amplifier in detail with circuit diagram.	(07)
Q-3 (b)	List out the different types of an amplifier with circuit diagram.	(03)
	OR	
Q-3	Describe the push pull amplifier in detail.	(10)
Q-4 (a)	Describe in detail the construction and working of RC phase shift oscillator.	(06)
Q-4 (b)	Write a note on Hartley Oscillator.	(04)
(~)	OR	
Q-4	Discuss in detail Wein bridge oscillator & derive it's expression.	(10)
Q-5	Derive the equation of the negative feedback in amplifier with necessary circuit diagrams.  OR	(10)
Q-5 (a)	Discuss in the detail about feedback in amplifier.	(07)
Q-5 (b)	List out the merits and demerits of negative feedback.	(03)
Q-6	Explain the circuit diagram of Astable multivibrator and derive it's frequency of oscillation.  OR	(10)
Q-6	Draw the functional block diagram of 555 timer Ic and explain it.	(10)

•





# SEAT No. Sardar-Patel University B.Sc. (semester-V) CBCS Examination Nov. – 2017 09/11/2017, Thursday

### 10∰0 am to 12∰0 pm

#### **Electronics & Communication**

#### US05CELC02: Introduction to 8-bit Microprocessor

Maximum Marks: 70

No. of Printed Pages: 2

Note: Figure to the right indicates full marks.

Q-1	Choose the correct An	swer.			[10]		
1.	There are ge	eneral purpose registers.					
	a) four	b) five	c)six	d) eight			
2.	IN 8085 Microprocesso	or, ADD instruction byte size	e is				
	a) 1 byte	b) 2 byte	c) 3 byte	d) 4 byte			
3.	The address bus of 808	35 contains bits.					
	a) 8	b)16	c)32	d)64			
4.	addressing m	ode use only register as opr	rends.				
* -	a) immediate	b) indirect	c) register	d) direct			
5.	Instruction is	used to rotate the contain t	the accumulator left throug	gh carry.			
	a) RLC	b) RAL	c) RAR	d) RRC			
6.	Instruction is used to transfer control to subroutine.						
	a)PUSH	b) POP	c) CALL	d) RETURN			
7.	A 16 bit microprocesso	or has word length equal to	•				
	a) 1 byte	b) 2 byte	c) 3 byte	d) 4 byte			
8.	Instruction is	used to rotate the contain	the accumulator right throu	ugh carry.			
	a) RLC	b) RAL	c) RAR	d) RRC			
9.	JMP 2345H is	_ instruction.					
	a) 1 byte	b) 2 byte	c) 3 byte	d) 4 byte			
10.	The address bus of 8085 is						
		b) bidirectional	c) multidirectional	d) none			
Q-2	Answer in short.(Any	ten)			[20]		
1.	Give the example of th	•					
2.	Explain the function of	f SID and SOD?					
3,	Give the function of a	system bus.					
4.	What is the difference	between ADD and ADI inst	ruction?				
5.	Why the data bus is bi	directional?					
6.	Explain Subroutine.						
				**			

8.		Compare RAR AND R	RC instruction.		
9.		What is stack? State t	the uses of stack pointer.		
10.		Compare RET and PO	P instruction.	·	
11		Differentiate INR and	INX instruction.		
12.		Explain the function of	of program counter?		
Q-3		Explain the pin out di	iagram of 8085 micropro	cessor.	[10]
				OR	
Q-3		Explain in detain arch	nitecture of 8085 micropr	ocessor.	[10]
Q-4 (a	a)	· -		mory location, save the carry in one register and display the left to the left of above program.	[10]
			•	OR	
Q-4 (a	a)	Explain the FLAG regi	ister in 8085.		[05]
(E	b)	Explain the Arithmeti	ic instruction with examp	lle.	[05]
Q-5		• •	•	FFH to 00H in system with 2MHz clock frequency time delay number at output port 01H.	[10]
				OR	
Q-5 (a	a)	Explain Call and RETU	JRN instruction with exar	mple.	[05]
(1	b)	Discuss in detail adva	anced subroutine concep	ts.	[05]
Q-6		Write a main prograr Return the result in t		tiply two unsigned numbers place in the register H and L.	[10]
			•	OR	
Q-6 (a	a)	Explain the following	; instruction with example	e.	[05]
		(1) LHLD	(2) SHLD	•	
(1	b)			outine to convert the binary number stored at 6000H into its	[05]

7.

Explain the function of ALE.

\*\*\*\*\*



[38] SEAT NO.

No. of Printed Pages : 2

### SARDAR PATEL UNIVERSITY

#### B.Sc. Fifth semester

### Electronics & Communication

### US05CELC03

### Measuring instrument and Signal generators Saturday, 11/11/2017

Time: - 10:00AM To 1:00 PM Marks: - 70

Q1	Choose the correct Ar	iswer.( Attempt all)			[10]
1.	What is the typical range of resistance measurement of Kelvin's bridge?				
	a) $0.1\Omega$ to $M\Omega$	b) $0.00001\Omega$ to $M\Omega$	c) $1\Omega$ to low $M\Omega$	d) None	
2.	What is the balance co	ndition for the ac bridge	?		
	a) Z1=Z2Z3 Z4	b) Z1Z4=Z2Z3	c) Z1=Z2Z3	d) None	
3.	Which of the following	is used for measuremen	t of resistance only?		
	a) Wheatstone bridge	b) Hay bridge	c) Maxwell bridge	d) None	
4.	What is the normal ope	rating temperature rang	ge of Thermistors?		
	a) -100°C to 300 °C	b) 0°C to 300°C	c) 100°C to 300°C	d) None	
5.	Low impedance compon	ent such as low value res	sistors, large capacito	ors are	
	measured by connecting	g them with measu	ıring circuit.		
	a) parallel-series	b) series	c)Parallel	d)None	
6.	What is the equation of the gage factor K?				
	α) k=R+ΔR	b) k=1-2 <i>μ</i>	c) k=1+2 <i>μ</i>	d) None	
7.	Capacitive transducer is a transducer.				
	a) passive	b) Active	c) analog	d) digital	
8.	Which of following is an electromechanical device containing a resistance				
	element that is contac	t by a movable slider?			
	a) RTD	b)	c) Potentiometric	d) None	
		Capacitivetransducer	transducer		
9.	Attenuator is used to	the power leve	l of a signal by fixed	amount.	
	a) Reduce	b) Increase	c) Keep constant	d) None	
10.	In which of following t	he oscillation frequency	is controlled by inpu	t voltage?	
	<ul> <li>a) voltage-controlled</li> </ul>	b)spectrum analyzer	c) Peak detector	d) None	
	oscillator or VCO				

Q2		Answer in short.(Any ten)	[20]
1.		What is the measurement errors associated in whetstone bridge?	
2.		Draw the neat and clean circuit diagram of Schering bridge.	
3.		Draw the general circuit diagram of kelvin bridge.	
4.		Why digital voltmeter is more versatile instrument than analog?	
5.		What is the input range and absolute accuracy of Digital Voltmeter?	
6.		Which three basic questions are asked while selecting transducers?	
7.		What is the main function of frequency synthesized function generator?	
8.		In what shapes and sizes thermistors are available?	
9.		Give the classification of transducers.	
10.		List the advantages and disadvantages of RTD.	
11		What are the applications of Peak detector?	
12.		Draw the block diagram of sine wave generator.	
Q3		"The Maxwell bridge is limited to measurement of medium Q-coils (1 <q<10)"< td=""><td>[10]</td></q<10)"<>	[10]
		Justify the statement. Explain working of Maxwell's bridge,	
		OR	
Q3	(a)	Explain Wheatstone bridge with necessary circuit diagram and equations.	[5]
	(b)	Write a note onHay bridge.	[5]
Q4	(a)	With necessary circuit diagram explain series and parallel Q-meter circuit.	[10]
<b>0</b> 4		OR	
Q4	(a)	With necessary equations and circuit diagram explain basic Q-meter circuit.	[5]
	(b)	Explain successive approximation type ADC in detail.	[5]
Q5	(a)	What are the Thermistors? Describe any two characteristics of Thermistor.	[5]
	(b)	Write a short note on piezoelectric transducer.	[5]
		OR	
Q5	(a)	Explain principle and construction of LVDT.	[5]
	(b)	Write a short note on a capacitive transducer.	[5]
Q6		Explain working of Hartley and Colitis oscillator in detail with circuit diagrams.  OR	[10]
Q6		With necessary circuit diagram explain function generator.	[10]
٦		The mesons of the same and an explain full entire generator.	[10]

[10]

Sardar Patel University

No. of Printed Pages: 21

B.Sc. (Semester-V) CBCS Examination October - 2017

Subject: Electronics & Communication

Date: 13/11/2017, Monday Time: 10:30 AM to 100 PM Paper: Digital communication system Code: US05CELC04:

Maximum Marks: 70

Note: Figure to the right indicates full marks.

-1	Choose the correct Answer.	
		lemodulation increases, the distortion is
	a) increase	c) depend on input
	b) decrease	d) remain constant
	Flat top sampling uses	
	a) chopping	c) multiplication
	b) switching circuit	d) sample & hold circuit
	Natural sampling uses	
	a) chopping	c) multiplication
	b) switching circuit	d) sample & hold circuit
	In, the frequency of the carrier is	switched depending on the input digital signal.
	a) ASK	c) PSK
	b) FSK	d) None
	Which of following used to avoid effect of u	inder sampling?
	a) interpoling	c) low pass
	b) pre-alise	d) high pass
	Companding is used for	· · · · · · · · · · · · · · · · · · ·
	a)To overcome quantization noise in PCM	c) in the PCM transmitter as amplitude limiter
	b) to protect the small signal in PCM for	d) to overcome impulse noise in reciever
	distortion	
	Quantization noise occurs in	
	a) FDM	c) TDM
	b) PWM	d) PCM
	In PAM, the bandwidth of the transmission	channel depends on
	a) width of the pulse	c) height of the pulse
	b) rise time of the pulse	d) frequency of the pulse
	Coherent detection is also known as	•
	a) Synchronous detection	c) product detection
	b) balance detection	d) None
	is used to maintain fame and sample	le synchronization between TM transmitter and
	receiver.	
	a) framing bit	c) signaling bit
	b) sign bit	d) CRC bit
		. '

Q-2	Answer in short. (Any ten)	[20]
1.	Give reason ideal sampling is not suitable for transmission purpose.	
2.	Give an account of regenerative repeater.	
3.	Write the time domain and frequency domain representation of flat-top sampled signal.	
4,4,8,5,5	What is quantizer? Explain its types.	
5.	What is aliasing? How it can avoid?	
6.	Explain ASK.	
7.	Define multiplexing and mention the different types of multiplexing.	
8.	Explain pulse time modulation.	
9.	Mention the difference between TDM and FDM.	
10.	What do you meant by coherent and non-coherent digital modulation techniques?	
11	Draw the FSK signal for the digital data 10110010.	
12.	What is DC wandering? When it is arising?	-
Q-3	Discuss, in detail, sampling techniques with necessary circuit diagram and waveform.	[10]
	OR	
Q-3	State and prove sampling theorem in time domain.	[10]
Q-4	What is PPM signal? Explain the generation and detection of PPM signal with necessary circuit diagram and waveform. Also mention the advantage and disadvantages.	[10]
	OR	
Q-4 (a)	Elaborate the term "Companding" in digital communication with suitable example.	[05]
(b)	Explain the basic element of PCM signal.	[05]
Q-5	What is BPSK signal? Explain the generation and detection of BPSK signal with necessary diagram and wave-from.	[10]
	OR	
Q-5	Explain in detail binary shift keying with necessary diagram.	[10]
Q-6 (a)	Explain Time Division Multiplexing(TDM), in detail.	[05]
(b)	Write a short note on: Frequency Division Multiplexing (FDM).	[05]
a jobne 🕡 s	OR	. 1
Q-6	Explain different types of line encoding formats used in telephone line transmission.	[10]

\*\*\*\*\*\*\*\*\*



SEAT No
---------

No. of Printed Pages: 02

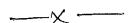
### [32]

### SARDAR PATEL UNIVERSITY V.V.NAGAR

B.Sc. (V<sup>th</sup> SEM.) ELECTRONICS & COMMUNICATION
15<sup>th</sup>NOVEMBER-2017 EXAMINATION
SUB. – ANTENNA AND ITS APPLICATION
SUB.CODE-US05CELC05

TIME	:-10:00 am to 1:00 pm		MARKS-70
Q-1	Choose correct answer		[10]
1.	Pitch angle for helical antenna is de	efined by	[]
	a) tan¹S/πC	c) tan <sup>-1</sup> S/πD	
	b) tan¹S/πL	d) none of above	
2.	The noise voltage for resistance R is	s given by	
	a) $V=2\sqrt{KTBR}$	c) $V = \sqrt{KTBR}$	
	b) $V=2\sqrt{KTB}$	d) none of above	
3.		nal power versus a level of noise power and	
٥.	is most often expressed as a measu	·	
	a) signal-to-noise ratio	c) phase margin	
	b) contrast to noise	d) none of above	
4.	•	of helical antenna is around	
	a) 30MHz to 3GHz	c) Above 3GHz	
	b) 10MHz to 10MHz	d) none of above	
5.	The intrinsic impedance of free spa	•	
	a) 377 π	c) 90π	
	b) 120π	d) none of above	
6.	The Radiation pattern of end-fire a	•	
	a) uni-directional	c) multidirectional	
	b) bidirectional	d) none of above	
7.	The Radiation pattern of broadside	array is	
	a) uni-directional	c) multidirectional	
	b) bidirectional	d) none of above	. ,
8.	Which expression involves the elec	trostatic field for far field due to alternating	
	current element?		
	a) Only E <sub>e</sub>	c) Both and b	
	b) Only E <sub>r</sub>	d) none of above	
9.	The arrangement consisting two ele	ectric poles are known as	
	a)dipole	c)array	
	b)monopole	d) none of above	
10.	The intrinsic impedance of free spa	ce is symbolized by	
	a) Z <sub>0</sub>	c) X <sub>o</sub>	
	b) Y <sub>0</sub>	d) none of above	
Q-2	Short answer type question. (ar		[20]
1.	Define: Effective area and effective	height of antenna.	
2.	State Helmholtz theorem.		
3.	Sketch the labelled diagram of horr		
4.	Evaluate radiation resistance of an	element of length L=2 m at /LKHz	

5.	Give only expression of far field due to alternating current element.	
6.	Obtain the pattern of broadside array of eight element spaced on half wavelength	
	and fed in phase using multiplication of pattern.	
7.	Explain slot impedance.	
8.	Evaluate radiation resistance of an element of length L=2 m at 20 KHz.	
9.	Give the expression for radiated power. Also find power radiated for r=1 cm, $I_{eff}$ = 0.7 amp. At SGHz.	
10.	Draw a labelled diagram of volcano smoke and conical antenna.	
11	Why signal to noise ratio is required?	
12.	Find the Tchebyscheff polynimialT <sub>4</sub> (x).	
Q.3(A)	Derive the complete expression for far field due to sinusoidal current distribution.	[05]
Q.3(B)	Obtain the near field due to sinusoidal current distribution.	[05]
	OR	
Q.3(A)	Explain the meaning of different terms of Ez, Ey and $H_{\Psi_{c}}$	[05]
Q.3(B)	Derive the expression for power radiated by current element and discuss its all cases.	[05]
Q.4(A)	Explain effective area of antenna in detail.	[05]
Q.4(B)	Write a short note on antenna terminal impedance.	[05]
	OR	
Q.4(A)	Explain in brief signal to noise (S/N) ratio.	[05]
Q.4(B)	Explain binomial array in detail.	[05]
Q.5(A)	Write a note on: Broadside array.	[05]
Q.5(B)	Write a note on: linear array.	[05]
,	OR	
Q.5	Explain multiplication of pattern in detail.	[10]
Q.6	Sketch and explain yagi-uda antenna.	[10]
-	OR	
Q.6(A)	Explain helical antenna with necessary diagram.	[05]
Q.6(B)	Derive optimum horn dimension for pyramidal horn.	[05]



[25]

#### SARDAR PATEL UNIVERSITY

B.Sc. Semester V Examination (CBCS)

### US05CELC06: Data Communication and Network

17<sup>th</sup> November 2017, Friday 10:00 am to 1:00 pm

**Total Marks: 70** 

Note: Figures to the right indicate maximum marks.

Assume data wherever necessary.

		Assume data wherever necessary.	
Q1	1	Choose the correct answer.  Terminators are used in topology.  a) Bus b) Star c) Mesh d) Ring	[10]
	2	Communication between a computer and a keyboard involves transma) Simplex b) half-duplex c) full-duplex d) automat	
	3	is a collection of many separate networks.  a) A WAN  b) An internet  c) A LAN  d) None of the above	ove
	4	A is a set of rules that governs data communication. a) Forum b) protocol c) standard d) none of the abo	ve
	5	For large network which topology is best? a) Bus b) Star c) Mesh d) Ring	
	6	is defined as the "Interface between data terminal equipment and data communications equipment using serial binary data exchange.  1) RS232 b) RS 449 c)X-21 d) RS532	
	7	A is a topology for a Local Area Network (LAN) in which all nodes an individually connected to a central connection point, like a hub or a switch.  a) Bus b) Star c) Mesh d) Ring	e.
	8	What does stand for CRC?  a) Cycle Redundancy Check b) Cyclic Reduction Check c) Cycle Reduction Check d) Cyclic Redundancy Check	
	9	Which of the following architecture uses CSMA/CD access method?  a) ARCnet b) Ethernet c) CSNET d) ARPANET	
	10	A is a data communication system within a building, plant, or campu between nearby buildings.  a) MAN b) LAN c) WAN d) none of the about the communication system within a building, plant, or campu between nearby buildings.	
Q2	1 2 3 4 5 6	Answer in short [ANY TEN] Explain single-bit error correction. Define Topology. Define Asynchronous transmission. What is Network and list out different types of network. Explain RS- 449 interface. Write down OSI model Layers.	[20]

	8 9 10 11 12	Give short note on mesh topology.  Differentiate error correction and error detection.  Define RS – 232 interface.  Explain protocol efficiency?  What is Ethernet?	
Q3		Explain Network application topologies and at least three Standard Creation Committees.  OR	[10]
Q3		Describe in detail the standard OSI layer with require diagram.	[10]
Q4 Q4	A B A B	Give the description about Transmission modes.  Give the short on Carrier sense multiple accesses with CD.  OR  Explain Aloha protocol and token passing.  Define Simplex, half duplex and full duplex.	[05] [05] [05] [05]
Q5		Discus in detail about Internet Virus and Worm.	[10]
Q5		Write note on CRC (Cycle Redundancy Check).	[10]
Q6		Discuss about Sliding window Protocol.  OR	[10]
Q6		Write a short note on Protocol Correctness.	[10]

Discuss Collision detection.

No. of Printed Pages ; 2

Total Marks: 70

#### SARDAR PATEL UNIVERSITY

B.Sc. (5<sup>th</sup> Semester) Examination

2017

Tuesday, 07<sup>th</sup> November 10:00 a.m. to 1:00 p.m.

**US05CELE01 - Discrete And Linear Circuits** 

Q.1 Choose the correct answer [10] 1. Voltage series feedback amplifier is also called . . . (a) Current amplifier (b) Voltage amplifier (c) Transresistance amplifier 2. In the feedback amplifier, sensitivity D is equal to \_\_\_\_\_. (b) 1+AB (b) Aβ (c) 1-AB 3. The operational amplifier has \_\_\_\_\_ output resistance and \_\_\_\_\_ input resistance. (a) Low, High (b) High, High (c) High, Low 4. In every practical oscillator the loop gain is \_\_\_\_\_ than unity. (a) Smaller (b) Equal (c) Larger 5. Radio frequency oscillator generates \_\_\_\_\_ range frequencies. (a) 20 KHz to 30 MHz (b) 20 Hz to 20 KHz (c) 30 MHz to 300 MHz 6. The conversion efficiency of class B amplifier is less than \_\_\_\_\_ amplifier. (b) Class A (c) Class C (a) Class AB 7. Distortion introduced by non-linearity of dynamic transfer characteristics can be eliminated by \_\_\_\_\_ (a) Push-pull amplifier (b) Audio amplifier (c) RC coupled amplifier 8. Power amplifier is \_\_\_\_\_ power converter. (b) DC to AC (a) AC to DC (c) DC to DC 9. Which regulator circuit gives high efficiency? (a) Switching (b) Series (c) Shunt 10. Dual tracking regulator provides \_\_\_\_\_ power supply. (a) Both positive & negative (b) Positive (c) Negative Q.2 Answer any TEN questions in brief [20] 1. What are the four possible topologies of a feedback amplifier? 2. Define Desensitivity. 3. Give a comparison of positive and negative feedback. 4. Define Rise time and Delay time.

5. Draw the circuit diagram of Colpitt's oscillator.

8. Write the advantage of push pull amplifier.

7. Draw the circuit diagram of class B push pull amplifier.

6. State Barkhausen criteria.

- 9. Define the conversion efficiency.
- 10. What is peak Inverse voltage?
- 11. Draw the block diagram of regulated power supply.
- 12. List the three reasons why unregulated power supply is not good enough for some applications.

Q.3	Explain the input and output regist	
Q.3	Explain the input and output resistance of feedback amplifier.  OR	[10]
Q.3	Explain the general characteristics of Negative feedback.	[10]
Q.4	Draw the circuit diagram of Phase shift oscillator and explain it in detail.	[10]
Q.4	OR	[10]
Q. <del>4</del>	Draw the circuit diagram of Wein bridge oscillator and explain it in detail.	[10]
Q.5	Draw the circuit diagram of Class A Push pull amplifier and explain it.	
0.5	OR	[10]
Q.5	Draw the circuit diagram of Class A large signal amplifier and explain it.	[10]
Q.6	What is SMPS? Explain its working with neat diagram.	
Q.6	OR .	[10]
<b>U.</b> 0	Explain the discrete voltage regulator circuit.	[10]

----- Best of Luck -----



No. of Printed Pages: 2

### SARDARPATEL UNIVERSITY, V.V. NAGAR

T.YB.Sc. Sem-V EXAMINATION

SUB. CODE:-US05CELE02

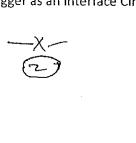
Digital Systems

DATE:-09/11/2017, Throsoday TIME:-10:00 am to 01:00 pm

MARKS-70

Q-1	Choose correct answer		[10]
1.	A serial-in-serial-out shift register can also work as		
	(A) Serial in Parallel Out	(C) Parallel in Parallel Out	
	(B) Parallel in Serial Out	(D) None of these	
2.	In bidirectional register data can be shift	ted from left to right and right to	
	(A)Left	(C) Right	
	(B) Up	(D) None of these	
3.	Carry generate function CG =		
	(A) A • B	(C) A + B	
	(B) A - B	(D) None of these	
4.	Low speed modem uses n	nodulation.	
	(A) FSK	(C)AM	
	(B) AM-FM	(D) None of these	
5.	Registers are made up of		
	(A) Flip-Flops	(C) Resistors	
	(B) Capacitor	(D) None of these	
6.	PAL means		
	(A) Programmable Array Logic	(C) Program Access Memory	
	(B) Programmable Alternate Loop	(D) None of these	
7.	In ring counter data is		
	(A) Circulated	(C) Shift-out	
	(B) Shifted in	(D) None of these	
8.	The carry propagate function mean		
	(A) A+B	(C) A - B	
	(B) A • B	(D) None of these	
9.	is a basic comparate	or.	•
	(A) X-NOR	(C) AND	
	(B) XOR	(D) None of these	
10.	The Schmitt Trigger has U.T.P. &		
	(A) LTP	(C) Zero-Pt	
	(B) NO Pt	(D) None of these	
Q-2	Short answer type question. (any ten)		[20]
1.	Draw the logic diagram of 4-Bit Parallel-	-in-Parallel-out simple shift register.	
2.	Explain UART Drawing Diagram		
3.	Draw the logic symbol of 74LS83 and la		
4.	Draw the logic diagram of 4 bit serial-in		
5.	Draw the logic diagram of 4 bit serial-in		
6.	Draw the logic diagram of 4 Stage Simp	le Ring Counter.	
7.	Draw the logic diagram of 4 Stage Johns		
8.		Circuit that produces CG & CP Functions.	
9.	Draw a neat block diagram of Serial Add	der and explain in short.	
10.	Explain FSK Drawing Diagram.		
11.	Draw the block diagram of digital data	transmission using MODEM.	
12.	Draw the figure of asynchronous data f	ormat.	

Q.3	List different types of data transmission in shift register and explain its working in detail drawing diagram.	[10]
	OR	
Q.3	Draw the neat logic diagram of 4-bit controlled buffer register and explain its working in detail.	[10]
Q.4	Give an account of 4-bit Bidirectional register and explain its working in detail.  OR	[10]
Q.4	List applications of shift register and explain any two in detail.	[10]
Q.5	Explain TWO's complements addition and subtraction using parallel adders.  OR	[10]
Q.5	Give an account of comparator.	[10]
Q.6	Give an account of CMOS-TO-TTL interfacing & Explain in brief TTL-TO-CMOS interfacing.	[10]
	OR	
Q.6	Give an account of Schmitt Trigger as an Interface Circuit.	[10]



SEAT No.\_\_\_\_

No. of Printed Pages 3 2

#### SARDAR PATEL UNIVERSITY

### B.Sc. (5<sup>th</sup> Semester) Examination Saturday, 11<sup>th</sup> November 2017 10:00 a.m. to 1:00 p.m.

US05CELE03 - 8-bit Microprocessor programming and its applications

					Total Marks: 70
Q.1	Choose the correct ans	swer			[10]
1.	8085 microprocessor h	as	·		
	(a) 40 pins (b)	14 pins	(c) 20 pins		
2.	Multiplex address and				•
	(a) A <sub>9</sub> - A <sub>15</sub> (b)	4 <sub>8</sub> - A <sub>15</sub>	(c) AD <sub>0</sub> - AD <sub>7</sub>		
3.	The address bus of mic	roprocessor 8	3085 contains	bits.	
	(a) 8 (b) 16				
4.	Which instruction is us	ed to comple	ment the content o	of accumu	lator ?
	(a) XRA A	(b) ANA A	(c) CMA		
5.	Instructions can be clas		•		
	(a) LSB & MSB	(b) Number	size (c) Wor	d size	
6.	In the data transfer ins	truction the c	data is copied from		
	(a) Source to destination	on (	b) Destination to s	ource	(c) None of these
7.	In 8085 SUI instruction	has	byte size.		
	(a) Three (b)	Four	(c) Two		
8.	Conditional JUMP instr	uction is	•		
	(a) JMP 16-bit address	(b)	JC 16-bit address		(c) None of these
9,	JNC is instruc	tion.			
٠	(a) 1-byte (b	) 2-byte	(c) 3-byte		
10	. Which instruction is us	ed to store th	ne content of accur	nulator to	the specified
	address?				
	(a) STAX D	(b) LDAX B	(c) STA	8500	
		·			
Q.2	Answer any TEN ques	tions in brief	·		[20]
1.	What are the inputs of	ALU?			
2.	How many flags are th	ere in 8085 m	nicroprocessor ? Na	ame them	•
3.	Write a program to loa	ıd 9FH in regi:	ster B, Add 9FH wit	th 4CH usi	ng ADI instruction,
	display the result at th	e OUT PORTO	).	•	
4.	Which instructions are	used to clear	r the content of ac	cumulator	٠ ٢
5.	How many types of da	ta transfer in:	structions are ther	e, list ther	n?
6.	How many addressing	modes are th	ere and name the	m.	
7.	How many types of bra	anch instructi	ons are there, list t	them?	

(I)

[P.T.O]

- 8. List the instructions which transfers the data from memory to microprocessor and microprocessor to memory.
- 9. Define program and software.
- 10. List the logical instructions.
- 11. What is the difference between INR R and DCR R instructions.
- 12. Write a program to load 25H in register B, Display 25H at port address O1H.
- Q.3 Discuss in detail the architecture of 8085 microprocessor. [10] OR Q.3 Draw the pin out diagram of 8085 microprocessor. Explain in detail the functions of each pin. [10] Q.4 (a) Classify the instruction according to their word size and explain it in detail. [05] (b) Explain register addressing mode and direct addressing mode. [05] OR (a) Explain Immediate addressing mode and Indirect addressing mode in details. [05] Q.4 (b) Explain the method of writing and executing a simple program. [05] Q.5 Discuss in detail the Arithmetic instructions. [10] Discuss in detail data transfer instructions. Q.5 [10]Discuss in detail the additional data transfer and 16 bit arithmetic instructions. [10] Q.6 Discuss in detail the programming techniques. Q.6 [10]

----- Best of Luck -----



	SEAT No.		9
		1	
+7		SAR	DAR
$\sim$		ß	.Sc

#### SARDAR PATEL UNIVERSITY

B. Sc. - V Semester

Date 13/11/2017, Monday

Instrumentation I

Session: Morning Time: 10:00 am to 1:00 pm

Course	Code
Course	Coue:

U S 0 5 C E L E 0 4

#### Q-1 Multiple Choice questions:

- 1. The Maxwell bridge is used to measure
  - (i) low value resistors
  - (ii) Capacitor
    - (iii) Inductor
    - (iv) High value resistor
- 2. At the null condition the current through the detector is
  - (i) minimum
  - (ii) maximum
  - (iii) zero
  - (iv) Infinite
- 3. The modification applied to Hay bridge to measure high Q coil is
  - (i) Connecting a resistor in series to capacitor in arm 1.
  - (ii) Connecting a resistor in parallel to capacitor in arm 1.
  - (iii) Connecting an inductor parallel to capacitor in arm 1.
  - (iv) None of above.
- 4. Dissipation factor is
  - (i) reciprocal of quality factor.
  - (ii) proportional to quality factor.
  - (iii) reciprocal of power factor.
  - (iv) none of the above.
- 5. Schering bridge is used to measure capacitance of a capacitor whose phase angle is
  - (i) is nearer to 90°.
  - (ii) is lower than 90°.
  - (iii) is equal to 90°.
  - (iv) none of above
- 6. Transducer forms a part of \_\_\_\_\_\_ in instrumentation system.
  - (i) Input device
  - (ii) Output device
  - · (iii) Processing device
    - (iv) All of these.
  - 7. In which transducer the emf voltage is generated across the junction of two dissimilar metals when junction is heated
    - (i) Thermistor
    - (ii) Photoconductive cell

,	(iii) (iv)	Thermocouple Solar cell		
9. 1	Γhermist (i) (ii) (iii) (iv)	cors are widely used in the temperature range -100°C to 100°C -100°C to 200°C -100°C to 300°C 100° C to 300° C		
10.	In the c (i) (ii) (iii) (iv)	onstruction of phototube the photosensitive material is used in construction of Anode Cathode Grid Source		
2. 3. 4. 5. 6. 7. 8. 9.	Draw Write Why K What Draw What Classif In what What	er any ten questions in brief.  simplified circuit of commercial Kelvin double bridge.  two differences between ac bridge and dc bridge.  Kelvin bridge is called double bridge? What is Kelvin bridge used for?  do you mean by Dissipation factor? What does it tell?  impedance triangle for inductor and capacitor?  are the important blocks of instrumentation system?  e Transducer.  fy transducer. Give two examples.  at shapes and size the thermistors are available in the market?  are thermistors? They are prepared from which material?  are the three important characteristics of thermistors?  e Acrystalline materials. Where Piezoelectric crystals are used?	20	
		cribe Kelvin double bridge in detail.		10
Q-3	B. The BC DA	OR plain why Maxwell bridge is unsuited for measurement of low Q coils. The ac bridge is in balance with following constants. Arm AB, R=450 $\Omega$ , arm C, R=300 $\Omega$ in series with capacitor C=0.256 $\mu$ F and arm CD unknown, arm $\Omega$ R=200 $\Omega$ in series with inductor L=15.9 mH. The oscillator frequency is 1	·	5 5
Q-4	KH Describ	oe in detail Hay bridge and show that it is suitable for the measurement of High (	Q coil.	10
Q-4		OR be in detail Schering bridge and show that the dial of Schering bridge can be cali y in terms of dissipation factor D.	brated	10
Q-5	Expla	in working of LVDT in detail.		10
Q-5	Write	OR a short note on Transducer.		10
Q-6	Describ	be working of Photoelectric transducer in detail.		10
Q-6	·Descri	OR be any two characteristics of Thermistors.  *********  ************************		10
		(2)		

Sc

SEAT No. \_\_\_\_\_ No. of Printed Pages: 02

SARDAR PATEL UNIVERSITY V.V.NAGAR

[33/A-13]

B.Sc. (V<sup>th</sup> SEM.) ELECTRONICS 15<sup>th</sup>NOVEMBER-2017 EXAMINATION SUB. - INDUSTRIAL ELECTRONICS- I SUB.CODE-US05CELE05

TIME	:-10:00 am to 1:00 pm	MA	ARKS-70		
Q-1	Choose correct answer		[10]		
1.	Thyristor mainly used for				
	(A) Power controlling	(C) Relaxation oscillation			
	(B) Rectification	(D) None of above			
2.	Recommended method to TURN-ON				
	(A) Triggering by A.C signal	(C) Triggering by D.C signal			
	(B) Triggering by pulse signal	(D) None of above			
3.	Dc motor consists of				
	(A) TRIAC	(C) Amplifier			
	(B) Stator	(D) None of above			
4.	Stepper motor used fors	• •			
	(A) Discrete	(C) Linear			
	(B) Constant	(D) None of above			
5.	SUS means	,			
	(A) silicon unilateral switch	(C) silicon uni-port switch			
	(B) silicon uni-junction switch	(D) None of above			
6.	UJT mainly used for	. ,			
	(A) relaxation oscillator	(C) amplification			
	(B) rectification	(D) None of above			
7.	TRIAC type of device.				
	(A) multi-directional	(C) bi-directional			
	(B) uni-directional	(D) None of above			
8.	Connection of SCR used for controlling very high current.				
	(A) series	(C) bi-directional			
	(B) parallel	(D) None of above			
9.	is a circuit which convert	t DC power in to AC power at desired			
	output voltage.	•	•		
	(A) Rectifier	(C) Filter			
	(B) Inverter	(D) None of above			
10.	Static equalising circuit is external co	ompensating circuit to produce uniform			
	voltage in operation of SCR.				
	(A) parallel	(C) normal			
	(B) series	(D) None of above	-		
Q-2	Short answer type question. (any to	en)	[20]		
1.	State function of gate in SCR.	•			
2.	Define string efficiency.				
3.	Differentiate between semiconduct	or and thyristor device.			
4.	What do you mean by power contro	ofling action?			
5.	Differentiate between D.C. motor a	nd stepper motor.			
6.	State principle of operation of stepp	er motor.			
7.	Define reverse recovery current IRR.				

8. 9. 10. 11. 12.	Define holding current I <sub>h.</sub> List application of thyristor device. Differentiate TRIAC and SCR. Briefly explain overvoltage protection circuit. State different type of inverter circuit.	
Q.3(A)	Discuss principle of operation and characteristics of SCR.	[06]
Q.3(B)	Explain TURN-OFF mechanism of SCR.  OR	[04]
Q.3(A)	Discuss different method of turning on a SCR with TURN-ON characteristics.	[06]
Q.3(B)	Briefly discuss gate control circuit in SCR.	[04]
Q.4	Discuss the series operation of SCR with necessary diagram and compensation circuits.	[10]
Q.4	OR  Discuss the parallel operation of SCR with necessary diagram and compensation circuits.	[10]
Q.5(A)	Discuss characteristics and operation of UJT with necessary diagram.	[06]
Q.5(B)	Write a note on UJT as a relaxation oscillator.  OR	[04]
Q.5(A)	Discuss triggering mode of TRIAC.	[06]
Q.5(B)	Explain application of SCR as static circuit breaker.	[04]
Q.6	Draw the schematic diagram of D.C motor and discuss its working in detail.  OR	[10]
Q.6(A)	Explain series inverter giving necessary diagram and waveforms.	[06]
Q.6(B)	Write a note on stepper motor	[04]



SC

SEAT No.

No. of Printed Pages ? O2

### [26]

### SARDARPATEL UNIVERSITY V.V.NAGAR

T.YB.Sc. Sem-V EXAMINATION

SUB. CODE:-US05CELE06

**Analog Communications** 

DATE:-17/11/2017 TIME:-10:00 am to 1:00 pm

MARKS-70

Q-1	Choose correct answer		[10]		
1.	A single transistor working as frequency mix				
	(A) converter	(C) mixer			
	(B) local oscillator	(D) None of these			
2.	IF frequency in case of of super-heterodyne	•			
	(A)455 khz	(C) 55 khz			
	(B) 45 khz	(D) None of these			
3.	Aspect ratio used in T.V receiver is				
	(A) 4:3	(C) 3:3			
	(B) 4:4	(D) None of these			
4.	The function of amplitude de-modulation	is opposite of			
	(A)Amplitude modulation	(C)Frequency modulation			
	(B) Phase modulation	(D) None of these			
5.	The function of IF amplifier in super hetero-	dyne receiver is			
	(A) Reject	(C) Attenuate			
	(B) select and amplify	(D) None of these			
6.	As different stations are tuned in Radio Rece	eiver o/p frequency of mixer is			
	(A) Constant	(C) Changes			
	(B) Zero	(D) None of these			
7.	Problem of flicker is solved by				
	(A) Interlaced Scanning	 (C) Cut-off			
	(B) Blanking	(D) None of these			
8.	is used to match impedance of	TV Receiver to input.			
	(A)BALUN	(C) RF TUNER			
	(B) IF TRAP	(D) None of these			
9.	In TV sound signal ismodulated.				
	(A) AM	(C) PM			
	(B) FM	(D) None of these			
10.	In TV picture signal ismodula				
	(A) AM	(C) PM			
	(B) FM	(D) None of these			
Q-2	Short answer type question. (any ten)		[20]		
1.	Draw the block diagram of mixer and explai	n its working.	[1		
2.	Explain interlaced scanning.				
3.	Explain principle of superhetrodyne.				
4.	Draw the circuit of linear diode detector.				
5.	Explain how AVC works.				
6.	List salient features of Broad Cast Receiver.				
7.	Give classifications of a Radio Receiver.				
		•			
8.	Explain Aspect Ratio.				
9.	Explain scanning.				
10.	Draw the circuit of BALUN.				
11.	Draw the block diagram of VHF Tuner in TV	•			
12.	Draw the diagram of IF trap.				

Q.3	Draw the circuit of RF Amplifier and explain its working .  OR	[10
Q.3	Draw the circuit of frequency converter using one transistor and explain its working	[10
Q.4	Draw the block diagram of super hetero-dyne Radio Receiver and explain function of each block.	[1
	OR	
Q.4	List basic functions of AM Receiver and explain each in detail.	[1
Q.5	Give an account of Composite Video Signal.	[1
	OR	
Q.5	Give an account of Image Orthicon Camera tube.	[1
Q.6	Draw a neat block diagram of Black & White Broadcast TV Receiver and label it.  OR	[1
Q.6	List different inter stage coupling methods. Explain any two in detail.	[1
	X	

[38/A27], SEAT No. SARDAR PATEL UNIVERSITY No. of Printed Pages: 2 SC

### T. Y. B.Sc, (FIFTH SEMESTER) EXAMINATION

2017

# Tuesday,7<sup>th</sup> November 10.00 am to 1.00 pm

### US05CENV01 Environmental Biotechnology

				Tota	d Marks	:70
answer and wr	ite in an	swer sh	eet		(1	0)
ronmental probl	ems with	i the use	of plant is cal	lled as	· ·	,
(b) Bioreme	diation	(c)	Phytoremedi	ation –	(d) Phyto	
11						
te clay has	perm	eability	which prever	nts dispe	ersal of oxy	/gen
ıı (h) Mediun	1	(a)	T 0***		(1) 17	
on of a contami	ı nant hv s	(c) a nlant is	LOW known ac		(a) Ex	cess
(b) Phytodegra	adation	(c) Phy	vtoextraction	(d) Pł	 vtostabiliz	aton
ally grown for th	ne purpos	se of		(4) 11	.,	aton
				•		
(b) Phytostabiliz	zation	(c) Ph	ytodegradatio:	n (d)	Phytoextra	action
reecfully utilize	nd for nor	marral af	P		1 1 .	
costumy utilize	tu ioi iei	novai oi	radioactive c	ompour	ids during	
(b) Sunflower		(c) Pon	lar	(d) I.	emna	
iferative mass o	f cells pr	oduced	from isolated	plant tis	sue is calle	ed
(b) callus		(c) exp	lant	(d) en	dosperm	
sed to produce h	aploid pl	lants	1	(1)		
number of plant	lets in sh	c) ant	ner through tiggu	(d) pro	otoplast	
(b) mass cultiv	zation	(c) Mic	ronronagatior	c Cantair (d) or:	o is afting	
the scissors of n	nolecular	genetic	s	. (4) 8.0	arting.	
(b) Lipases		(c) Exo	nucleases	(d)Re	striction	
oro :	urand as t	a a 1 d a duu	· · · · · · · · · · · · · · · · · · ·			
,are   h) bacteria	usea as a	ooi to tra	ansier the gen	es (d) m	aolona	
(o) odoloma		(0) 1 143	imu	(a) III	icieus	
nv Ten)					,	(20)
	ediation				·	(20)
SS						
minar Air Flow	7					
g of Autoclave						
Endonucleases ?	)					
ication of rDN	A techno	logy				
			purpose			
	conmental proble (b) Bioremed  te clay has l (b) Medium on of a contami (b) Phytodegra ally grown for the (b) Phytostabilize cessfully utilize (b) Sunflower iferative mass o  (b) callus ted to produce h (b) embryo number of plant (b) mass cultive the scissors of n (b) Lipases  (b) Lipases  (c) bacteria  are to (c) bacteria  (d) extrinsic biorem (extrinsic biorem (	ronmental problems with (b) Bioremediation  te clay has permediation  te clay has permediation (b) Medium on of a contaminant by a (b) Phytodegradation ally grown for the purpose (b) Phytostabilization  tecessfully utilized for rer (b) Sunflower iferative mass of cells pr (b) callus ted to produce haploid pl (b) embryo number of plantlets in sh (b) mass cultivation the scissors of molecular (b) Lipases  are used as te (b) bacteria  my Ten) extrinsic bioremediation  as remediation attor species ? tion Explain aminar Air Flow g of Autoclave Endonucleases ? ication of r DNA technological production  and the scissors of molecular contains a product of the scissors of molecular contains	ronmental problems with the use (b) Bioremediation (c)  te clay has permeability (b) Medium (c) on of a contaminant by a plant is (b) Phytodegradation (c) Phytologradation (d) Phytostabilization (e) Phytostabilization (f)	(b) Bioremediation (c) Phytoremediate clay has permeability which prevent [1]  (b) Medium (c) Low on of a contaminant by a plant is known as (b) Phytodegradation (c) Phytoextraction ally grown for the purpose of (b) Phytostabilization (c) Phytodegradation (c) Phytodegradation (c) Phytodegradation (c) Phytodegradation (c) Sunflower (c) Poplar iferative mass of cells produced from isolated (d) callus (c) explant (e) explant (e) embryo (c) anther number of plantlets in short time through tissue (b) mass cultivation (c) Micropropagation (c) Exonucleases (d) Lipases (e) Exonucleases (e) Exonucleases (e) Exonucleases (f) bacteria (f) Plasmid (f	answer and write in answer sheet ronmental problems with the use of plant is called as (b) Bioremediation (c) Phytoremediation  the clay has permeability which prevents disped to be clay has permeability which prevents disped to phytodegradation (c) Low on of a contaminant by a plant is known as (b) Phytodegradation (c) Phytodegradation (d) Phytodegradation (d) Phytodegradation (d) Phytodegradation (d) Contaminant of composition (e) Poplar (d) Low coessfully utilized for removal of radioactive compount (b) Sunflower (c) Poplar (d) Low coessfully utilized for removal of radioactive compount (e) Contamination coefficient (d) entire through tissue culture (d) produce haploid plants (e) explant (d) entire through tissue culture (d) mass cultivation (c) Micropropagation (d) grather scissors of molecular genetics (e) Exonucleases (d) Reference (e) Plasmid (d) not the scissors of molecular genetics (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Plasmid (d) not premediation (e) Phytodegradation (e) Phytode	ronmental problems with the use of plant is called as  (b) Bioremediation (c) Phytoremediation (d) Phyto  re clay has permeability which prevents dispersal of oxy  (b) Medium (c) Low (d) Exponsion of a contaminant by a plant is known as (b) Phytodegradation (c) Phytoextraction (d) Phytostabilization for the purpose of  (b) Phytostabilization (c) Phytodegradation (d) Phytostabilization (e) Poplar (d) Lemna (ferative mass of cells produced from isolated plant tissue is called (e) Company (e) Compa

(PTO)

<ul><li>Q.3.(a) What is Bioremediation? Explain any two methods of In situ bioremediation.</li><li>(b) Explain Land farming and composting</li></ul>	(05) (05)
OR	
<ul><li>Q.3.(a) Discuss two methods of Ex situ bioremediation in brief</li><li>(b) What is Biosparging and bioslurping? Explain</li></ul>	(05) (05)
<ul><li>Q.4 (a) Discuss the Phytovolatization process</li><li>(b) What is Rhizofiltration?</li></ul>	(05) (05)
OR	
<ul><li>Q.4 (a) Discuss the Role of Hyperaccumulator Plants</li><li>(b) State in brief the advantages of Phytodegradation</li></ul>	(05) (05)
Q.5.(a) Give a brief account on General process of Plant Tissue culture	(05)
(b) Write about the media composition for Plant tissue culture	(05)
OR	
Q.5. (a) Discuss various methods of Organ culture	(10)
<ul><li>Q.6. (a) Explain the recombinant DNA technology with suitable diagrams</li><li>(b) State Role of Plasmid in r DNA technology</li></ul>	(05) (05)
OR	
<ul><li>Q.6. (a) Discuss the Role of Endonucleases in r DNA technology</li><li>(b) What are transgenic animals? Discuss one example</li></ul>	(05) (05)

No. of Printed Pages: 2

#### SARDAR PATEL UNIVERSITY

### T. Y. B. Sc. (FIFTH SEMESTER) EXAMINATION

#### 2017

### THURSDAY, 9th NOVEMBER

Time: 10.00 a.m. to 01.00 p.m.

### **US05CENV02 (ENVIRONMENTAL SCIENCE)**

### (PESTICIDE, HERBICIDE, FUNGICIDE TOXICOLOGY)

- Note: 1. Answers of all the questions (including multiple choice questions) should be written in the provided answer book only
  - 2. Figures to the right indicate the full marks of sub question
  - 3. Draw neat and labelled diagrams wherever necessary

Maximum Marks: 70

Q. 1.	Multiple choice qu	estions		(10)
1.	The major biotransfo	ormation site in vertebr	ates is	
	a. Skin		<b>c.</b> Kidneys	d. Lungs
2.	If respiratory tract is	affected by toxicants t		<del>-</del>
	a. Ingestion		c. Absorption	
3.	Access of some toxi	cants to foetus is preve		
		b. GI tract		
4.	Which of the following	ng is an organochlorine	insecticide?	
	a. Malathion	<b>b.</b> Aldrin	c. Pyrethroids	d. Chloroquine
5.	Pesticides that kill th	e target species by va		
		rbicides <b>c.</b> Fur		
6.		f cross linked fatty acid		
		<b>b.</b> Epi- cuticular wax		d. Nucleus
7.				o plasmadesmata is called
	a. Non polar entry	<b>b.</b> Polar entry	c. None	d. Both
8.	The meaning of mole	ecular fate is	·	
	a. Bioagumentation	b. Bioaccumulation	c. Bioconcentration	d. Detoxification
9.		is a type of fu		
	a. Sulphur	b. Copper	c. Mercury	d. Carboxyamide
10		not used on acidic soi		•
	a. Captan	<b>b.</b> Fyton	c. Ziram	d. Benomyl

Q. 2.	Answer the following questions in short (Any Ten)	20)
	1. Define: Bioassay	
	2. Write a note on Dose-response curve	
	3. Enlist the storage depots of toxicants in human body	٠
	4. What is class resistance? What is cross resistance?	
	5. Draw structure of Aldrin and Dialdrin insecticides	
	6. What is axonic poison? What is neurosynaptic poison?	
	7. Define: Symplastic system	
	8. Enlist the morphological response to herbicide by the plants	
	9. Write about Biotic factors affecting foliar herbicide absorption	
	10. Define: Protective fungicide & Curative fungicide	
	11. Write about properties of Ziram & Thiram	
	12. Draw the structural formula of Dichlone fungicide	
Q. 3.	a. Write in detail about sites of absorption of toxicants in the body	(06)
	<b>b.</b> Write a short note on excretion of toxicants from the body of organism	(04)
0.0	OR	(40)
Q. 3.	Describe in detail about Phase II reaction of biotransformation of Xenobiotics	(10)
Q. 4.	a. Write a detailed note on classification of pesticides	(06)
	b. Write a short note on: DDT in environment	(04)
	OR	(0.0)
Q. 4.	<ul><li>a. Write in detail on impact of pesticides on various organisms</li><li>b. Write a note on: Pesticides Resistance</li></ul>	(06) (04)
Q. 5.	a. Explain Soil Application and draw the hypothetical diagram representing herbicide	(06)
α. σ.	absorption into roots	` '
	b. Write a note on the effect of Dichlobenil herbicide on the growth and plant structure OR	(04)
Q. 5.	a. Explain Absorption and translocation of triazine and its effect on the growth and plant structure	(05)
	<ul> <li>b. Draw the labelled schematic representation of plasma membrane affected by various toxic molecules</li> </ul>	(05)
Q. 6.	<ul><li>a. Write a note on different types of Copper fungicides along with its subtypes</li><li>b. Write about Benomyl fungicides</li></ul>	(05) (05)
	OR	
Q. 6.	a. Write a detail note on Dichlone Fungicide	(05)
	b. Describe Organophosphorous fungicides	(05)

# SARDAR PATEL UNIVERSITY B.Sc. FIFTH SEMESTER EXAMINATION

2017

#### SATURDAY, 11<sup>th</sup> NOVEMBER

#### 10:00 am to 01:00 pm

#### **USO5CENV03**

#### (ENVIRONMENTAL POLLUTION)

Marks: 70

Note: 1. Answers of all the questions (including multiple choice questions) should be written in the provided answer book only

2. Draw neat and labeled diagrams wherever necessary

Q.1. Select the corr	ect answer and write it in the	e answer sheet.	[10]
1technique	is suitable for separation of la	orger particles having a	size more than 10μ
(a) Filtration	(b) Sedimentation	(c) Impingement	(d) Freeze out
2 method is	usually employed when the g	iven pollutant is extrer	nely low in concentration
(a) Adsorption	(b) Absorption	(c) Sedimentation	(d) Freeze out
3 pollutant	are organic in nature and conta	ain mainly carbon and	hydrogen
(a) Inorganic	(b) Organic	(c) Solid particles	(d) Liquid particles
4. The study of soil	s commonly known as	_	
(a) Pedology	(b) Edaphalogy	(c) Both of these	(d) None of these
5 are readil	y biodegradable in nature		
(a) Traizine	(b) Carbamates	(c) Dalapon	(d) DDT
6. 75% radiation cau	ised DNA damage is due to w	hich free radical?	
(a) OH	(b) НОН	(c) O	(d) HO
7. Number of vibrat	ions per second is known as _		
(a) Frequency	(b) Wave length	(c) Amplitu	de (d) Sone
8. A °C rise	in temperature doubles the to	xic effect of potassium	cyanide
(a) 5°C	(b) 10°C	(c) 15°C	(d) 20 °C
9 has high	octane and relatively clean cor	mbustion characteristic	es.
(a) Petrol	(b) CNG	(c) Ethanol	(d) Diesel
10 is a stand	ard measure of the performan	ce of an engine or avia	tion fuel
(a) Octane number	(b) Oxygen	(c) Combus	tion (d) Fuel

Q.2. Answer the following questions in other (Any 10)	[20]	
1. Define Air pollution according to the Bureau of Indian Standards		
2. Write a note on indoor air pollution		
3. Write note on effects of SO <sub>X</sub> on materials		
4. Enlist sources of soil pollution		
5. What is radioactive fallout?		
6. Discuss the natural sources of radiation pollution		
7. State the principle of sound level meter		
8. Name the different parts of human ear		
9. Define thermal pollution and enlist its sources		
10. Describe carbon monoxide as a vehicular pollutant		
11. Define vehicular pollution		
12. Write note on particulate matter		
Q.3. (a) Describe Packed tower scrubbers and Plate tower scrubbers for the control of gaseous pollutants		
(b) Explain the measurement of pollutants by dust-fall determination	[04]	
OR	[05]	
Q.3. (a) Describe the sampling of particulate matter		
(b) Classify air pollutants	[05]	
Q.4. (a) Discuss the adverse effects of fertilizers and pesticides in soil		
(b) Explain the procedures to control soil pollution		
OR		
Q.4. (a) Discuss the effects of radiation pollution	[05]	
(b) Explain the methods for prevention and control of radiation pollution	[05]	
Q.5. (a) Describe the sources of noise pollution	[05]	
(b) Discuss the control measures of noise pollution	[05]	
OR		
Q.5. (a) Discuss the effects of thermal pollution	[05]	
(b) Explain the measures to control thermal pollution	[05]	
Q.6. Discuss the scenario of air pollution in India due to automobile emission OR	[10]	
Q.6. Describe biodiesel as an alternative fuel and abatement of vehicular pollution	[10]	

	۲.	•			•		٠	 7
(39]		SI	TAT	No		- Andrew Company	sedē	
		era saka	F					

No. of Printed Pages: 9\_

# SARDAR PATEL UNIVERSITY T.Y.B.Sc.5<sup>th</sup> SEMESTER EXAMINATION 13<sup>th</sup> November 2017, Monday 10.00 AM to1.00 PM Earth System Sciences (US05CENV04)

Multiple Choice Questions (one mark each)  Drainage is topographic region from which a stream receives runoff.  (a)density (b)texture (c)basin (d)frequency process involving uplift or warping of large parts of the Earth's crust.  (a)Epeirogenic(b)Orogenic(c) Both of the above(d) None  Lusture of diamond is known as (a)Metallic (b) Pearly (c) Vitreous (d) Adamantine  Change in colour of mineral due to oxidation at surface is known as (a) Lusture (b)Tarnish (c) Iridescence (d)False colour If the system can exchange energy,but not mass with its surroundings is called as (a)Closed system (b) Isolated system (c) Phase (d) Open system Earthquakes are measured using observations from
Drainage is topographic region from which a stream receives runoff.  (a)density (b)texture (c)basin (d)frequency process involving uplift or warping of large parts of the Earth's crust.  (a)Epeirogenic(b)Orogenic(c) Both of the above(d) None Lusture of diamond is known as  (a)Metallic (b) Pearly (c) Vitreous (d) Adamantine Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b)Tarnish (c) Iridescence (d)False colour If the system can exchange energy,but not mass with its surroundings is called as
(a)density (b)texture (c)basin (d)frequency process involving uplift or warping of large parts of the Earth's crust.  (a)Epeirogenic(b)Orogenic(c) Both of the above(d) None  Lusture of diamond is known as  (a)Metallic (b) Pearly (c) Vitreous (d) Adamantine  Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b)Tarnish (c) Iridescence (d)False colour  If the system can exchange energy, but not mass with its surroundings is called as  (a)Closed system (b) Isolated system (c) Phase (d) Open system
process involving uplift or warping of large parts of the Earth's crust.  (a)Epeirogenic(b)Orogenic(c) Both of the above(d) None  Lusture of diamond is known as  (a)Metallic (b) Pearly (c) Vitreous (d) Adamantine  Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b)Tarnish (c) Iridescence (d)False colour  If the system can exchange energy, but not mass with its surroundings is called as  (a)Closed system (b) Isolated system (c) Phase (d) Open system
crust.  (a)Epeirogenic(b)Orogenic(c) Both of the above(d) None  Lusture of diamond is known as
Lusture of diamond is known as  (a) Metallic (b) Pearly (c) Vitreous (d) Adamantine Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b) Tarnish (c) Iridescence (d) False colour If the system can exchange energy, but not mass with its surroundings is called as  (a) Closed system (b) Isolated system (c) Phase (d) Open system
(a) Metallic (b) Pearly (c) Vitreous (d) Adamantine Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b) Tarnish (c) Iridescence (d) False colour If the system can exchange energy, but not mass with its surroundings is called as  (a) Closed system (b) Isolated system (c) Phase (d) Open system
(a) Metallic (b) Pearly (c) Vitreous (d) Adamantine Change in colour of mineral due to oxidation at surface is known as  (a) Lusture (b) Tarnish (c) Iridescence (d) False colour If the system can exchange energy, but not mass with its surroundings is called as  (a) Closed system (b) Isolated system (c) Phase (d) Open system
(a) Lusture (b) Tarnish (c) Iridescence (d) False colour  If the system can exchange energy, but not mass with its surroundings is called as  (a) Closed system (b) Isolated system (c) Phase (d) Open system
If the system can exchange energy, but not mass with its surroundings is called as  (a)Closed system (b) Isolated system (c) Phase (d) Open system
If the system can exchange energy, but not mass with its surroundings is called as  (a)Closed system (b) Isolated system (c) Phase (d) Open system
said industries site interserve sonig observations non
(a) Tsunameter (b)Creepmeters (c) Seismometers (d)Magnetometer
zone is least active seismic zone.
(a)Zone II (b)Zone III (c) Zone IV (d) Zone V
Taifu is the term used for tropical cyclone in
(a)Japan (b)America (c)Australia (d) China
Time lag between successive tsunami waves ranges between
(a)10 to 20 minutes(b)20 to 40 minutes(c)10 to 50 minutes(d)20 to 60 minutes
In Japan tropical cyclones are termed as
(a)Bagulo (b)Typhoons (c) Taifu (d)Willy willy
(a)Baguio (b) 1 yphoons (c) Taitu (d) Willy willy
Answer any ten
Give a short note on denudational processes.  Discuss in brief diastrophism and sudden movement.  How erosion and deposition takes place in weathering?  What is meant by Pseudochromatic colour for minerals?  Define (i) Streak (ii) Cleavage.
(P)

6 7	Enlist Mineral forming processes.  Define Mineral Stability.	
8	Define Earthquake.	
9	Explain in brief classification of volcano based on the frequency of eruption.	
- 10	What are the different types of floods? Explain	
]]	Write a note on properties of hurricanes. Briefly discuss 2 types of Tsunami.	
12	Drietty discuss 2 types of 1 stillaint.	
Q.3a.	Write a note on drainage pattern of river.	5
b.	Give a detailed note on chemical weathering processes.	5
	OR	
Q.3a.	Explain in detail biological process of weathering.	5
у. <i>эа.</i> b.	Give a detailed note on endogenic geomorphic processes.	5
Q.4a.	Discuss specific gravity.	5
Ь.	Write a note on hardness of minerals.	5
	OR	5
Q.4a.	Write a note on property of cleavage in minerals.	5 5
b.	Explain in detail Water Cycle.	,
Q.5a.	Describe in detail volcanic hazards and its impacts.	5
b.	Discuss Gujarat Earthquake case study in detail.  OR	5
Q.5a.	Explain types of volcanoes and their characteristics.	5
b.	Write a detailed note on prediction and causes of Earthquake.	5
Q.6	Write a note on Pre-Tsunami disaster stage. Discuss characteristic feature of	10
	Tsunami waves.	
	OR .	
Q.6 .	Discuss characteristic features of tropical cyclones.	10

,		•	 •	٠.
	SEAT	No	 <del></del>	

No. of Frinted Pages: 02

### <u>[34]</u>

# SARDAR PATEL UNIVERSITY T.Y.B.Sc.5th SEMESTER EXAMINATION 15<sup>th</sup> November 2017, Wednesday 10.00 AM to 1.00 PM

Natural Resources, US05CENV 05

Q.1 1.	Multiple Choice Questions (one mark each)  Forests arenatural resources.	Marks-70 10
*	(a)biotic (b)abiotic (c) cyclic (d) potential	
2.	energy is derived from water.	
	(a)solar (b)wind (c) hydropower (d) geothermal	
3.	Wind energy is energy resource	
4.	NALCO is concerned with	
5.	(a)zinc (b) aluminium (c) gold (d) manganese Rain water harvesting is a method of conservation.  (a)soil (b) water (c) food (d) forest  Overgrazing causes	
6.	Overgrazing causes (c) food (d) forest	
7.	(a)pollution (b) floods (c)desertification (d)all of the above Dams are built for	
8.	(a) power generation(b) flood prevention (c) both (d) none  is a soil conservation practice.  (a) planting of trees (b) grazing (c) erosion (d) None of these	
9.	is alternative use of insecticide which is environmentally safe  (a) rodenticide (b) biocide (c) pesticide (d) none	
10.	The World Food Conference was held in the year  (a)1986 (b) 1976 (c) 1990 (d)2000	
Q.2 1 2 3	Answer any ten Give a note on geothermal energy. Write a short note on pros and cons of nuclear power. Write down the renewable natural resources.	20

4 5 6 7 8 9 10 11	Define soil erosion. Which are the causes of desertification? Write a note on Manganese. Discuss physical properties of water. Write down uses of Dams. Define drip irrigation. Enlist three methods of water conservation. Explain various aquatic food resources	
12	Explain the uses of biocides.	
Q.3	<ul><li>(a) Give a classification of Natural Resources</li><li>(b) Write about conservation of Natural Resources</li></ul>	[66] . [04]
	OR	
Q.3.	Write Short notes: (a)Hydropower (b)Tidal energy	[06] [04]
Q.4.	(a) Write about the causes and agents of soil erosic (b)Explain the land management practices.	on. [06 [04]
	OR	
Q.4.	Write Short notes: (a) Iron ore (b) Zinc	[06] [04]
Q.5.	<ul><li>(a) Write about benefits and problems of dams.</li><li>(b) Explain chemical properties of water.</li><li>OR</li></ul>	[05] [05]
Q.5.	<ul><li>(a) Write a short note on sources of water</li><li>(b) Write in detail about drip irrigation.</li></ul>	[05] [05]
Q.6.	Give a detailed account on World Food Problem.  OR	[10]
Q.6.	Write all the plant and animal food resources.	[10]

**-** × -

No. of Printed Pages : 02

### SARDAR PATEL UNIVERSITY T.Y.B.Sc.(FIFTH SEMESTER) EXAMINATION

#### 2017

#### Friday, 17<sup>th</sup> November 10.00 am to 1.00 pm

### US05CENV06 Population Dynamics and Biostatistics

•	US05CENV06 Populati	on Dynamics will	Total Marks -70
Q.1. Select the c	orrect answer and write	it in the answer sh	(10)
<ol> <li>Demographic</li> <li>Industrial</li> </ol>	transition is an important fa (b) Commercial	actor in (c) Population	growth (d) Economic
<ul><li>2. Ethics is deriv</li><li>(a) Ethos</li></ul>	red from Greek word (b) Etuis	(c) Evokes	(d) Educes
(a) Hepatotoxic	ainly affected by (b) Nephrotoxic	(-)	
4.Hyperkeratosis (a) Barium	is a pathological effect of (b) Copper	(c) Lead	metal (d) Arsenic
(a) RPMT	cholerae was isolated by (b) GeorgesI.	(c) 1. William	(d) Robert Koch
(a) 4.3	hector bio-producti (b) 1.2	(•)	
(a) 26	(b) 30	(0)	pounds CO <sub>2</sub> in atmosphere (d) 56
8.Frequency dis	stribution table can be com (b) Binominal	pared with (c) Both of ab	standard distributions bove (d) None of above
9. Map diagran (a) Geographic	n shows dis al (b)Comparative	stribution of variou (c) Imaginary	s frequencies (d)Tabular
10. Quota sam (a) two	pling has mainly (b) three	_types (c) four	(d) five

CP. T.O.)

Q.2. Answer the following in brief (Any Ten)	(20)
1. What is the need of Environmental Ethics? Explain.	
2. Give abrief account on environmental issue: "Diminishing of biological capital"	
3. Discuss short term factors which affect the population	
4. Write short note on effects of Mercury on public health	
5. Give a note on Polio disease	
6.Define the terms: (i) Health (ii) Disease	
7.Discuss sustainable ways to control air pollution	
8. Write a note on biophysical measurements for sustainable development	
9. Give a brief account on conservation of energy in home and yard sustainably	
10. Write in short about arithmetic mean	
11.Discuss cluster sampling	
12. What is sample and sampling?	
Q.3 (a) Write in brief about exponential growth of population study	(05)
(b) Discuss: Howguiding principles helps to follow environmental ethics?	(05)
OR	
Q.3 (a) Give a note on basic directives of environmental ethics	(05)
(b)Writea note onLogistic growth curve for population studies	(05)
Q.4 (a) Discuss effects of severe environment	(05)
(b) Give a note on HIV with appropriate diagram	(05)
OR	
Q.4 (a) Write a note on types of stress	(05)
(b) Discuss the effects of disease causing pathogens in water	(05)
Q.5 (a) Discuss imperative for sustainable development	(06)
(b) How can one have improvement of vehicle fuel efficiency sustainably?	(04)
OR	
Q.5 (a) Write a note on control of noise pollution	(06)
(b) Give a brief note on environmental degradation and conservation issues	(04)
Q.6 Give a detailed note on methods to present statistical data	(10)
OR	. /
Q.6 Calculate mean, median, mode and standard deviation of given data	(10)
Gl. 10 20 20 20 20 40 40 50 50 C0 C0 70 70 90 90	. 00

Class	10-20	20-30	30-40	40-50	50-60	60-70	70-80	80-90
Frequenc	32	29	21	23	46	10	35	41



Se	eat No.: No. of Printed Pages: 2	
(3	SARDAR PATEL UNIVERSITY	
	B. Sc. (Genetics) – Fifth Semester Examination (CBCS)	
	Tuesday, 7 <sup>th</sup> November 2017	
	10:00 a.m. to 1:00 p.m. US05CGEN01: Instrumental Methods of Analysis	
	Total Marks: 70	
N	ote: (1) Figures to the right indicate marks. (2) Draw a neat and labeled diagram, wherever necessary.	
 Q. 1		 [10]
_	. Amplitude is responsible for of light.	[xv]
ı	(a) Colour (b) Intensity (c) Power (d) Bending	
	· · · · · · · · · · · · · · · · · · ·	
ii	A A C A L	
	of the magnified images.	
	(a) Condenser lens (b) Objective lens (c) Diaphragm (d) Eye piece	
iii	· · · · · · · · · · · · · · · · · · ·	
	(a) Fluorescence Microscope (b) Transmission Electron Microscope	
•	(c) Scanning Electron Microscope (d) Phase Contrast Microscope	
iv	3.0 A C C C C C C C C C C C C C C C C C C	
	(a) Poise (b) Swedburg (c) RPM (d) RCF	
V		
	(a) RPM (b) Sedimentation coefficient (c) RCF (d) Sedimentation velocity	4
vi	• • • — — —	
	(a) Sunlight (b) Prism (c) Hydrogen lamp (d) Tungsten filament	
vii		
	injector and	
	(a) Pump (b) Analytical column (c) Recorder (d) Detector	
viii	O I V	
	(a) Liquid-solid chromatography (b) Ion exchange chromatography	
	(c) Gel permeation chromatography (d) Liquid-liquid chromatography	
ix	•	
	(a) IEF (b) Cellulose acetate electrophoresis (c) PFGE (d) Agarose gel	
X	·	
	(a) Silver nitrate (b) CBB (c) Ehidium bromide (d) BPB	
	~ ~ ~	

<u>P.T.O.</u>

	Q.Z	Answer any <u>TEN</u> from the following:	[20]
	i.	Define the terms wave length and frequency.	
	ii.	Enlist steps for tissue processing in electron microscope.	
	iii.	Write applications of fluorescence microscope.	
	iv.	State Beer- Lambert's law.	
	V.	Write applications of colorimeters.	
	vi.	Differentiate between rate zonal and isopycnic centrifugation.	
	vii.	Define the term partition coefficient.	
	viii.	Enlist different types of column matrix materials.	
	ix.	Write principle of chromatography.	
÷	x.	What is contour - clamped homogeneous electric field?	
	xi.	Enlist applications of PFGE.	
	xii.	"Electrophoresis is a half electrolytic process" Justify.	
Q.3	(a)	Explain principle and working of phase contrast microscope.	[6]
	(b)	Briefly explain magnification and resolution in terms of microscope.	[4]
		OR	
Q.3	(a)	Write short notes on: (i) Polarization (ii) Refraction	[6]
	(b)	Differentiate between SEM and TEM.	[4]
Q.4	(a)	Write a note on electromagnetic radiations.	[4]
	(b)	Discuss types of rotors.	[6]
		OR	
Q.4	(a)	Write a note on Density gradient centrifugation.	[4]
	(b)	Differentiate between visible and UV spectrophotometer.	[6]
Q.5	(a)	Write a note on Thin layer chromatography.	[6]
	(b)	Explain applications of gas liquid chromatography.	[4]
		OR	
Q.5	(a)	Differentiate between anionic and cationic exchangers.	[4]
	(b)	Describe affinity chromatography.	[6]
Q.6		Explain SDS poly acrylamide gel electrophoresis in detail.	[10]
		OR	
Q.6		Explain agarose electrophoresis and isoelectric focusing.	[10]

\*\*\*\*\*\*

Page 2 of 2



(49)

#### SARDAR PATEL UNIVERSITY

#### B.Sc. (Genetics) - Fifth Semester Examination (CBCS)

Thursday, 09th November 2017 10:00 a.m. to 1:00 p.m.

#### **US05CGEN02:** Molecular and Microbial Genetics

Total Marks: 70 Note: (1) Figures to the right indicate marks. (2) Draw a neat and labeled diagram, wherever necessary. Choose the most appropriate answer from the four alternatives given: Q. 1 [10] The first successful transformation of rDNA molecule into a bacterium was carried out by..... (a) Nathan, Arber and Smith (b) Boyer and Cohen (c) Watson, Crick and Wilkins (d) None of them The most popular and widely used engineered plasmid vector is .................. (a) pBR 322 (b) pUC vector (c) pSC 101 (d) pUC 19 iii. Following is not an example of plasmid (b) R factor (a) F factor (c) Col factor (d) None of them iv. Retrotransposons differ from DNA transposons in-(a) The ability to transpose (b) Having no RNA intermediate (c) Having an RNA intermediate (d)Inability to show transposition A recent Drosophila transposable element is represented by: (a) LINES & SINES (b) Transposons (c) P elements (d) Crossover suppressors vi. What is the source of the repair template during excision repair? (a) There is none, since repair is by direct reversal (b) There is none, since repair is bypassed. (c) The strand complementary to the damaged DNA. (d) The homologue of the damaged DNA. In eukaryotes, the vast majority of DNA synthesis occurs during ...... of the cell cycle (a) G1 phase (b) G2 phase (c) S phase (d) None of them viii. Reassociation kinetics of the genome depends on...... (a) Genome Size (b) Types of nucleotide in genome (c) Types of sequence in genome (d) All the above ix. What is the role of eukaryotic RNA polymerase I? (a) Transcription of mRNA only (b) Transcription of mRNA, rRNA and tRNA. (c) Transcription of 'small' RNAs including tRNAs, 5S RNAs and snRNAs. (d) Transcription of the major rRNA transcript.

(d) Operator and regulator gene

(c) Operons

X. Essential components of eukaryotic cistron are......

(b) Exons

(a) Intron

Q.2	Answer any <u>TEN</u> from the following:	[20]				
	i. What do you mean by competence factor?					
	ii. Describe function of F plasmid in conjugation.					
	iii. What is template phage?					
	iv. What do you mean by site specific recombination?					
	v. What is transposable element?					
	vi. Write a short note on Yeast "Ty" element.					
	vii. What do you mean by C <sub>0</sub> t analysis?					
•	viii. When DNA is replicated in eukaryotes?	:				
	ix. What is C-value of a genome?					
	x. What is transcription unit? Is it the same thing as a gene?					
•	xi. Describe any two post translational modifications.					
	xii. What do you mean by splicing of tRNA?					
Q.3 (a)	What do mean by conjugation? Write a detail note on F factor and u-tube experiment.	[06]				
(b)	Discuss in detail about plasmid and it's any two types.					
	OR					
Q.3 (a)	Give an account on specialized transduction in bacterial cells.					
(b)	Write the brief note on life cycle of virus.	[04]				
Q.4	Write a detail note on models of DNA recombination studied by you.	[10]				
	OR					
Q.4	Discuss in detail about of mismatch repair mechanism for DNA and site specific recombination.	[10]				
Q.5	Explain in detail about DNA reassociation kinetics (Cot curve analysis).  OR					
Q.5	Write a detail note on organization of chloroplast genome.	[10]				
Q.6 (a)	Explain in detail about posttranscriptional modification of rRNA in eukaryotes.	[05]				
(b)	Discuss the initiation factors with its function for Eukaryotic translation.	[05]				
	OR					
(a)	Give an account on initiation of transcription of mRNA in eukaryotes.	[06]				
<b>(b)</b>	Write a note phosphorylation of amino acid for posttranslational modification.	[04]				
	**************					

SEAT No. No of Printed Pages: 02 SARDAR PATEL UNIVERSITY 5<sup>th</sup> Semester B.Sc Examination Saturday, 11<sup>th</sup> November 2017 10:00 a.m. to 01:00 p.m. Subject: Genetics Paper Code: US05CGEN03 (Introduction to Genetic Engineering) Note: i) Attempt all questions. **Total Marks: 70** ii) Marks are indicated on the right hand side. Q.1Answer the following Multiple Choice Questions. All are compulsory 1X10=10 The DNA Polymerase enzyme is also known as: 1. A) DNA directed DNA polymerase B) RNA directed RNA polymerase C) RNA directed DNA polymerase D) DNA directed RNA polymerase 2. Which of these chemical is used for long term storage of DNA? A) 70% Ethanol B) Iso amyl alcohol C) Tris-EDTA D) CTAB Which of the following bonds are joined by DNA Ligase: 3. A) Hydrogen bonds B) Glycosidic bonds C) Covalent bonds D) Phosphodiester bonds The size of the bacteriophage  $\lambda$  genome is: 4. A) 37 KB B) 52KB C) 49 KB D) 200KB Which is the most characteristic feature of a shuttle vector: 5. A) It can replicate in single host B) It can replicate in unique host C) It can replicate in different host D) None of the above Transfer of T-DNA from Ti plasmid into plant cell is mediated by 6. A) mob gene B) vir gene C) nif gene D) octopine gene RNA in a DNA-RNA hybrid is digested by 7. A) S1 nuclease B) RNase A C) RNase H D) RNase T1 Transfer of DNA from agarose gel to nylon membrane is: 8. A) Southern blotting B) Western blotting C) Northern blotting D) All of the above 9. Hairpin loop formation is a feature in: A) Genomic library B) cDNA Library C) colony hybridization D) Autoradiography 10. "GI" stands for: A) Gene Indexing B) Geographical Indication C) Genetic Identity D) Genomic Identity 2X10=20 Q.2Attempt Any ten of the following Mention various properties of DNA ligase. 1. Enumerate precaution to be taken during DNA isolation. 2. 3. Define STAR activity and Restriction site. What are the main features of Ti plasmid? 4. 5. Define vectors. Mention various phage based vectors. Briefly mention advantages of pUC over PBR322 6. Enumerate at least four transformation methods. 7. 8. Mention various features of Cartagena protocol? Mention applications of GMO in agriculture. 9. What is probe? Enumerate various kinds of probes used frequently. 10. What are the advantages of cDNA library. 11. P.T.O Define Genomic library and mention its applications 12.

Q.3 A	Explain Plasmid DNA isolation in detail with the rationale of chemicals used.	10
	OR	
Q.3 A	Give a comparative account of all three classes of Restriction Endonucleases.	05
В	Mention various applications of Polynucleotide Kinase and DNA polymerase.	05
Q.4 A	Write a short note on pTi as vector and its significance.	05
$\mathbf{B}$	What are the desirable features of vectors to be used in genetic engineering?	05
	OR	
Q.4 A	Explain with diagram λ replacement vector.	05
$\mathbf{B}$	Explain with diagram yeast plasmid based vectors in brief.	05
Q.5 A	Write a note on Biosafety regulation for GMO	05
В	What are the applications of gene cloning?	05
	OR	
Q.5 A	Write a note on Transformation methods.	05
В	Explain any two methods of recombinant selection.	05
Q.6 A	Give a comparative account of c- DNA library and Gene library.	05
В	Briefly explain biotin mediated probing method.	05
	OR	
Q.6	Mention various blotting techniques with relevant diagrams.	10

\*\*\*\*\*\*\*\*



## "SARDAR PATEL UNIVERSITY EXAMINATION TYBSc Fifth Semester, 13th November, 2017, Monday

Subject: Genetics; Course: US05CGEN04; Paper---Plant Biotechnology

Tim	e: 10.00am to 1.00pm	Total mark	s70
Q1	MULTIPLE CHOICE QUESTIONS. Attempt all questions each carr i. In plant tissue culture, roots and shoots induction can accomplish (A) using tissue of a certain minimum size (B) using a particular auxin-cytokinin ratio (C) using a specific concentration of sucrose in culture medium (D) manipulating physical factors such as light, pH and temperature.	ned by:	[10]
	ii. Cellular totipotency is demonstrated by  (A) only carrot cells  (B) all plants cells  (C) all eukaryotic cells  (D) only bacterial cell		
	iii. Protoplast isolation by enzymatic digestion of cell wall was given [A] Murashige [C] Cocking [B] Miller [D] Haberlandt	by	
	iv. protoplast fusions can be achieved by  (A) Ficoll (B) Polyethylene glycol (C) High voltage X-rays (D) Surface protein of Hepatitis B	s virus	
·	v. Greatest benefit of shoot tip/meristem culture is  (A) Development of somaclonal variations  (B) Development of transgenic plants  (C) Production of virus free plants  (D) All the above		
	vi. Glutamine Synthase is competitively inhibited by  (A) glufosinate (B) ammonia  (C) glyphosate (D). phosphoenolpyruvate  vii. Explant is disinfected through  (A) Autoclaving (B) UV-irradiations (C) Surface sterilization (D)	D) Dry heat	
	viii. Biolistic technique is used in  (A). Tissue culture process (C). Hybridization process (D). Germplasm conservati  ix. Which of the following cannot be used as a vector?	on process	
	(A) Phage (B) Plasmid (C) Bacterium (D) All can be used	as vectors	

- x. Function of Vir A in Ti plasmid is to-----
  - **A** activate Vir G for transcription
  - B transfer T-DNA to plant cell
  - recognizes and nicks at T-DNA border sequence  $\mathbf{C}$
  - **D** protects T DNA from nucleases of

Q2.	<ul> <li>Short questions. Attempt any TEN each carry Two marks.</li> <li>a. Define open culture and closed culture.</li> <li>b. Enlist the composition of MS medium</li> <li>c. Define cybrid and somatic hybrid.</li> <li>d. Enlist the applications of plant tissue culture technique.</li> <li>e. Explain the term selectable markers.</li> <li>f. Expand the abbreviations—IAA,CAT,GUS,PAT</li> <li>g. Explain microinjection technique.</li> <li>h. Describe the meaning and significance of edible vaccine.</li> <li>i. What do you mean by the term 'flavr savr' tomato?</li> <li>j. Differentiate between the TDNA of Ti and Ri plasmid.</li> <li>k. Enlist all the methods of gene transfer in plants.</li> <li>l. Define cytoplasmic male sterility?</li> </ul>	[20]
Q3a Q3b	Define callus. Explain suspension culture in detail. Explain what is laminar air flow?	[07] [03]
Q3a Q3b.	OR Define sterilization. Explain in detail the different ways of using heat for sterilization? Explain the protocol for artificial seed preparation.	[07]
Q30.	Explain the protocol for artificial seed preparation.	[03]
Q4.	Define androgenic haploids. Discuss in detail the methods of getting haploid through pollen culture.  OR	[10]
Q4a. Q4b.	How would you screen somatic hybrids explain any two methods. How nurse culture helps in raising haploids?	[07] [03]
Q5a. Q5b.	Explain in detail how Ti plasmid transfers its T-DNA to a plant. Discuss the use of liposome in gene transfer.	[07] [03]
Q5	OR Give a detailed account on markers used in plant transformation.	[10]
Q6a. Q6b.	Explain various approaches used to make herbicide resistant plants Define edible vaccine and give its significance.  OR	[07] [03]
Q6a. Q6b.	Explain the importance and making of 'Golden Rice'. Write a short note on barnase/barstar system.	[07] [03]

No. of Printed Pages: 2

#### SARDAR PATEL UNIVERSITY

B.Sc. (Genetics) – Fifth Semester Examination (CBCS)
Wednesday, 15<sup>th</sup> November 2017
10:00 a.m. to 01:00 p.m. **US05CGEN05: Immunogenetics** 

	Choose the most ap	propriate answer from t	he four alternatives giv	en:	
i.	The most abundant is (a) IgG	mmunoglobulin in human l (b) IgA	body is (c) IgM	(d) IgE	
ii.	_	ods of diagnosis utilize la immunosorbent Assay	beled antibodies excep	t?	
	(b) Agglutination in	hibition test			
	(c) Radio immunoa	ssay			
	(d) Immunofluoresc	ence			
iii.	Lysis of foreign cell is (a) Ig D and Ig E	is mediated through (b) Ig M and Ig G	(c) Ig M only	(d) Ig A only	
iv.	Hybridoma is biotec	hnique which involves fusio	on of		
	(a) B cell with T –cell		(b) T-cell with spleen cell		
	(c) Spleen cell with	myeloma cell	(d) Myeloma c	ell with B-cell	
v.	The MHC antigens serve as essential elements in the regulation of				
	(a) Cell-cell interactions		(b) Tissue –ce	ll interactions	
	(c) Organ –cell inte	eractions	(d) None of the	nese	
vi.	The T cell receptor are				
	(a) Secretary	(b) Membrane bound	(c) Both (a) and (b)	(d) None of above	
vii.	MHC class I is a cell	surface molecules present	o <b>n</b>		
	(a) B cell	(b) All nucleated cell	(c) APCs	(d) T cell	
iii.	An excess of antibod	y inhibits precipitation and	agglutination reaction is	called	
	(a) Prozone effect	(b) Polyvalent Ab	(c) Both a and b	(d) None of above	
ix.	Which category of h	ypersensitivities is IgE-med	liated?		
	(a)Type I	(b) Type II	(c) Type III	(d) Type IV	

Q.2		Answer any <u>TEN</u> from the following:	20]
	•	i. Differentiate between antigen and immunogen.	
		ii. What do you mean by epitopes?	
		iii. Write characteristics of antigen-antibody reactions.	
		iv. Write a short note on BCR.	
		v. What do you mean by monoclonal antibody?	
		vi. Write a short note on clonal- selection theory.	
		vii. Write a short note on inflammation and its signs.	
		viii. Give different strategies for Immunosuppression.	
		ix. What is MHC?	
		x. Write a short note on hypersensitivity.	
		xi. Define non-ediable vaccine.	
		xii. What do you mean by immunodeficiencies?	
Q.3	(a)	Discuss in detail about molecular structure of antibodies.	[06]
	(b)	Give a detail account on radioimmunoassay techniques.	[04]
		<u>OR</u>	
Q.3	(a)	Write a detail note on Immunoglobulin structure and its classes.	[06]
	(b)	Give an account on Immunofluorescence.	[04]
Q.4		Give an overview of genetics basis of antibiotic diversity.	[10]
		<u>OR</u>	
Q.4		Give an account on rearrangement of immunoglobins genes.	[10]
		- 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	[10]
Q.5		DAPIGNI III double double double of both pro-	[IU]
		OR CARLO 1 or I	<b>[10]</b>
Q.5		Give structure and types of MHC. Write a note on structure and functions of MHC class I and Class II.	[10]
Q.6		What do you mean by autominianty. Disease in action of the control of	[10]
		autoimmune disorders.  OR	10.67
Q.6	(a)	Give a detail account on AIDS.	[06]
	(b)	Write a note on types of vaccines.	[04]
		******************	

SEAT No.

No. of Printed Pages 102

Sardar Patel University

#### Ty. B.Sc. Fifth Semester Examination-2017 **Subject-Genetics**

Course-US05CGEN06 - Human Genetics

Time: 10.00am to 1.00pm Date: 17 /11/2017

Total Marks-70

Q.1	Multiple Choice Questions (one mark each) Attempt all
1	Gene for beta chain of the hemoglobin molecule, located on chromosome
	a) 29 (b) 13 (c) 11 (d) 24
2	LINEs , SINEs, and LTR elements are example of a) Retrotransposons (b) transposable elements (c) RNA transposons (d) All of above
3	Mode of inheritance for Tay Sach's Syndrome (a)Autosomal recessive (b) Autosomal dominant
	(c) Sex linked (d) None of the above
4	Triple marker test do not have following
	a) AFP b) uE3 c) Inhibin A d) none of above
5	Data of recombinants is required for mapping
	a)Physical b) Genetic c) (c)Both a and b (d) none of above
6	Guthrie Bacterial inhibition assay is used to diagnose
	a)Thalassemia b)PKU c) Sickle cell anemia d) None of above
7	Plaques and tangles are characteristics of
	a)Parkinson's b) Alzheimer's c) Down syndrome d) All of above
8	Hemophilia is caused due to deficiency of
	a)Beta chains b) Alpha chains c) Both A and B d)Factor VIII
9	Chorionic villus sampling and amniocentesis are examples oftechnique
	a)Invasive b) non invasive c) Prenatal d) All of above
10	Human genome project was completed in
	a)2001 b) 2003 c) 2004 d) Not competed till date
Q.2	Short questions: attempt any ten 20 M
1	Give history and scope of human genetics
2	What are gene families?
3	What are microsatellite?
4	Write about restriction maps.
5	Give differences between physical map and genetic map.
	$CDT\Lambda$

6		Give importance of mapping of human genome .	
7		What is Marfan syndrome?	
8		Give an account of galactosemia .	
9		Write brief about PKU.	
10		What are types of tests used for genetic screening?	
11		What is ferric chloride test?	
12		Give importance of prenatal screening.	
Q.3		Write an elaborative note on HGP	10M
٧,٥	1	OR	
Q.3		Write an elaborative note on organization and characteristics of human genome.	10M
Q.4	a.	Write an elaborative note on genetic mapping of human genome.	6M
~C· ·	b.	Write a note on BAC libraries	4M
		OR	
Q.4	a.	Give an account of physical mapping	6M
-	b.	Write a note on chromosomal walking	4M
Q.5	a.	Write a note on diabetes mellitus	5M
Q.5	b.	Write a note on Alzheimer's disease	5M
	ν.	OR	
Q.5	a.	Write a note on Parkinson's disease	5 M
<b>4</b>	b.	Write a note on obesity	5 M
Q.6	a.	Give commonly used blood and urine test used for screening of genetic disorders	5M .
	b.	Write short note on QUAD test	5M
		OR	
Q.6	a.	Write the context of genetic testing.	5M
	b.	Write short note on SAGE	5M
		X	

5.4

					-
	١.	1	0_	Λ	18)
(	u	V	1.6	H-	TD.
-		-	~		-/

SEAT No.\_\_\_\_

### No. of Printed Pages: 3

#### Sardar Patel University B. Sc. (Semester – V) Examination

Date: 07-11-2017, Tuesday

Time: 10:00pm to 01:00pm

**Industrial Chemistry** 

COURSE NO: US05CICH01 (Organic Chemistry - II)

Notes: Figures to the right indicate full marks.

Total marks: 70

Q.1 Answer the following Multiple Choice Questions. (All are compulsory)

(10)

. /	iswer the following Martiple enouge quest	(10)
1.	Pyridine undergoes nucleophilic substitut	tion with NaNH, at 100°C to form
	A. 2-Aminopyridine	
	B. 3-Aminopyridine	D. None of these
2.	Which of the following reagents will react	
	A. HCOOH	C. H <sub>2</sub> O <sub>2</sub>
	B. CHCl₃/KOH	D. (CH <sub>3</sub> C0) <sub>2</sub> 0/SnC1 <sub>4</sub>
3.	Furan reacts with ammonia in the presen	
	A. Pyridine	C. Pyrrole
	B. Furfural	D. Furoic acid
4.	Naphthalene undergoes nitration with	
	A. 1-Nitronaphthalene	C. 2-Nitronaphthalene
	B. 1,2-Dinitronaphthalene	D. 1,5-Dinitronaphthalene
5.	All carbon atoms in Anthracene are	= U.St.a - HERODYBYCHEC MARKOSENNAL AS
	A. sp hybridized	C. sp <sup>3</sup> hybridized
	B. sp <sup>2</sup> hybridized	D. None of these
6.	Sodium borohydride is an important	reagent.
	A. Bromination	C. Reducing
	B. Oxidizing	D. Methylating
7.	Aldehyde having undergo Aldol con-	densation
	A. ά- hydrogen	C. γ – hydrogen
	B. β – hydrogen	D. δ – hydrogen
8.	compounds doesn't undergo Aldol	condensation reaction.
	A. CH₃CHO	C. CH₃CH2CHO
	B. HCHO	D. CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CHO
9.	Signal pattern of the CH3 protons in the	NMR spectra of the CH <sub>3</sub> CH <sub>2</sub> Br <sub>2</sub> and CH <sub>3</sub> COOH
	is	
	A. Triplet & Singlet	C. Triplet & Doublet
	B. Doublet & Singlet	D. None of them
10	. Number of signals (ignoring the splitting the $\dot{p}$ -xylene and 2-Propanol is	patterns) would you see in the NMR spectra of
	A. Two & Three	C. Three & Four
	B. Three & Two	D. None of them

		ă.			. 4	•	
			- 7	Control gite			
Q.2 /	Answer the following	short questions (#	iny Ten)		•	•	(20)
_ 1	. Give synthesis of l	Pyrrole.	÷ .				, ,
_ 2	. Outline the rules for	or naming mono &	di heterocycl	ic compound	•		
3	<ul> <li>Write order of rela</li> </ul>	tive basicity of RC	CH2NH2 and R	LC≡N.			
4			tra Parasia. Na manasian	TEN BURGA			
	. Give resonating str		cene.			•	
	. Write a reaction of	of Naphthalene w	ith conc. sul	phuric acid a	t 165°C?	``,	
7	•					-	
8	. Write preparation a	and properties of C	)smium tetrao	xide.			
9	. Define a term rearr	rangement.					
1	0. Write information	obtained from H <sup>1</sup> 1	IMR-spectros	copy.			
1	1. The NMR spectrum	n of compound $C_2$	H <sub>6</sub> O, shows o	ne signal only	y, a singlet. I	Deduce the	
	structure of it.	3.19		,		* T	
1	2. Giving a formula formula	or calculation of D	BE, calculate	d a value for	$MF C_7H_4N_2$		
	:	Net of			111		
		MARINE.				•	
	Vrite notes on follow				1.	4.4	(10)
	Properties of Pyrid					4 A.	
В	. Electrophilic substi	itution in Thiopher	ne.		:	4	
			OR	and the second of the second		. •	
	iscuss the following.	A T		• •	ili. Na na aku 11		(10
	<ul> <li>Structure of Pyrrole</li> </ul>			and the second			
B	. Nucleophilic substi	tution in Pyridine.		ing a filological section of the sec		1	
	Vrite notes on followi						(10)
Α	. Electrophilic substi	tution reaction in	Napthalene.		a grant that		,,
	Synthesis of Phena		***	*: * *	Carlos Albania	521.4	
	3 - 3 - 3 81	s s fe	OR		4" " ·		
Q.4 D	iscuss the structure o	of Anthracene with	iustification	of their react	ions also wri	te synthes	is of
it	•	11.0	ara Pa	144			.5 0.
					编码 海流区		(10)
Q.5 W	Vrite preparation, pro	perties and uses o	of following re	agents	. Angli and an		(10)
	Aluminum isopropo			agerits.	and the second		(10)
	SeO₂.						•
	0002.		OR	•			
O 5 W	/rite notes on Pinaco	l_Pinacolone and I			-4		/401
Q, -> V,	THE HOLES ON A MIREO	i i iliacololic allu i	Selizinc Aciu N	earrangemer	π.		(10)
0.6 E	rom the following set	s of NIMP IP and	d IIV data ais	~ = ==================================			
follow	rom the following set	S OLIVING, IN ALL	a ov data, givi	e a structure	consistent w	ith each of	
	-			1.4%			(10)
	Molecular weight:	oogm/moi; %age:	L=85./%, H=1			•	õ
	(doublet 30.0sq) δ			n to the	alport Styles	*:1	
2.	Molecular weight:	130gm/mol; %age	: C=73.84%, H	=13.84% <sup>-</sup> and	O=12.34%;	UV: λmax:	
	200nm; NMR: δ 1.1	(singlet for all pro	otons).			•	
			OR	applications			

Characteristic Infrared Absorption Frequencies.

	Characteristic Infrared Absorption Frequences:  Bond Compound type Frequency range cm <sup>-1</sup>					
Compound	2850-2960, 1350-1470.					
Alkanes.	3020-3080 (m), 675-1000.					
Alkenes.	3000-3100 (m), 675-870.					
Aromatic rings.	3300-3100 (11// 010 01					
Alkynes.	•••					
Alkenes.	1640-1680 (v)					
Alkynes.	2100-2260 (1)					
Aromatic rings.	1500, 1600 (v)					
Alcohols, Ethers, Carboxyllc acids, Esters.	1080-1300					
Aldehyde, Ketones, Carboxylic acids, Esters.	1690-1760					
Monameric alcohols, Phenols	3610-3640 (v)					
Hydrogen bonded alcohols, Phenols.	3200-3600 (broad)					
	2500-3000 (broad)					
1	3300-3500 (m)					
1	1180-1360.					
	2210-2260 (v)					
	1515-1560, 1345-1385					
	Alkenes. Aromatic rings. Alkynes.					

Double	Bonds
Structure unit	Frequency cm <sup>-1</sup>
C=C	1620-1680
C=O Aldehydes and ketones Carboxylic acids Acid anhydrides Acyl halides Esters Amides	1710-1750 1700-1725 1800-1850 & 1740-1790 1770-1815 1730-1750 1680-1700
Substituted derivatives of Benzene Mono substituted	730-770 & 690-710
Ortho-disubstituted Meta-disubstituted Para-disubstituted	735-770 750-810 & 680-730 790-840

Characteristic Proton Chemical Shift

Characteristic Proton Chemical Strict					
Type of Proton	Chemical shift δ, ppm	Type of Proton	Chemical shift $\delta$ , ppm $3.4-4$		
Cyclopropane Primary R-CH <sub>3</sub> Secondary R <sub>2</sub> CH <sub>2</sub> Tertiary R <sub>3</sub> CH Vinylic C=C-H Acetylenic Ar-H Benzylic Ar-C-H Allylic C=C-C-Fluorides H-C-G Bromides H-C-G Iodides H-C-I	6 - 8.5 2.2 - 3 H 1.7 4 - 4.5 I 3 - 4 2.5 - 4	Alcohols H-C-OH Ethers H-C-OR Esters RCOO-C-H Esters H-C-COOR Acids H-C-COOH Carbonyl compounds H-C-C=O Aldehydic RCH=O Hydroxylic RO-H Phenolic ArO-H Enolic C=C-O-H Carboxylic RCOO-H Amino R-NH2	3.3 - 4 3.7 - 4.1 2 - 2.2 2 - 2.6		

er og state det etter fra  $\lambda_{\rm eff} = \lambda_{\rm BH} \approx 3.4$  . enterioren errorea. Prodestanten errorea. West (III)

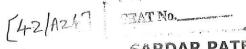
• Butter Very IV seed Historia (B. 1948) - Berlin Berlin (B. 1948)

(50	SEAT No.	No.	of Printed Pages	ş; <u>9</u>
	Sardar Pat	el Universi	its	
Doto		er – V) Examination		
Date.	09-11-2017, Thursday	al Chemistry	Time: 10:00am	to vi:vupm
		•	Manufaatuuni	
Notes	US05CICH02 (Unit Proc Figures to the right indicate full marks.	ess in Organic	40.00	م المصادة الم
140162	. Figures to the right indicate full marks.		101 	al marks: 70
			garanti kantan da wates	este pe her i
Q.1 Ar	nswer the following Multiple Choice Ques	stions. (All are com	pulsory)	ar and 14 (10
			' Yan ing dalam kalim r>Na kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim kalim	e and
1.	A vortex type of agitation is done in		oth of them	
	A. Biazzi nitration	C. Bo	oth of them	
	B. Schimidnitration	D. No	one of them	
2.	Amines can be produced by reduction of		in a la Maria de la companya de la contraction d	e a martina d
	A. Nitro group	C. Az	zo group	1 ''
	B. Nitroso group	D. Al	ll of them	
3.	Which of the following is reducing agent	?		
	A. Fe + Acid	C. Ca	aro's Acid	ing second second second
	B. Alkaline H2O2	D. KI	MNO <sub>4</sub>	and San San San
4.	The usual form of oxidation with dichron	nate is in the presen	ice of	
	A. Sulfuric acid		ydrochoric acid	
	B. Acetic acid	D. Al	ll of these	, fet .
5.	Manufacturing of acetic acid is an exami	ole of		*
	A. Liquid phase oxidation with O2			7
	B. Gas phase oxidation with H2O2	a te tag	and Albandynas of A	and the second of the second
	C. Both of these	7	AND AND AND AND AND AND AND AND AND AND	
	D. None of them.			
6.	In BHC (Benzene Hexachloride), Which	h isomer is Insectic	ally active ?	
	A. Alpha-Isomer	C. G	amma-Isomer	
	B. Delta-isomer	D. Be	eta-Isomer	
7.	In chlorobenzene manufacturing process	, type of rea	ector is used.	:
	A. Lead or Glass lined			
	B. Polymer lined			
	C. S.S lined	en en en en en en en en en en en en en e		
	D. All of these			
8.	Aromatic diazo compounds react with hy	drogen fluoride and	d boron trifloride to f	orm
	A. Aromatic fluorides	_	en en en en en en en en en en en en en e	
	B. Aliphatic fluorides		e juga di sekara di kacamatan di kacamatan di kacamatan di kacamatan di kacamatan di kacamatan di kacamatan di Kacamatan di kacamatan di kacama	
	C. Heterocyclic fluorides			
	D. None of these			
. 9.	The temperature of esterifying column is	n Ethyl acetate mani	ufacturing is	
	A. 80 °C	C. 18		
	B. 8 °C	D. 28		
10	KCN + H2O HCN + KOI			
	A. Hydrolysis	C. E	sterification	
	B. Hydrogenation		one of these.	
	• 0		<del></del> -	

了-微

Q.2 Answer the following short questions. (ANY TEN)		(20)
1. Define term partial reduction.		
2. List different factors affecting the Bechamp Reduction.		
3. Give, suitable example for "Ammonolysis".	gradina in Name and American	
4. Define the terms Sulfonation.		14
5. State the principal Sulfonating agents.		
6. Enlist the various peroxide used as an oxidizing agents.		
7. Define term Hydrogenation.	Maria Maria Maria	
8. Enlist the catalyst used for Hydrogenation reaction.		
9. Write reactions for Chlorination by substitution reactions.	and the second transfer of	
10. Define term Esterification giving suitable examples.	aafsus teel op eliks Alexandro teks	
11. Define term "Trans esterification".		
12. Write a reaction for hydrolysis of ester.		
to injurorysis of ester,	en en en en en en en en en en en en en e	. :
etteri, ilien etter	garas Hostolikas Mad	. I
Q.3 Write a notes on following:	$\mu^{\mu}$ , $\gamma = \gamma^{\pm}$	
1. Manufacturing of Nitrobenzane by continuous	e in the space for	
2. Commercial manufacture of Aniline by Bechamp reduction.		
$\cap$ D	ing despite the contract of th	
Q.3 Enlist various methods of REDUCTION giving suitable examples.	in and the second second second second second second second second second second second second second second se	(10)
en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	androne de la companya di serie de la Companya di serie di serie di serie de la companya di serie de la companya di serie de la companya di serie de	
Q.4 Giving suitable examples, discuss different type of Oxidation reaction OR	<b>is.</b> — — — — — — — — — — — — — — — — — — —	(10)
Q.4 Discuss the commercial manufacturing process of Benzene sulfonic a		
•	and the second s	(10)
Q.5 Write a notes on commercial manufacturing process of Chlorobenzen	<b>e.</b> 1	/10\
OR		
Q.5 Giving definition of hydrogenation reaction explain Hydrogenation of	f vegetable oil.	(10)
	e e e e e e e	
Q.6 Write a notes on manufacturing procedure of Ethyl acetate.	pant du	
	geograph is	(10)
Q.6 Discuss the following:		/1A\
1. Mechanism of Hydrolysis reaction.	matery (III) — en vine (III) A service de la company (III)	
2 Feterification of Conhamment 1.1.1.1.1	Paragora (Huasari A) Historia di Harabarak	14
	e de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la  · (	
superior graduate stagement in the second first of the		
	- th	
是 <sup>1</sup> 种类。	1.4	4
	The second secon	

The standard of the standard o



No. of printed pages:02

### SARDAR PATEL UNIVERSITY

#### B.Sc. - Semester-V (Industrial Chemistry) (CBCS) Petroleum Technology US05CICH03

Date: 11-11-2017, Saturday Total Marks: 70 Time: 10.00am to 1.00 pm 10 Multiple Choice Questions. Answer the questions in your answer sheet. Demusification process is used in crude oil for removal of 1 d. sulphur c.salt b. lead a.water Which one of the following is the cracking catalyst? 2 c. alumino silicate catalyst d. nickel b. calcium Which of the following treatment used for removal of sulphur from fuels? 3 d. Alkali washing c. Hydrofining b. Dewaxing a. Sulphuric acid Saybolt Viscometer is used for the determination of d. Moisture content c. carbon content a. kinematic viscosity b. Aniline point Aniline Point is used for the determination of \_ 5 d. kinematic viscosity c. aromatic portion a. carbon content b. moisture content Chloroform is manufactured from chlorination of \_  $c.C_2H_6$ d.C<sub>3</sub>H<sub>8</sub> a.CH<sub>4</sub>  $b.C_2H_2$ Hydrogenation of benzoic acid used in plaldium gives the product d. Maleic anhydride c. Pthlalic anhydride a. Caprolactum b. Salicylic acid is used in manufacture for unsaturated polyster manufacture. 8 d. H<sub>2</sub>O<sub>2</sub> c. CS<sub>2</sub> a. Maleic anhydride b. HCN The raw materials used in manufacture of Vinyl acetate are b. Acetic acid and ethane a. Acetic acid and acetylene d. Acetic anhydride and acetylene c. Acetic anhydride ethanol Oxidation product of cumin is 10 d. maleic acid c. phenol & acetone b. phthalic acid a. Benzoic acid 20 Answer the following in short.(ANY TEN) Q-2 Write the detail of inorganic theory of petroleum formation. Explain importance of petroleum refining before separation. 2 Explain the conversion of paraffin during thermal cracking. 3 What are the different chemicals derived from C2 fraction? 4 Discuss on Flash and Fire point test methods with diagram for testing petroleum products. 5 Draw diagram for manufacturing carbon disulphide. (CS2). 6 Sketch the flow sheet diagram for manufacture of iso propyl benzene. 7 Write the reaction conditions for manufacture of maleic anhydride. 8 Give the important uses of Hydrogen Cyanide. 9 Discuss applications of Vinyl acetate. 10 Discuss the area where Acetaldehyde can be used. 11 Discuss the area where Ethyl alcohol can be used.

12

Q-3			
(a)	Write a note on Girbitol Process.		
(b)	Explain demulsification and desalting of crude oil.	To graduate the second	05
	OR	1000 · 1	05
Q-3		A SHELFE MICHAEL AND A SHELF	15.1.1.1.1
(a)	Write a note on Gasoline and its additives.	•	and the second
(b)	Construction & working of bubble cap tray.	A second	05
1		The state of white manager	05
Q-4		A Committee of the Comm	
(a)	Heing flow diagram		
(b)	Using flow diagram explain separation of C <sub>8</sub> aromatic Explain the hypersorber methods of ethylene separation	fraction using fractional crystallization	1. 05
(~)	Explain the hypersorber methods of ethylene separation	n from cracker gas.	05
Q-4	the second secon		05
(a)	Using azeotropic distillation, explain assertion is		
(b)	Using azeotropic distillation, explain separation of tolu Write note on separation of butane by extractive distill	ene by using methanol.	05
	Write note on sepration of butane by extractive distillat fraction using acctone.	ion from butane from butane-butene	05
_		The state of the s	.)
Q-5			١.
(a)	Give manufacture of caprolactum with necessary diagra	ams.	
Q-5	gar varia		10
Q-5 (a)		A CONTRACTOR OF THE SECOND SEC	
(a)	Discuss manufacture of isopropyl benzene.	i di	10
Q-6		en en en en en en en en en en en en en e	10
(a)	***	Martin Day v. C	
(b)	Discuss manufacture of ethylene oxide from ethylene. Discuss manufacture of Ethyl alcohol.	e e 114 e gastralador (1 <sub>940</sub> e co.). Se e como de como do como do como do como do como do como do como do como do como do como do como do como do c	05
		有 10 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 grant 1 g 1 grant 1	05
Q-6	OR		
a)	Explain on oxo synthesis- hydroformylation of ethylene		
b)	Discuss the manufacture of phenol through cumene.		05
	i and again dufficile.		05
	All The Best		
	**** **** ***** ****		
		The second section with the second section of the section of the second section of the section o	
		The same of the second of	

starting of the second

Service Agent Control of the Control



(41 & A-12)

Date: 13/11/2017, Monday

Time: 10:00 a.m. to 01: 00 p.m.

(OWT VV/) made No. Off. that were a . . . 2

Subject code: US05CICH04

Total Marks: 70

# SARDAR PATEL UNIVERSITY B.Sc. Industrial Chemistry (5<sup>th</sup>Semester) Examination BUSINESS ORGANISATION

0.1	What are the assumptions behind directing strategy?	
Q.1	Select the correct option	10
I.	Formation, of which of the following form of organisation is easy?	
	a. Proprietorship b. JSC c. Cooperative d. Partnership	
II. <sub>20</sub>	In which form of ownership, there is a restriction on transfer of a partner's interest?	nis.
	a. Partnership b. JSC c. Government company d. None	
Ш.	The management of cooperative organisation is:	
	a. Oligarchic b. Democratic c. Hierarchy d. Dictatorship	
IV.	The nature of management in joint stock companies is a second management in joint stock companies is	
	a. Democratic b. Oligarchic c. Dictatorship a bond. None work	
V.		
	a. JS company b. JH family c. Proprietorship d. Partnership	
VI.		
-1.00.4		
VII	a. 10 b. 2 c. 7 d. 5  Henry Fayol suggested as Principle of management.	
7.0	a. Scalar Chain of Command b. Unity of Direction	
1/111	d. All of them	
VIII.	Integrating people within the organisation is refered as:	
	a. Planning b. Directing c. Decision Making d. Organising	
IX.	Which of the following is a tangible property? The psychottal acquire and publical	
	a. cost b. Quality of the c. Leadership and d. Public relations	
Χ.	Panel interview is one of the following types of interview:	
	a. Planned by the b. Informal control c. Formal body and d. Group	

(P.T.O.)

1

Q.2	Answer the following in short (ANY TWO)	20
l.	Define the term partner by estoppels?	
11.	Define cooperative society.	•
Ш,	What is the ownership? Name the different forms of ownership.	
IV.	List the re- organisations required in expansion of partnership to JSC.	
$V^{*}$	Name different clauses in the memorandum of association.	Variable Control
VI.	Explain unity of command as discussed in Fayol's principles of management.	1.
VII	List the steps of management by objective.	
VIII	State the principles of planning.	
IX.	What are the assumptions behind directing strategy?	
Χ.	Differentiate staffing from recruitment.	
XI.	Give the advantages and disadvantages of manpower from internal sources.	
XII.	What are the intangible properties?	
Q.3a		05
b	Discuss the different type of cooperative societies. 28 (20) 200 (20) (20) (20)	05
	OR on the same of	
Q.3a	Discuss the characteristics of an ideal organization.	05
Ъ.	Write the advantage and disadvantage of cooperative organization.	05
Q.4a	How a Joint Hindu Family does differ from Partnership organisation? Explain	05
b.		05
	aligners of the state of the OR and the State of the Stat	
Q.4a	Discuss the memorandum of association and Article of association.	05
b.	Write a note on the advantage and disadvantage of Joint Stock Company?	05
Q.5a	Discuss the principles of directing.	05
b.		05
	OR The state of th	
Q.5a		05
b.	Explain briefly the steps involved in decision making,	05
Q.6	Discuss the scientific selection process of manpower.	10
	OR the section of the	
Q.6	Discuss various method of establishing control standards also explain the	10
	control process.	



,

ges O

### [36/A-14]

#### SARDAR PATEL UNIVERSITY

T. Y. B.Sc. Industrial Chemistry

(Semester - 5th) EXAMINATION

15th November 2017

Course No.: US05C1CH05

(Mechanical Operations.)

Total	Marks: 70 Time: 10.00 am to 1:	uupm
Q.1	Answer the given multiple choice questions.	[10]
1.	Filter media nylon cloth is used when slurry is	
	a) acidic c) neutral	
	b) alkaline d) all of the above	
2.	The slurry is forced against basket sides due to centrifugal force in	
	a)Plate filter b) Suspended batch centrifugal c) Leaf filter d) All of these	
3.	Filters which are operated with less than atmospheric pressure is known as	
	a) Vaccum filter b) atmospheric filter c) absolute filter d) none of these	
4.	In a cyclone separator separation is based on	
	a)centrifugal force b) gravitational force c) viscous force d) Inertia force	
5.	The separation of solids from suspension in liquid by gravity settling is called	
	a) Filtration b)Sedimentation c) Mixing d)Evaporation	
6.	For the transportation of ultrafine particle, the equipment used is	
	a) Pneumatic Conveyor c) Apron Conveyor	
	b) Screw conveyor d) Belt Conveyor.	
7.	Highly viscous liquids and pastes are agitated by	
	a)Propeller b) Turbine agitator c) Multiple blade paddles d) None of these	×
8.	Ultrafine grinders are commonly reduce the size of solids by	
	a)compression b) Impact c) Attrition d) All of these	
9.	The crusher in which movable jaw is pivoted at the top is	
	a) Blake aw crusher b) Roll crusher c) Dodge jaw crusher d) Gyratory crusher.	
10.	To get a fine talc powder from its granules, the equipment used is	
	a)Roll crusher b) Ball mill c) Jaw crusher d) Gyratory crusher	
Q.2	Attempt any ten.	[20]
i.	Write the difference between cake filter and clarifying filters.	
ii.	List any two desired characteristics of filter media.	
iii.	Discuss classification of filters.	
iv.	Explain: Terminal settling velocity and Hindered settling.	
٧.	Explain: Diamagnetic and Paramagnetic materials.	
vi.	Name any two devices work on magnetic separation technique?	
vii.	Discuss about different types of impeller.	
viii.	Explain the function of idlers and tripper with belt conveyer.	
ix.	List different purpose of mixing.	
х.	Write (1) Rittinger's law (2) Kick's law	
xi.	What is the difference between Ball mill and Tube mill.	
vii	Discuss classification of size reduction equipments	740

Q.3 a) b)	Write a note on: Plate and Frame Filter press.  Discuss about Rotary filter.	[5] [5]
Q.3 a) b)	OR With the help of diagram explain working of Leaf filter. Discuss Suspended Batch centrifuge.	[5] [5]
Q.4 a) b)	Write a note on: Magnetic drum separator Discuss about Hydraulic Jig.  OR	[5] [5]
Q.4 a) b)	Discuss working of Gravity Settling Tank. Explain working of Cyclone Separator.	[5] [5]
Q.5 a)	Write about different equipments used for mixing viscous masses.	[10]
Q.5 a) b)	OR  Explain swirling and vortex .Also discuss about different preventive measures.  Write a note on: Belt conveyor.	[5] [5]
Q.6 a) b)	Write a note on: Jaw Crusher. List any five differences between open circuit grinding and closed circuit grinding.  OR	[5] [5]
Q.6 a) b)	Derive an equation for angle of nip while selection of crushing rolls  Explain working of Gyratory crusher with the help of diagram.	[5] [5]
	and the second of the second o	

#### 29 \$ A-12

#### SARDAR PATEL UNIVERSITY

T. Y. B.Sc. Industrial Chemistry

(Semester – 5th) EXAMINATION

17th November 2017

Course No.: US05CICH06

(Fluid Mechanics and Heat Transfer)

Tota	al Marks: 70	Time: 10.00 am to 1:0	00pm
Q.1	Answer the given multiple choice questions.	•	[10]
1.	Fluid which does not offer resistance to flow is known as	Commence of the Hills	(20)
	a) Ideal fluid. c) Hydroliquid		:
	b) Real fluid d)Non-ideal fluid.		
<i>2.</i> .	If the density of fluid is affected appreciably with the change in	temperature and	
	pressure, the fluid is		y).
	a)compressible b) ideal c) incompressible d) real		
3.	A fluid is a substance which is		
	a)Capable to flow c)No definite change		+ _ V _ L
	b)Undergoes deformation d) All of these		
4.	Volute converts the energy of the liquid imparted by the	e impeller to pressure en	iergy.
	a)Mechanical b) Kinetic c) Potential d)Translation	nl , we have substituted by	G/
5.	is used to change the direction of flowing fluid.	$\label{eq:constraints} \varphi_{i} = \Phi_{i} = -\frac{1}{4\pi} \frac{1}{\sqrt{2}} \left( \partial \theta_{i} - \partial \theta_{i} - \partial \theta_{i} \right) = 0$	
	a)Plug b)Reducer c) elbow d) none of these.	And the state of t	
6.	When a wall is formed from series of layers of different materials it is	s called	
_	a) Hot plate b) Composite wall c) conductive mantel d) insulate	or mantel	•
<i>7</i> .	The thermal conductivities of solids may with increase in t	emperature	
	a) increases c) decreases		
8.			46多篇
0.	The ability of a material to conduct heat is a)volatility b) thermal conductivity c) viscosity d) none of the		\$
9.	The well suited heat exchanger for in paraffin wax is	nese	
<i>7</i> .	a)Scraped surface b)Graphite c) Finned tube d) None of t	hoso	
10.	The centre to center distance between two tubes is known as	nese.	
	a) Pitch c) Clearance		
	b) Economy d) Efficiency		
Q.2	Attempt any ten.		[20]
i.	Define: 1) Laminar flow 2)Turbulent flow		
ii.	Write briefly on Inclined Manometer.		
iii.	Explain Hydrodynamics and Aerodynamics.		
iv.	Discuss about different types of pipe fittings.		
ν.	Explain: Capacity and Overall efficiency of pump.		
vi.	Explain Priming and NPSH.		
vii.	What do you mean by fouling factor?		
viii.	Explain parallel flow heat exchanger and counter flow heat exc	hanger ?	
ix.	Explain Natural convection and Forced Convection.		
х.	Write any two difference between single and multipass shell and tube	e heat exchanger.	
xi.	Explain when plate type heat exchanger is used.	<b>.</b>	
rii	When kattle type reheiler is used?		

Q.3a) b)	Derive Bernoullie's theorem. Also write its limitations.  Write a note on: U-Tube Manometer.	[5] [5]
ĺ	OR	101
Q.3 a) b)	Derive an equation for loss of head due to sudden enlargement.	[5] [5]
Q.4 a) b)		[5] [5]
	OR	er teger fill
Q.4 a) b)	Write a note on: Different types of valves. Explain working of Reciprocating Pumps.	[5] [5]
Q.5 a) b)	Derive an equation for heat flow through sphere.  Derive an equation for compound resistance in series.	[5]
	OR CALL	and the second
Q.5 a) b)	- · · · · · · · · · · · · · · · · · · ·	
	Data: Specific heat of thermic fluid =0.65 Kcal/Kg $^{0}$ C. Specific heat of water =1.0 Kcal/Kg $^{0}$ C.	
Q.6 a)	Write notes on: a) Graphite Heat Exchanger. b) Finned Tube Heat Exchanger.	10] 
Q.6 a) b)	OR  Discuss construction of Shell and Tube Heat Exchanger.  Write classification of heat exchange equipments based on its function.	[5] [5]
	X - X - X - X - X - X - X - X - X - X -	Variation of the Control of the Cont

and the second of the second o

### [47 A19]

### No. of Printed Pages: 2

# Sardar Patel University B. Sc. (Semester – V) Examination

NAMES OF TAXABLE PARTY STATES OF TAXABLE CO.	D. DC. ( DCIII
Date: 07-11-2017, Tuesday	

Time: 10:00am to 01:00pm

Industrial Chemistry Vocational

Note	US05CICV01 s: Figures to the right indicate full marks.	(Organic Chemistry)	Total marks: 70
	-	cestica) and Asamange ment	
Q.1 A	Answer the following Multiple Choice Qu	estions. (All are compulsory)	(10) Editar Automation of
1	. The decreasing order of "s" character in	the three hybrid orbitals is	S office proparation of
	A. $Sp>sp^2>sp^3$	$C = \operatorname{sn}^3 > \operatorname{sn}^2 >$	sprimde anal sa UCI . P
	B. $sp^2>sp>sp3$	D. None of the	an la sal sa sa sa sa sa sa sa sa sa sa sa sa sa
2.	atoms/group is	ectron-withdrawing inductiv	e effect) of certain
	A. $NO_2>F>COOH>C_6H_5$	C. $NO_2 > C_6 H$	I <sub>5</sub> >COOH> F
_	B. $NO_2 > COOH > F > C_6H_5$	D. None of the	nem,
3.	the following carbocation has	the least stability?	
	A. Methyl	C. Tert-butyl	-
	B. Ethyl	D. Isopropyl.	
14.	Aldehydes having ά- hydrogen on warn called reaction.	ing with mild base to give β	- hydroxy aldehyde
	A. Aldol	C. Diels – Al	der "
	B. Friedel - Craft months box notes	D. Fries rearr	angement de la la la la la la la la la la la la la
٥.	····· compounds doesn	t undergo Aldol condensatio	on.
	A. HCHO	C. CH <sub>3</sub> CH <sub>2</sub> C	
	B. CH <sub>3</sub> CHO	D. CH <sub>3</sub> CH <sub>2</sub> C	H <sub>2</sub> CHO.
6.	Lead tetra acetate is an important	roomant	
	A. Oxidizing	C. Methylatin	B. Pinacol Pinacolon.g
	B. Acetoxylatingc.	1) All of thes	P
7.	Lithium Aluminium Hydride is an impo	tont	
	A. Reducing	O D MARKET	A. Birroille Arie Reime B. Tres Reamaneance
	B. Oxidizing	D. Methylatin	B. I mes Itean sagemone, g.
8.	Selenium dioxide is an important r		
	A. Oxidizing		
	B. Reducing	<ul><li>C. Brominatir</li><li>D. Methylatin</li></ul>	ig madatawananti ki d
9.	The NMR spectrum of compound (A) C	H.O. shows one signal auto	g.
	structure of (A) is	1160, shows one signal only	a singlet. The possible
	A. Dimethylether	C. Ethanol.	A. Lead your account on
	D 14 .1 1 1 1 1	D. None of the	em, months in mainland . A
10	. How many signals (ignoring the splitting	natterns) would visit at '	em,
	2-Propanol	patterns) would you see in t	ne NIVIK spectra of the
70	A. One signal.		
60	B. Two signals	C. Three signs	
	- San San San San San San San San San San	D. Four signal	S.

Q.2 Answer the following short questions. (Any Ten)		(20)
1. Which type of reagent are produce by heterolytic cleavage	e of bonds?	
2. Define term "Nucleophiles"	**	
3. Define term free radicals, How they are generated?	ngha Khile •	
4. Write a reaction for Cross-Aldol condensation reaction.		
5. Differentiate terms Reaction and Rearrangement.		
6. Give preparation of N- Bromosuccinimide reagent.	A STATE STATE OF THE STATE OF T	
7. Enlist various uses of Selenium dioxide.		
8. Give preparation of Sodium Borohydride.	- 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
9. Define term elimination reaction.		and grade
	on for ME Call No	
10. Giving a formula for calculation of DBE; calculated a value of the latest the made of the latest the latest the made of the latest the		
11. Calculate the molecular formula for the data "MF: 107.5g	m/moi; %age: C=34	.3%, H=5.6%,
N=13.0%, Cl=3.0%."	0.1. 0.1. 0	
12. Predict the signal pattern of the CH <sub>3</sub> protons in the NMR s	spectra of the CH <sub>3</sub> C	ООН
compounds.	The second second	1498 - 1498 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 1499 - 149
Q.3 What are different types of reaction intermediates formed by a covalent bond? Explain with examples.	homolytic and heter	
A DOYARON DUNG: LADIANI WILL CAMINIES.	13.5%	<i>→</i> (10
	4 + 9 - 2	(10
OR OR	nd Addition reaction	
	nd Addition reaction	
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar	- 15. - 15.	. (10
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following:	- 15. - 15.	. (10
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein–Ponndorf–Verley Reduction.	esta.	100 je <b>j</b> 1410 je <b>j</b> 1410 jej
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein–Ponndorf–Verley Reduction. B. Pinacol–Pinacolone Rearrangement	esta.	. (10 
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR	ing and the state of the state	. (10
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein–Ponndorf–Verley Reduction. B. Pinacol–Pinacolone Rearrangement OR Q.4 Discuss the following:	esta.	. (10
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein–Ponndorf–Verley Reduction. B. Pinacol–Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement	ing and the state of the state	. (10
OR Q.3 Giving suitable examples, discuss the Elimination reaction ar Q.4 Write a notes on following: A. Meerwein–Ponndorf–Verley Reduction. B. Pinacol–Pinacolone Rearrangement OR Q.4 Discuss the following:	ing and the state of the state	. (10)
OR Q.3 Giving suitable examples, discuss the Elimination reaction and Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement.		. (10
Q.3 Giving suitable examples, discuss the Elimination reaction and Q.4 Write a notes on following:  A. Meerwein-Ponndorf-Verley Reduction.  B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following:  A. Benzilic Acid Rearrangement  B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following:		(10
Q.4 Write a notes on following:  A. Meerwein-Ponndorf-Verley Reduction.  B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following:  A. Benzilic Acid Rearrangement  B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following:  A. Aluminum isopropoxide and  B. N- Bromosuccinimide	enter de la companya	
Q.4 Write a notes on following:  A. Meerwein-Ponndorf-Verley Reduction.  B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following:  A. Benzilic Acid Rearrangement  B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following:  A. Aluminum isopropoxide and  B. N- Bromosuccinimide	enter de la companya	
OR Q.3 Giving suitable examples, discuss the Elimination reaction and Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement. Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide OR		(10
Q.4 Write a notes on following:  A. Meerwein-Ponndorf-Verley Reduction.  B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following:  A. Benzilic Acid Rearrangement  B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following:  A. Aluminum isopropoxide and  B. N- Bromosuccinimide		(10)
Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement. Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide OR Q.5 Write properties, preparations and uses of following: A. Lead tetra acetate and B. Lithium aluminum hydride		
Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide  OR  Q.5 Write properties, preparations and uses of following: A. Lead tetra acetate and B. Lithium aluminum hydride		
Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement. Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide OR Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide OR Q.5 Write properties, preparations and uses of following: A. Lead tetra acetate and	The second secon	(10
Q.4 Write a notes on following:  A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement  OR  Q.4 Discuss the following:  A. Benzilic Acid Rearrangement  B. Fries Rearrangement.  Q.5 Write properties, preparations and uses of following:  A. Aluminum isopropoxide and  B. N- Bromosuccinimide  OR  Q.5 Write properties, preparations and uses of following:  A. Lead tetra acetate and  B. Lithium aluminum hydride  Q.6 Write the principle of IR spectroscopy and discuss the applic	The second secon	(10  17 30 4 4  18 17 30 4 4  18 17 30 4 4  18 18 18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18 18  18 18 18
Q.4 Write a notes on following: A. Meerwein-Ponndorf-Verley Reduction. B. Pinacol-Pinacolone Rearrangement OR Q.4 Discuss the following: A. Benzilic Acid Rearrangement B. Fries Rearrangement. Q.5 Write properties, preparations and uses of following: A. Aluminum isopropoxide and B. N- Bromosuccinimide OR Q.5 Write properties, preparations and uses of following: A. Lead tetra acetate and B. Lithium aluminum hydride Q.6 Write the principle of IR spectroscopy and discuss the applic	The second secon	(10  11 3 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4

(B)

Aluminum oxide

#### SARDAR PATEL UNIVERSITY

#### T.Y.B.Sc. Industrial chemistry (VOC.) SEMESTER -V**EXAMINATION -2017**

#### **HEAVY AND FINE INORGANIC CHEMICALS** SUBJECT CODE: HS05CICV02

		November 2017	JUJCI	TIME: 10:00 AM TO 1:00 PM				
DAY: Thursday		day	TOTAL MARKS: 70					
Q. 1	Cho	ose the correct answer.			[10]			
(1)								
	(A)	120°C	(C)	135°C				
	(B)	140°C		130°C				
(2)	the	rield of the phosphorous based on conten						
	(A)	Silica	(C)					
	(B)	Coke	(D)	Aluminum oxide				
(3)	Arse	nic is removed from the crude phosphoric	acid I	by the action of				
	(A)	CO <sub>2</sub>	(C)	H <sub>2</sub> O				
	(B)	H₂S	(D)	Na <sub>2</sub> CO₃				
(4)	In Na	aCl manufacturing, at which concentration	ı calcii	Im sulfate completely precipitates out				
	by so	plar evaporation?		ompletely precipitates out				
	(A)	25 <sup>o</sup> Be	(C)	27 <sup>o</sup> Be				
	(B)	20 <sup>o</sup> Be	(D)	13.5°Be				
(5)	In Gl	auber's salt preparation, the salt cake is d	ed in hot water to form					
	In Glauber's salt preparation, the salt cake is dissolved in hot water to formsolution.							
	(A)	36°Be	(C)	32 <sup>0</sup> Be				
	(B)	28 <sup>o</sup> Be	(D)	None of above				
(6)	In so	dium sulfate manufacture, at which temp	e salt enriched bring leaving the well					
	is chi	In sodium sulfate manufacture, at which temperature salt enriched brine leaving the well is chilled?						
	(A)	0°F-5°F	(C)	10 <sup>0</sup> F-15 <sup>0</sup> F				
	(B)	15°F-20°F		20°F-25°F				
(7)	The c	conjugated base of oxalic acid is known as	<b>(-</b> )	201231				
	(A)	Dicarboxylic acid	(C)	 Oxalate				
	(B)	Oxalyl chloride	(D)	None of these				
(8)	Perch	loric acid forms with water.	1-,	Tronc of these				
	(A)	Perchlorate	(C)	Hydrate				
	(B)	Azeotrope	(D)	None of these				
(9)		Which of the following chemical is used as an anesthetic agent?						
	(A)	Ethanol	(C)	Carbon tetrachloride				
	(B)	Chloroform	(D)	None of above				
10)	Which	n is the most widely used packing materia	ر–، Lin ad	sorption chromatography?				
	(A)	Starch		Inulin				

Inulin

(D) Calcium carbonate

(C)

Q.2	Ans	wer the following.(attempt ten)	[20]					
(1)	Enlist the raw materials required to manufacture of triple superphosphate.							
(2)	Enlist various methods for concentrating nitric acid.							
(3)	<ul><li>Write uses of ammonia.</li><li>Define catalyst with examples.</li></ul>							
(4)								
(5)								
(6)	Give the uses of sodium sulfite.							
(7)		te properties of sodium borohydride.						
(8)	8) Write the uses of oxalic acid. 9) Explain preparation of Fehling solution No.1.							
(9)								
(10)								
(11)		at is Nujol?						
(12)		t the various methods for production of methanol and give its uses.						
Q.3	(A)	With the help of flow diagram explain the manufacture of ammonia.	[05]					
	(B)	Discuss the manufacture of triple super phosphate.	[05]					
^ -		OR						
Q.3	(A)	Write a note on manufacture of lime.	[05]					
	(B)	With the help of flow diagram explain the manufacture of super phosphate.	[05]					
Q.4	(A)	Explain the manufacture of sodium chloride in detail.	[05]					
	(B)	Write a complete note on Raney Nickel catalyst.	[05]					
		OR	[03]					
Q.4	(A)	With the help of flow diagram explain the manufacture of bromine.	[05]					
	(B)	Explain manufacturing process of sodium thiosulfate.	[05]					
Q.5		Explain the leblanc process for soda ash manufacture.	[40]					
		OR	[10]					
Q. 5		With the help of flow diagram explain the manufacture of sodium borohydride	[10]					
Q. 6	(A)	With the help of flow diagram explain the manufacture of methanol using catalytic	[05]					
		hydrogenation of carbon monoxide.						
	(B)	Write a note on potassium bromide	[05]					
_		OR	1					
Q. 6	(A)	With the help of flow diagram explain the manufacture of ethanol using indirect hydration.	[05]					
	(B)	With the help of flow diagram explain the manufacture of chloroform.	[05]					

N

# [43] SEAT No.\_\_\_\_

No. of Printed Pages 12

## SARDAR PATEL UNIVERSITY

# B.Sc. INUSTRIAL CHEMISTRY VOCATIONAL (5<sup>th</sup>Semester) Examination TECHNOLOGY OF PETROLEUM AND PETROLEUM PRODUCTS

	TECHNOLOGY OF PET SATURDAY, 11/11/2017 10:00 a.m. to 01:00 p.m Select the correct optic	•	AND PE		RODUCTS ubject code: USO5CICV03 Total Marks: 70	
1.	Demulsification process a. water b. S		ude oil fo c. Salts	or removal of	d. None of these	
11.	Which of the following to a. Sulphuric acid b. D		ed for rei c. Hydrol		ur from fuels? d. Alkali washing	
III.	mg/lit, tar pres a. 10 - 15 b. 5		r gasolin c. 1 – 4	e.	d. 16-20	
IV.	For production of HCN fi a. pt - Rh alloy b. C	rom methane Copper based	e ca l c. Silica	talyst is used. I	d. Alumina	
V.	For the removal of unreaproduction of HCN. a. H <sub>2</sub> SO <sub>4</sub> b. V			used as scrub · n carbonate	bing agent in the	
VI.	Carbon disulphide is pro a. Oxygen b. S		e reactio c. a & b b		vith at High tempe d. Carbon.	
VII.	The raw material for ma a. Acetylene / Acetic acid c. Acetylene / ethylene	d	b. Benze	tate are ne / ethylene benzene/ met	•	
VIII.	The raw material for ma a. Acetylene / Acetic acid c. Acetylene / ethylene					
IX.	are used as cat a. Molibdenum& tungest c. Alúmina			3 & AlCl3	on of isobutene	
Χ.	Is the ideal structur a. Butadiene c. Pentadiene	e for rubber	b. Buta			

Q.2	Answer the following in short (ANY TEN)	20
I.	Write the detail of modern theory of petroleum formation.	
П.	Name the various reactions taking place in catalytic cracking.	
111,	Explain signification of salts removal from crude oil.	
IV.	Write a short note on properties & uses of HCN.	
V.	Write a short note on properties & uses of CS <sub>2</sub> .	
VI.	Write a short note on properties & uses of Methanol.	
VII	Write properties and uses of acrylonitrile.	
VIII.	Give the outline the different routes to manufacture glycerine.	
IX.	Give the reaction of styrene synthesis from benzene.	
X.	Explain the limitation of molecular sieve.	
XI.	Name various chemical obtained from butanes.	
XII.	Give the outline how the various chemical obtained from propane.	
Q.3a	Discuss the theories of formation and composition of petroleum.	05
b.	Explain the working of Bubble Cap Tray in petroleum distillation.  OR	05
Q.3a	Justify significance of the demulsification & desalting of crude oil by giving	05
	the process details.	
b.	Explain the process of distillation & refining of light petroleum products.	05
Q.4a	With the help of flow diagram explain the manufacturing of Methanol.	05
b.	What are the different chemicals derived from Ethane – Ethene fraction?  OR	05
Q.4.	Discuss the raw materials used, synthesis, manufacture, properties and uses of HCN and CS <sub>2</sub> .	10
Q.5a	With the help of flow diagram explain manufacture and use of ethylene glycol from ethylene oxide.	05
b.	Describe the mfg of vinyl acetate from acetic acid and ethylene.	05
	OR	
Q.5a	Describe the process of industrial manufacture of Acrylonitrile.	05
Ь.	Describe the process of manufacturing Ethylene Oxide from Ethylene.	05
Q.6	Describe manufacturing method of Butadiene from Butane.  OR	10
Q.6a	Describe the method of production of BTX (Benzene, Toluene, Xylene).	05
b.	Write the composition, form & general application of Chrome Alumina catalyst.	05



SARDAR PATEL UNIVERSITY, VALLABH VIDYANAGAR
B.Sc. (5<sup>th</sup> Semester) External Examination
. Monday 13<sup>th</sup> November 2017
Subject: US05CICV04 (Industrial Management & Economics) **Industrial Chemistry Vocational** 

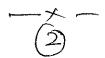
Time: 10:00a.m. to 01:00 p.m.

**Total Marks: 70** 

<b>Q.1</b>	Attempt All Questions	(Multiple Choice Que	stions)		[10]
1.	Identify the different forms of business organization is of ideal form of the organization.				
	· (A) Characteristics	(B) Features	(C) Objectives	(D) Goal	
2.	Lawful business is a	of business o	organization.		
	(A) Characteristics	(B) Features	(C) Objectives	(D) Goal	
3.	Prompt Decision Makin	g is a	of Sole Proprietorship	o.	
	(A) Advantage	(B) Disadvantage	(C) Features	(D) Objectives	
4.	Scientific management	meansle	eading to increased indus	strial efficiency.	
	(A) improve Model	(B) improve Method	(C) improve Process	(D) None	
5.	Managed development of all the to (A) Scientific	echnical and mechanica	I facilities of industry.	chanical factor and the (D) Behavior	
6.	How many methods usually advocated by the exponents of scientific management in industry?				
	(A) 3	(B) 4	(C) 5	(D) 6	
7.	Interviewing is a	of HRM.			
	(A) Objectives	(B) Function	(C) Process	(D) None	
8.	The wage packet is acce	epted as	there will be no mot	ivation.	
	(A) Fair	(B) Just	(C) Fair & Just	(D) None	
9.	How many Factors invo	olved in Project Cost Est	timation.		
	(A) 10	(B) 11	(C)12	(D) 14	
10.	Cost Reporting is a	of cos	t accounting.		
	(A) Method	(B) Element	(C) Process	(D) None	

Q.2	Short Questions (Attempt Any Ten)	[20]
1.	What are the objectives of co-operative society?	
2.	What are the advantage and disadvantage of sole proprietorship?	
3.	What are the characteristics of JHF?	
4.	Define scientific management.	
5.	What is controlling in organization?	
6.	What is staffing?	
7.	What are the functions of HRD?	
8.	Describe employee evaluation.	
9.	Describe HR Manager's responsibilities.	
10.	Define finished product with example.	
11.	What is man power? How can we implement man power in Organization?	
12.	Define complexity of project with example.	
Q.3	(A) What are characteristics of Partnership? Explain in details.	[05]
	(B) What are characteristics of Joint Stock Company? Explain in Details.	[05]
	OR	
Q.3	(A) Explain the advantages and disadvantages of Joint Stock Company?	[05]
	(B) Explain the advantages and disadvantages of Co-operative society?	[05]
Q.4	(A) Explain the functions of management with real life example.	[05]
	(B) What are the functions of management? Explain in details.	[05]
	OR	
Q.4	(A) Explain the Principles of Scientific management.	[05]
	(B) "Suppose you are manager of any organization and you have to handle one plant	
	then what is your preparation for handling it" Comment on it	[05]
Q.5	(A) Explain the Expert System and Executive Information system.	[05]
	(B) What is Fourth Language? Explain in detail.	[05]
	OR	
Q.5	(A) Explain in details about Functions of MIS	[05]
	(B) Describe about Components of Safety Services.	[05]
Q.6	Explain about Cost Components in detail.	[10]
	OR ·	
Q.6	Explain in details about Factors involved in Project Cost Estimation	[10]





<u>SC</u>

the second secon	41 <b>5</b>	•
SEAT No	Sardar Patel University T.Y. B. Sc. (Semester-V) Industrial Chemistry Vocatio US05CICV05 (Pharmaceutic	) onal
Q1. Answer the following multiple	choice questions.	(10)
1. Formulations must provide  (a) Efficacy (b) Safety (c)  (c) Fig. 19 (c) (c) (c) (d) (c) (d) (d) (d) (d) (d) (d) (d) (d) (d) (d	) Convenience (d) All o	
2. The First edition of Indian Phar (a) 1955 (b) 1996	(c) 1975	(d) 1985
route of drug adm     (a) Subcutaneous (b) Epicuta		
4. Cat gut is an example of(a) Absorbable (b) Non		(d) Adsorbable
5 is used to seal th		None of these
6. Most of the potent pyrogens at (a) Gram Negative Bacteria		
7. Micro crystalline cellulose is  (a) Bulking agent (b) Disinte		(d) Both (a) & (b)
8. Vanillin is a agen  (a) Flavoring (b) Lubricating	t.	
9. Seedless varieties of fruits are o	obtained by	of these (d) Both (a) & (b)
10. Characteristic behavior of Bha		

(b) Floats (c) Settles (d) Suspends

(a) Solubilize

Q.2 Answer the following short questions (Any ten).	(20)
1. Enlist various types of formulations.	
2. Write role of pharmacopoeia.	
3. Write advantages of parentral route of drug administration.	
4. Write uses of plaster of paris.	
5. Give examples of thermoset plastics.	
6. Define sutures and ligatures.	
7. What is the function of Lubricant in formulation? Give one example.	
8. Define Glidant. Give one example.	
9. Explain the role of disintegrant with an example.	
10. Define "Bhasma" and give it's examples.	
11. Write the importance of pH range in soil.	
12. Define the term phytopharmaceuticals.	
Q.3 A. Write a short note on oral route of drug administration.	(05)
B. Write a short note on British Pharmacopeia.	(05)
<u>OR</u>	
Q.3A. Explain the importance of dosage forms.	(05)
B. Write note on Indian pharmacopoeia.	(05)
Q.4A. Write a short note on CAT GUT.	(05)
B. Write advantages and disadvantages of plastic as packaging material.  OR	(05)
Q.4A. Write advantages and disadvantages of glass as packaging material.	(05)
B. Write note on densities of powder and their measurement.	(05)
Q.5A Write short note on suspending agents.	(05)
B. Define Diluents. Write structure, properties and uses of Lactose.	(05)
<u>OR</u>	
Q.5A. Write a short note on colouring agents.	(05)
B. Define Binder. Write structure, properties and uses of Acacia.	(05)
Q.6 Discuss different methods of classification of crude drugs.	(10)
<u>OR</u>	
Q.6 Describe different methods of cultivation and discuss the effect of soil and soil fertility.	(10)
**************************************	

2/2

TEAT No.\_\_

No. of Printed Pages; OL

## [30]

## SARDAR PATEL UNIVERSITY

# B.Sc. Industrial Chemistry (5<sup>th</sup>Semester) Examination SEPARATION TECHNIQUES

	17/11/2017 : 10:00 a.m. to 1: 00 p.m.	Subject code: US05CICV06 Total Marks: 70
Q.1	Select the correct option	10
l.	is the temperature at which a liquid mixture	e of known composition
	start to vaporize as temperature increases.	
	(a) Bubble point (b) Dew point	
	(c) None of these. (d) Saturation temp.	
11.	The driving force for separation of constituents of	liquid mixture distillation
	is(a) Solubility of commonants(b) Poletice	volatility of avatam
	(a) Solubility of components (b) Relative (c) Distribution of mixture (d) None of	· ·
Ш.	Reflux ratio is the ratio of	uiese
111.	<del></del> -	b) Reflux to distillate (L/D)
	(c) Reflux multiply by distillate (L.D)	
IV.	Gas absorption is also called as:	.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
	(a)Scrubbing (b) Drying	
	(c) Adsorption (d) None of	these
V	What is the tower diameter to packing size	e ratio for minimizing the
	channeling?	
	(a)>8 (b) <8	
	(c) $=8$ (d) None of	
VI.	Which extractor is used in the petroleum industry	
	`,	fixer settler
		Pulse column
VII	In crystallization mass transfer occurs from solution	
	( )	Solid
1/111	• • • • • • • • • • • • • • • • • • • •	None of these
VIII.		Slurry
		lone of these
IX.	For drying paper sheet dryer is employ	
17.51	(a)Drum (b) T	ray
	• • • • • • • • • • • • • • • • • • • •	Rotary
X.	In falling rate period, rate of drying	
	, ,	Decreases
	(c) Constant (d)	Random

Q:.2	Answer the following in short (ANY TEN)	20
l.	Define Reflux ratio.	
II.	Write in brief on Bubble cap tray	
Ш.	Give limitations of McCabethile method	
IV,	Differentiate between distillation and extraction.	
V.	Define the following terms with reference to extraction I. Raffinate II.Extract.	
VI.	Write the characteristics of tower packing.	,
VII	Explain the mechanism of crystallization	
VIII.	Give Classification of crystallizers.	
IX.	Define the term Magma.	
Χ.	Give comparison between Evaporation and Drying.	
XI.	Give classification of dryer.	
XII.	Explain equilibrium moisture and critical moisture content	
Q.3a	Write in brief with a sketch the Simple distillation.	05
Ь.	Write a note on Steam distillation.	05
	OR	
Q.3a	Write note on Azeotropic distillation.	05
b.	Explain the Flash distillation & draw its sketch.	05
Q.4a	Give the construction and working of continuous operated mixer-settler with neat sketch.	05
b.	Write briefly on mechanically agitated vessels used for gas absorption.  OR	05
Q.4a	Draw the neat sketch of packed column used for gas absorption and write its construction and functioning.	05
b.	Write in brief on rotating-disk contactors.	05
Q.5a	Explain the construction and working of Agitated crystallizer with suitable diagram.	05
b.	Explain the construction and working of Swenson-walker crystallizer with suitable diagram.	05
	OR	
Q.5a	Explain the construction and working of vacuum crystallizer with suitable	05
•	diagram.	
b.	Explain the working of Basket extractor with suitable diagram.	05
Q.6	Explain the construction & working of Tray dryer with neat diagram also give its advantages and disadvantages	10
	OR	
Q.6	Explain the construction & working of Spray dryer with neat diagram and also give its advantages and disadvantages.	10

## SARDAR PATEL UNIVERSITY

Vallabis Valya**naga**r

## B.Sc. (5th Sem) Examination - 2017

07<sup>th</sup> November, 2017 (Tuesday) 10:00 AM - 01:00 PM

## **US05CINS01** (Instrumentation)

8085 Microprocessor Architecture and Programming - 1

Maximum Marks: 70

Que 1	Each question below gives a multimost appropriate one.	ple choic	ce of answers. Choose the	[10]
1	: Group of Program.			
	a) Firmware	b) Hard	ware	
	c) Software	d) Mach		
2	Each Manufacturer of a Micropre	•		
	Symbolic Code for each Instruction.		• • • • • • • • • • • • • • • • • • • •	
	a) Mnemonic	<b>b)</b> Firm		
	c) Machine Language	•	nbly Language	
3	Assembly Language is in	-,	man named	
-	a) Binary	b) Octal		
	c) Hexadecimal	•	sh-Like Languages	
4	The Microprocessor (MPU) Uses _	Bus	to Identify a Peripheral or	
_	Memory Location.		+	•
	a) Data	b) Contr	rol	
	c) Address	•	ess and Data	
5	The Eight Data Lines Enable the MP			
	From			
	a) $OO_H$ to $FF_H$	<b>b)</b> 0000	0001 <sub>B</sub> to 11111111 <sub>B</sub>	
	c) 0000 <sub>H</sub> to FFFF <sub>H</sub>	<b>d)</b> 0001	H to FFFFH	
6	: Comprised of Various Single	Lines th	nat Carry Synchronization	
	Signals.			
	a) Data Bus	b) Addre	ess Bus	
	c) Control Bus	d) Data	Bus and Control Bus	
7	: Non - Maskable Interrupt.			
	a) INTR	b) RST	5.5	•
	c) RST 7.5	d) TRAF		
8	: Associated With DMA.			
	a) INTR	b) REAI		
	c) HLDA	d) RESE	ET OUT	
9	: Flag Not Affected By INR Inst			
	a) Parity	b) Zero		
	c) Sign	d) Carry	7	
10	: No Flags Affected.	<b>.</b>		
	a) IN	b) ADD		
	c) ANA	d) XRI		

Que 2	Sho	rt Questions (Attempt any TEN)	[20]
1 2 3 4 5 6 7 8 9 10 11	Define: Program and Software. What is the Use of Flag Register? Explain Briefly Use of Accumulator. Data Bus Is Bidirectional. Justify. Give Relation Between Width of Address Bus and Memory Size. Calculated Memory Size of 8085 Microprocessor. Explain: Encoder. Explain: SID and SOD. Explain: ALE and CLK (OUT). Write on DMA. Write Assembly Language Program to Load 37 <sub>H</sub> in Register B. Display the Number at Out Port 1. Write Assembly Language Program to Add 93 <sub>H</sub> (in Register C) and B7 <sub>H</sub> (in Register D). Write Assembly Language Program to Subtract 40 <sub>H</sub> (in Register H) From 8C <sub>H</sub> (in Register B).		
Que 3	[A] [B]	Write a Note on 8085 Programming Model With Necessary Diagram. Give Classification of 8085 Instructions on the Basis of Instruction Word Size.	[ <b>0</b> 5] [ <b>0</b> 5]
	[C] [D]	Write a Note on 8085 Hardware Model With Necessary Diagram. Discuss 8085 Instruction Classification on the Basis of Various 8085 Operations.	[05] [05]
Que 4	[A] [B]	Write a Brief Note on Tri - State Devices and Buffer. Explain Peripheral - Mapped and Memory - Mapped I/O.  OR	[05] [05]
	[C] [D]	Discuss Bidirectional Buffer and Decoder. Give an Account of Microprocessor - Initiated Operations and 8085 Bus Organization.	[05] [05]
Que 5	[A]	Write a Note on 8085 Microprocessor.  OR	[10]
	[B] [C]	Draw Schematic of Latching Low - Order Address Bus. Explain it.  Draw Schematic to Generate Read/Write Control Signals for Memory and I/Os. Explain it.	[ <b>05</b> ] [ <b>0</b> 5]
Que 6	[ <b>A</b> ]	Write Assembly Language Program to Mask Higher Nibble from $BC_H$ (in Register L) and $CD_H$ (in Register C). AND Lower Nibbles. Store	[05]
	[B]	Result at C500 <sub>H</sub> .  Write Assembly Language Program to Add DF <sub>H</sub> (in Register B) and E5 <sub>H</sub> (in Register H). If Sum is Greater Than FF <sub>H</sub> , Store CC <sub>H</sub> at D500 <sub>H</sub> Otherwise Store Sum at D500 <sub>H</sub> OR	[05]
	[C]	Write Assembly Language Program to Mask Lower Nibble from AB <sub>H</sub> (in Register D) and EF <sub>H</sub> (in Register H). XOR Lower Nibbles. Store	[05]
	[D]	Result at D000 <sub>H</sub> . Explain: ANI and ADD With Suitable Examples.	<b>[0</b> 5]

SEAT NO.\_

No. of Printed Pages : 2

## SARDAR PATEL UNIVERSITY

T.Y.B.Sc. Examination, FIFTH Semester
Thursday, 9<sup>TH</sup> November 2016
Time: 10.00 am To 1.00 pm
Instrumentation Course Code: USO5CINS02

0

Total Marks: 7  Q-1 Write answers to the following multiple choice questions in your answer [10] book by selecting the proper option.  (1) The basis for measuring the thermodynamic property like temperature is law of thermodynamics.  (a) zeroth (b) second (c) first (d) third  (2) The principle of working of the constant volume thermometer is based on law.  (a) Boyle (b) Newton (c) Charles (d) Gauss  (3) cannot be used for negative pressure measurement.  (a) piezometer (b) U-tube manometer (c) Pirani gauge (d) Bourdon gauge  (4) The piezometer is used for measuring pressure.  (a) agauge (b) absolute (c) vacuum (d) total  (5) A well type manometer is used in preference to a simple U-tube manometer to obtain better (a) accuracy (b) volume (c) precision (d) sensitivity  (6) 1 micro bar = dyne/cm².  (a) 1 (b) 2 (c) 3 (d) 4  (7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification?  (a) C - type (b) spiral (c) helical shaped (d) none of these  (8) The equation used in capacitive method is (a) C = K AD (b) C = D A/K (c) C = K D/A (d) C = K A/D  (9) 1 micron is equal to mm of Hg.  (a) 10² (b) 10² (c) 10³ (d) 10⁵  (10) The unit of pressure is (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³  Q-2 Answer the following questions in brief. (Answer any Ten Questions)  (1) Enlist the various temperature measurement methods.  (2) Explain about the Temperature scales in brief.  (3) Define: Thermocouple.  (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.  (5) State any four characteristics of manometric liquid.  (6) Enlist the advantages of Pirani gauge.  (7) Draw the schematic diagram of Bourdon gauge.  (8) State the limitations of thermal conductivity gauges.			Course Title: Process Measurement Technique - I	
book by selecting the proper option.  (1) The basis for measuring the thermodynamic property like temperature is law of thermodynamics.  (a) zeroth (b) second (c) first (d) third  (2) The principle of working of the constant volume thermometer is based on law.  (a) Boyle (b) Newton (c) Charles (d) Gauss  (3) cannot be used for negative pressure measurement.  (a) piezometer (b) U-tube manometer (c) Pirani gauge (d) Bourdon gauge  (4) The piezometer is used for measuring pressure.  (a) gauge (b) absolute (c) vacuum (d) total  (5) A well type manometer is used in preference to a simple U-tube manometer to obtain better  (a) accuracy (b) volume (c) precision (d) sensitivity  (6) 1 micro bar =				irks : 7
(1) The basis for measuring the thermodynamic property like temperature is law of thermodynamics.  (a) zeroth (b) second (c) first (d) third  (2) The principle of working of the constant volume thermometer is based on law.  (a) Boyle (b) Newton (c) Charles (d) Gauss  (3)	Q-1	Write	answers to the following multiple choice questions in your answer	[10]
law of thermodynamics.				
(a) zeroth (b) second (c) first (d) third  (2) The principle of working of the constant volume thermometer is based on		(1)	The basis for measuring the thermodynamic property like temperature is	
(2) The principle of working of the constant volume thermometer is based on law.  (a) Boyle (b) Newton (c) Charles (d) Gauss  (3) cannot be used for negative pressure measurement.  (a) piezometer (b) U-tube manometer (c) Pirani gauge (d) Bourdon gauge  (4) The piezometer is used for measuring pressure.  (a) gauge (b) absolute (c) vacuum (d) total  (5) A well type manometer is used in preference to a simple U-tube manometer to obtain better (a) accuracy (b) volume (c) precision (d) sensitivity  (6) 1 micro bar = dyne/cm².  (a) 1 (b) 2 (c) 3 (d) 4  (7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification?  (a) C - type (b) spiral (c) helical shaped (d) none of these  (8) The equation used in capacitive method is (a) C - k A D (b) C = D A/K (c) C = K D/A (d) C = K A/D  (9) 1 micron is equal to mm of Hg.  (a) 10² (b) 10⁴ (c) 10³ (d) 10⁵  (10) The unit of pressure is (a) 10² (b) 10⁴ (c) 10³ (d) 10⁻ (b) 10² (m) 10			The state of the s	
law. (a) Boyle (b) Newton (c) Charles (d) Gauss (3)				
(a) Boyle (b) Newton (c) Charles (d) Gauss	•	(2)		
(a)				
<ul> <li>(a) piezometer (b) U-tube manometer (c) Pirani gauge (d) Bourdon gauge</li> <li>(4) The piezometer is used for measuring pressure.</li></ul>		<b>(-)</b>		
<ul> <li>(4) The piezometer is used for measuring pressure. (a) gauge (b) absolute (c) vacuum (d) total</li> <li>(5) A well type manometer is used in preference to a simple U-tube manometer to obtain better (a) accuracy (b) volume (c) precision (d) sensitivity</li> <li>(6) 1 micro bar = dyne/cm². (a) 1 (b) 2 (c) 3 (d) 4</li> <li>(7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification? (a) C - type (b) spiral (c) helical shaped (d) none of these</li> <li>(8) The equation used in capacitive method is (a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D</li> <li>(9) 1 micron is equal to mm of Hg. (a) 10² (b) 10⁴ (c) 10³ (d) 10⁵</li> <li>(10) The unit of pressure is (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions)</li> <li>(1) Enlist the various temperature measurement methods.</li> <li>(2) Explain about the Temperature scales in brief.</li> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(3)		
(a) gauge (b) absolute (c) vacuum (d) total  (5) A well type manometer is used in preference to a simple U-tube manometer to obtain better  (a) accuracy (b) volume (c) precision (d) sensitivity  (6) I micro bar = dyne/cm².  (a) 1 (b) 2 (c) 3 (d) 4  (7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification?  (a) C - type (b) spiral (c) helical shaped (d) none of these  (8) The equation used in capacitive method is,  (a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D  (9) I micron is equal to mm of Hg.  (a) 10⁻² (b) 10⁻⁴ (c) 10⁻³ (d) 10⁻⁵  (10) The unit of pressure is,  (a) N/m² (b) N²/m³ (c) D/cm (d) D/cm³  Q-2 Answer the following questions in brief. (Answer any Ten Questions)  (1) Enlist the various temperature measurement methods.  (2) Explain about the Temperature scales in brief.  (3) Define: Thermocouple.  (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.  (5) State any four characteristics of manometric liquid.  (6) Enlist the advantages of Pirani gauge.  (7) Draw the schematic diagram of Bourdon gauge.		(4)		
<ul> <li>(5) A well type manometer is used in preference to a simple U-tube manometer to obtain better</li></ul>		(4)		
obtain better		/m\		
<ul> <li>(a) accuracy (b) volume (c) precision (d) sensitivity</li> <li>(6) 1 micro bar = dyne/cm².</li> <li>(a) 1 (b) 2 (c) 3 (d) 4</li> <li>(7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification?</li> <li>(a) C - type (b) spiral (c) helical shaped (d) none of these</li> <li>(8) The equation used in capacitive method is</li> <li>(a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D</li> <li>(9) 1 micron is equal to mm of Hg.</li> <li>(a) 10⁻² (b) 10⁻⁴ (c) 10⁻³ (d) 10⁻⁵</li> <li>(10) The unit of pressure is</li> <li>(a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions)</li> <li>(1) Enlist the various temperature measurement methods.</li> <li>(2) Explain about the Temperature scales in brief.</li> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(5)		
<ul> <li>(6) 1 micro bar = dyne/cm². (a) 1 (b) 2 (c) 3 (d) 4</li> <li>(7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification? (a) C - type (b) spiral (c) helical shaped (d) none of these</li> <li>(8) The equation used in capacitive method is (a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D</li> <li>(9) 1 micron is equal to mm of Hg. (a) 10⁻² (b) 10⁻⁴ (c) 10⁻³ (d) 10⁻⁵</li> <li>(10) The unit of pressure is (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions) (1) Enlist the various temperature measurement methods.</li> <li>(2) Explain about the Temperature scales in brief.</li> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>	·	•		
<ul> <li>(a) 1 (b) 2 (c) 3 (d) 4</li> <li>(7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification? <ul> <li>(a) C - type (b) spiral (c) helical shaped (d) none of these</li> </ul> </li> <li>(8) The equation used in capacitive method is</li></ul>		(6)		
<ul> <li>(7) Which of the following type of Bourdon tube shapes has a small tip travel and necessitates amplification? <ul> <li>(a) C - type (b) spiral (c) helical shaped (d) none of these</li> </ul> </li> <li>(8) The equation used in capacitive method is</li></ul>		(0)		
necessitates amplification?  (a) C - type (b) spiral (c) helical shaped (d) none of these  (8) The equation used in capacitive method is  (a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D  (9) 1 micron is equal to mm of Hg.  (a) 10 <sup>-2</sup> (b) 10 <sup>-4</sup> (c) 10 <sup>-3</sup> (d) 10 <sup>-5</sup> (10) The unit of pressure is  (a) N/m <sup>2</sup> (b) N <sup>2</sup> /m <sup>3</sup> (c) D/ cm (d) D/cm <sup>3</sup> Q-2 Answer the following questions in brief. (Answer any Ten Questions)  (1) Enlist the various temperature measurement methods.  (2) Explain about the Temperature scales in brief.  (3) Define: Thermocouple.  (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.  (5) State any four characteristics of manometric liquid.  (6) Enlist the advantages of Pirani gauge.  (7) Draw the schematic diagram of Bourdon gauge.		(7)		
<ul> <li>(a) C - type (b) spiral (c) helical shaped (d) none of these</li> <li>(8) The equation used in capacitive method is (a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D</li> <li>(9) 1 micron is equal to mm of Hg. (a) 10<sup>-2</sup> (b) 10<sup>-4</sup> (c) 10<sup>-3</sup> (d) 10<sup>-5</sup></li> <li>(10) The unit of pressure is (a) N/m<sup>2</sup> (b) N<sup>2</sup>/m<sup>3</sup> (c) D/ cm (d) D/cm<sup>3</sup></li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions) (1) Enlist the various temperature measurement methods.</li> <li>(2) Explain about the Temperature scales in brief.</li> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(*)		
<ul> <li>(8) The equation used in capacitive method is</li></ul>			·	
<ul> <li>(a) C = K A D (b) C = D A/K (c) C = K D/A (d) C = K A/D</li> <li>(9) 1 micron is equal to mm of Hg.</li></ul>		(8)		
<ul> <li>(9) 1 micron is equal to mm of Hg.     (a) 10<sup>-2</sup> (b) 10<sup>-4</sup> (c) 10<sup>-3</sup> (d) 10<sup>-5</sup> </li> <li>(10) The unit of pressure is     (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions)     (1) Enlist the various temperature measurement methods.     (2) Explain about the Temperature scales in brief.     (3) Define: Thermocouple.     (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.     (5) State any four characteristics of manometric liquid.     (6) Enlist the advantages of Pirani gauge.     (7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		( )	·	
<ul> <li>(10) The unit of pressure is (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions) (1) Enlist the various temperature measurement methods. (2) Explain about the Temperature scales in brief. (3) Define: Thermocouple. (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure. (5) State any four characteristics of manometric liquid. (6) Enlist the advantages of Pirani gauge. (7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(9)	1 micron is equal to mm of Hg.	
<ul> <li>(10) The unit of pressure is (a) N/m² (b) N²/m³ (c) D/ cm (d) D/cm³</li> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions) (1) Enlist the various temperature measurement methods. (2) Explain about the Temperature scales in brief. (3) Define: Thermocouple. (4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure. (5) State any four characteristics of manometric liquid. (6) Enlist the advantages of Pirani gauge. (7) Draw the schematic diagram of Bourdon gauge.</li> </ul>			(a) $10^{-2}$ (b) $10^{-4}$ (c) $10^{-3}$ (d) $10^{-5}$	
<ul> <li>Q-2 Answer the following questions in brief. (Answer any Ten Questions) <ol> <li>Enlist the various temperature measurement methods.</li> <li>Explain about the Temperature scales in brief.</li> <li>Define: Thermocouple.</li> <li>Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>State any four characteristics of manometric liquid.</li> <li>Enlist the advantages of Pirani gauge.</li> <li>Draw the schematic diagram of Bourdon gauge.</li> </ol> </li> </ul>		(10)	The unit of pressure is	
<ol> <li>Enlist the various temperature measurement methods.</li> <li>Explain about the Temperature scales in brief.</li> <li>Define: Thermocouple.</li> <li>Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>State any four characteristics of manometric liquid.</li> <li>Enlist the advantages of Pirani gauge.</li> <li>Draw the schematic diagram of Bourdon gauge.</li> </ol>			(a) $N/m^2$ (b) $N^2/m^3$ (c) D/ cm (d) D/cm <sup>3</sup>	
<ol> <li>Enlist the various temperature measurement methods.</li> <li>Explain about the Temperature scales in brief.</li> <li>Define: Thermocouple.</li> <li>Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>State any four characteristics of manometric liquid.</li> <li>Enlist the advantages of Pirani gauge.</li> <li>Draw the schematic diagram of Bourdon gauge.</li> </ol>		_	of the first the fact (Annual Tan Orankiana)	נססז
<ul> <li>(2) Explain about the Temperature scales in brief.</li> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>	Q-2			լշսյ
<ul> <li>(3) Define: Thermocouple.</li> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(1)	Enlist the various temperature measurement metrious.	
<ul> <li>(4) Draw diagram for the relation between absolute gauge pressure and atmospheric pressure.</li> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		-		
atmospheric pressure.  (5) State any four characteristics of manometric liquid.  (6) Enlist the advantages of Pirani gauge.  (7) Draw the schematic diagram of Bourdon gauge.				
<ul> <li>(5) State any four characteristics of manometric liquid.</li> <li>(6) Enlist the advantages of Pirani gauge.</li> <li>(7) Draw the schematic diagram of Bourdon gauge.</li> </ul>		(4)		
<ul><li>(6) Enlist the advantages of Pirani gauge.</li><li>(7) Draw the schematic diagram of Bourdon gauge.</li></ul>		(5)		
(7) Draw the schematic diagram of Bourdon gauge.			•	
/			<u> </u>	
(1.0)		• •	· · · · · · · · · · · · · · · · · · ·	(PTO)
	•	(0)	Sact the amended of the mar conductivity gadges.	( )

	(12)	Draw the schematic diagrams of float and shaft type level measurements.	
Q-3	(a) (b)	Write a detailed note on bimetallic thermometer.  Explain the laws of intermediate temperatures and intermediate metals.  OR	[6] [4]
Q-3	(a) (b)	Explain the electrical resistance thermometers with typical NTC diagram in brief. Write a short note on Liquid-in-glass thermometer.	[6] [4]
Q-4	(a) (b)	Derive the equation of U-tube double column manometer.  Define Static pressure and Total pressure.  OR	[6] [4]
Q-4	(a) (b)	Write a detailed note on single-column manometer.  Discuss the Ring balance manometer.	[6] [4]
Q-5	(a) (b)	Describe the working of Ionization gauge with the help of necessary diagram.  Write a short note on the Bourdon Gauge.	[6] [4]
Q-5	(a) (b)	OR Discuss the Thermal Conductivity Gauge with necessary diagram. Write a short note on McLeod gauge.	[6] [4]
Q-6		s and derive the expression for capacitive and ultrasonic method in brief and also s their advantages and disadvantages.	[10]
Q-6		OR as about the 'Direct Method' and Explain hydrometer method and pressure and of level measurement in briefly.	[10]

(9) State the advantages of ionization gauges.

(11) State the names of non-electrical and electrical methods.

(10) Define Density and its unit.

(T)

[44]

No. of Printed Pages; 2

## SARDAR PATEL UNIVERSITY

T.Y.B.Sc. Examination, FIFTH Semester Saturday, 11<sup>TH</sup> November 2017
Time: 10.00 am To 1.00 pm

Instrumentation Course Code: US05CINS03

: 70

		Course Title: Introduction to Control System
		Total Marks : 7
1-Q		answers to the following multiple choice questions in your [10] or book by selecting the proper option.
		A process control loop regulates some variables in a process.
	• ,	(a) dynamic (b) static (c) multi (d) single
	(2)	The selection of what controller mode to use in a process is a function of the of the process.
	<b>(m)</b>	(a) characteristics (b) features (c) constants (d) variables
	(3)	The deviation or error of the controlled variable from the set point is given by $e = \underline{\hspace{1cm}}$ .  (a) $r + b$ (b) $r \cdot b$ (c) $r - b$ (d) $r/b$
	(4)	A permanent error taking place in the operating point of the controlled variable due to the change in load is called  (a) two-set (b) one-set (c) onset (d) offset
	(5)	In the proportional mode if the error is, the output is a constant and equal to $p_0$ .  (a) 3 (b) 2 (c) 1 (d) 0
	(6)	Compressor controllers or regulators are used to vary the of the compressor.  (a) humidity (b) delivery (c) pressure (d) temperature
	(7)	The decrease in the volume of air in a compressor results in the of water.  (a) precipitation (b) evaporation (c) boiling (d) melting
	(8)	Refrigerated air dryers use to cool the air to a lower temperature.  (a) water (b) refrigerant (c) coolant (d) cooling agent
	(9)	The part of the valve through which the valve plug stem moves is known as assembly.  (a) Bonnet (b) cabinet (c) window (d) opening
	(10)	

			the following questions in brief. (Answer any Ten Questions)	20]
Q-2	Answe	er t	ovide a list of various control system parameters.	
	(1) (2)		. II - Laute A MI MA CHITEH Siuna to the man	
	( )	باه	ement and has $p=25\%$ , then find the corresponding	
	(3)		ive a brief introduction to controllers.  ive a brief introduction to continuous controller modes.	
	(4) (5)		nlist the characteristics of Derivative Control Mode.	
	(6)	C	tate the characteristics of PI mode.	
	(7)	3.4	with a short note on water-cooled intercoolers.	
	(8)	E	nlist the problems caused by water in the control lines.  nlist important factors for designing of an Instrument Air System.	
	(9) (10)		What is cavitation?	
	(10)		Vrite a brief note on control valves.	
	(12)	V	What is erosion of a valve?	
	-			[6]
Q-3	(a)	) E	explain the floating control mode in detail.  Discuss the applications of floating control mode in detail.	[4]
				[6]
Q-3	3 (a)	<b>)</b> [	Discuss the various process characteristics based on management	[0]
		- (	controller mode is selected. With the help of necessary figure and equation explain the	[4]
	(b)	)	Two-Position Control Mode.	
				[10]
Q-	4 With	th	e help of a suitable example explain the Proportional Controller Mode	<b></b>
			il with its applications and limitations.  OR	rel
Q-	л (a	١)	Explain the Proportional-Integral(PI) Controller Mode in detail.	[6] [4]
Q.	- (b	·)	Discuss about the Three Mode(PID) Controller.	רבז
	•			[6]
Q-	·5 (a	a)	Write a note on liquid piston compressor.	[4]
	(1	b)	Explain the reciprocating type compressor in detail.	
		_	OR  Discuss about the sizing criteria, pressure level and air supply source	[6]
Q	-5 (	a)	required for the designing of an instrument air system.	
	,	h)	Write a note on sliding vane rotary compressor.	[4]
	(	b)		[6]
O	9-6 (	a)	Discuss the three-way valve in detail.	[4]
~~		b)	- u it- insiduc foatilites of Didilitianity valves in war-	r ~ 7
_		(a)	Discuss about the double port globe valve in detail.	[6] [4]
ζ,		(a) (b)	and the second that the second	F1

## SARDAR PATEL UNIVERSITY

Vallabh Vidyanagar - 388120

B.Sc. (5<sup>th</sup> Sem) Examination - 2017 13<sup>th</sup> November, 2017 (Monday) 10:00 AM - 01:00 pm

US05CINS04 (Instrumentation)

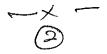
Programmable Logic Controller (PLC) - I

Maximum Marks: 70

### Que 1 Each question below gives a multiple choice of answers. Choose the most appropriate [10] one.

1	: Useful For Delaying Turn - ON Eve	ents.
	a) Timer	b) Counter
	c) Delay - ON Time Delay Relay	d) Delay - OFF Time Delay Relay
2	: MTW.	
	a) Magnetic Tool Winding	b) Microwave Transmission Workstation
	c) Machine Tool Wire	d) Movable Tool Workstation
3	: Includes Features such as Higher	Math Functions, PID Control Loops, and
	Optional Programming Commands.	, , , , , , , , , , , , , , , , , , , ,
	a) Input Module	b) Personal Computer
	c) PLC Processor	d) Programming Unit
4		nputs to the PLC, Nor Can They Be Used as
	Outputs.	
	a) Relay	b) Transformer and Push Button
	c) PLC Physical Components	d) Internal Relay
5		Math Functions and Other Data Manipulation
	Functions That Are Controlled By a Transit	
	a) Oscillator	b) PLC Software Components
	c) Internal Relay	d) Relay
6	: Color Lamp of Control Panel Indic	cates Conditions That Are Important But Not
	Dangerous.	
	a) Red	b) Green
	c) Amber	d) Pink
7	: Excellent For Applications Requiring	g Time to be "Streched".
	a) Relay	b) Counter
	c) Delay - ON Time Delay Relay	d) Delay - OFF Time Delay Relay
8	Latch Circuit Cannot Be Obtained By Using	g a Switch.
	a) Pushbutton	b) Selector
	c) Limit	d) Maintained
9	: THHN.	:
	a) Thermocouple Heat - Resistant Nylon	b) Thermoplastic Heat - Resistant Nylon
	Coated	Coated
	c) Thermoplastic Humidity - Resistant	d) Thermistor Heat - Resistant Nylon
	Nylon Coated	Coated
10	: Provides Data and Power Connec	tions to the Processor an Modules Via the
	Backplane.	
	a) PLC Output Module	b) Transformer and Wiring
	c) PLC Input Module	d) PLC Mounting Rack
		C P.T.
	_	

1 2 3 4 5 6 7 8 9 10 11 12	What Do You Mean By Normally Open (N/O) and Normally Closed (N/C) Contacts? Draw Ladder Diagram For X-OR and X-NOR Logic Gates. Explain Function of Input and Output Modules. Draw the Ladder Logic Rung For a Normally Open IN <sub>1</sub> AND'ed With a Normally Closed IN <sub>2</sub> Driving a Coil CR <sub>1</sub> .  Draw Ladder For $Y = \overline{AB + (C + D)}$ .  Delay - OFF Timer (TOF) Relays are Used in Outdoor Lighting Control Motion Sensors. Justify.  Explain Basic Difference Between Delay - ON Timer (TON) and Delay - OFF Timer (TOF) Relays.  Enlist Any Four PLC Programming Languages.  Draw Ladder Diagram For Universal Gates.  Draw Ladder For $Y = \overline{(A + B) \cdot C}$ What is PLC?  What Do You Mean By Disagreement Circuit?					
Que 3	[A] [B]	Write on PLC Contactor With Necessary Diagram. Explain Selector Switches Used in PLC.	[05] [05]			
	[]	OR	[03]			
	[C]	What is Timer? Explain Delay - ON Timer (TON) Relay With Timing Diagram.	[05]			
	[D]	Discuss Pushbutton Switches Used in PLC.	[05]			
Que 4	[A] [B]	Explain Importance of Reference Designators in PLC. Give Significance of Latching Contacts in PLC.	[05] [05]			
	OR					
	[C] [D]	Draw Ladder Diagram For AND OR and OR AND Logic. Give an Account of PLC Wiring.	[05] [05]			
Que 5	[A]	Write a Detailed Note on PLC Configuration.	[10]			
		OR				
	[B]	Discuss "Solve The Ladder" in Terms of PLC.	[10]			
Que 6	[A]	Give an Account of Always-ON and Always-OFF Contacts.	[05]			
	[B]	Explain PLC Internal Relay.	[05]			
		OR	٠			
	[C]	Write Note on PLC Physical Components Vs. Program Components.	[10]			



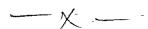
SC

70

	1	No. of Printed Pages: 02  SARDAR PATEL UNIVERSITY  T.Y.B.Sc. Examination, FIFTH Semester  Wednesday, 15 <sup>TH</sup> November 2017  Time: 10.00 am To 1.00 pm  Instrumentation Course Code: USO5CINS05	@ @
,		Course Title: Industrial Electronics - 1	•
		Total Ma Inswers to the following multiple choice questions in your answer book by selecting	
th		per option.	
	(1)	The primary coil of a transformer is also called	
		(a) main coil (b) low voltage coil	
	(2)	(c) high voltage coil (d) medium voltage coil	
	(2)	a a more definition of the date	
		(a) hysteresis loss (b) eddy current loss (c) stray loss (d) capacitive loss	
	(3)		
	(3)	If the windings surround a considerable part of the core the transformer is called (a) shell type (b) oil-filled (c) core-type (d) air-tight	
	(4)		
	1.1	(a) capacitor (b) inductor (c) rectifier (d) inverter	
	(5)	The machine which converts mechanical power to electrical power is called	
	V=1	(a) generator (b) motor (c) converter (d) inverter	
	(6)		
		masses quickly the dc motor is used.	
		(a) shunt (b) series (c) parallel (d) inverted	
	(7)	A MA THE SECOND STREET STREET	
		their constancy of	
		(a) speed (b) armature current	
		(c) armature torque (d) back emf	
	(8)	, , , , , , , , , , , , , , , , , , , ,	
	19407 1976	(a) slip (b) split (c) spot (d) end	
	(9)	In a two phase induction motor the resultant revolving magnetic flux produced by the	
		stator windings has a constant value	
		(a) $\Phi_{\rm m}$ (b) $2\Phi_{\rm m}$ (c) $\Phi_{\rm m}/2$ (d) $3\Phi_{\rm m}/2$	
	(10)	The quantity $(N_s - N)$ is called speed.	
		(a) actual (b) virtual (c) sleep (d) split	
Q-2 Ar	CWOR	the following questions in brief (Annual Trans.	
ų-∠ Ai			[20]
	(+)	Write a short note on classification of transformer on the basis of the cooling methods employed.	t
	(2)	Discuss the theory of an ideal transformer in brief.	
ī		Write a short note on voltage-transformation ratio.	
		Explain the motor action using the principle of working of a dc motor.	
	(5)	Write a short note on mechanical losses in a dc generator.	
	(6)	Discuss the general principle of operation of a dc motor.	PTO
	9		

		The state of speed regulation,	
	(8)	The design determines of the motors.	
	(9)	Enlist the applications of series dc motors.	
	(10)	and the first of de motors with regards to their principle of operation	
	(11)	State the advantages of induction motors.	
	(12)	Explain how does the rotor of an induction motor rotate?	
Q-3	(a)	Explain the construction of Core type and Shell type transformer in detail.	<b>.</b>
	(b)	Derive the expressions for the equivalent reactance and impedance of transformer.  OR	[6] [4]
Q-3	, ,	Discuss the step-by-step procedure for obtaining the single-line equivalent circuit of a transformer.	[6]
	(b)	Derive the formula for the equivalent resistance of a transformer.	[4]
Q-4	Descril rectify	be the construction and working of a simple-loop dc generator in detail. Also discuss the ng action of split-ring.	[10]
0.4	(-)	OR	
Q-4	(a)	Give a detailed account of the total losses in a dc generator.	[6]
	(b)	Write a detailed note on armature torque of a dc motor.	[4]
Q-5	(a)	Discuss the different types of characteristics of a shunt dc motor in detail.	[6]
	(b)	Derive the expression for the speed of a dc motor.	[4]
		OR	F - 4
Q-5	(a)	What are compound dc motors? With the help of necessary figures explain the construction and working of cumulatively and differentially compound dc motors in detail.	[6]
	(b)	Write a detailed note on the applications of dc motor.	[4]
Q-6	(a)	In case of a two-phase induction motor, explain how the rotating magnetic field is generated by its stator windings.	[5]
	(b)	Write a detailed note on the frequency of rotor currents.  OR	[5]
Q-6	(a)	In case of a three-phase induction motor, explain how the rotating magnetic field is generated by its stator windings.	[5]
	(b)	With the help of necessary diagrams explain the construction of induction motors in	[5]

(7) Write a short note on speed regulation.



detail.

[31]

### SARDAR PATEL UNIVERSITY

T.Y.B.Sc. Examination, FIFTH Semester

Friday, 17<sup>TH</sup> November 2017 Time: 10.00 am To 1.00 pm

Instrumentation Course Code: USO5CINS06 Course Title: Analytical Instrumentation

Total Marks: 70

- Q-1 Write answers to the following multiple choice questions in your answer book by [10] selecting the proper option.
  - (1) The useful pH range for glass electrode lies between \_\_\_\_
    - (a) pH 7 and pH 11
- (b) pH 1 and pH 11
- (c) pH 7 and pH 111
- (d) pH 11 and pH 1
- (2) For electron capture detector which is the best carrier gas?
  - (a) hydrogen (b) oxygen (c) argon (d) helium
- (3) Which part of GC is considered as the heart of chromatography?
  - (a) column (b) recorder (c) gas cylinder (d) Owen
- (4) Full form of HPCL is
  - (a) High Pressure Lead Chromatography
  - (b) High Pressure Liquid Chromatography
  - (c) High Pump Liquid Chromatography
  - (d) High Pump Lead Chromatography
- (5) The cell in spectrophotometric detectors may be made up with \_\_\_\_ (a) Teflon (b) aluminum (c) nickel (d) copper
- (6) pH of pure water is \_\_\_\_
  - (a) 5 (b) 7 (c) 4 (d) 2
- (7) Which type of gas is used as carrier gas in gas chromatography? (a) H<sub>2</sub> (b) inert gas (c) CO<sub>2</sub> (d) O<sub>2</sub>
- (8) The length of column in gas chromatography is between \_\_\_
  - (a) 1 and 30 m (b) 1 and 40 m (c) 1 and 50 m (d) 1 and 60 m
- (9) The displacement of the membrane in magnetic wind analyser can be commonly measured by using \_\_\_\_
  - (a) LEDT (b) LDVT (c) LVDT (d) LEVT
- (10) Full form of TLC is
  - (a) Thermoelectric Lead Chromatography
  - (b) Thin Layer Chromatography
  - (c) Temperature Leading Chromatography
  - (d) Thick Layer Chromatograph

PTO

Q-2		r the following questions in brief. (Answer any Ten Questions)	[20]
		Define pH. Give its formula.	
	(2)	Enlist the names of electrodes.	
	(3)	Why column is considered as Heart of gas Chromatography?	
	(4)	Enlist the types of Gas Analyzers.	
	(5)	Give classification of liquid chromatography.	
	(6)	Write a note on adsorption detector.	
	(7)	Enlist the names of pH meters.	
	(8)	Give classification of chromatography.	
	(9)	What is the use of pressure regulator and flow regulator?	
	(10)	Enlist the thermal conductivity meter.	
	(11)	Write a brief note on TLC.	
,	(12)	Enlist the detectors of the liquid chromatography.	
Q-3	(a)	Explain Chopper Amplifier type pH meter.	[5]
		How to calculate?	[5]
	• •	OR	ורו
Q-3	(a)	Explain Buffer solution.	[5]
	(b)	Write a note on glass electrode.	[5]
Q-4		ne chromatography. With necessary block diagram explain gas matography.	[10]
	CIIIO	OR	
Q-4	(a)	Why do we need to connect detector with strip chart recorder?	[5]
~ -	(b)	Write a brief note on capillary column and packed column.	[5]
			ren
Q-5	(a)	With necessary diagram explain magnetic wind analyzer.	[5]
	(b)	Write a note on conductivity cell.	[5]
		OR	[5]
Q-5	(a)	Write a note on IR gas analyzer.	[5]
	(b)	Write a note on thermal conductivity analyzer.	[3]
Q-6	(a)	Write a note on detection system.	[5]
Q-u	(b)	3 1 3	[5]
	(D)	OR	
Q-6	(a)	Write a note on column.	[5]
Q C	(b)		[5]
	(-)		
		$\longrightarrow \chi \longrightarrow$	

<u></u>	43]	i	SEAT No
		ı	

No. of Printed Pages: 2

### SARDAR PATEL UNIVERSITY

## T.Y.B.Sc. (Semester - V) EXAMINATION November - 2017

US05CINT01: Relational Database Management System-I

Date: 0	7/11/2017, Tyesday	Time: 10:00	Time: 10:00 to 01:00 PM			
Q.1	Multiple choice ques	stions:		[10]		
1.	Actual value of a col	umn is also called		. ,		
	[A] Tuple	[B] Derived Attribut	e			
	[C] Attribute	[D] Domain				
2.	Which SQL statemen	it is used to update da	ta in a database?			
• •	[A] SAVE AS	[B] MODIFY				
3.	To commit a transac	tion means making the	changes			
	[A] Temporary					
	[C] Parsed	[D] Rolled Back				
4.	Which of the following Comparisons	ng SQL operations den ?	nands the use of wild o	cards		
	[A] IN	[B] BETWEEN	[C] EXISTS	[D] LIKE		
5.	Which SQL keyword	is used to sort the resu	ılt-set?			
	[A] Sort	[B] Order by	[C] Sort by	[D] Order		
6.	Business rules, which are enforced on data being stored in a table, are called					
•		[B] Constraint				
	[C] Unique	[D] Protocol				
7.	The data held across	the primary key colum	in must be			
		[B] Repetitive		[D] Simple		
8.	The function c	onverts char, a CHARA	CTER value expressing	g a number, to a		
	NUMBER data-type.					
	[A] TO_NUMBER	[B] TO_CHAR				
	[C] TO_DATE	[D] TO_NUM				
9.	clause im	poses a condition of th	e group by clause.			
	[A] Group by	[B] Sub Query	[C] Having	[D] Where		
10.	Thestate	ment provides various	types of access to dat	tabase object.		
	[A] Select			-		
	[C] Revoke	[D] Grant				

(PTO)

Q.2	Write answer in short (Any Ten)	[20
	1. Draw ER- Diagram for E-shopping system.	_
	2. Define Attribute , Tuple	
	3. What is normalization? List its phases.	
	4. List any two SQL * PLUS commands with example.	
	5. Explain concept of DUAL table.	
	6. List any four arithmetic operator and give any two examples on dual table.	
	7. List the restriction available on ALTER table.	
	8. How UNIQUE key can apply at table level?	
	9. List all scalar functions available in oracle.	
	10. What is sub query? Why it is used for?	
	11. List the diff. types of joins.	
	12. What is synonym?	
Q.3(a)	Differentiate DBMS and RDBMS.	[4]
(b)	Explain types of relationship and mapping cardinality with example	[6]
Q.3(a)	List out Codd rules and explain any four.	[4]
(b)	Define Normalization. Explain 1NF, 2NF and 3NF by taking suitable example.	[c]
()		[6]
Q.4(a)	Explain DDL, DML, and DCL.	[6]
(b)	Explain CREATE Statement.	[4]
Q.4(a)	Discuss pattern matching predicate in detail.	[6]
(b)	Explain SELECT statement.	[4]
Q.5	Define Primary key and foreign key concept with appropriate illustration	[10]
Q.5	List all Aggregate functions available in oracle and explain any Three of them with	[10]
7,0	appropriate syntax and example.	[10]
Q.6(a)	What is transaction processing? Explain the commit, rollback.	[6]
(b)	Explain group by & having with example.	[4]
		[4]
Q.6(a)	What is view? Why it is created, explain it syntax & example.	[6]
(b)	What is index? Explain creation of simple & composes index.	[4]



Page 2 of 2

# SEAT No. \_\_\_\_\_ No SARDAR PATEL UNIVERSITY T.Y. B.Sc. EXAMINATION, SEM – V

## INFORMATION TECHNOLOGY

**US05CINT02: DATA AND FILE STRUCTURE** 

Date:	09/11/	/2017 Time: 10:00am to 01:00pm Max.Marks	:70
Q-1	Multi	dod ple Choice Question	[10]
	i)	An array is a data structure.	
		a. Composite. b. Unordered. c. Non-composite. d. Heterogeneous.	
	ii)	An array is a data structure.	
		a. Non-composite. b. Non- Linear. c. Ordered. d. Heterogeneous.	
	iii)	The number of elements in an array is called the of the array.	
		a. Size. b. Base. c. Type. d. Index.	
•	iv)	In preorder traversal of a binary tree, root node is always located at	
		position.	
		a. Second last c. First	
		b. Second d. Last	
	v)	Preorder traversal of a binary tree can be visited manner.	
		a. Left sub-tree, Root, Right sub-tree.	
		b. Root, Left sub-tree, Right sub-tree.	
		c. Root, Right sub-tree, Left sub-tree.	
		d .Left sub-tree, Right sub-tree, Root.	
	vi)	Inorder traversal of a binary tree can be visited manner.	
		a. Left sub-tree, Root, Right sub-tree.	
		b. Root, Left sub-tree, Right sub-tree.	
		c. Root, Right sub-tree, Left sub-tree.	
		d. Left sub-tree, Right sub-tree, Root	
	vii)	K-way merging is known as	
		a. Multiple merge c. Simple merging	
		b. Selection sort d. Binary merging	
	viii)	If n denotes the sum of the sizes of the two sub tables to be merged, then the	
		timing performance of merge sort algorithm is	
		a. $O(n^2)$ c. $O(n^3)$	
		b. O (n) d. None of these	
	ix)	A is a collection of information items about a particular entity.	
	,	a. Record b. Database c. Entity d. None of these	
	x)	is a unit of meaningful information about an entity.	
		a. Record b. Entity c. Item d. None of these	
$\dot{\mathbf{Q}} - 2$	Answ	er the following in short. (Any 10)	[20]
_	i)	Write difference between Linear and Non-Linear data structure	[20]
	ii)	List out Characteristics of algorithm for data structure	
	iii)	Define Hierarchical structure of data structure.	
	iv)	Define tree with an example.	
	v)	Define doubly link list.	
	vi)	Write difference between Singly link list and Doubly link list	
ż	vii)	List application of searching.	
	viii)	List application of sorting.	
	ix)	Define searching.	
	x)	Define: Key, Sequence Key.	
	xí)	Define: File, Database.	
	xii)	What do you mean by Transaction?	

Q-3		Write detail note on Two-Dimension array.	[10]
		OR	
Q-3		What is Stack? Write algorithms for PUSH(), POP(), PEEP() and CHANGE() operations on stack	[10]
Q-4	a)	Explain insertion operation in lexically ordered binary tree.	[5]
	b)	Write Algorithm to Insert element in Doubly link list OR	[5]
Q-4	a)	Write Algorithm to insert element in sorted ordered list.	[5]
	<b>b</b> )	Explain Deletion operation in lexically ordered binary tree.	[5]
Q – 5		List sorting techniques and explain any two of them with algorithm.  OR	[10]
Q-5		List and explain searching techniques with algorithms.	[10]
Q-6		Write a detail note on structure of index sequential file. OR	[10]
Q-6	a)	Write a detail note on processing of Direct file.	[5]
	b)	Write a short note on Multiple buffering.	[5]
		2	

[45] SEAT NO.\_\_\_\_

## SARDAR PATEL UNIVERSITY

T.Y.B.Sc : SEMESTER - V INFORMATION TECHNOLOGY

		US05CINT03: Visual Programming Through VB .NET	
Date		11-2017 Time: 10:00am to 01:00nm May Mayle	s· 70
Q – 1	Sci Muli	ruoday	. , ,
ω – 1	i)	tiple Choice Question	[10]
	''	Dynamic Arrays can Redimensioned using Keyword.  A) Dim B) Redim	
	ii)		
	111)	A Procedure that does not return a Value is called as  A) Sub Procedure B) Method	
	iii)	C) Constructor D) Function Toolbox in Visual Studio Consist of	
	****/		
		=/ =iot of t of this	
	iv)	- / =	
	147	is default event for form.  A) Click  B) DhlClick	•
		_/	
	v)	-	
	•,	specifies the interior spacing of the control.  A) Padding  B) Shifting	
		-/	
	vi)	,	
	,	is used when all condition becomes false in selectCase statement.	
		A) End Select B) Case Else	
		C) Else Case D) Select End	
	vii)	statement enables exception handling .	
	,	A) Option Explicit B) On Error Goto	:
		C) Option Strict D) None of the above	
	viii)	Which property can not be used with multiline property of textbox.	
	,	A) Name  B) Passwordchar	
		C) Font D) Margin	
	ix)	DataSets are loaded from the database using the:	
		A) Load method B) Read method	
		C) Fill method D) None of the above	
	x)	Stored procedures are invoked using:	
		A) SqlCommand object and its ExecuteReader method.	
		B) SqlConnection and its ExecuteReader method.	
		C) SqlCommand object and its ExecuteNonQuery method	
		D) SqlParameter object and its Execute method.	
· •	A		
J – Z		npt any 10 questions:	[20]
	i)	Define Boxing.	~ -
	· ii)	What is JIT?	
	iii)	What is a variable?	
	iv)	What is the use of 'By Val' & 'By Ref' keyword?	
	vi)	What is procedure? How it is created in VB .NET?	
	vi)	Write down the syntax of With End Width.	
	vii)	Explain the use of Menu.	
	viii) ivi	What is the use of Groupbox?	
	ix) v\	Explain any four properties of DateTimePicker.	
	x) xi)	Explain System.Data Namespace. What are the features of ADO Not?	
		TIME OF THE LEADINGS OF ALL STATES	

xii) What are the applications of ADO .net?

Q - 3	a) b)	Explain .NET Architecture. Write a note on Class Library.	[5] [5]
Q – 3	a) b)	OR Write a short note on Solution Explorer Write a note on CTS?	[5] [5]
Q – 4	a) b)	Explain Form life cycle in detail. Explain Input box with example.	[5] [5]
Q – 4		OR Explain the following structures with example. 1. For Next 2. While End While 3. If End If	[10]
Q – 5	a) b)	Explain try Catch final in detail with example. Write a note on HScorllBar and VScrollBar.	[5] [5]
Q – 5		OR Explain following controls with its properties, methods and events 1) Treeview 2) RichTextBox 3) ColorDialogBox	[10]
Q – 6		Explain use of "ExecuteScaler", "ExecuteNonQuery" and "ExecuteReader" method in detail.	[10]
Q – 6	a)	OR Explain the steps to bind the application with the Database in ADO .net.	[5]
	b)	Explain the connected architecture of ADO.NET in brief.  ———————————————————————————————————	[5]

## SARDAR PATEL UNIVERSITY

B.Sc. (Information Technology) Examination

5<sup>th</sup> Semester (CBCS) (Regular & NC)

Monday, Date: 13<sup>th</sup> November, 2017

Session: Morning Time: 10:00 A.M. TO 01:00 PM

Course T	Code: US05CINT04  Total Marks: 70  Title: Operating Systems	[10]				
Q1.	Multiple Choice Questions.(Attempt All)	լւսյ				
1.	A Program in execution is known as  C. Record					
	A. The					
	D, Hoceas					
2.	is <b>not</b> a valid process state.					
	A. RUNNING C. NEW D. STOP					
_	B. READY Round Robin scheduling is essentially the preemptive version of					
3.	C BUNS					
	D. None of these.					
4.	scheduling algorithm gives minimum Page Faults.					
٦.	A FIFO C. Optimal					
	D. None of the these					
5.	Which of the following memory allocation strategy is fastest?  C. Worst-fit					
	n. Dost itt					
	D. Pilot iii					
6.	A process produces information that is consumed by a					
	consumer process.  Cooperating C. Producer					
	D. Independent					
7.	Section in EXT2 file system represents permissions on a file.					
, ·	A node C, block					
	D. i-node					
8.	Option of <b>ls</b> command in LINUX will sort output according to file size.  Cs					
	N, 1					
9.	B1 In LINUX the case statement ends with					
9.	A. end C. end case					
	D. esac					
10.	command is used to is change a permission of a file in LINUX.  C. chmod					
	A. Chiode					
	b. mode	[20]				
Q2.	Answer the following short questions (free may all 7 me and Pesponse Time?					
1.	What are Turnaround Time and Response Time?					
2.	Draw the diagram of PCB.					
3.	Define Operating system. Give 2 examples.					
4.	Explain Best-fit memory allocation technique in brief.					
5.	List different types of fragmentation.					
6.	List out all page replacement algorithms.					
7.	Justify "Linux is a Secure Operating System".	Justify "Linux is a Secure Operating System".				
8.	List three requirements for solution to Critical section problem.					
9.	Explain when does Race condition arise?					
10.	1 1 - 1 - aut					
11.	1 -1 with its ontions					
12.	Inhiam ****					

Explain the functions performed by Operating System in brief. Q3. a.

[05]

Explain Round Robin scheduling with example. Draw Gantt chart and [05] find average waiting time for 4 processes given below:

(Time Quantum:03 millisec)

PROCESS	BURST TIME (millisec)
P1	21
P2	3
P3	6
P4	2

OR

Q3. a. List and explain various Process states with labeled diagram.

[05]

[05]

Explain FCFS Scheduling algorithm in brief. Draw Gantt chart and find average waiting time for 4 processes given below:

PROCESS	BURST TIME (millisec)
P1	21
P2	3
P3	6
P4	2

Q4.a. Explain FIFO page replacement in brief and Calculate Page faults using [05] IFO page replacement algorithm for following reference string:

Reference string = 1, 2, 3, 4, 2, 1, 5, 6, 2, 1, 2, 3, 1,3,1

What is Fragmentation? Explain any one type of Fragmentation in detail. [05]

Q4.a. Explain Paging techniques with advantages and disadvantages. [05]

b. Explain Optimal page replacement Algorithm in brief and Calculate Page [05] faults using Optimal page replacement algorithm for following reference string: (Number of Frames = 3)

Reference string = 1, 2, 3, 4, 2, 1, 5, 6, 2, 1, 2, 3, 1,3,1

Q5.a. Explain concept of Process Synchronization in brief. [05]

Explain Critical-section problem in details. Explain algorithm 3 for [05] solving critical section problem for two-process.

What do you mean by Deadlock? Explain all necessary conditions for [05] occurrence of deadlock.

b. What is LINUX? Explain basic features of LINUX Operating System. [05]

Q6. Explain following LINUX commands in brief: [10]i) ls ii) cp iii) who iv) rmdir v) mv OR

Q6. Explain for and while loop statements in LINUX. [10]

No. of Printed Pages: 02

# SARDAR PATELUNIVERSITY B.Sc. EXAMINATION, 5<sup>th</sup> SEMESTER 15<sup>th</sup> November Wednesday, 2017

10:00a.m. to 1:00 p.m.

## US05CINT05

[System Analysis Design and Software Engineering] Maximum Marks: 70

Q-1	1)	-	oice Question.[Each Question carried outputs of subsystem are known		<del>-</del>	[10]
	*)	_	he other, is called			
		A.	Boundary	В.	Subsystem	
		. C.	Interface	D.	Black box	
	2)	Operational of	evaluation consider			
	,	Α.	Response time	В.	Easy of use	
		. C.	Reliability of computation	D.	All of these	
	3)		is defined as the activity that make	s possible	the transformation of	
	,	input to outp		٠٠,		
		A.	Input	В.	Process	
		C.	Output	D.	Goal.	
	4)	The	is contains a list of terms and	their defin	itions for all data items	
		and data stor	es of a system.		•	
		A.	Data Dictionary	В.	DFD	
		C.	Decision Table	D.	Decision Tree	
	5)	Salary of per	sonnel is called			
		A.	Recurring Cost	В.	Tangible Cost	
		C.	Intangible Cost	D.	One Time Cost	
	6)	Structure Ar	nalysis does not use			
		<b>A.</b> •	Data Dictionary	В.	Structured Diagram	
		C	E-R Diagram	D.	Decision Tree	
	7)		means reports contain misleadi	ng informa	tion.	
		A.	Problem of Reliability	В.	Problem of Validity	
		C.	Problem of Timeliness	D.	Problem of Capacity	
	8)	To conduct	a detailed feasibility study, firstly a	ın expert co	ommittee called	
		A.	Steering Committee	В.	Review Committee	
		C.	Evaluation Committee	D.	None of these	
	9)		model provides better risk man	agement ar	nd cost of each phase.	
		A.	Spiral	В.	Prototype	
		C.	Iterative enhancement	D.	Waterfall	
	10)		part requires major effo	rts.		
•		Α.	Testing	В.	Maintenance	
		C.	Coding	D.	Design	

Q-2		Give Answers for the following:(Any ten)	[20]
	1.	Draw the general model of a system.	
	2.	List subsystem and interface of computer system.	
	3.	Define System with example.	
	4.	List Input, Process, Output and Feedback of any educational University system.	
	5.	Write difference between natural and artificial system.	٠
	6.	List methods of feasibility study.	
	7.	What is SSADM? List all methodology of SSADM.	
	8.	Explain system survey.	
	9.	Explain hardware study.	
	10.	Write a short note on maintenance phase.	
	11.	Explain advantages of spiral model.	
	12.	Define: Software, Software Engineering.	
Q -3	A) B)	Explain any two Characteristics of System.  Explain System analysis.	[5] [5]
		OR	
Q -3	A)	Explain role of system analyst.	[5]
	B)	Explain Interface.	[5]
Q -4	A)	Explain Problem Identification.	[5]
	B)	Write difference between system analysis and system design.  OR	[5]
Q -4	A)	Explain system design specification and programming.	[5]
	B)	Write difference between technical and operational feasibility.	[5]
Q -5	A)	Explain need for structured analysis and design(SSADM).	[5]
	B)	Write difference between tangible and intangible benefit.  OR	[5]
Q -5	A)	Explain structure analysis of SSADM.	[5]
	B)	Write advantages of SSADM.	[5]
Q -6		Explain waterfall model in detail.	[10]
		OR	£1 03
Q -6		List phases of software development and explain any two in detail.	[10]

SEAT No.

No. of Printed Pages: 02

10

[32]

## SARDAR PATEL UNIVERSITY

External Examination (CBCS)

B. Sc. (CA & IT) - V<sup>th -</sup> Semester

US05CINT06 - Computer Architecture & Microprocessor

17<sup>th</sup> November, Friday - 2017

Time:	10:00 am to 1:00 pm		Total Marks :70
Q-1	Select an appropriate option.		10
1.	CPU stands for  (a) Central Processing Unit  (c) Central Programming Unit	(b) Control Processing (d) Control Programmin	Jnit g Unit
2.	The store intermediate dat (a) Register Set (b) CPU	a used during the executi (c) CU	on of instruction. (d) ALU
3.	RPN Stands for	(b) Reverse Programm (d) None of these	ng Notation
4.	The register specifies the register (a) Word Count (b) Address	number of words that mus (c) Data	st be transferred. (d) Control
5.	The contains an address (a) Address Register (c) Control Register	(b) Word Count Register	ation in memory.
6.	The memory unit that communicate (a) Main Memory (c) Secondary Memory	(b) Auxiliary Memory	s called the
7.	Theunit provides the necessoperations in the microcomputer.  (a) Control (b) Input		
8.	The group of eight bit is called (a) Bit (b) Byte	(c) Nibble	(d) Kilo Byte
9.	LED Stands for  (a) Light Energy Data  (c) Light Emitting Diode	(b) Light Emitting Da (d) Liquid Emitting D	
10.	PSW stands for  (a) Program Status Word  (c) Program Set Word	(b) Program Store Word (d) None of these	b

Q-2	Answer the following questions. (Attempt any TEN)	20
1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11.	Explain PUSH and POP operation of Stack. Define Infix Notation with example. Define I/O command. Explain Bus Request (BR) and Bus Grant (BG). What is Cache memory? Define Multiprogramming. Define Multiprocessor. Define Machine Level Languages. What is Operating System? Define Data Bus. Define Program Counter. Write full form of EPROM and EEPROM.	
Q-3 (a) (b)	Explain Components of CPU. Explain Two Address Instruction with example.	5 5
	OR	
Q-3 (a) (b)	Explain General Register Organization. Explain RPN with example.	5 5
Q-4 (a) (b)	What is DMA? Explain DMA Controller. Explain Binary Multiplication using Register Method.	5 5
	OR	
Q-4 (a) (b)	What is Memory? Explain memory hierarchy in computer system.  Explain Virtual Memory with Address Space and Memory Space.	5 5
Q-( (a) (b)	Explain characteristics of Multiprocessor.	5 5
	OR	
Q- (a) (b)	Explain Microprocessor in detail.	5 5
Q.	-6 Explain Memory Classification in detail.	10
	OR	
Q	-6 Write short note on: (i) Internal Data Operation (ii) Input Output Devices	10

(PTO)

No. of Printed Pages: 2

## SARDAR PATEL UNIVERSITY

## **B.Sc.** Fifth semester

# Instrumentation (Vocational) US05CINV01

## **Process Measurement Techniques-1** Tuesday, 07/11/2017

Time	e: - 1(	0:00AM To 1:00 PM		<b>Marks: - 70</b>	
Q.1	Cho	ose the correct answer (Attemp	ot all)	(10)	
(1)		e measuring temperature which of tl s temperature condition?	he follo	owing is used to reduce effect of	
	(a)	Liquid filled thermometer.	(c)	Gas filled thermometer.	
	(b)	Vapour filled thermometer.	(d)	Dual filled thermometer.	
(2)	Wha	t is the principle of operation of Bi A	Netalli	c strip type thermometer?	
	(a)	Thermal electricity.	(c)	Thermal radiation.	
	(b)	Thermal expansion.	(d)	Thermal conduction.	
(3)		ch of the following scale is accepted	as inte	ernational scale for temperature	
		surement?			
	(a)		(c)	Celsius.	
	(b)	Rankin.	(d)	Kelvin.	
(4)	Whi	ch of following sensor offers highest		,	
	(a)	Thermistors.		Thermocouples.	
	(b)	RTD.	(d)	Bimetallic strip.	
(5)	Which method is best suited for measurement of temperature of remotely placed object?				
	(a)		(c)	Mercury thermometers.	
	(b)	Filled system thermometers.	(d)	•	
(6)	• •	t is relationship between The energy			
	temp	oerature?			
	(a)	Directly proportional.	(c)	Inversely proportional.	
	(b)	Square of temperature.	(d)	Forth power of temperature.	
(7)	Which of the following manometer offers highest sensitivity?				
	(a)	U tube.	(c)	Inclined U tube.	
	(b)	Inverted U tube.	(d)	Piezo meter.	
(8)	Wha	t is the principle of operation of the	e manoi	meter?	
	(a)	Electrostatic balance.	(c)	Hydrostatic balance.	
	(b)	Electromagnetic balance.	(d)	Pneumatic balance.	
(9)	Whi	ch of following cannot be used to me	asure	vacuum?	
	(a)	Piezo meter.	(c)	Pirani gauge.	
	(b)	U tube manometer.	(d)	Bourdon gauge.	
(10)		ch type of bourdon gauge has small t	ip trav	vel?	
•	(a),		(c)		
	(b)	Helical.	(d)	Twisted.	

Q2	Answer the following questions (Any Ten)	(20)
(1)	Explain the principle of the liquid in glass thermometer.	
(2)	Explain why the vapor filled thermometers are better than liquid filled.	
(3)	Define temperature and explain importance of temperature measurement industrial processes.	nt in
(4)	What is the difference between the industrial and Laboratory type RTD senso	rs?
( <del>ד)</del> (5)	Discuss the See back's experiment of the thermoelectricity.	
(6)	Write a note on thermo pile? And list its characteristics features.	
(7)	Explain the practical setup used to measure the atmospheric pressure.	
(8)	Differentiate Atmospheric and Absolute pressure.	
(9)	Calculate the pressure excreted at the bottom of the water tank filled meter level.	to 1.5
(10)	List the advantages and disadvantages of the Ring balance pressure gau	ge.
(11)	List the advantages and disadvantages of the bourdon gauge.	
(12)	List the characteristics features required in materials of the Diaphragm.	
		44.00
Q3	Describe the physical quantity temp. List and discuss the various	(10)
	measurement scales (units) used for the temperature measurement and	
	also write interconversion formulas.	
	OR	
Q3	With necessary circuits explain the principles of working of bi-metallic	(10)
	strip thermometer.	
Q4	Write a note on total radiation Pyrometer method for temperature	(10)
α.	measurement.	` ,
	OR	
Q4	What are thermocouples? Explain construction of thermocouple probe	(10)
<b>α</b> ,	with necessary diagrams and also discuss its characteristics curve.	` ,
		(40)
Q5	With necessary diagrams write a note on Inclined U tube manometers.	(10)
	OR	(4.0)
Q5	Explain various terms used in the pressure measurements and explain	n (10)
	the method for the measurement of atmospheric pressure.	
Q6	Explain working principle and construction of Pirani gauge.	(10)
Q,	OR	
Q6	Explain the principles of Bourdon gauge and draw the diagrams o	f (10)
٠,٠	various types of bourdon gauge and discuss their features.	
	X =	

Answer the following questions (Any Ten)

(20)

## SARDAR PATEL UNIVERSITY

## B.Sc. ( $5^{TH}$ Sem) Examination - 2017

09th November, 2017 (Thursday)

10:00 AM - 01:00 PM

US05CINV02 (Instrumentation - Vocational)

Control Technique - I

Maximum Marks: 70

## Que 1 Each question below gives a multiple choice of answers. Choose the [10] most appropriate one.

Determined by the Error.  a) Two - Position  c) Floating  d) None of These  : The Elapsed Time Between the Instant a Deviation Occurs and the Corrective Action First Occurs.  a) Control Lag  b) Process Lag  c) Control Parameter Range  : Control Mode, the Output of the Control Element Changes at a Fixed Rate when the Error Exceeds the Neutral Zone.  a) Multiposition  b) Single Speed  c) Two - Postion  d) Multi Speed  : Control Mode Also Refered as Ress. Action.  a) Proportional  c) Derivative  d) Two - Position  : Control Mode Also Reffered as Anticipatory Control.  a) Derivative  c) Proportional - Derivative  d) Integral  d) Integral  c) Proportional - Integral  d) Integral  c) Proportional - Integral  i) Proportional - Integral  c) Proportional - Integral  d) Intergral  c) Source  d) Compressor  psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100  b) 100 - 125  c) 125 - 170  d) 150 - 200  s) Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity  c) Conductometric Method  . Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.  a) Flame Ionization  c) Thermal Conductivity  d) Galvanometric	1	: Control Mode, the Specific	Output of Controller is Not Uniquely
c) Floating d) None of These  The Elapsed Time Between the Instant a Deviation Occurs and the Corrective Action First Occurs.  a) Control Lag b) Process Lag c) Control Parameter Range d) Dead Time  3		5	1.0
		·	· · · · · · · · · · · · · · · · · · ·
Corrective Action First Occurs.  a) Control Lag c) Control Parameter Range d) Dead Time  Control Mode, the Output of the Control Element Changes at a Fixed Rate when the Error Exceeds the Neutral Zone.  a) Multiposition b) Single Speed c) Two - Postion d) Multi Speed  C) Two - Postion d) Multi Speed  C) Derivative Derivative c) Proportional - Derivative d) Two - Position c) Proportional - Derivative d) Integral c) Derivative c) Proportional - Derivative d) Integral fin Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero. a) Proportioal - Integral d) Integral c) Proportional - Integral d) Compressor a) For 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity b) Flame Ionization c) Conductometric Method Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	2	, ,	
c) Control Parameter Range d) Dead Time  : Control Mode, the Output of the Control Element Changes at a Fixed Rate when the Error Exceeds the Neutral Zone.  a) Multiposition b) Single Speed  c) Two - Postion d) Multi Speed  : Control Mode Also Remered as Research Action.  a) Proportional b) Integral  c) Derivative d) Two - Position  : Control Mode Also Reffered as Anticipatory Control.  a) Derivative b) Proportional - Integral  c) Proportional - Derivative d) Integral  for In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero.  a) Proportial b) Derivative  c) Proportional - Integral d) Integral  : Criteria not Important while Designing Instrument Air System (IAS).  a) pH b) Sizing  c) Source d) Compressor  a) 50 - 100 b) 100 - 125  c) 125 - 170 d) 150 - 200  Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Flame Ionization b) Pramagnetic Susceptibility	Zi	Corrective Action First Occurs.	ne instant a Deviation Occurs and the
		a) Control Lag	b) Process Lag
Fixed Rate when the Error Exceeds the Neutral Zone.  a) Multiposition b) Single Speed c) Two - Postion d) Multi Speed	_	c) Control Parameter Range	<b>d)</b> Dead Time
a) Multiposition b) Single Speed c) Two - Postion d) Multi Speed c: Control Mode Also Renered as Resc. Action. a) Proportional c) Derivative d) Two - Position  i: Control Mode Also Reffered as Anticipatory Control. a) Derivative b) Proportional - Integral c) Proportional - Derivative d) Integral for In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero. a) Proportial b) Derivative c) Proportional - Integral c) Criteria not Important while Designing Instrument Air System (IAS). a) pH b) Sizing c) Source d) Compressor  solution c) Source d) Compressor a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200 c) Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity c) Conductometric Method d) Potentiometric Method c) Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	3	: Control Mode, the Output	of the Control Element Changes at a
c) Two - Postion  —: Control Mode Also Renered as Research Action.  a) Proportional  c) Derivative  —: Control Mode Also Reffered as Anticipatory Control.  a) Derivative  —: Control Mode Also Reffered as Anticipatory Control.  a) Derivative  b) Proportional - Integral  c) Proportional - Derivative  d) Integral  for In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero.  a) Proportial  b) Derivative  c) Proportional - Integral  d) Intergral  :: Criteria not Important while Designing Instrument Air System (IAS).  a) pH  b) Sizing  c) Source  psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100  b) 100 - 125  c) 125 - 170  d) 150 - 200  [Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity  b) Flame Ionization  c) Conductometric Method  —: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.  a) Flame Ionization  b) Pramagnetic Susceptibility			
a) Proportional c) Derivative d) Two - Position  : Control Mode Also Reffered as Anticipatory Control. a) Derivative b) Proportional - Integral c) Proportional - Derivative d) Integral 6 In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero. a) Proportial b) Derivative c) Proportional - Integral d) Integral : Criteria not Important while Designing Instrument Air System (IAS). a) pH b) Sizing c) Source psig: Pressure Level of Liquid Piston Compressor. a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity c) Conductometric Method : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	4		
c) Derivative  d) Two - Position  : Control Mode Also Reffered as Anticipatory Control.  a) Derivative  b) Proportional - Integral  c) Proportional - Derivative  d) Integral  In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero.  a) Proportial  b) Derivative  c) Proportional - Integral  d) Intergral  : Criteria not Important while Designing Instrument Air System (IAS).  a) pH  b) Sizing  c) Source  psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100  psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100  c) 125 - 170  d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity  b) Flame Ionization  c) Conductometric Method  : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.  a) Flame Ionization  b) Pramagnetic Susceptibility	4		
<ul> <li></li></ul>			, .
a) Derivative b) Proportional - Integral c) Proportional - Derivative d) Integral In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero. a) Proportial b) Derivative c) Proportional - Integral d) Intergral: Criteria not Important while Designing Instrument Air System (IAS). a) pH b) Sizing c) Source d) Compressor psig: Pressure Level of Liquid Piston Compressor. a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  g: Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity c) Conductometric Method: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	_	•	•
c) Proportional - Derivative d) Integral  In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero.  a) Proportial b) Derivative c) Proportional - Integral d) Integral: Criteria not Important while Designing Instrument Air System (IAS).  a) pH b) Sizing c) Source d) Compressor psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200: Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	5		
In Control Mode, If the Error is Zero, the Output Stays Fixed at a Value Equal to what It was when the Error went to Zero.  a) Proportial b) Derivative c) Proportional - Integral d) Intergral: Criteria not Important while Designing Instrument Air System (IAS).  a) pH b) Sizing c) Source d) Compressor psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  9: Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility			b) Proportional - Integral
Value Equal to what It was when the Error went to Zero.  a) Proportial  b) Derivative  c) Proportional - Integral  : Criteria not Important while Designing Instrument Air System (IAS).  a) pH  b) Sizing  c) Source  psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100  b) 100 - 125  c) 125 - 170  d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity  b) Flame Ionization  c) Conductometric Method  : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.  a) Flame Ionization  b) Pramagnetic Susceptibility	_		
a) Proportial b) Derivative c) Proportional - Integral d) Intergral c: Criteria not Important while Designing Instrument Air System (IAS). a) pH b) Sizing c) Source d) Compressor a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity b) Flame Ionization c) Conductometric Method c) Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	0		
<ul> <li>Criteria not Important while Designing Instrument Air System (IAS).</li> <li>a) pH</li> <li>b) Sizing</li> <li>c) Source</li> <li>d) Compressor</li> <li>psig: Pressure Level of Liquid Piston Compressor.</li> <li>a) 50 - 100</li> <li>b) 100 - 125</li> <li>c) 125 - 170</li> <li>d) 150 - 200</li> <li>Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.</li> <li>a) Thermal Conductivity</li> <li>b) Flame Ionization</li> <li>c) Conductometric Method</li> <li>Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.</li> <li>a) Flame Ionization</li> <li>b) Pramagnetic Susceptibility</li> </ul>			
a) pH b) Sizing c) Source d) Compressor  psig: Pressure Level of Liquid Piston Compressor. a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility		c) Proportional - Integral	d) Intergral
a) pH b) Sizing c) Source d) Compressor  psig: Pressure Level of Liquid Piston Compressor. a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	7	: Criteria not Important while I	Designing Instrument Air System (IAS).
psig: Pressure Level of Liquid Piston Compressor.  a) 50 - 100 b) 100 - 125 c) 125 - 170 d) 150 - 200  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States. a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility			
<ul> <li>a) 50 - 100</li> <li>b) 100 - 125</li> <li>c) 125 - 170</li> <li>d) 150 - 200</li> <li>gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.</li> <li>a) Thermal Conductivity</li> <li>b) Flame Ionization</li> <li>c) Conductometric Method</li> <li>d) Potentiometric Method</li> <li>gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.</li> <li>a) Flame Ionization</li> <li>b) Pramagnetic Susceptibility</li> </ul>		r ·	
c) 125 - 170  : Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.  a) Thermal Conductivity  b) Flame Ionization  c) Conductometric Method  : Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.  a) Flame Ionization  b) Pramagnetic Susceptibility	8		
<ul> <li>Gas Analyzer Works on the Principle of No Interaction of Intramolecular Energy States.</li> <li>a) Thermal Conductivity</li> <li>b) Flame Ionization</li> <li>c) Conductometric Method</li> <li>d) Potentiometric Method</li> <li>Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.</li> <li>a) Flame Ionization</li> <li>b) Pramagnetic Susceptibility</li> </ul>		•	<b>b)</b> 100 - 125
Intramolecular Energy States.  a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method C: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. b) Pramagnetic Susceptibility		•	•
a) Thermal Conductivity b) Flame Ionization c) Conductometric Method d) Potentiometric Method C: Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States. a) Flame Ionization b) Pramagnetic Susceptibility	9		the Principle of No Interaction of
c) Conductometric Method d) Potentiometric Method  : Gas Analyzer Works on the Principle of Excitation of Intramolecular  Energy States. a) Flame Ionization b) Pramagnetic Susceptibility			i
<ul> <li>Gas Analyzer Works on the Principle of Excitation of Intramolecular Energy States.</li> <li>a) Flame Ionization</li> <li>b) Pramagnetic Susceptibility</li> </ul>		a) Thermal Conductivity	<b>b)</b> Flame Ionization
Energy States.  a) Flame Ionization  b) Pramagnetic Susceptibility			
a) Flame Ionization b) Pramagnetic Susceptibility	10		rinciple of Excitation of Intramolecular
, , ,		<u> </u>	
c) Thermal Conductivity d) Galvanometric		•	
		c) Thermal Conductivity	d) Galvanometric

Que 2	Sho	ort Questions (Attempt any TEN)	[20]
1	Exp	Explain Briefly Process Equation.	
2		Define: Transient and Process Lag.	
3	The Temperature has a Range 400 to 540 K. Set Point 484 K. Find the Percent of Span Error when the Temperature is 479 K.		
4	Enl	Enlist Characteristics of Proportional Control Mode.	
5	Wh	What is Continuous Control Mode?	
6	Wh	What are the Characteristics of Proportional - Integral Control Mode?	
7	Enl	Enlist the Characteristics of Air Used in IAS.	
8	Wh	What are the Important Factors for Designing of IAS.	
9	Exp	Explain the Need of IAS.	
10	Wha	What is the Importance of Industrial Gas Analyzer?	
11	Give	Give Classificaiton of Gas Analyzer.	
12	Explain Principle of Infrared Gas Analyzer.		
Que 3	[A]	What is Controller? Explain Two - Position Control Mode. What is the Importance of Neutral Zone?	[05]
	[B]	Write a Note on Single Speed Floating Control Mode.	[05]
	OR		
	[C]	What is Discontinuous Controller? Explain Multi Position Control Mode.	[05]
	[D]	A Liquid Level Control System Linearly converts a Displacement of 2 - 3 meters into a 4 - 20 mA Control Signal. A Relay serves as the Two - Position Controller to Open or Close an Inlet Valve. The Relay Closes at 12 mA and Opens at 10 mA. Find  i) The Relation between Displacement Level and Current, and ii) The Neutral Zone in meters.	[05]
Que 4	[A]	Write a Note on Proportional Control Mode.  Give an Account of Proportional - Integral - Derivative (PID) Control	[ <b>0</b> 5]
	[13]	Mode.	լսել
	OR		
	[C]	Explain Integral Control Mode.	[05]
	[D]	Discuss Proportional - Derivative (PD) Control Mode.	[CE]
Que 5	[A]	Wrtie a Note on Reciprocating Type Compressor.	[05]
	[B]	What is Dryer? Explain Refrigeration Type Dryer.	[05]
		OR	
	[C]	Explain Sliding Vane Rotary Compressor.	<b>[0</b> 5]
	[D]	Give an Account of Compressor Cooling.	[05]
Que 6	[A]	Write a Note on Paramagnetic Oxygen Analyzer.	[10]
		OR	
	[B]	Discuss Magnetic Wind Instruments.	[10]
		<del>-</del> x-	

#### **SARDAR PATEL UNIVERSITY V.V.NAGAR**

B.Sc.(V<sup>th</sup> SEM.) INSTRUMENTATION(V) NOVEMBER-2017 EXAMINATION CONTROL SYSTEM COMPONENTS SUB.CODE-US05CINV03

DATE:-11/11/2017.	Saturday	TIME:- 10:00 am	to 1:00 nm
JM   L - 11   11   201   .	, <b></b>	HIVIL. TO.OU alli	IO TIOU DIII

MARKS-70

Q-1	Choose correct answer		[10]
1.	Which of the following is not a timing relay?		
	(A) OFF-delay relay	(C) On-delay relay	
	(B) Latching relay	(D) none of above	
2.	Accelerometer is used for measuremen	nt of	
	(A) velocity	(C) acceleration	
	(B) position	(D) none of above	
3.	Contactors are proffered in application	involving	
	(A) Relay logic circuit	(C) logic power application	
	(B) High switching frequency	(D) none of above	
4.	Which of the valve is used for large flo	ws with low pressure drops?	
	(A) Diaphragm valve	(C) Butterfly valve	
	(B) Globe valve	(D) none of above	
5.	Which of the following is an advantage	of stepper motor?	
	(A) High accuracy	(C) Full torque at standstill	
	(B) Open loop control possible	(D) all of above	
6.	Valve is proffered in control	ing gas flow.	
	(A) Butterfly	(C) Diaphragm	
	(B) Needle	(D) none of above	
7.	Diaphragm and spring type actuators a	re example of actuators.	
	(A) Hydraulic	(C) Electro mechanical	
	(B) Pneumatic	(D) none of above	
8.	The dropped level of current after the	relay is energized is called	•
	(A) sealed current	(C) Drop out current	
	(B) saturation current	(D) none of above	
9.	In velocity servos is used in fee	dback line.	
	(A) Position sensor	(C) Accelerometer.	
	(B) Tachometer	(D) none of above	
10.	Which of the following is not an advan	tage of solid state relay?	
	(A) Finite on-off resistance	(C) Long life	
	(B) Higher switching speed	(D) none of above	
Q-2	Short answer type question. (any ten		[20]
1.	Explain brief : Relay logic	•	[1
2.	Briefly explain diaphragm valve.		
3.	List advantage and disadvantage of ste	pper motor.	
4.	Draw block diagram of acceleration se		
5.	Briefly explain Linear output actuator.		
6.	List features of single port valve.	•	
7.	Draw diagram of relays as AND and O	R logic.	
8.	List requirements of servo amplifiers.		
9.	Briefly explain potentiometer as a pos	tion sensor.	
10.	Briefly explain term: Rangeability.		
11.	Enlist advantage and limitations of soli	d state relay.	
12.	State features of Butterfly valve.	,	
O 3(V)	Write a detailed note on electromecha	anical relay	[10]

OR

Q.3(A)	Write a detailed note on solid state relay.	[06]
Q.3(B)	Explain working of magnetic motor starter with necessary diagram.	[04]
Q.4(A)	Explain the working of synchro torque transmitter.	[06]
Q.4(B)	Write a note on AC rate generator.	[04]
	OR	
Q.4(A)	Explain the working of basic servo system.	[06]
Q.4(B)	Discuss variable reluctance type stepper motor.	[04]
Q.5(A)	Explain flow rate V/s flow characteristics.	[05]
Q.5(B)	Write a note on solenoid actuators.	[05]
	OR	
Q.5(A)	Explain double port globe valve in detail.	[05]
Q.5(B)	Explain construction of diaphragm valves with necessary diagram.	[05]
Q.6(A)	Explain spring and diaphragm type pneumatic actuators in detail with necessary diagram.	[10]
	OR	
Q.6(A)	Write a note on hydraulic actuators.	[06]
Q.6(B)	Explain piston type actuators in detail.	[04]



(45 & A-15)

(3)

(4)

(5)

SEAT No.

No. of Printed pages: 2

### SARDAR PATEL UNIVERSITY Vallabh Vidyanagar

## Examination : B. Sc. Instrumentation (Vocational) → V S ► M US05CINV04 – Analytical Instrumentation

		Monday, 13 <sup>th</sup> November, 2017, 10:00 am – 1:00 pm Total Marks	s: 70
		Note: The figures to the right indicate maximum marks.	
Q-1.	(1)	Multiple Choice Questions- Standard buffer tablet of pH is available in the market.  (a) 7 (b) 15 (c) 2 (d) 8	[10]
	(2)	pH value is dependent on	
	(3)	If identical solutions are placed inside & outside the bulb of glass electrode, there exists an EMF of few mV is called as  (a) symmetric potential (b) asymmetry potential (c) equilibrium (d) none	*:
	(4)	The term "Plug" is used in process.  (a) detection (b) sample injection (c) heating (d) none	
	(5)	Response time of any detector should be (a) very High (b) very Low (c) high (d) none	
	(6)	The name of chromatography is defined from its  (a) two phases (b) stationary phase (c) three phases (d) mobile phase	
	(7)	TCD is sample technique.  (a) destructive (b) non destructive (c) volatile (d) none	
	(8)	If the molecular weight is < 2000, method selection for liquid chromatography is (a) Ionic/nonionic (b) soluble/insoluble (c) high/low (d) aqueous/non aqueous	
	(9)	RI detector depends on	
	(10)	Analysis of pharmaceutical products is carried out better by detector.  (a) Refractive Index (b) Fluoresence (c) Conductivity (d) Thermal	
Q-2.		Short answer type (attempt any ten)	[20]
	(1)	Why pH 7 is considered as neutral?	
	(2)	Enlist different types of pH meter.	

What should be the property of carrier gas in GC?

What are advantages of helical tube column?

Draw neat labeled diagram of Combination electrode.

	(6)	What do understand by Paper Chromatography?	
	(7)	Enlist drawbacks of large diameter column?	
	(8)	What are precautions for sample injection in Chromatography?	
	(9)	Draw neat diagram of Syringe Injector for Liquid Chromatography.	
	(10)	What are limitations of Laser detectors?	
	(11)	Enlist requirements for detectors.	
	(12)	Enlist different types of Laser detectors.	
Q-3.	(a)	Explain Null type pH meter in length.	[6]
	(b)	Give the basic principle of pH measurement and write the Nernst's equation with	[4]
		appropriate interpretation.  OR	
Q-3.	(a)	Draw labeled diagram of reference electrode and explain it in detail.	[6]
	(b)	Define Buffer solution. State precautions to be taken to handle it.	[4]
Q-4.	(a)	Draw the block diagram of Gas chromatography (GC) and explain it.	[5]
	(p)	Discuss Temperature control circuit of Oven in Gas Chromatography.  OR	[5]
Q-4.	(a)	Discuss Flame ionization detector (FID) used in GC and write its limitations.	[5]
	(p)	Explain working principle of Electron Capture Detector (ECD).	[5]
Q-5.	(a)	Explain HPLC system with flow measurement and control.	[6]
	(p)	Discuss Reciprocating Piston pump used in LC.  OR	[4]
Q-5.	(a)	What do you mean by Gradient Elution? Discuss its mode and list sequence of program entries.	[6]
	(b)	Explain Syringe type pump.	[4]
Q-6.	(a)	Describe working of Mass detector.	[5]
	(b)	Explain Fluorescence Detector.  OR	[5]
Q-6.	(a)	Discuss UV absorbance detector.	[5]
	(b)	Explain working of Refractive Index detector.	[5]



5'C

No. of Printed Pages: 02

#### SARDAR PATEL UNIVERSITY V.V.NAGAR

[40]

B.Sc.(V<sup>th</sup> SEM.) INSTRUMENTATION (VOC.) 15<sup>th</sup> NOVEMBER-2017 EXAMINATION

## 8-BIT MICROPROCESSOR PROGRAMMING AND APPLICATION-1 SUB.CODE-US05CINV05

<b>Q-1</b> 1.	Choose correct answer The content of accumulator is A5 H, after e	xecution of CMA instruction it becomes	[10]
	 (A) 55 H	(C) A5 H	
	(A) 55 H (B) AA H	(D) none of above	
2.	Flag is affected during data transfer	• •	
۷.	(A) Carry	(C) Zero	
	(B) Sign	(D) none of above	
3.	In 8085 up SUB instruction is	, ,	
٥.	(A) 1	(C) 3	
	(B) 4	(D) none of above	
4.	The address buses of 8085 µp contain		
٦,	(A) Four	(C) Sixteen.	
	(B) Eight	(D) none of above	
5.	is machine control instruction.		
0.	(A) RET	(C) JNC	
	(B) NOP	(D) none of above	
6.	CALL and RET are type instruction.	( )	
0.	(A) Advance	(C) Logical	
	(B) Branch	(D) none of above	
7.	JNC is byte instruction.	,	
	(A) One	(C) Three	
	(B) Two	(D) none of above	
8.	Which of following is two bytes instruction?	?	
	(A) MVI B,0A H	(C) JMP 2001 H	
	(B) MOV B,A	(D) none of above	
9.	Following instructions transfer the data from	m memory to microprocessor.	
	(A) STA 16-bit	(C) DCX, B	
	(B) LDAX, B	(D) none of above	
10.	is the 16-bit register in 8085 μp.		
	(A) Stack pointer	(C) Accumulator	
	(B) Flag register	(D) none of above	
Q-2	Short answer type question. (any ten)		[20]
1.	Define static and dynamic debugging.		
2.	State any 2 byte and 3 byte instructions.		
3.	State characteristics of logical instruction.		
4.	State function of ALU.		
5.	Define programme and software,		
6.	Define looping and counting technique.		
7.	Differentiate between DCR and DCX instr	uction.	
8.	Why data bus is bi-directional in 8085 μp?		
9.	List pins of interrupt control section of 808		
10.	State meaning of RRC and RLC with illust	tration.	
11.	State different addressing mode of 8085 µ	JD QL	

12.	What do you mean by NOP instruction in 80885up?	-
Q.3	Draw the architectural block diagram of 8085 µp and discuss function of each section of it.	[10]
	OR	
Q.3	Discuss the concept of bus timing and de-multiplexing the bus AD <sub>0</sub> - AD <sub>7</sub>	[10]
Q.4	Describe classification of instruction according to operation perform and according to word size.	[10]
	OR	
Q.4(A)	Explain the method of writing, assembling and executing a simple program in 8085 µp.	[07]
Q.4(B)	Differentiate between op-code and operand.	[03]
Q.5(A)	Discuss different logical instructions with suitable illustration.	[06]
Q.5(B)	Write a programme to load two numbers in two registers now subtract one number from other such that carry flag will set and display the answer at output port.  OR	[04]
Q.5(A)	Explain different data transfer instructions with suitable example.	[06]
Q.5(B)	Write a programme: to load 7C H and 3B H in register C and D respectively. Now increment content of C than add both the number and display the sum at output port.	[04]
Q.6(A) Q.6(B)	Discuss different 16 –bit arithmetic instructions with illustration. Write a program to load 9B H and A7 H in register B and C respectively. Now add both the numbers, if the sum is greater than FF H display 01 at output port 0 otherwise display the sum.	[05] [05]
	OR .	
Q.6(A)	Describe conditional and un-conditional jump instructions giving suitable examples.	[05]
Q.6(B)	Discuss method of static and dynamic debugging a programme.	[05]

SEAT No.\_\_\_\_

#### [33]

#### SARDARPATEL UNIVERSITY V.V.NAGAR

T.YB.Sc. Sem-V EXAMINATION

SUB. CODE:- USO5CINV06

Signal Conditioning & Communication

DATE:-17/11/2017 TIME:-10:00 am to 1:00 pm

MARKS-70

Q-1	Choose correct answer		[10]
1.	Tristate switch has	output states.	(10)
	(A) 1	(C) 3	
	(B) 2	(D) None of these	
2.	Registers are made of		
	(A) Flip-flop	(C) Resistor	
	(B) Capacitor	(D) None of these	
3.	In ring counter data is		
	(A) Rotated	(C) Not Rotated	
	(B) Steady	(D) None of these	
4.	Flash type A/D Converter is		
	(A) Fastest	(C) Slow	
	(B) Very slow	(D) None of these	
5.	Successive approximation type	A/D Converter converter.	
	(A) Fast	(C) Slowest	
	(B) Very fast	(D) None of these	
6.	In V/F type A/D converter		
	(A) Frequency	(C) Time	
	(B) Voltage	(D) None of these	
7.	Modem is also called		
	(A) DCE	(C) Capacitor	
	(B) DTE	(D) None of these	
8.	The schmitt trigger has U.T.P. a		
	(A) L.T.P.	(C) No Point	
	(B) M.T.P.	(D) None of these	
9.	Schering Bridge is used for the	measurement of unknown	
	(A) Resistance	(C) Inductance	
	(B) Capacitance	(D) None of these	
10.	Maxwell bridge is used for mea		
	(A) Resistance	(C) Inductance	
	(B) Capacitance	(D) None of these	
Q-2	Short answer type question. (a		[20]
1.	Explain FSK with necessary dia		[]
2.	Explain Asynchronous & Synch		
3.	Explain low speed Modem ope	·	
4.	Draw the operation diagram of	successive approximation converter.	
5.	Draw the figure of 3-bit paralle	comparator type A/D converter.	
6.	Draw the inter connecting circuit of samples & hold circuits and A/D converter.		
7.	Draw the circuit of weighted resistor type -4 bit D/A converter.		
8.	Write a truth - table tristate sw		
9.	Explain linearity, sensitivity and	d accuracy.	
10.	Draw the circuit diagram for ha		
11.	Draw the circuit diagram for M		
12.	State limitations of whetstone	<del>-</del>	

Q.3	Explain working of R-2R ladder type D/A converter.	
0.0	OR	
Q.3	Give an account of Tristate switch.	
Q.4	Give an account of Schmitt trigger as an interface circuit.	
	OR	
Q.4	Give an account of Modems & interfaces.	
Q.5	Explain in detail voltage to frequency type A/D converter.	
	OR	
Q.5	Explain in detail working of counter type A/D converter.	
Q.6	Give an account of Wheastone bridge.	
	OR	
Q.6	Give an account of Wein bridge.	
	X	

[458A23]

#### SARDAR PATEL UNIVERSITY

T. Y. B.Sc. (V Semester) Examination
TUESDAY, NOV. 7, 2017 Time: 10:00 a.m. to 1:00 p.m.
US05CMIC01(MICROBIOLOGY)
FUNDAMENTALS OF MOLECULAR BIOLOGY

		Maximum Marks: 70
Q.1.	Each question below gives a multiple appropriate one.	choice of answers. Choose the most [10]
1	Which histone is not a part of the nucleoson	in the control of the
	(a) H1	(b) H2A tests had believed at the first
	(c) H2B	(d) H3
2	Overall conclusion of Hershey & Chase exp	
	(a) DNA was responsible for heredity	(b) Protein & DNA was responsible for heredity
	(c) Ratio of Adenine to thymine was always the same	(d) Phage DNA was similar to bacterial DNA
3	Which of the following enzyme is not involved	ved in DNA replication?
	(a) DNA Polymerase	(b)Primase
1:1	(c) DNA Helicase	(d) Reverse Transcriptase
4	Which of the following eukaryotic DNA pol	ymerase has intrinsic Primase activity
	(a) DNA pol β	(b) DNA Pol σ
	(c) DNA Pol δ	(d) DNA Pola
5	What is the function of SSB protein in DNA	replication
14 <sup>25</sup> 1	(a) They unwind a DNA double helix to form a separate single strands	(b) They bind to a single stranded DNA & assist in the reformation of double
6	(c) They bind to single stranded DNA and prevent the single strands from reforming base pairs.  Genetic code was deciphered by	stranded DNA (d) None of these
	(a) Marshall Nirenberg	(b) Hargobind Khorana
	(c) James D Watson	(d) Francis C. Crick
7	The first m RNA codon to specify an amino	
in William Paragraphia	(a) UAA	(b) AUG
	(c) UAG	(o) NOG
8	Translation in prokaryotes begins by the for the:	(d) TAC mation of 30 S initiation complex between
	(a) 30 S ribosomal subunit, mRNA, IF & N-fmet-t-RNA	(b) 30 S ribosomal subunit, tRNA, &IF
	(c) 30 S ribosomal subunit, & mRNA only	(d) 30 S ribosomal subunit, mRNA & IF
9	The lac operon is a cluster of:	
	(a) Three structural gene	(b)Three structural genes &their promoter
	(c) A regulator gene, an operator & a promoter	(d)A regulator gene, an operator, a promoter & three structural genes

10	The S	Sigma is a component of:	
	(a) D	NA Polymerase (b) DNA Ligase	
	(c) R1	NA Polymerase (d) Endonuclease	
Q. 2	Short	t Questions (Attempt any TEN)	[20]
1	Distir	nguish between B and Z form of DNA	
. 2		the structure of Adenine and Cytosine	
3	What	do you mean by the terms Nucleosome and Solenoid?	
4		powers the action of helicase and what is the function of SSB protein?	
5	What	t is θ model of replication?	N.
6 7		t is Ori C and what is its significance in DNA replication?	
8	What	t is the role of gag, pol and env genes in retroviruses?	44.2
9	Uau	PNIA notymerase recognizes the promoter region of DNA template?	•
10	Enlis	t the components and their role involved in initiation of protein synthesis.	af )
11	Enlis	t the features of Genetic Code.	
12	Draw	va neat labeled structure of 70 S ribosome and what is its function?	
		en grave sedena egis di samura di samura di samura di samura di samura di samura di samura di samura di samura	10#1
Q. 3	[A] [B]	Explain the Watson and Crick Model of DNA  Explain with the help of an experimental evidence that the use of radioisoto proved DNA is a genetic material	[05] pes [05]
		OR	
Q.3	[A]	Explain in detail the organization of Eukaryotic chromosome	[06
•	[D]	Write an extensive note on t-RNA	[04
		and Automatical Control of the Contr	
Q. 4	[A]	Write a note on Rolling Circle model of Chromosome replication	[05]
	[B]	Explain the Replication of Eukaryotic DNA	[05
0.4	[A] :	OR  DNA replication is Semi-conservative – Justify	[04
Q.4.	[A] [B]	Explain in detail Initiation and elongation of DNA replication in Prokaryotes.	[06
Q. 5		Discuss the regulation of gene expression (Lac operon) in detail.	[10]
Q.5.		Explain in detail the molecular mechanism of Transcription	[10
Q. 6	[A] [B]	Write a note on Wobble hypothesis.  Explain post translational modification of proteins  OR	100
Q.6.	[A] [B]	Explain Activation and Initiation and elongation of Protein synthesis Explain in detail how genetic code was deciphered	[06 [04



grand states and the grand agreements of the con-

(a) control components are near 1 (d);
 (a) control the control of the control A(b);

[55/AZ6]

No. of Printed Pages : 2

#### SARDAR PATEL UNIVERSITY B.Sc.(5<sup>th</sup> Semester) EXAMINATION 2017 Thursday, November 9<sup>th</sup>, 2017 10:00 a.m. TO 1:00 a.m. SUBJECT: MICROBIOLOGY US05CMIC02

	(Bioinstrumentation)	*** <b>*</b> **
NY	TOTAL MAR  (1) All the questions are compulsory.	KS: 70
Note:	<ul><li>(1) All the questions are compulsory.</li><li>(2) Figures on the right indicate marks.</li></ul>	:
Q-1	Select the correct answer for each question from the option given below	[10]
1.	Which of the following spectroscopic technique based on the principle of	41
	"Bond Vibration"?	*
	(A) UV-Visible (B) Atomic absorption (C) Infrared (D) Nephlometry	
2.	Which of the following is NOT a part of monochromatic system?	11.5
	(A)Entrance slit (B) Prizm (C) Exit slit (D) Sample Cell	11
3.	Which of the following act as a cross linker in PAGE?  (A)TEMED (B) Bis acrylamide (C) Ammonium persu; phate (D) Acrylamide	
4.	Which centrifugal technique separates analytes with same size but different	
***	density?	
ya s	(A) Differential (B) Isopycnic (C) Rate zonal (D) None of these	
5.	Which of the following chromatographic technique can be used for the	
- 1 S	determination of relative molecular size of proteins?	
	(A)Adsorption (B) Ion exchange (C) Gel permeation (D) Affinity	5 a y
6.	Which of the following detector is used in Gas Liquid Chromatography to	,,
	detect chlorinated compounds?	
	(A)Nitrogen phosphorous (B) Electron capture	
	(C) Thermal ionization (D) Flame ionization	· ! .
7.	Which of the following tool is an example of homology and similarity?	
	(A)PROSPECT (B) EMBOSS (C) RAMSOL (D) BLAST: 10 10 10 10 10 10 10 10 10 10 10 10 10	
8.	An immobilized biological material which can specifically interact with an	
	analyte and produce physical, chemical or electrical signals that can be	
es i	measured is called	
_	(A) Transducer (B) Biosensor (C) Thermistor (D) Transponder	
9.	Which of the following technique is 'denaturing PAGE' with discontinuou	S
	buffer system?  (A) Notice (P) Pulse Field (C) SDS (D) Aggregation (C) SDS (D)	
40	(A) Native (b) Fulse Field (C) SDS (D) Agaiose	
10.		) Selection
odi. National	system:	
• •	(A)Pump (B) Photomultiplier (C) Injector (D) Chopper	
	(I) CP	COT
}	som vid sagadigad för att som att stamfarga av sist i som gjör	
\.\frac{1}{2}	THE WAR IN THE CAMPAGNATE OF THE MARKET OF A COMPAGNATION OF THE PARTY.	

and the first of the register to extend a least distribution of the energy team of the extent of gainst.

<b>Q-2</b> [1]	Write	about sequence of events taking place in the flame during flame emission	[20]
[2] [3] [4]	Write What: Write	about quodrupole mass analyser with diagram in mass spectroscopy.  is nephlometry? How can it be used to analyse suspensions?  about the role of (1) Ammonium persulphate and (2) TEMED in ophoresis.	·
[5] [6]	Write Enlist centrit	about principle of RNA denaturation during electrophoresis. ideal characteristic of gradient material used in density gradient fugation.	
[7] [8] [9]	Define What	about specific and non specific elution in affinity chromatography.  e (1) Partition Coefficient (2) Theorytical Plate do you mean by column bleeding in Gas Liquid Chromatography? How can aken care of during separation of analytes?	
[10] [11] [12]	Write Write Defin	the working of luminescent biosensor using luciferase enzyme.  about the aim of bioinformatics  e the terms isotopes and radioisotopes and give one example of each.	•
Q-3 (		Discuss principle, sample preparation and applications of Infra Red	[06]
(	(B)	What are atomizers? Discuss the role of atomizers in Atomic Absorption Spectroscopy.  OR	[04]
Q-3		Give detail account on principle, working and applications of UV-Vis Spectroscopy.	[10]
		The state of the part of the state of the st	F1 03
Q-4		Give detail account on- Poly Acrylamide Gel Electrophoresis (PAGE).	[10]
Q-4	17	Discuss in detail about different types of density gradient centrifugations in terms of its principle, working and applications.	[10]
		i de la compresión de l	•
Q-5	(A)	Discuss about pumping system, sample injection and applications of High Performance Liquid Chromatography (HPLC).	
	(B)	What are the criteria for selection of ion exchanger in ion exchange chromatography? Discuss about various steps involved in the ion exchange chromatography.  OR	[]
Q-5		Write a note on- Gel Permeation Chromatography	[06]
	(B)	Draw neat and labeled diagram of Gas Liquid Chromatography (GLC) and write about different types of detectors.	[04]
Q-6	(A)	Discuss about the scope of bioinformatics.	[05]
~ ~	(B)	Enlist different type of radioactive decay and discuss any one in detail.  OR	[05]
Q-6		Define biosensor and giving one example each explain different types of biosensors.	[]
****	*****	: 眼水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水水	<b>平平平平平</b>



[47/A27]

No. of Printed Pages : 3

#### SARDAR PATEL UNIVERSITY

B.Sc.- ISEM, NOVE-17 Examination Subject Code: USO5CMICO3

Day	:Sa	1/11/2017 turday ures on the right indicate marks.		Time:10.00a.m. to1.00 p.m. Total marks: 70	
Q.1		Answer the following Multiple compulsory)	Choice	Questions.(All questions are	10
	. 1	In which of the following process during nutrient transport?  (a) Facilitated diffusion.	1	ASHRINDBIN III-IS ONLY	
		(c) Binding Protein transfer.	7.3.1 N. S.	Group translocation.  All of the above.	
	2	Signal peptide contains cleavage aminoacid location?	site f	or peptidase enzyme at which	
		(a) 18 <sup>th</sup> aminoacid.		17 <sup>th</sup> aminoacid	
		(c) 20th aminoacid.	(d)	None of the above.	
	3	Polymyxins affects which of the fo	ollowina	hrocese 2	
		(a) Inhibitscellwall synthesis.		process ? Inhibits protein synthesis.	
		(c) Damages cell-membrane.		None of the above.	
	4	Which of the following is an exam	ple of s	emi-synthetic nenicilling	7
		(a) Amphicillin.	(b)	Amoxycillin.	
		(c) Oxacillin.		All of the above.	
	5	During germination of endospo		tivation process is triggered	
		(a) Heat	(b)	Low PH was and and torily	
		(c) By reducing agents		All of the above	
	6	Which of the following bond/s is protein?	/are ob	served in tertiary structure of	
		(a) hydrogen	(b)	ionic	
		(c) disulfide		All of these	

(P.TO.)

	7	Those enzymes which can catalyse same reaction but having p	
		different properties are known as	
		(a) Apoenzyme. (b) Coenzyme.	•
		(c) Holoenzyme. (d) Isoenzyme	
	8	Km values lies in the range ofmol/lit.	
		(a) $10^{-1}$ to $10^{-5}$ (b) $10^{-1}$ to $10^{-6}$	
		(c) $10^{-2}$ to $10^{-6}$ (d) $10^{-3}$ to $10^{-6}$	
	9	How many digits are there in each enzyme commission number?	
		(a) 3 (b) 4	
		(c) 6 (d) 8 amount of the second of the seco	
·	10	Which of the following is/are mechanism type of multisubstrate reactions?	
		(a) Ordered mechanism (b) Random mechanism	
		(c) Ping pong mechanism (d) All of these	
		and a support of the state of t	
		en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	
Q.2	144,1	Give short answers to the following questions. (Attempt Any Ten) 12	
	1	Write in brief importance of siderophores.	
	2	Give example of various components involved in group translocation process.	
	3	Give examples of types of Permeases.	
	4	Write few drawbacks of Streptomycin.	
	5	Write mode of action of Tetracycline.	
	6	Enlist types of Bacterial movement. Also give examples of each.	
	7	Write about temperature coefficient or Q10.	
	8	List out the bonds/interactions that maintain tertiary structure.	
	9 .	Define: Zymogens and allosteric enzyme.	
•	10	Draw the Lineweaver-Burk plot for uncompetitive and mixed inhibition.	
	11	What are the significance of double reciprocal plot?	
	12	What is covalent modification?	
		The state of the s	
		(m)	
	;		

Q.3		Write short notes on:-	
	(A)	Write note on-Na-K Pump.	05
	(B)	Siderophores.	05
	` ′	OR	00
Q.3		Write short notes on:-	
	(A)	Write short note on:- Signal peptides.	06
	(B)	Write short note on :- "Group translocation".	04
	(-)	The state the state of the stat	<del>-</del>
Q.4		Write short note on:-	
	(A)	Antibiotics causing damage to cell membrane.	05
	(B)		05
		OR	
Q.4	(A)	Write mode of action of following antimicrobial agents:-	06
		1. Penicillin. 2. Chloremphanicol. 3. Sulfonamides.	
	(B)	Ideal characteristics of Chemotherapeutic agents.	04
Q.5		Write short note on:-	
	(A)	Factors affecting enzyme action.	06
	(B)	Active site of Enzyme.	04
		OR	
Q.5		Write short note on:-	
	(A)	Mechanism of enzyme action.	05
	(B)	IUB system of Classification of enzyme.	05
Q.6		Derive Michaelis-Menten equation by using Briggs-Haldane assumption	10
٠,٠		and explain the significance of Km.	10
•		OR	
Q.6		Describe in detail about enzyme inhibition.	10
<b>-</b>		The second design of the property of the second sec	10
		***********	

.s.) are the control of a post of the control of the con a whereast to the state of the The second of th

No. of Printed Pages : 9 [47/A27]

SARDAR PATEL UNIVERSITY

B.Sc.-I.SEM, NOVE-17 Examination

Subject Code: USO5CMICO3

	Subject code.or	200011	12000		
	Subject Title:-Microbial ph	ysiolo	gy & Enzymolo	y Xp	
ay:Sat	./11/2017 Turday Tres on the right indicate marks.		Time:10.00a.n Total marks:	•	m.
.1 Iveitiun	Answer the following Multiple compulsory)	Choice	Questions.(All	questions	are 10
1 .	In which of the following process during nutrient transport?		Marie Carried Company		ient
	<ul><li>(a) Facilitated diffusion.</li><li>(c) Binding Protein transfer.</li></ul>	(b)	Group transloca All of the above		
2 (mpT	Signal peptide contains cleavage aminoacid location?	e site		nzyme at W	hich
	<ul> <li>(a) 18<sup>th</sup>aminoacid.</li> <li>(c) 20th aminoacid.</li> </ul>	(p)	17 <sup>th</sup> aminoacid None of the ab	ove.,	1 S
4	Polymyxins affects which of the (a) Inhibitscellwall synthesis. (c) Damages cell-membrane. Which of the following is an exa (a) Amphicillin. (c) Oxacillin.	(b) (d) mple of (b)	Inhibits protein None of the ab semi-synthetic p Amoxycillin.	ove. penicillin?	6 4 6 6 1
5	During germination of endos by	pore, ( (b) (d)	Low P <sup>H</sup> All of the abov	ess is trig ve tiary structu	
15	(c) distilline			CP.TR	0.)

	Those enzymes which can catalyse same reaction but having p different properties are known as  (a) Apoenzyme. (b) Coenzyme. (c) Holoenzyme. (d) Isoenzyme
<sub>:</sub> 8	Km values lies in the range ofmol/lit.
9	How many digits are there in each enzyme commission number?  (a) 3  (b) 4
10	Which of the following is/are mechanism type of multisubstrate reactions?  (a) Ordered mechanism (b) Random mechanism (c) Ping pong mechanism (d) All of these
).2 · ·····	Give short answers to the following questions. (Attempt Any Ten) 12
7 8 9 10	Write in brief importance of siderophores.  Give example of various components involved in group translocation process.  Give examples of types of Permeases.  Write few drawbacks of Streptomycin.  Write mode of action of Tetracycline.  Enlist types of Bacterial movement. Also give examples of each.  Write about temperature coefficient or Q <sub>10</sub> .  List out the bonds/interactions that maintain tertiary structure.  Define: Zymogens and allosteric enzyme.  Draw the Lineweaver-Burk plot for uncompetitive and mixed inhibition.  What are the significance of double reciprocal plot?  What is covalent modification?

Q.3		Write short notes on:-		
•	(A)	Write note on-Na-K Pump.		05
	(B)	Siderophores.	I	05
	. ,	OR		
Q.3		Write short notes on:-		
	(A)	Write short note on:- Signal peptides.		06
	(B)	Write short note on :- "Group translocation".		04
Q.4		Write short note on:-	•	
•	(A)	Antibiotics causing damage to cell membrane.		05
		Sporulation of Endospore.		05
	<b>\-</b> ,	OR	,	
Q.4	(A)	Write mode of action of following antimicrobial agents:- 1. Penicillin. 2. Chloremphanicol. 3. Sulfonamides.		06
	(B)	Ideal characteristics of Chemotherapeutic agents.		04
Q.5		Write short note on:-	d <b>#</b>	
	(A)	the state of the s		06
	(B)	_ <del>_</del> _		04
	` '	OR	i ,	•
Q.5		Write short note on:-	j	_
	(A)	Mechanism of enzyme action.		05
	(B)	A mile and a side of		05
Q.6		Derive Michaelis-Menten equation by using Briggs-Haldane and explain the significance of Km.	assumption	10
		OR		4.0
Q.6		Describe in detail about enzyme inhibition.	·	10
		*************************************	programme	

p.T.

*.		٠
1.1	e de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	
	en en en en en en en en en en en en en e	
:7	is the same of the	
	and with the stage of the stag	
Ayri,	en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la comp	
	· · · · · · · · · · · · · · · · · · ·	
QA.		
	Marie Committee Electronia Approximate Electronia (Assert	
\$ 11 -	territorio de la Companya de Marena de la Companya	
,		
	· · · · · · · · · · · · · · · · · · ·	
44	in the second of	. '
N. 1		
		ı
10	and the second of the second o	
%()	The state of the s	
the second	The second second security of the second second second second second second second second second second second	
ı		
		1 1
	and the second of the second o	
		ż

-0.3

|--|

No. of Printed Pages 2

No. of Pages 02

Sardar Patel University T.Y.B.Sc V<sup>th</sup>Sem Examination - 2017 Monday 13<sup>th</sup> Nov 2017 Subject: Microbiology Course Code US05CMIC04 (Immunology) Time: 10:00 am - 01:00 pm

.1.	Total Marks 70 Select the correct answer for each question from the option given below. [10]
1.	among the following is an antigen presenting cell.
	(a) dendritic cell (b) erythrocytes (c) basophils (d) eosinophils
2.	Acute phase proteins are produced by
	(a) spleen (b) thymus (c) liver (d) kidney
3.	Penicillin is an example of
	(a) antibody (b) CD (c) TLR (d) hapten
1.	Fas – FasL pathway is also known as pathway.
	(a) CD95 (b) CD8 (c) CD4 (d) CD28
5.	The phenomenon of immune exclusion is exhibited by
	(a) Ig G (b) IG M (c) Ig A (d) Ig E
6.	Post zone area in zone phenomenon is an area of
	(a) antigen excess (b) antibody excess (c) both in equal (d) none of these
7.	Widal test for typhoid is an example of agglutination test.
	(a) tube (b) slide (c) gel (d) passive
8.	In chemically induced tumors the TSTA on the membrane of tumor cells is
	·
• •	(a) tumor specific (b) virus specific (c) both (a) & (b) (d) none of these
9.	Type hypersensitivity is called cytolytic or cytotoxic reaction.
	(a) I (b) II (c) III (d) IV
10.	is a systemic autoimmune disease.
	(a) SLE (b) myasthenia gravis (c) Hashimoto's disease
	(d) autoimmune thrombocytopenia
	(1)

Q.2.	Write short answers of the following questions. (Any ten)	[20]
1.	Write the function of spleen and thymus.	
2.	What is the roll of NK cells in protecting the host?	
3.	What is the significance of toll like receptors?	
4.	Explain what is humoral and cell mediated immunity.	
5.	Write the structure and function of MHC class I molecule.	
6.	Explain naturally acquired active and passive immunity.	
7.	Write the passive agglutination test.	
8.	Differentiate between precipitation and agglutination reaction.	
9.	Write the importance of monoclonal antibodies.	
10.	Explain the terms neoantigens and sequestered antigens.	
11.	Define hypersensitivity and how is it classified?	
12.	Which are the different approaches attempted in immunotherapy of cancer?	
Q.3.	Write a detail note on cytokines.	[10]
	to the OR and the Control of the Con	[-~]
Q.3.(a)	Write a note on acute inflammatory response.	[05]
(b)	Define complement and explain the alternative complement pathway.	[05]
Q.4.	Discuss the activation of T – cells.	[10]
	OR	[-~]
Q.4.(a)	Define antigens and write a note on properties of antigens.	[06]
(b)	What are T independent antigens and write the T independent activation of B – cells.	[04]
Q.5.(a)	Explain the structure and functions of different types of immunoglobulin.	(06)
(b)	Write a note on general features of antigen antibody reactions.	(04)
` ,	OR	(04)
Q.5.(a)	Write a note on precipitation reactions in gel.	(06)
(b)	Explain the tenets of clonal selection theory.	(04)
Q.6.	Discuss the mechanism of allograft rejection reaction and mention the factors favoring the survival of allograft.	[10]
	OR	
Q.6.	Write an essay on immunodeficiency diseases.	[10]
-	y a subset.	[ r v ]

SHAT NO.

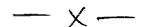
#### [41/A-16]

No. of Printed Pages: 02

# SARDAR PATEL UNIVERSITY TYBSc EXAMINATION SEMESTER-5 MICROBIOLOGY - US05CMIC05 Microbial Diversity & Ecology

Date: 15/11/2017 Time: 10:00 am to 01:00pm Wednesday Total marks: 70 N.B: Figures on the right indicate marks. Q.1 M.C.Q. (01 - mark each) 10 1 is used in brewing and baking industries. (a) Gibberella fujikuroi (b) Aspergillus oryzae Penicillium notatum (c) (d) ... Saccharomyces cerevisiae 2 is the genetic material of HIV (a) Two single(+sense)RNA (b) ds DNA (c) ds RNA (d) Two single(+sense)DNA Mycoplasma are placed in taxonomic class 3 Mollicutes and a second of the (b) **Tenericutes** (c) Gracilicutes (d) **Basidiomycetes** All of the following are examples of positive interactions among microbes except\_ Commensalism (a) (b) Mutualism (c) Synergism (d) Predation 5 are produced within the sac like structures in fungi (a) Chlamydospores (b) Sporangiospores (c) Conidiospores (d) Arthrospores are the infectious agents of potato spindle tuberdisease . 6 (a) Viruses Bacteria (b) (c) Viroids (d) Algae 7 are Gram Positive filamentous bacteria. Mycoplasma (a) 4 Actinomycetes (b) (c) Rickettsia (d) Chlamydia 8 Mycorrhizae is an example of\_ \_ relationship. (a) Commensalism (b) Amensalism Mutualism (c) (d) Neutralism 9 The causative agent of small pox is \_ (a) Variola virus essential in spatial of (b) heart Vaccinia virus (c) Hepatitis virus ं (d) HIV . . . is the intracellular absorptive structure developed by plant parasitic fungi. 10 (a) Appressorium Parenthosome (b) (c) Sclerotium (d) Haustorium

Q.2	2	Give short answers to the following questions. (Answers	wer Any Ten)	*	20
	1	What is an appressorium and infection peg in fungi?	PAY 1		
	2	What are micro and macroelements in the nutrition of fungi?	<b>3</b> 名		
	3 ,	What is conidiophore and conidiospore?	÷ **	apa H	
	4	Draw neat and labelled diagram of HIV	erite data Tanah ang tanggan ang tanggan ang tanggan ang tanggan ang tanggan ang tanggan ang tanggan ang tanggan ang tang		
	- 5	Give the examples of diseases caused by viroids and prions			
11.0	6	Enlist physical properties of Pox viruses.	graph Carrier galax	, NA	:13
	7	How does Rickettsia differ from Chlamydia?	was to the		
	8	Give names of four genera that metabolize sulfur compounds.			
	9	Describe the habitat of anoxygenic phototrophic bacteria.	<ul> <li>A second popular special della communitation of the communi</li></ul>		
	10	Define the terms synergism and syntrophism.	* * * * * * * * * * * * * * * * * * * *		
	11	What is rhizosphere effect?	A William Control of the Control of		
	12	Define the terms ecology and ecosystem.	gradient Wagen with		
			gs (4) — g		
Q.3	(a)	Explain the asexual reproduction in fungi.	er en en en en en en en en en en en en en	was a	lac
	(b)	Explain in short about the three steps plasmogamy, karyogamy			(06)
		in fungi.	y and meiosis in sexual repro	ouction	(04)
		OR	e and property and profession of the	Se Qu	<b>X</b>
Q.3	(a)	Explain the types of organic acids produced by fungi with suita	ble examples.	: : : : : : : : : : : : : : : : : : :	(04)
	(b)	Write a note on modes and mechanism of nutrition in fungi.		44	(06)
		en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	tin a constated while		: (00)
Q.4	(a)	Describe the structural genes of HIV.	en engage edit i		(06)
	(b)	Write about the resistance of HIV.	je njegovišta dije	4 11.5	(04)
		in the same of the		• *	(04)
Q.4	(a)	How does Variola and Vaccinia virus differ?	+ + 12	45	Incl
	(b)	Describe the morphology of pox virus with the help of labeled of	diagram.		(06)
					(04)
<b>Q.5</b>	(a)	Write notes on Mycoplasma.	en en en en en en en en en en en en en e		
	(b)	Write names of disease caused by Rickettsia spp and arthropod		7 :	(06)
			a paragraph and same		(04)
2.5	(a)	Describe characteristics of purple sulfur bacteria.	n de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co	1::1	
•	(b)	Write notes on characteristics of genera Spirocheates	e de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co		(06)
	. ,	on and actionates of Benefia SpiroLiteates		4.5 3.5 5.47	(04)
Չ.6		Justify that negative interactions act to maintain and	· ·	4	•
<b>√.</b> ∪		Justify that negative interactions act to maintain ecological bala	ince.		(10)
<b>).6</b>		Write an exhaustive note on commercial and the same of			N.C
4.U		Write an exhaustive note on commensalism and synergism inte	raction.		(10)



#### No. of Printed Pages : 02

#### [34 & A-15] SARDAR PATEL UNIVERSITY

# B.Sc. Examination Semester- 5 Microbiology – US05CMIC06

Fermentation Technology Date: 17/11/2017 Time: 10:00 AM to 1:00 PM Day: Friday Total marks: 70 N.B: Figures on the right indicate marks: Multiple Choice Questions. (01 - mark each) The second of the second Q.110 1 Process that produces Single Cell Protein as product is called (a) Enzyme production (b) Biomass production (C) Transformation process (d) None of the above A medium lacking growth factors is called (a) Enrichment medium (b) Minimal medium (C) Complete medium (d) Complex medium Which of the following compound(s) can provide more energy than Glucose on per weight basis? (a) Oil & fats (b) Hydrocarbons (C) Both (a) & (b) (d) None of the above Phenylacetic acid is a \_\_\_\_\_ which is used in Penicillin production. (a) Precursors (b) Inducers (C) Inhibitors (d) All of the above PEG is used to induce recombination in the state of the s Sexual cycle (b) Parasexual cycle (C) Protoplast fusion (d) None of the above In which of the following reactor, impellers are used? (a) CSTR (b) Fluidized bed reactor (C) Air lift fermenter (d) None of the above Oxygen electrodes measure \_\_\_\_\_ during the fermentation process. Partial pressure of dissolved oxygen (b) Dissolved oxygen concentration (C) Mole fraction of oxygen 1 gad (d) None of the above Which of the following is the most common method used in fermentation industries for air sterilization? and the search of the analysis for the sign of the Application of the .Use of heat (a) (b) "Use of chemical" which (C) Use of filters (d) None of the above An enhancement of closed batch fermentation process is known as \_\_\_\_\_. Solid substrate fermentation (b) Fed batch fermentation (C) Continuous fermentation (d) None of the above 10 Which of the following factors are considered for scale-up? (a) Inoculum development. (b) Sterilization. (c) Environmental parameters. (d) All of the above. (P. T. O.)

Q.2	•	Give short answers to the following questions. (02 - marks each) (Any Ten)	20
		in the second second second second second second second second second second second second second second second	_,
	1	Differentiate between Trophophase & Idiophase.	
	2	Enlist the range of fermentation processes, giving suitable examples.	71
:	3 -	Why, the medium used for screening organic acid producers be sufficiently buffered?	ij.
	5	Write on the substrates used as nitrogen source in fermentation medium.  Enlist the ideal characteristics of a fermentation medium	:
	6	'Strain improvement strategy is an important aspect of a fermentation industry'-	
	v	justify.	
	7	Describe in brief on impellers.	
	8	Write on the control of temperature during a fermentation process.	
	9	Write on the problems associated with SSF	
	10	What is Maillard type browning reaction?	
	11	What is the importance of "sampling" during a fermentation process?	
	12	Describe scale-up in brief.	
		and the state of the state of the state of the state of the state of the state of the state of the state of the	
.3		Write Notes on	
	a	Primary screening of antibiotic producers.	0
	b	Primary screening of growth factor producers	0
•		UK	
.3		What is secondary screening? Discuss the significance of secondary screening.	1
A		White Name and the state of the	
.4	•	Write Notes on the way to the formation of the state of t	_
	a b	Isolation of auxotrophic mutants for primary metabolite.  Role of X-radiation & UV radiation in strain improvement.	04
	U	OR	0
.4		What is recombination? Explain protoplast fusion in detail.	•
•		The state of the s	C
5		Write Notes on	
	a	Ideal characteristics of a fermenter	0:
	b	Air-lift fermenter	0:
		OR OR OR OF THE PROPERTY OF TH	
.5		Write Notes on	
	a	Monitoring and control of foam during fermentation.	0.
	b	Enlist the components of fermenter involved in aeration & agitation. Describe in brief	0:
		spargers.	
_			
6		what is $K_L a$ ? Describe the factors that affect $K_L a$ .	16
6		$\mathbf{O}\mathbf{K}$	
6		Define sterilization. Discuss in detail on the methods of medium sterilization.	10
		× ×	

[8] For a function  $f: R \to R$ , if  $\lim_{x \to 0} f(x) = f(0) + 1$  then f is

[C] exist but they are not equal [D] cannot exist together

[A] continuous at 0 [B] discontinuous at 0 [C] not defined at 0 [D] none

[9] If f is continuous on an interval I then

[A] do not exist

[A] f is uniformly continuous on I

f is not necessarily uniformly continuous on I

f may have some points of discontinuities in I [C]

none

[10] The condition that f is monotonic increasing at c is [D]  $f'(c) \leq 0$ [C]  $f'(c) \geqslant 0$ [B]  $f'(c) \neq 0$ [A] f'(c) = 0



(PTO)

5

5

5

5

- Q: 2. Answer ANY TEN of the following.
  - [1] Prove that supremum of a set S of numbers, if it exists, is unique.
  - [2] Can the greatest and smallest member of a set be same? justify your answer.
  - [3] Find the g.l.b and l.u.b. of  $\left\{1 + \frac{1}{n^2} / n \in N\right\}$  if they exist.
  - [4] Give an example of a set whose derived set and interior are identical.
  - [5] Find the largest open subset of  $(1,2) \cup (4,8)$ .
  - [6] Prove that intersection of two neighbourhoods is also a neighbourhood
  - [7] Evaluate :  $\lim_{x\to 0} \frac{\sqrt{4+x}-2}{x}$
  - [8] Define: (i) Limit of a function (ii) Removable Discontinuity
  - [9] Is the function  $f(x) = |x+1|, x \in R$  continuous at x = -1? Justify.
  - [ 10] Is the following function derivable at x = 1?

$$f(x) = \begin{cases} x & \text{if } 0 \leqslant x < 1\\ 1 & \text{if } x \geqslant 1 \end{cases}$$

- [11] Prove that the function  $x^2$  is uniformly continuous on [-1,1].
- [ 12] Prove that a function which is uniformly continuous on an interval is continuous on that interval.
- Q: 3 [A] Prove that none of  $\sqrt{2}$  and  $\sqrt{8}$  is a rational number.
  - [B] State and prove the Archimedean property of R and deduce that if a is a positive real number a and b any real number then there exists a positive integer n such that na > b.

OR

- **Q:** 3 [A] Is Q an order complete field? Prove your claim.
  - [B] In usual notations prove that  $E(x) = e^x$ ,  $x \in R$ .
- Q: 4 [A] State and prove the Bolzano-Weiestrass theorem for sets.
  - [B] If S and T are sets of real numbers then prove the following (i)  $S \subset T \Rightarrow S' \subset T'$  (ii)  $(S \cup T)' = S' \cup T'$  5

OR

- Q: 4 [A] Prove that interior of a set S is the largest open subset of S.
  - [B] Prove that derived set of a set is closed.

5

Q: 5 [A] If a function f is continuous on [a, b] and f(a) and f(b) are of opposite signs, then prove that there exists at least one point  $\alpha \in (a, b)$  such that  $f(\alpha) = 0$ .

5

[B] If a function f is continuous at an interior point c of [a, b] and  $f(c) \neq 0$ , then prove that, there exists  $\delta > 0$  such that f(x) has the same sign as f(c) for every  $x \in (c - \delta, c + \delta)$ .

5

OR

Q: 5 [A] Show that a function  $f:[a,b] \to \Re$  is continuous at point c of [a,b] iff

$$\lim_{n \to \infty} c_n = c \Longrightarrow \lim_{n \to \infty} f(c_n) = f(c)$$

5

[B] Prove that the function f defined on  $\mathbb{R}$  as follows is discontinuous at every point.

 $f(x) = \begin{cases} 1 & \text{when x is irrational} \\ -1 & \text{when x is rational} \end{cases}$ 

5

Q: 6 [A] State and prove the Darboux's theorem for derivable function.

5

[B] Define Uniformly Continuous function and prove that the function  $\frac{1}{x}$  is not uniformly continuous on (0,1].

5

OR

Q: 6 [A] If f'(c) > 0, then prove that f is a monotonic increasing function at point x = c.

5

5

[B] Show that  $\log(1+x)$  lies between  $x-\frac{x^2}{2}$  and  $x-\frac{x^2}{2(1+x)}$ ,  $\forall x>0$ 

ー ケ ー (a)

.

[56/A25]

Sardar Patel University, Vallabh Vidyanagar

B.Sc. Examinations: 2017-18 - Y Sem

Subject: Mathematics

US05CMTH02

Max. Marks: 70

Real Analysis-II

Date: 09/11/2017, Thursday

Timing: 10.00 am - 01.00 pm

Q: 1. Answer the following by choosing correct answers from given choices.

10

[1] Every convergent sequence is

[A] oscillating

[B] bounded

[C] unbounded [D] none

- [2] The sequene  $\{(-1)^{n^2+1}\}$ [A] is convergent [B] diverges to  $\infty$  [C] diverges to  $-\infty$  [D] oscillates finitely
- [3] A sequence  $\{S_n\}$ ; where

 $S_n = \begin{cases} 2 & \text{; if } n = 1 \text{ or even} \\ p & \text{; where } p \text{ is the smallest prime factor of n.} \end{cases}$ 

[A] convergent [B] divergent [C] oscillates finitely [D] oscillates infinitely

- [4] A positive term series  $\sum_{n=1}^{\infty} \frac{1}{n^p}$  is convergent if and only if [A] p < 1 [B] p > 1 [C]  $p \le 1$  [D]  $p \ge 1$
- [5] The positive term series  $1+r+r^2+r^3+\ldots+r^n+\ldots$  converges for [A] r>1 [B]  $r\geqslant 1$  [C] r<1 [D]  $r\leqslant 1$ [A] r > 1
- [6] If  $\sum_{i=1}^{\infty} u_i$  is a positive term series and  $\sum_{i=1}^{n} u_i < 100, \forall n$  then the series  $\frac{1}{2} |u_i| < 100$ [B] diverges to  $+\infty$  [C] diverges to  $-\infty$ [D] none [A] is convergent

[7]  $\lim_{x \to 1} \lim_{y \to 1} \frac{x^2 + y^2}{x + y} =$ [D] none

[8]  $\lim_{(x,y)\to(6,\pi)} x^2 \tan \frac{y}{x} =$ [D]  $3\sqrt{12}$ [C]  $12\sqrt{3}$ [B]  $36\sqrt{3}$ [A] 36

[9] The necessary condition for a function f to have an extreme value at (2,4) is [A]  $f_x(2,4) = 0$ ,  $f_y(2,4) \neq 0$  [B]  $f_x(2,4) \neq 0$ ,  $f_y(2,4) = 0$  $\text{[C]} \ f_x(2,4) \neq 0, \ f_y(2,4) \neq 0 \quad \text{(D)} \ f_x(2,4) = 0, \ f_y(2,4) = 0$ 

[10] For a function f whose domain contains a neighbourhood of (2,-1), if f(x,y) - f(2,-1) assumes positive as well as negative signs in every neighbourhood of (2,-1) then at (2,-1), f has [A] no extreme value [B] a minimum [C] a maximum

5

5

- [1] Define: (i) Monotonic Sequence (ii) Finitely Oscillating sequence
- [2] Show that  $\lim_{n\to\infty} \frac{3+2\sqrt{n}}{\sqrt{n}} = 2$
- [3] Show that  $\lim_{n\to\infty} \sqrt[n]{n} = 1$
- [4] Show that the necessary condition for convergence of an infinite series  $\sum_{n=1}^{\infty} u_n$  is that  $\lim_{n\to\infty} u_n = 0$
- [5] If  $\sum_{n=1}^{\infty} u_n = u$  and  $\sum_{n=1}^{\infty} v_n = v$  then prove that  $\sum_{n=1}^{\infty} (u_n v_n) = u v$
- [6] Show that the series  $1 + \frac{1}{2!} + \frac{1}{3!} + \dots$  is convergent.
- [7] Show that :  $\lim_{(x,y)\to(0,0)} \frac{x\sin(x^2+y^2)}{x^2+y^2} = 0$
- [8] Evaluate:  $\lim_{(x,y)\to(1,1)} \frac{e^{(x-y)}-1}{x-y}$
- [9] If  $f(x,y) = x^2y + e^{xy^2}$ , then find  $f_x$  and  $f_y$
- [10] State the necessary conditions for a function z = f(x, y) to attain extreme values at a point (a, b)
- [11] State Maclaurin's theorem
- [12] Show that  $y^2 + x^2y + x^4$  has a minimum at (0,0).
- Q: 3 [A] If  $\{a_n\}$  and  $\{b_n\}$  are two sequences such that  $\lim_{n \to \infty} a_n = a$  and  $\lim_{n \to \infty} b_n = b$ , then prove that  $\lim_{n \to \infty} (a_n b_n) = ab$ 
  - [B] Prove that a sequence  $\{S_n\}$  defined by the recursion formula  $S_{n+1} = \sqrt{7 + S_n}$ , where  $S_1 = \sqrt{7}$ , converges to the positive root of  $x^2 x 7 = 0$

OR

- Q: 3 [A] If a sequence of closed intervals  $[a_n, b_n]$  is such that each member  $[a_{n+1}, b_{n+1}]$  is contained in the preceding one  $[a_n, b_n]$  and  $\lim_{n \to \infty} (b_n a_n) = 0$  then prove that there is one and only one point common to all the intervals of the sequence.
  - [B] State and prove the Bolzano-Weierstarss theorem for sequence 5
- Q: 4 [A] State and prove the comparision test of first type in limit form.
  - [B] State and prove Cauchy's general principle for convergence of a series. and prove that the series  $\sum \frac{1}{n}$  does not converge.

	5
	5
[B] Test the series $\sum \frac{n^2-1}{n^2+1}x^n$ for convergence. Q: 5 [A] Define Limit of a function and by using the definition of limit prove that :	5
$[B] \begin{tabular}{ll} For the following function show that the repeated limits exist but the double limit does not when $(x,y) \to (0,0)$ \\ f(x,y) = $\left\{ \frac{x^2 - y^2}{x^2 + y^2} \; ; \; \text{when } x \neq y \\ 0 \; ; \; \text{when } x = y \end{tabular} \right. $$	5
OR	
<ul> <li>Q: 5. If z = f(x,y) is a function of independent variables x, y and if x, y are changed to new independent variables u, v by the substitution x = φ(u, v); y = ψ(u, v), to new independent variables u, v by the substitution x = φ(u, v); y = ψ(u, v), to new independent variables u, v by the substitution x = φ(u, v); y = ψ(u, v), to new independent variables x, y and if x, y are changed to new independent variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y and if x, y are changed variables x, y are changed variables x, y and if x, y are changed variables x, y are changed variables x, y and if x, y are changed variables x, y are changed variable</li></ul>	10 5 5
OR  Q: 6 [A] Prove that the first four terms of Maclaurin's expansion of $e^{ax}\cos by$ are $1 + ax + \frac{a^2x^2 - b^2y^2}{2!} + \frac{a^3x^3 - 3ab^2xy^2}{3!}$ [B] Define Extreme Value and investigate the maxima and minima of the function $f(x,y) = x^3 + y^3 - 63(x+y) + 12xy$ — $\times$ — $\times$ — $\times$ — $\times$	5 on 5

and the first of the second se

and the second s

· '''
.

SARDAR PATEL UNIVERSITY  Language Standard Pages : 3  SARDAR PATEL UNIVERSITY  Language Standard Stan		No. of Printed Pages : 3	
Time: 10:00 TO 01:00 P.M  USOSCMTH03/Metric Space  Maximum Marks: 70  Q.1 Answer the following by selecting correct answer from the options.  1 The set of all cluster points of $\mathbb{Q}$ is  (a) $\mathbb{N}$ (b) $\mathbb{Q}$ (c) $\mathbb{R} - \mathbb{Q}$ (d) $\mathbb{R}$ 2 Let $d: M \times M \to \mathbb{R}$ be a metric on $M$ . Then which of the following is also a metric on $M$ ?  (a) $d_1(x,y) = min\{1, d(x,y)\}$ (b) $d_1(x,y) = max\{1, d(x,y)\}$ (c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (a) $d_1(x,y) = max\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (e) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (g) $d_1(x,y) = max\{0, d(x,y)\}$ (h) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{1, d(x,y)\}$ (g) $d_1(x,y) = max\{1, d(x,y) = max\{1, d(x,y) = max\{0, d(x,$		SARDAR PATEL UNIVERSITY	
USOSCMTH03/Metric Space  Time: 10:00 TO 01:00 P.M  Maximum Marks: 70  Q.1 Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following by selecting correct answer from the options.  [10]  Answer the following be a metric on M.  [10]  Answer Any Ten of the following.  Answer Any Ten of the following are open set in $(M, d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset E$ .  Is arbitrary union of closed set is closed? Justify!	2	[48] A26] B.S. SEM-V EXAMINATION	era en
Time: 10:00 TO 01:00 P.M  Q.1 Answer the following by selecting correct answer from the options.  1. The set of all cluster points of $\mathbb{Q}$ is		12 November 2017, Saturday	Seature 3.
Q.1 Answer the following by selecting correct answer from the options. [10]  1. The set of all cluster points of $\mathbb{Q}$ is	8	USUSCMTH03/Metric chace	
Q.1 Answer the following by selecting correct answer from the options. [10]  1 The set of all cluster points of $\mathbb{Q}$ is		Ilme: 10:00 TO 01:00 P.M	
(a) $\mathbb{N}$ (b) $\mathbb{Q}$ (c) $\mathbb{R} - \mathbb{Q}$ (d) $\mathbb{R}$ 2 Let $d: M \times M \to \mathbb{R}$ be a metric on $M$ . Then which of the following is also a metric on $M$ ?  (a) $d_1(x,y) = min\{1, d(x,y)\}$ (b) $d_1(x,y) = max\{1, d(x,y)\}$ (c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (e) $(0,1) \cup (1,2)$ (f) $(0,1) \cup (1,2)$ (f) $(0,1) \cup (1,2)$ (f) $(0,1) \cup (1,2)$ (g) $(0,1) \cup (1,2)$ (h) $(0,1) \cup (1,2)$ (h) $(0,1) \cup (1,2)$ (f) $(0,1) \cup (1,2)$ (g) $(0,1) \cup (1,2)$ (h) $(0,1) \cup (1,2)$ (l) $(0,1) \cup (1,2)$ (o) $(0,1) \cup (1$	Q		
(a) $\mathbb{R}$ (b) $\mathbb{Q}$ (c) $\mathbb{R} - \mathbb{Q}$ (d) $\mathbb{R}$ 2 Let $d: M \times M \to \mathbb{R}$ be a metric on $M$ . Then which of the following is also a metric on $M$ ?  (a) $d_1(x,y) = min\{1, d(x,y)\}$ (b) $d_1(x,y) = max\{1, d(x,y)\}$ (c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (g) $d_$		The set of all cluster points of $\mathbb{O}$ is	[10]
on $M$ ?  (a) $d_1(x,y) = min\{1, d(x,y)\}$ (b) $d_1(x,y) = max\{1, d(x,y)\}$ (c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (e) $d_1(x,y) = min\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (h) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (g) $d_1(x,y) = max\{$	*	(a) N (b) O	250 A. <del></del>
(a) $d_1(x,y) = min\{1, d(x,y)\}$ (b) $d_1(x,y) = max\{1, d(x,y)\}$ (c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{0, d(x,y)\}$ (e) $d_1(x,y) = min\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (g) $d_1(x,y) = max\{0, d(x,y)\}$ (h) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{0, d(x,y)\}$ (g) $d_1(x,y) = max\{0, d(x,y)\}$ (h) $d_1(x,y) = max\{0, d(x,y)\}$ (f) $d_1(x,y) = max\{1, d(x,y)\}$ (g) $d_1(x,y) = max\{1, d(x,y)\}$ (f) $d_1(x,y) = max\{1, d(x,y)\}$ (g) $d_1(x,y) = max\{1, d(x,y)\}$ (f) $d_1(x,y) = max\{1, d(x,y)\}$ (g) $d_1(x,y) = max\{1, d(x,y)\}$ (f) $d_1(x,y) = max\{1, d(x,y)\}$ (g) $d_1(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y) = max\{0, d(x,y) = max\{1, d(x,y)$	2	Let $d: M \times M \to \mathbb{R}$ be a metric on $M$ . Then which of the $s$ $u$	
(a) $(0,1)$ (b) $(1,2)$ (d) $[0,1] \cup [1,2]$ 4 In the metric space $M = [0,1]$ with usual metric, $B \begin{bmatrix} 1 \\ 7 \end{bmatrix}$ , $30 \end{bmatrix} = $ (a) $[0,1]$ (b) $I_4^1$ , $I_1$ (c) $[0,\frac{1}{4}]$ (d) $(0,1)$ 5 $A_1$ and $A_2$ are two connected subset of a metric space $M$ . $A_1 \cup A_2$ is connected if (a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (c) $A_1 \cap \overline{A_2} = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (0, \infty)$ of $\mathbb{R}^1$ is (a) bounded (c) neither bounded nor totally bounded (d) none image of Cauchy sequence under uniform continuous function is sequence (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ function (a) fis not a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous  Q.2 Answer ANY TEN of the following.  1 If $\rho$ is a metric space. Then $\emptyset$ and $M$ are open set in $(M, d)$ .  Every function from Rd (into a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ . Is arbitrary union of closed set is closed? Justify!		on $M$ ?	tric
(a) $(0,1)$ (b) $(1,2)$ (d) $[0,1] \cup [1,2]$ 4 In the metric space $M = [0,1]$ with usual metric, $B \begin{bmatrix} 1 \\ 7 \end{bmatrix}$ , $30 \end{bmatrix} = \frac{1}{2}$ (a) $[0,1]$ (b) $[\frac{1}{4},1]$ (c) $[0,\frac{1}{4}]$ (d) $(0,1)$ 5 $A_1$ and $A_2$ are two connected subset of a metric space M. $A_1 \cup A_2$ is connected if $\frac{1}{2}$ (a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (c) $A_1 \cap \overline{A_2} = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (0, \infty)$ to totally bounded (e) neither bounded nor totally bounded (f) none image of Cauchy sequence under uniform continuous function is sequence  (a) a Cauchy (b) not a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1]$ (c) $(0,1]$ function (a) f is not a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous (e) $f^{-1}$ is not a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ . Is arbitrary union of closed set is closed? Justify!		(a) $d_1(x, y) = min\{1, d(x, y)\}$ (b) $d_1(x, y) = min\{1, d(x, y)\}$	
(a) $(0,1)$ (b) $(1,2)$ (d) $[0,1] \cup [1,2]$ 4 In the metric space $M = [0,1]$ with usual metric, $B \begin{bmatrix} 1 \\ 7 \end{bmatrix}$ , $30 \end{bmatrix} = \frac{1}{2}$ (a) $[0,1]$ (b) $[\frac{1}{4},1]$ (c) $[0,\frac{1}{4}]$ (d) $(0,1)$ 5 $A_1$ and $A_2$ are two connected subset of a metric space M. $A_1 \cup A_2$ is connected if $\frac{1}{2}$ (a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (c) $A_1 \cap \overline{A_2} = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (0, \infty)$ to totally bounded (e) neither bounded nor totally bounded (f) none image of Cauchy sequence under uniform continuous function is sequence  (a) a Cauchy (b) not a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1]$ (c) $(0,1]$ function (a) f is not a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous (e) $f^{-1}$ is not a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ . Is arbitrary union of closed set is closed? Justify!		(c) $d_1(x,y) = min\{0, d(x,y)\}$ (d) $d_1(x,y) = max\{1, d(x,y)\}$	
(a) $(0,1) \cup (1,2)$ (b) $(1,2)$ (d) $[0,1] \cup [1,2]$ In the metric space $M=[0,1]$ with usual metric, $B\begin{bmatrix} 1\\ 7\\ 30\end{bmatrix} = $ (a) $[0,1]$ (b) $I_4^1$ , $I_1$ (c) $I_4^1$ (d) $I_4^1$ (d) $I_4^1$ (d) $I_4^1$ (d) $I_4^1$ (e) $I_4^1$ (for $I_4^1$ (	3		17/1
In the metric space $M=[0,1]$ with usual metric, $B\begin{bmatrix} 1\\ 7\\ 7\\ 30\end{bmatrix}= \begin{bmatrix} (a) & [0,1] & (b) \begin{bmatrix} 1\\ 4\\ 1\end{bmatrix} & (c) & [0,\frac{1}{4}] \end{bmatrix}$ (d) $(0,1)$ A <sub>1</sub> and A <sub>2</sub> are two connected subset of a metric space M. A <sub>1</sub> $\cup$ A <sub>2</sub> is connected if  (a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M=R_d$ and $A=(0,\infty)$ then diameter of A=  (a) 0 (b) 1 (c) -1 (d) $\infty$ 7 Subset $(0,\infty)$ of $\mathbb{R}^1$ is  (a) bounded  (c) neither bounded nor totally bounded (d) none  8 Image of Cauchy sequence under uniform continuous function is  sequence  (a) a Cauchy (b) not a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact.  (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then  (a) f is not a continuous  (b) $f^{-1}$ is a continuous  (c) $f^{-1}$ is not a continuous  (d) f is a continuous  Q.2 Answer ANY TEN of the following.  1 If p is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		(a) (0,1)	
(a) $[0,1]$ (b) $[\frac{1}{4},1]$ (c) $[0,\frac{1}{4}]$ (d) $(0,1)$ $A_1$ and $A_2$ are two connected subset of a metric space M. $A_1 \cup A_2$ is connected if  (a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (c) $A_1 \cap \overline{A_2} = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (c) - 1$ (d) $\infty$ 7 Subset $(0, \infty)$ of $\mathbb{R}^1$ is (a) bounded (b) totally bounded (d) none  8 Image of Cauchy sequence under uniform continuous function is sequence (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1]$ (c) $(0,1]$ (d) $(0,1)$ function (a) f is not a continuous (c) $f^{-1}$ is not a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous  Q.2 Answer ANY TEN of the following.  1 If $p$ is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  Every function from Rd (into a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ . Is arbitrary union of closed set is closed? Justify!	a	(d) [0, 1] 1 [1, 2]	
$(a) \ [0,1] \qquad (b) \ [\frac{1}{4},1] \qquad (c) \ [0,\frac{1}{4}] \qquad (d) \ (0,1)$ $A_1 \text{ and } A_2 \text{ are two connected subset of a metric space } M. \ A_1 \cup A_2 \text{ is connected } if \\ (a) \ A_1 \cap A_2 = \emptyset \qquad (b) \ \overline{A_1} \cap A_2 = \emptyset \qquad (d) A_1 \cap A_2 \neq \emptyset$ $6 \ M = R_d \text{ and } A = (0, \infty) \text{ then diameter of } A = \\ (a) \ 0 \qquad (b) \ 1 \qquad (c) -1 \qquad (d) \infty$ $7 \ \text{Subset } (0, \infty) \text{ of } \mathbb{R}^1 \text{ is} \qquad (a) \text{ bounded} \qquad (b) \text{ totally bounded} \qquad (c) \text{ neither bounded nor totally bounded} \qquad (d) \text{ none} \qquad (e) \text{ neither bounded nor totally bounded} \qquad (d) \text{ none} \qquad (e) \text{ mage of Cauchy sequence} \qquad (e) \text{ not a Cauchy} \qquad (f) \text{ none} \qquad (g) \text{ converges to two distinct point} \qquad (f) \text{ none} \qquad (g) \text{ more sequence} \qquad (g) \text{ none} \qquad (g) \text{ more sequence} \qquad (g) \text{ none} \qquad (g) \text{ more sequence} \qquad (g) \text{ none} \qquad (g) \text{ for a Cauchy} \qquad (g) \text{ for a Cauchy} \qquad (g) \text{ none} \qquad (g) \text{ for a Cauchy} \qquad (g)  fo$	4	In the metric space $M = [0, 1]$ with usual metric $R\begin{bmatrix} 1 & 20 \end{bmatrix}$	
$A_1 \text{ and } A_2 \text{ are two connected subset of a metric space M. } A_1 \cup A_2 \text{ is connected if } \\ (a) \ A_1 \cap A_2 = \emptyset \\ (c) \ A_1 \cap \overline{A_2} = \emptyset \\ (c) \ A_1 \cap \overline{A_2} = \emptyset \\ (d) \ A_1 \cap A_2 \neq \emptyset \\ (d) \ A_1 \cap A_2 \neq \emptyset \\ (d) \ A_1 \cap A_2 \neq \emptyset \\ (d) \ A_1 \cap A_2 \neq \emptyset \\ (e) \ A_1 \cap A_2 \neq \emptyset \\ (f) \ A_1 \cap A_2 \neq \emptyset \\ (f) \ A_1 \cap A_2 \neq \emptyset \\ (f) \cap A_1 \cap A_2 \neq \emptyset \\ (g) \cap A_1 \cap A_2 \neq \emptyset \\ (g) \cap A_1 \cap A_2 \neq \emptyset \\ (g) \cap A_1 \cap A_2 \neq \emptyset$ $(g) \cap A_1 \cap A_2 \cap A_1 \cap A_2 \cap A_2 \cap A_1 \cap A_2 \cap A_2 \cap A_1 \cap A_2 \cap A_2 \cap A_1 \cap A_2 \cap A_2 \cap A_1 \cap A_2 \cap A_1 \cap A_2 \cap A_1 \cap A_2 \cap A_1 \cap A_1 \cap A_2 \cap A_1 \cap A_1 \cap A_2 \cap A_1 \cap A_1 \cap A_2 \cap A_1 \cap A_1 \cap A_2 \cap A_1 \cap A_1 \cap A_1 \cap A_2 \cap A_1$		(a) $[0,1]$ (b) $\begin{bmatrix} 1 \\ 1 \end{bmatrix}$ (c) $\begin{bmatrix} 1 \\ 1 \end{bmatrix}$	
(a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (0, \infty)$ (d) $A_1 \cap A_2 \neq \emptyset$ 7 Subset $A_1 \cap A_2 = \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ 8 Judy (a) bounded (b) totally bounded (c) neither bounded nor totally bounded (d) none large of Cauchy sequence under uniform continuous function is sequence  (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $A_1 \cap A_2 = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (b) $A_1 \cap A_2 \neq \emptyset$ (c) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (h) totally bounded (d) none large of the larg	5	$A_1$ and $A_2$ are two connected $A_2$ and $A_3$ are two connected $A_4$ and $A_5$ are two connected $A_4$	
(a) $A_1 \cap A_2 = \emptyset$ (b) $\overline{A_1} \cap A_2 = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ 6 $M = R_d$ and $A = (0, \infty)$ then diameter of $A = (0, \infty)$ (d) $A_1 \cap A_2 \neq \emptyset$ 7 Subset $A_1 \cap A_2 = \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ 8 Judy (a) bounded (b) totally bounded (c) neither bounded nor totally bounded (d) none large of Cauchy sequence under uniform continuous function is sequence  (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $A_1 \cap A_2 = \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (b) $A_1 \cap A_2 \neq \emptyset$ (c) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_2 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (d) $A_1 \cap A_2 \neq \emptyset$ (e) $A_1 \cap A_2 \neq \emptyset$ (f) $A_1 \cap A_2 \neq \emptyset$ (h) totally bounded (d) none large of the larg		if $A_1 \cup A_2$ are two connected subset of a metric space M. $A_1 \cup A_2$ is connected	
6 $M=R_d$ and $A=(0,\infty)$ then diameter of $A=(a)$ 0 (b) 1 (c) -1 (d) $\infty$ 7 Subset $(0,\infty)$ of $\mathbb{R}^1$ is (a) bounded (c) neither bounded nor totally bounded (d) none  8 Image of Cauchy sequence under uniform continuous function is  sequence (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f:R_d\to d$ defined by $f(x)=x$ all $x$ in R then function (a) $f$ is not a continuous (c) $f^{-1}$ is not a continuous (d) $f$ is a continuous  Q.2 Answer ANY TEN of the following.  1 If $p$ is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  2 Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		(a) $A_1 \cap A_2 = \emptyset$	
(a) 0 (b) 1 (c) -1 (d) $\infty$ 7 Subset $(0, \infty)$ of $\mathbb{R}^1$ is  (a) bounded (b) totally bounded (c) neither bounded nor totally bounded (d) none  8 Image of Cauchy sequence under uniform continuous function is  sequence  (a) a Cauchy (b) not a Cauchy (d) none  9 Metric space with absolutes value metric is compact.  (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ (a) f is not a continuous  (b) $f^{-1}$ is a continuous  (c) $f^{-1}$ is not a continuous  (d) f is a continuous  (e) $f^{-1}$ is not a continuous  (f) $f^{-1}$ is not a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous  (g) $f^{-1}$ is a continuous		$(c) A_1 \cap \overline{A_2} = \emptyset $ $(b) A_1 \cap A_2 = \emptyset$	
7 Subset $(0, \infty)$ of $\mathbb{R}^1$ is (a) bounded (b) totally bounded (c) neither bounded nor totally bounded (d) none [mage of Cauchy sequence under uniform continuous function is sequence (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1]$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f: R_d \to d$ defined by $f(x) = x$ all $f: R_d \to d$ defined by $f: R_d \to d$ def	6	$M = R_d$ and $A = (0, \infty)$ then diameter of $A$	
Subset $(0, \infty)$ of $\mathbb{R}^1$ is  (a) bounded (c) neither bounded nor totally bounded (d) none  Image of Cauchy sequence under uniform continuous function is  sequence (a) a Cauchy (c) converges to two distinct point (d) none  Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f$ is not a continuous (c) $f^{-1}$ is not a continuous (d) $f$ is a continuous  Q.2 Answer ANY TEN of the following.  If $f$ is a metric for a set $f$ , then show that $f$ is metric on $f$ .  Let $f$ is a metric space. Then $f$ and $f$ is metric on $f$ is a continuous on $f$ in usual notations prove that $f$ is a continuous on $f$ in usual notations prove that $f$ is a continuous on $f$ in usual notations of closed set is closed? Justify!		(a) U (b) 1	
(a) bounded (c) neither bounded nor totally bounded (d) none  Image of Cauchy sequence under uniform continuous function is  sequence  (a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ (a) $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f: R_d \to d$ defined by $f(x) = x$ all $f: R$ in R then function (b) $f: R_d \to d$ defined by $f(x) = x$ all $f: R$ in R then function (c) $f: R_d \to d$ defined by $f(x) = x$ all $f: R$ in R then function (d) $f: R_d \to d$ defined by $f(x) = x$ all $f: R$ is a continuous (c) $f: R_d \to d$ defined by $f(x) = x$ all $f: R$ is a continuous  Q.2 Answer ANY TEN of the following.  If $f: R_d \to d$ is a metric for a set $f: R_d \to d$ and $f: R_d \to d$ are open set in $f: R_d \to d$ .  Every function from Rd (into a metric space) is continuous on Rd In usual notations prove that $f: R_d \to d$ lustify!	7	Subset $(0, \infty)$ of $\mathbb{R}^1$ is $(0, \infty)$	
(a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f: f:		Lates the first the property of the property o	
(a) a Cauchy (c) converges to two distinct point (d) none  9 Metric space with absolutes value metric is compact. (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f: f:		(c) neither bounded nor totally bounded (d) none	[8]
(a) a Cauchy (b) not a Cauchy (c) converges to two distinct point (d) none  Metric space with absolutes value metric is compact.  (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) f is not a continuous (b) $f^{-1}$ is a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous  Q.2 Answer ANY TEN of the following.  If p is a metric for a set M, then show that $2p$ is metric on M.  Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!	8	Image of Cauchy sequence under uniform continuous function is	
(c) converges to two distinct point (d) none  Metric space with absolutes value metric is compact.  (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) $f$ is not a continuous (c) $f^{-1}$ is not a continuous (d) $f$ is a continuous  Q.2 Answer ANY TEN of the following.  If $p$ is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  2 Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		sequence	
9 Metric space with absolutes value metric is compact.  (a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f:R_d \to d$ defined by $f(x) = x$ all $x$ in R then function  (a) f is not a continuous  (b) $f^{-1}$ is a continuous  (c) $f^{-1}$ is not a continuous  (d) f is a continuous  Q.2 Answer ANY TEN of the following.  1 If p is a metric for a set M, then show that 2p is metric on M.  2 Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  3 Every function from Rd (into a metric space) is continuous on Rd  4 In usual notations prove that $E \subset \overline{E}$ .  5 Is arbitrary union of closed set is closed? Justify!		(-) a caucity	
(a) $[0,1]$ (b) $[0,1)$ (c) $(0,1]$ (d) $(0,1)$ 10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function (a) f is not a continuous (b) $f^{-1}$ is a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous  Q.2 Answer ANY TEN of the following.  1 If p is a metric for a set M, then show that $2p$ is metric on M.  2 Let $(M, d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M, d)$ .  3 Every function from Rd (into a metric space) is continuous on Rd 4 In usual notations prove that $E \subset \overline{E}$ . 5 Is arbitrary union of closed set is closed? Justify!	0	(d) none	
10 $f: R_d \to d$ defined by $f(x) = x$ all $x$ in R then function  (a) f is not a continuous  (b) $f^{-1}$ is a continuous  (c) $f^{-1}$ is not a continuous  (d) f is a continuous  Q.2 Answer ANY TEN of the following.  If $p$ is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  Let $(M, d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M, d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!	121	with absolutes value metric is compact	
(a) f is not a continuous (b) $f^{-1}$ is a continuous (c) $f^{-1}$ is not a continuous (d) f is a continuous (d) f is a continuous (d) f is a continuous (e) $f^{-1}$ is a metric for a set M, then show that $f^{-1}$ is metric on M. Let $f^{-1}$ is a metric space. Then $f^{-1}$ and $f^{-1}$ is a continuous (20)	10	(0) (0.1)	
(a) Fishot a continuous (b) $f^{-1}$ is a continuous (c) $f^{-1}$ is not a continuous (d) $f$ is a continuous  Q.2 Answer ANY TEN of the following.  If $p$ is a metric for a set $M$ , then show that $2p$ is metric on $M$ .  Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  Every function from Rd (into a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!	10	(x) = x and $(x) = x$ and $(x) = x$ then function	
Q.2 Answer ANY TEN of the following.  If $\rho$ is a metric for a set $M$ , then show that $2\rho$ is metric on $M$ .  Let $(M,d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M,d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		(a) i is not a continuous	
Q.2 Answer ANY TEN of the following.  If p is a metric for a set M, then show that 2p is metric on M.  Let $(M, d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M, d)$ .  Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		(d) f is a continuous	
Let $(M, d)$ be a metric space. Then $\emptyset$ and $M$ are open set in $(M, d)$ .  Every function from Rd (into a metric space) is continuous on Rd In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!	Q.2	Answer ANY TEN of the following sequence and the following to enguence	
Every function from Rd (into a metric space) is continuous on Rd  In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!		If p is a metric for a set M, then should be a	[20]
In usual notations prove that $E \subset \overline{E}$ .  Is arbitrary union of closed set is closed? Justify!	C 25 To 1	Let $(M, d)$ be a metric space. Then $d$ and $M$	· nerminal di
Is arbitrary union of closed set is closed? Justify!		Every function from Rd (into a metric space) is a set in $(M, d)$ .	
Is arbitrary union of closed set is closed? Justify!		In usual notations prove that $E \subset \overline{E}$	
	5	Is arbitrary union of closed set is closed? bushful	16.0
	9		CPTAN

	U	Let M a methic space and let $A \subseteq D \subseteq M$ . If A is a defise in D and if D is define in				
		M, then prove that $A$ is dense in $M$ .				
	7	Show that every subset of $R_d$ is bounded.				
	8	Prove or disprove that the empty set φ and singleton set are assumed to be connected				
-(11	9	Let T: $[0, 1/3] \rightarrow [0, 1/3]$ be defined byT $(x) = x^2$ , $\forall x \in [0, 1/3]$ . Prove that T is a contraction on $[0, 1/3]$ .	•			
	10	Any compact subset A of a metric space $(M, d)$ is closed and bounded.				
	11 12	Let f be a real valued continuous function on [a, b]. Then prove that f is bounded. Show that $f: \mathbb{R} \to \mathbb{R}$ defined by $f(x) = x$ is uniformly continuous.				
	Q-3					
		Lat (M. a) has a matric chase of (C. ) is a convergent convence of points of M. then	(E)			
	[A]	Let $(M, p)$ be a metric space. If $\{S_n\}$ is a convergent sequence of points of M, then $\{S_n\}$ is Cauchy. Is converse true? Justify!	[5]			
	[B]	The inverse image under f of any open ball B[f (a); $\epsilon$ ] about f (a) contains an open ball B[a; $\delta$ ] about a.	[5]			
		and the second of the second o				
	[C] ·	Let (M, d) be a metric space. Show that a function defined by	[5]			
	,	$d_1(x,y) = \frac{d(x,y)}{1+d(x,y)}$ is a metric for M.	* ··· *			
	[D]	Let M be a metric space, and let f and g be real valued functions which are	[5]			
•		continuous at a $\in$ M and g(a) $\neq$ 0. Prove that $\frac{f}{g}$ is continuous at a.				
	0.4					
	Q-4	Draws that fits continuous life the inverse image of every one set is one	[E]			
٠	[A]	Prove that f is continuous iff the inverse image of every open set is open. [5]  The subset 4 of ID is connected iff whenever a C 4 b C B with a C b then				
	[B]	The subset A of $\mathbb{R}$ is connected iff whenever $a \in A$ , $b \in B$ with $a < b$ , then  [5]				
	•	$c \in A$ for any $c$ such that $a < c < b$ .				
	[0]	OR	re3			
	[C]	Let $(M, \rho_1)$ and $(M, \rho_2)$ be metric spaces and let $f: M_1 \to M_2$ . Then $f$ is continuous on $M_1$ if and only if $f^{-1}(F)$ is closed subset of $M_1$ whenever $F$ is a	[5]			
	[D]	closed subset of $M_2$ .	r – 1			
	[D]	Let $E$ be any subset of the metric space $M$ , then show that $\overline{E}$ is closed.	[5]			
		and the first of the control of the control of the control of the control of the control of the control of the	·. ·			
		· 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年 - 1987年				
	Q-5	Let (M, ρ) be a metric space. The subset A of M is totally bounded iff every	[10]			
**	Q-J					
	(14.1		V.			
:		Define: Complete metric space. State and prove generalized nested interval	[10]			
		theorem.				
		- Community Community (1994年) 1994年 日本地は、大田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田田				
			F .			
		and the second of the second o				

If the metric space M has the Heine- Borel property, then M is compact.	[5]
Let f be a continuous real valued function on the compact metric space M, then f attains a maximum value at the same point of M. Also f attains a minimum value at some points of M.	[5]
OR	
Prove that continuous image of compact metric space is compact.	[5]
If f is a one- one continuous function form the compact metric space $M_1$ onto a metric space $M_2$ then prove that f is a homeomorphism of $M_1$ onto $M_2$ .	[5]
	If the metric space M has the Heine- Borel property, then M is compact. Let f be a continuous real valued function on the compact metric space M, then f attains a maximum value at the same point of M. Also f attains a minimum value at some points of M.

came, that we contribute the contribution of t

mente de la composition de la companie de la companie de la companie de la companie de la companie de la compa La composition de la companie de la companie de la companie de la companie de la companie de la companie de la

### SARDAR PATEL UNIVERSITY

## B.Sc.SEM-V EXAMINATION 13<sup>th</sup> November 2017, Monday 10.00 a.m. to 01.00 p.m.

US05CMTH04 (Abstract Algebra-I)

Maximum Marks: 70

000001111101	(122212000 1228			
-	the correct option in the fo	ollowing questions,	mention the correct	[10]
(1) $O(2)$ in $\mathbb{Z}$	is			
(a) 3	(b) 0	(c) infinite	(d) 2	
(2) Additive i	nverse of 2 in $\mathbb{Z}_6$ is	, ,		
• •	(b) 2	(c) 3	(d) 1	
\ /	lic group of order i			
	(b) prime		(d) 1	
	nite cyclic group has exact		ators.	1.31
(a) 3	(b) 1	(c) 2	(d) 4	
	ation $\sigma$ is said to be odd p		ature of $\sigma$ is	
		(c) 1	(d) -2	:
(6) În Klein 4	l-group $G = \{e, a, b, c\}$ , $ab$			
(a) $a$	(b) b	(c) c	(d) e	
(7) Centre of	<b>Z</b> is			1
(a) Z	Z is(b) 2	(c) N	(d) 1	
(8)	is generator of group $\{\frac{1}{8}\}$	$\{\frac{1}{2}, \frac{1}{2}, \frac{1}{2}, \frac{1}{2}, \frac{1}{4}, \frac{2}{8}, \dots\}$		v
(0)	is generator or group [	3' 4' 2' -, -, -, -,		
(a) $\frac{1}{6}$		(c) 1	(d) $\frac{1}{4}$	
O	of every transposition is .		4	
	= (b) -1		(d) $-2$	7 .
	oup has atleast nor		( )	
$\begin{array}{ccc} \text{(a)} & 3 \end{array}$		(c) 2	(d) 1	
\ /	the short questions. [any			[20]
-	roup and Cyclic group.		and the second s	
(2) Let $G$ an	d G' be two isomorphic g	roups, $G$ is abelian	then prove that $G'$ i	ıs
also abeli		T	in etini. Seeka eta	
(3) In group	G, prove that every eleme	ent of $G$ has unique		
	at product of two permuta			1 4
(5) State Eu	ler's theorem.		$\mathcal{F}^{\mathcal{G}} = \{ \{ \{ \} \} \} \}$	1
(6) Let H be	any subgroup of group G	. Then prove that $\alpha$	$aH = H \Leftrightarrow a \in H.$	
(7) Prove that	at every subgroup of abelia	an group G is norm	ial in G.	
(8) Express 1	the inverse of cycle (124	53) as a product of	f transpositions.	
(9) Define Si	mple group and Quotient	group.		
(10) Let $S_n$ be	e a group of all permutation	ons. Prove that ord	ler of $S_n$ is $n!$ .	
	gnature of the permutatio		[PTC	)]

- , (h (12) Find generators of  $\mathbb{Z}$ . Q.3(a) Let G be a semigroup. Assume that, for all a,b $\in$  G, the equations ax = b [5] and ya = b have unique solutions in G. Then prove that G is a group. (b) Prove that group G is abelian iff G = Z(G). [5] Q.3(c) Let H and K be finite subgroups of group G such that HK is a subgroup of [5] G. Then prove that  $O(HK) = \frac{O(H)O(K)}{O(H \cap K)}$ (d) Prove that a non-empty subset H of a group G is subgroup iff [5]  $ab^{-1} \in H \ \forall \ a, b \in H.$ Q.4(a) Let G be a finite cyclic group of order n, then prove that G has  $\phi(n)$ [5] generators iff (m, n) = 1. (b) Let G be a group and  $a, b \in G$  such that ab = ba. If O(a) = n, O(b) = mwith m, n relatively prime, then prove that O(ab) = mn.  $\mathbf{OR}$ Q.4(c) Prove that any subgroup of an infinite cyclic group is also an infinite cyclic [5] group. (d) If G is cyclic group of order n and  $a^m=e$  for some  $m\in\mathbb{Z}$  then prove that Q.5(a) Let G = (a) be a finite cyclic group of order n. Then prove that the mapping  $\theta: G \to G$  defined by  $\theta(a) = a^m$  is an automorphism of G iff m is relatively prime to n. (b) Prove that a homomorphism  $\theta: G \to G'$  of G to G' is one-one iff  $Ker\theta = \{e\}$ . OR Q.5(c) State and Prove First isomorphism theorem. [6] (d) Prove that a subgroup H is normal in group G iff  $xH = Hx \ \forall \ x \in G$ . 4 Q.6(a) Prove that G is direct product of subgroups H and K iff (i) every  $x \in G$  can be uniquely expressed as x = hk,  $h \in H$ ,  $k \in K$  (ii) hk = kh,  $h \in H$ ,  $k \in K$ . (b) Prove that the external direct product of two cyclic groups each of order 2 is [4]
- Q.6(c) State and prove Cayley's theorem. [6]
  (d) Prove that the set  $S_n$  of all permutation on n symbols forms a non-commutative [4] group.

OR

the Klein 4-group.



## [42/A-15]

## SARDAR PATEL UNIVERSITY B.Sc. SEM-V EXAMINATION 15th November 2017, Wednesday

No of printed pages:02

US05CMTH05/Number Theory
Time: 10:00 TO 01:00 P.M Maximum Marks: 70

Q.1	Answer the following by selecting correct answer from the options.	[10]
1	$(5'-1,5^{34}-1)=$ . The Highmonian is an expectation for all the sections $(5,5)$	•
	. (a) 1 (b) 4 (c) 5 (d) 25 kg Algarang general ng general ng ang garang general na	1
2	If n is an even integer then $3^n+1$ is divisible by	
11	$_{ij}$ (a) $_{5}$ (b) $_{2}$ (c) $_{3}$ (d) $_{4}$ - $_{ij,ij,ij,ij,ij,ij,ij,ij,ij,ij,ij,ij,ij,i$	
3	The number of multipliers of 7 among the integers from 200 to 500 is	
	(a) 71 (b) 28 (c) 43 (d) 63	
4	The highest power of 2 in 50! is	
	(a) 4/ (b) 23 (c) 15 (d) 22	
5	If a is a square number then S(a) is	1
HAA :	(a) even (b) odd (c) prime (d) zero	1.7
6	If $ca \equiv cb \pmod{n}$ then $a \equiv b \pmod{n}$ only if	
	(a) $(c,a)=1$ (b) $(c,a)=b$ (c) $(c,b)=1$ (d) $(c,n)=1$	
7	Prove that every number containing more than two digits can be divided by 4	
	iff the number formed by digits can be divided by 4	
	(a) last two (b) last three (3) first two (d) first three	
8	$(m,n)=$ then $\emptyset(mn)\neq\emptyset(m)+\emptyset(n)$ .	
	(a) (3,6) (b) (2,2) (c) (3,4) (d) (4,3)	
9	If m is not prime then $\emptyset(m)$ m-1.	
	$(a) > (b) < (c) = (a) \neq (a) \Rightarrow (b) = (a) \Rightarrow (b) $	
10	$\sum_{d/m} \emptyset(d) \mu(d) = 0$ iff m is	A.
•	(a) even (b) odd (c) prime (d) 1	
1 11		
<b>Q.2</b>	Answer ANY TEN of the following.	201
1.	Prove that $[a, b, c] = \frac{abc}{(ab,bc,ca)}$ ; $\forall a, b, c > 0$ .	
2		
3	If $(a-s)/(ab+st)$ . Then prove that $(a-s)/(at+bs)$ .	
4	Every square number is of the form 9k or 3k+1; where k is any integer.	
5	Prove that any two distinct Mersenne numbers are relatively prime.	
. 19	If m is a composite number then prove that $2^m-1$ is also composite.	1
6 · · · · · · · · · · · · · · · · · · ·	1 Tove that $[x] \cdot [y] \leq [x+y] \leq [x] + [y] + 1$ .	
	Find the positive integer solution of the Indeterminate equation $x^2 + xy - 6 = 0$ .	
8 ·	If $a_1 \equiv b_1(modn) \& a_2 \equiv b_2(modn)$ then $a_1^m \equiv b_1^m(modn)$ ; $\forall m \in N$ by using	
^	mathematical induction method.	
9	Find all relatively prime solutions of the equation $x^2 + y^2 = z^2$ with $0 < z < 30$ .	
10	If $(a, p) = 1$ ; p is prime then $a^{p-1} \equiv 1 \pmod{p}$ .	
11	Prove that $\emptyset(p^k) = p^k \left(1 - \frac{1}{p}\right)$ ; where $p$ is prime.	
	· • • • • • • • • • • • • • • • • • • •	

12	Find order of 2 modulo 7.	
<b>Q-3</b> [A]	State and Prove Unique factorisation theorem .	[5]
[B]	If 'm' is a composite integer and $n_m=111(m-times)$ then prove that $n_m$ is also composite number.	[5]
ii ral	or the control of the	Fed9
[C] [D] <b>Q-4</b>	State & prove fundamental theorem of Divisibility.  Prove that the prime numbers of the form (4n-1) is infinite.	[5] [5]
[A] [B]	Prove that Möbious function is multiplicative function. Prove that $\sum_{i=1}^n {u_i}^2 = u_n u_{n+1}$	[5] [5]
[C] [D] <b>Q-5</b>	Prove that every prime factor of $F_n(n>2)$ is of the form $2^{n+2}t+1$ ; for some $t\in Z$ . Prove that any prime factor of $M_p(p>2)$ is greater than p.	[5] [5]
[A]	Prove that the positive integer solution of the equation $x^{-1} + y^{-1} = z^{-1}$ ; $(x,y,z) = 1$ has and must have the form $x = a(a+b)$ ; $y = b(a+b)$ ; $z = ab$ ; where $a,b > 0$ & $(a,b) = 1$ .	[10]
[B]	Prove that the linear indeterminate equation $ax+by=c$ has solution iff $d/c$ ; where $(a,b)=d$ . If $x=x_0$ & $y=y_0$ is a particular solution of $ax+by=c$ . Then prove that general solution can be written as	[10]
	$x = x_0 + \frac{b}{d}t$ ; $y = y_0 - \frac{a}{d}t$ ; $t \in \mathbb{Z}$ . Find the positive integer solution of the	11
0.6	indeterminate equation $7x + 19y = 213$ .	
<b>Q-6</b> [A] [B]	If $a^n \equiv 1 \pmod{m}$ and $d$ is the order of $a'$ modulo $m$ . Then prove that $a'$ Solve the system of Congruences $a' \equiv 2 \pmod{3}$	[5]
	$x \equiv 3(mod5)$ $x \equiv 2(mod7)$ $\mathbf{OR}$	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
[C]	Show that the sum of $\phi(m)$ positive integers less than $m(m>1)$ and relatively prime to $m$ is $\frac{m}{2}\phi(m)$ . State & Prove Chienese – Remainder Theorem.	[5]
[D]	State & Prove Chienese - Remainder Theorem.	[5]
	The state of the s	

2

active the transfer of the section of the state of the st

The production of the state of

Maximum Marks: 70

[35 & A-14]

Time: 10:00 TO 01:00 P.M

### SARDAR PATEL UNIVERSITY B.Sc. SEM-V EXAMINATION 17th November 2017, Friday US05CMTH06/Mechanics - I

No of printed pages:02

Q.1 Answer the following by selecting correct answer from the options. [10] Acceleration = 1 (a)  $\frac{d^2s}{dt^2}$  (b)  $\frac{dv}{dt}$  (c)  $v\frac{dv}{ds}$  (d) all of above Dimension of angular velocity is \_\_\_\_\_ 2 (a) $M^0L^1T^{-1}$  (b)  $M^0L^0T^{-1}$  (c)  $M^1L^{-3}T^0$  (d)  $M^0L^0T^1$ 1 pound = grams. 3 (a) 465.3 (b) 435.6 (c) 456.3 (d) 453.6 Three equal forces acting at a point are in equilibrium then angle between any 4 two forces are (a) zero (b) equal (c) different (d) none of these In moment of a vector negative sign is taken for \_\_\_\_\_\_direction. 5 (b) clockwise (c) circular (d) anti-clockwise and the medianest and Lamina is the \_\_\_\_\_ of a rigid body. 6 (a)section (b) particle (c) portion (d) part The mass centre of the particle is denoted by (a) (x,y) (b)  $(\bar{x},\bar{y})$  (c)  $(x_0,y_0)$  (d)  $(x_m,y_m)$ The mass centre of the area in the first quadrant of the curve  $x^2 + y^2 = a^2$ 8 (a)  $(\frac{4a}{3\pi}, \frac{4a}{3\pi})$  (b)  $(\frac{4a}{\pi}, \frac{4a}{\pi})$  (c) (4a, 4a) (d)  $(\frac{3a}{4\pi}, \frac{3a}{4\pi})$ Intrinsic equation for common catenary is \_\_\_\_\_ 9 (a)  $S = c \tan \theta$  (b)  $S = \tan \theta$  (c)  $S = c^2 \tan \theta$  (d)  $S^2 = c \tan \theta$ The tangential component of the reaction for a cable in contact with a smooth 10 curve is \_ (b) N (c)  $\frac{N}{k}$  (d)  $\frac{N}{k}$ (a)k

## Q.2 Answer ANY TEN of the following.

[20]

- 1 Define gradient of vector.
- 2 Explain additions of vectors.
- 3 State newton's third law with illustration.
- 4 When a particle said to be in equilibrium?
- 5 State lamy's theorem.
- 6 Define equipollent system of forces.
- 7 Define work done.
- 8 Describe forces which do not work.
- 9 Define potential energy.

For catenary prove that  $S^2 = y^2 + 2yc$ . 10 In usual notations prove that  $S = c \sinh \frac{x}{c}$ 11 12 Explain the term Hodograph. Q-3 [A] State & prove Equation of motion of a particle moving in a straight line. [5] [B] If two forces of magnitude  $\vec{P} \& \vec{Q}$  makes an angle  $\theta$  then prove that the [5] magnitude of resultant is given by  $R^2 = P^2 + Q^2 + 2PQ \cos \theta$ [C]Two forces acting in opposite direction on a particle have resultant 34lbwt if they act at right angle to each other their resultant would have magnitude of 50 lbwt find magnitude of the forces. A scalar field v is given by over a plane &  $v = \frac{x^2 + y^2}{2x}$  then what are the level [D] curves? Show that at the point with polar co-ordinates  $(r, \theta)$ , grad v is inclined to the x-axis at an angle  $2\theta$  and its magnitude is  $\frac{1}{2} \sec^2 \theta$ . Q-4 [A] Find condition for the equilibrium of the body. State & prove theorem of triangle of forces. [B] OR In usual notations prove that M = xY - Xy. [C][5] If point 'O' is the circum center of  $\triangle ABC$  & the forces  $\overrightarrow{P}$ ,  $\overrightarrow{Q}$ , &  $\overrightarrow{R}$  are acting along [D]  $\overrightarrow{OA}$ ,  $\overrightarrow{OB}$ , &  $\overrightarrow{OC}$  respectively. If  $\overrightarrow{P}$ ,  $\overrightarrow{Q}$ , &  $\overrightarrow{R}$  are in equilibrium then show that  $\frac{p}{a^2(b^2+c^2-a^2)} = \frac{p}{b^2(a^2+c^2-b^2)} = \frac{p}{c^2(a^2+b^2-c^2)}$ Q-5 A light rigid rod of length 2b terminated heavy particles of weight w and W is placed inside a smooth hemispherical bowl of radius 'a' which is fixed with its own rim horizontal. If particle of weight 'w' is just rest below the rim then prove that  $wa^2 = W(2b^2 - a^2)$ . OR Show that there exist mass centre of a system of particles and it is unique. **Q-6** Derive the general formula for the cable hanging free. [A] Show that the equation of suspension bridge represents a parabola. [B] For common catenary prove that  $y = c(\cosh\left(\frac{x}{c}\right) - 1)$  where  $c = \frac{H}{W}$ [C]Find tangential and normal component of the reaction for a cable in contact with [D]

rough curve.

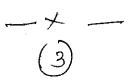
# SARDAR PATEL UNIVERSITY Fifth Semester B. Sc. Examination – 2017 Tuesday, 7<sup>th</sup> Nov. 2017 Time: 10:00 AM to 1:00 PM

PHYSICS: US05CPHY01 (Classical Mechanics)

N. B. All	the nota	tions and symbols have their usual meanings.	Total Mar	ks: 70
Que-1	(1)	Choose correct option to answer the question. According to Newton's law of gravitational force $F_g\alpha$		[10]
		(a) $m_1^2$ (b) $q_1q_2$ (c) $\frac{1}{r^2}$ (d) $m_1r_1$	(80)	
	(2)	A conservative force field $\vec{F} = \frac{\vec{F}}{\vec{F}}$		
		(a) $\vec{\nabla} g$ (b) $\Delta F$ (c) $-\vec{\nabla} V$ (d) $-V^2$		
	(3)	Generalized coordinates may have		
	(4)	(a) any dimensions (b) always dimensions of length (c) always had dimensions time (d) always have the dimensions of time and length force of friction is force.		
		(a) non-conservative (b) linear (c) conservative (d) always consta	ant	ŷ .
	(5)	Virtual work done by applied forces is		
		(a) infinite (b) imaginary (c) zero (d) negative		
	(6)	Frames of reference moving with constant velocity with reference other are calledframes of reference.	e to each	
		(a) non-inertial (b) active (c) rotational (d) inertial		
	(7)	The distance between any two particles of ais constant.		
		(a) rigid body (b) flexible body (c) liquid (d) gas		
	(8)	In a space system of N particle is denoted by single poir	its.	
		(a) configuration (b) momentum (c) velocity (d) mass		
	(9)	According to variational principle, the integral $J = \int_{x_1}^{x_2} f(y, y', x) dx$ should have value.		
		(a) infinite (b) neither minimum nor maximum (c) extremum (d) always minimum		
	(10)	Newton formulation is equivalent to formulation.		
		(a) Einstein (b) Jacobian (c) Schrodinger (d) Lagrangian	### ###	
		Page 1 of 3	(	(P.T.O.)

Que-2		Answer briefly <u>any ten</u> of the following questions.	[20]
	(1)	State Coulomb's law of electrostatic force and write down its equation. Also	
	(2)	give an equation for electric field in terms of a scalar potential.	
	(2)	In the case of two electrons, compare electrostatic and gravitational forces.	
	(3)	What are lines of force and equipotential surfaces?	
	(4) (5)	What are advantages of Lagrangian formulation over Newtonian formulation?  Define cyclic coordinate and show that the linear momentum conjugate to	
	(()	cyclic coordinates is constant.	
	(6)	Explain the following terms;	
	C mar 's	(i) Constraints (ii) degree of freedom.	
	<b>(7)</b>	What is meant by pseudo force? Give an example.	
	(8)	Explain briefly weightless condition of an observer in a satellite.	
	(9)	Is it possible to have a perfectly rigid body? Why?	•
	(10)	Briefly explain configuration space.	
	(11)	State Hamilton's principle.	
	(12)	Explain briefly advantages of Hamilton's formulations over Newtonian formulations.	<i>7</i>
Que-3	(a)	For equivalent one body problem obtain, $\mu \vec{r} = \vec{F}^{int}$ for a two particle system.	[06]
	(b)	State Kepler's three laws of planetary motion and prove the third law.	[04]
		OR	
Que-3	(a)	Derive Gauss' law (field equation) as;	[06]
<b>(</b>	()		[oo]
		$\int \overrightarrow{E} \cdot d\vec{\sigma} = 4\pi \gamma q$	
		Where $\gamma = \frac{1}{4\pi\epsilon_0}$ . Also write down Poisson's and Laplace's equations.	
	(b)	Give the defining equations of center of mass and reduce mass. Show that if mass of the one body is very greater than mass of the other body then the center of mass of the system coincides with the center of mass of the heavier body.	[04]
Que-4	(a)	Obtain Lagrange's equation of motion from D' Alembert's principle as; $\frac{d}{dt} \left( \frac{\partial L}{\partial \dot{q}_i} \right) - \frac{\partial L}{\partial q_i} = 0$	[06]
		Also modify the equation when potential energy is zero.	
	(b)	Derive equation of motion of a bead sliding along a uniformly rotating wire.	[04]
		OR English	
Que-4	(a)	Define Rayleigh dissipation function and equation of motion as; $\frac{d}{dt} \left( \frac{\partial L}{\partial \dot{q}_j} \right) - \frac{\partial L}{\partial q_j} + \frac{\partial \mathcal{F}}{\partial \dot{q}_j} = 0$	[06]
	(b)	Discuss Atwood machine and obtain an expression for the positions of masses.	[04]

[10] For rotating and fixed coordinate system obtain the relation; Que-5  $\left(\frac{d\vec{v}}{dt}\right)_{fir} = \left(\frac{d\vec{v}}{dt}\right)_{rot} + \left(\vec{\boldsymbol{\omega}} \times \vec{\boldsymbol{V}}\right) \quad \text{for } \vec{V}.$ Also obtain the relation;  $\left(\frac{d\overrightarrow{r'}}{dt}\right)_{fix} = \left(\frac{d\overrightarrow{R}}{dt}\right)_{fix} + \left(\frac{d\overrightarrow{r}}{dt}\right)_{rot} + (\overrightarrow{\omega} \times \overrightarrow{r})$ When origin  $\boldsymbol{o}$  of rotating coordinate system is moving with respect to origin  $m{O}'$  of fixed coordinate system,  $\vec{R}$  is the distance between  $m{O}$  and  $\vec{O}'$ .  $\vec{r'}$  and  $\vec{r}$ are the position vectors drawn from  $\boldsymbol{0}'$  and  $\boldsymbol{0}$ . [10] State Euler's and Charles' theorems. Obtain the expression; Que-5  $T=\frac{1}{2}I\omega^2$ Where T is the kinetic energy and I is the angular momentum. [06] Obtain Euler - Lagrange's equation. Que-6 (a) [04] Discuss brachistochrone (shortest time) problem. (b) OR Using Hamilton's principle, obtain Lagrange's equation of motion. [06] Que-6 (a) Using variational principle show that the shortest distance between two [04] (b) points is a straight line.



, 441.°

### SARDAR PATEL UNIVERSITY

### **BSc Examination 2017**

Semester: V

Subject: Physics Course: US05CPHY02

### **Mathematical Physics**

Thursday, Date: 09-11-2017

Time: 10.00 am to 1.00 pm

**Total Marks: 70** 

### INSTRUCTIONS:

- Attempt all questions. 1
- The symbols have their usual meaning. 2
- Figures to the right indicate full marks. 3

### Multiple Choice Questions: [Attempt all] Q-1

10

- A square matrix  $A = \begin{bmatrix} a & h & g \\ h & b & f \\ g & f & c \end{bmatrix}$  is called \_\_\_\_\_ matrix. (i)
  - (a) Unit

(b) Null

Symmetric (c)

- (d) Skew-symmetric
- For matrices A and B, product AB is possible only if two matrices are\_ (ii)
  - $A_{m \times n}$  and  $B_{m \times n}$ (a)

(b)  $A_{m \times n}$  and  $B_{n \times p}$ (d)  $A_{m \times n}$  and  $B_{m \times p}$ 

 $A_{m \times m}$  and  $B_{n \times n}$ (c)

- For position vector r of any point P in curvilinear coordinates  $\frac{\partial r}{\partial u} \times \frac{\partial r}{\partial v} = \underline{\hspace{1cm}}$ (iii)

- $\frac{h_1h_3}{h_2}\frac{\partial r}{\partial w}$   $\frac{h_2h_3}{h_1}\frac{\partial r}{\partial w}$   $\frac{h_2h_3}{\partial w}\frac{\partial r}{\partial w}$   $\frac{h_3h_2}{\partial w}\frac{\partial r}{\partial w}$   $\frac{h_3h_2}{\partial w}\frac{\partial r}{\partial w}$
- $\frac{d^2y}{dx^2} 2x\frac{dy}{dx} + 2vy = 0 \text{ is called } \underline{\hspace{1cm}}$  (Here v is a parameter)
  - (a)
- Legendre's differential equation (b) Bessel's differential equation
  - Hermite differential equation (d) None of these (c)
- For Legendre's equation, \_ (v)
  - k=1 or k=-1(a)
- (b) k = n or k = -n
- k = n or k = n 1(c)
- (d) k = n or k = -n 1

(vi)  $P_n(-\mu) = \underline{\hspace{1cm}} P_n(\mu)$ .

(a)

(b) -

(c)  $(-1)^n$ 

(d)  $(-\mu)^{\dagger}$ 

(vii) In complex representation of a Fourier series,  $\alpha_n =$ 

(a)  $(a_n + a_{-n})$ 

(b)  $(a_n-a_{-n}) = (a_n-a_{-n}) =$ 

(c)  $i(a_n + a_{-n})$ 

(d)  $i(a_n-a_{-n})$  . The second  $i(a_n-a_{-n})$ 

(viii) The heat equation is

(a)  $\frac{\partial u}{\partial t} = h^2 \nabla u$ 

(b)  $\frac{\partial u}{\partial t} = h \nabla^2 u$ 

(c)  $\frac{\partial u}{\partial t} = h^2 \nabla^2 u$ 

(d)  $\frac{\partial^2 u}{\partial t^2} = h^2 \nabla^2 u$ 

(ix)  $y = ax^2 + bx + c$  is the equation of \_\_\_\_\_.

(a) Parabola

(b) Ellipse

(c) Straight Line

(d) None of these

(x) The shift operator E is defined as

(a) Ef(x) = f(x)

(b) Ef(x) = f(x+h)

(c) Ef(x) = f(x - h)

(d) Ef(x) = f(h-x)

Q-2 Answer the following questions in short. (Attempt any ten)

20

Define curvilinear co-ordinates.

(2) Write condition of orthogonality for curvilinear coordinates.

(3) Define transpose of a matrix and inverse of a matrix.

(4) For Legendre's function, prove that:  $nP_n(\mu) = (2n-1)\mu P_{n-1}(\mu) - (n-1)P_{n-2}(\mu)$ .

(5) For Bessel's function, prove that:  $xJ'_n(x) = nJ_n(x) - xJ_{n+1}(x)$ .

(6) Write generating function for Hermite polynomials  $H_n(x)$ .

(7) Write sine series for f(x) when  $0 \le x \le \pi$ . (Note: derivation is not required)

(8) Give any two physical applications of Fourier series. (Note: derivation is not required)

(9)	٧	Vrite telegraphy equation.		
(10)	. <sub></sub> , C	Convert $y = ax^b$ in to equivalent equation of a straight line.		
(11)	Define and discuss interpolation.			
(1 <u>2</u> )	·F	for shift operator $E$ , prove that $\Delta = E - 1$ .		
Q-3	(a)	Derive expression of divergence in terms of orthogonal curvilinear system.	6	
	(b)	If $u = x + 4$ ; $v = 3y + 1$ ; $w = 2z - 3$ , show that $u, v, w$ orthogonal and find $ds^2$ and	4	
		the metrical coefficients: h <sub>1</sub> , h <sub>2</sub> , h <sub>3</sub> .		
		or the state of th	***	
Q-3	(a)	Discuss cylindrical co-ordinates as a special curvilinear system.	6	
	(b)	If $x = uv \cos w$ , $y = uv \sin w$ , $z = \frac{1}{2}(u^2 - v^2)$ ; find $h_1, h_2, h_3$ and show that	4	
		$ds^2 = (u^2 + v^2)(du^2 + dv^2) + uv  dw^2.$		
Q-4	(a)	Derive the series solution of Legendre differential equation in the form of	6	
		descending power of x.		
	(b)	Derive the generating function for Legendre's polynomials using relation	4	
		$V = (1 - 2\mu h + h^2)^{-\frac{1}{2}}.$		
		OR OR		
Q-4	(a)	Derive the series solution of Bessel's differential equation in the form of ascending	6	
		power of $x$ .		
	(b)	For Hermite polynomial using equation $H_n(x) = e^{x^2}(-1)^n \frac{d^n}{dx^n} (e^{-x^2})$ , calculate	4 '	
		Hermite polynomials $H_0(x)$ , $H_1(x)$ .		
		Thermite polynomials $H_0(x)$ , $H_1(x)$ .		
Q-5	(a)	Write the Fourier series for a periodic function $f(x)$ defined in the interval	6	
	` '	$[-\pi, \pi]$ . Derive the coefficients $a_0, a_n$ and $b_n$ of the series.		
	(b)	For one dimensional flow of electricity in a long insulated cable, derive one	4	
		dimensional diffusion equation.		
		OR		
Q-5	(a)	Find the Fourier series for a function $f(x)$ in the interval $(-\pi, \pi)$ .	6	
		Where $f(x) = \pi + x$ when $-\pi < x < 0$ ,		
		$f(x) = \pi - x \text{ when } 0 < x < \pi.$		
	(b)	Discuss Fourier series involving phase angles.	4	

Q-6 Derive Newton's backward difference interpolation formula for interpolation of any function y = f(x) with equal spaced values of x.

Evaluate f(45) from the following table of values.

x	10	20	30	40	50	
y = f(x)	46	66	81	93	101	

OR

Q-6 Write Simpson's 1/3 rule for integration.

10

Using Simpson's 1/3 rule, find the approximate value of  $y = \int_0^{\pi} \sin x \, dx$  by dividing the range of integration into ten equal parts.

What is the analytical value of  $y = \int_0^{\pi} \sin x \, dx$ .



ganter i selecció aporte, organisti en el como en la como en la como en la como en la general.

gallerians of the first angle of the engineers of the engineers of the engineers of the engineers of the engineers.

englander i de la companya de la companya de la companya de la companya de la companya de la companya de la co La companya de la companya de la companya de la companya de la companya de la companya de la companya de la co

and the second of the second o

organização de la completa de la propriação de la completa del completa de la completa de la completa de la completa del completa de la completa del la completa del la completa del la completa del la completa de la completa del la completa del la completa del la completa del la completa del la completa del la completa d

the page of the theory of the

the state of the s

n de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la companya de la companya de la companya de la companya de la companya de la companya de la co

> 。 1975年 - 李紹子《新春》(李紹寶)(1975年)(明寶)(1985年)

Sardar Patel University, Vallabh Vidyanagar – 388120 B Sc [Semester-V<sup>th</sup>] Subject Physics Course Code No: US05CPHY03

Subject/Course Title: Solid State Physics

Saturday, Date 11-11-2017

Time: 10.00 am to 01.00 pm

Total Marks-70

	Total M	41 K3-70
O 1	Multiple Choice Question [Attempt all]	10
Q-1	What is the wavelength of X-ray?	
<b>.</b>	(a) 1 A (c) 1 cm	
	(b) 1m (d) 1nm	
2	Electron diffraction are used to determine	-
L	(a) Structure properties (c) Surface properties	
	(b) Strength (d) Color of the material	
3	When the crystal is made rotate the direct and reciprocal lattice	
3	(a) Rotates (c) Not rotate	
1. 3.1.	(b) One rotates while other (d) Both are stationary	
	not rotates	:
4	Drude model of free electron theory cannot explain	٠,
	(a) thermal conductivity (c) Not sure	
	(b) Electrical conductivity (d) Electron heat capacity & paramagnetic	•
	susceptibility	٠
5	Type – II superconductor are known as	•
_	(a) Soft superconductors (c) No name	
	(b) Hard superconductors (d) Only the name	•
6	Fermi-Dirac distribution function is applicable to	
	(a) Spin particle (c) Spin with nair integral	·
	(b) Spin with opposite sign (d) Zero spin	
7	At a equilibrium conditions, the rate of generation of electron-hole pair and ra	te
	of recombination are?	
	(a) Unpredictable (c) Infinite	
	(b) Same (d) Zero	
8	p-type semiconductors can be made usingimpurity	
	(a) Tetravalent (c) Pentavalent	
	(b) Divalent (d) Trivalent	
9	Nanotechnology is the engineering of the structure less than size	
	(a) 100 nm (c) 100 Amp	
	(b) 100 Cm (d) 100 C	
10	The colour of the nano materials changes with the thickness because of	
	(a) Surface to volume ratio (b) mass to electron ratio (c) Density to volume ratio (d) Gravity to volume ratio	,
	(b) mass to electron ratio (d) Gravity to volume ratio	
Q-2	Attempt any ten questions in brief.	20
-		
•	1. When electron diffraction techniques are used.	
	2. When neutron diffraction techniques are used.	
	3. What is studied in powdermethod?	A 0 == - ×
	4. What is band—effective mass?	COTOS

		<ol> <li>What is type-I superconductor?</li> <li>What is Fermi energy?</li> <li>What are extrinsic semiconductors?</li> <li>What is photoelectric effect?</li> <li>What is photovoltaic effect?</li> <li>What is Nano technology?</li> <li>State Moore's first and second law.</li> <li>Define: Self-Healing Structures.</li> </ol>	
Q-3	(a)	Describe rotating crystal method for determination of crystal structure in detail.	6
	(b)	Explain the geometrical construction of reciprocal lattice.	4
Q-3	(a)	Discuss the Ewald construction of X-ray diffraction.	6
	(b)	When X-ray diffraction techniques are used?	4
Q-4	(a)	What is Wiedmann-Franz law, derive it.	6
	(b)	Discuss the effect of temperature on Fermi-Dirac distribution function.	4
Q-4	(a)	Derive the energy levels of free electron using Schrodinger equation in three dimensions.	6
	(b)	Explain the effect of heat capacity on superconductors.	4
Q-5	(a)	Discuss in detail intrinsic and extrinsic semiconductors.	10
		OR	
Q-5	(a)	Derive necessary equation for free carrier concentration in semiconductors.	10
Q-6	(a)	What is dip pen lithography.	5
	(b)	What is atomic force microscopy?	5
Q-6	(a) (b)	OR What are smart materials? What are nano sensors?	5 5
	•	_X - (2)	

#### T.Y.B.Sc EXAMINATION. 5<sup>TH</sup> SEMESTER

Subject: PHYSICS

Sub.Code: US05CPHY04

### Thermodynamic & Statistical physics

Date: 13/11/2017

Session: Morning

Day: Monday

Time:10.00am TO 1.00pm

Total marks: 70

### Q-1. Multiple choice questions

10

- 1. In first order phase transition process
- (A) No change in heat entropy & volume
  - (B) Change in heat entropy & volume
  - (C) Change in heat but no change in entropy & volume
  - (D) No change in heat but change in entropy & volume
- 2. Ehrenfest's equation is.....

$$(A)\frac{dP}{dT} = \frac{\alpha_{2-\alpha_{1}}}{K_{2-k_{1}}} \quad (B) \frac{dP}{dT} = \frac{K_{2-K_{1}}}{\alpha_{2-\alpha_{1}}} \quad (C) \frac{dP}{dT} = \frac{K_{1-\alpha_{1}}}{K_{2-\alpha_{2}}} \quad (D) \frac{dP}{dT} = \frac{K_{2-\alpha_{2}}}{K_{1-\alpha_{1}}}$$

- 3. Gibbs function is given by
  - (A) H=U+W
- (B) G=h-TS
- (C) h=U+PV (D) F=U-TS

- 4. The stirling formula is ℓn N!=
  - (A) N  $\ln \left(\frac{e}{N}\right)$

- (B) N  $\ln N-N$  (C)N  $\ln n$  (D)e  $\ln \left(\frac{N}{e}\right)$
- 5. Which is the factor for Gibbs paradox
  - (A) PNK ln (P) (B) PK ln (N)

- (A) PNK  $\ell$ n (P) (B) PK  $\ell$ n (N) (C) PN  $\ell$ n (K) (D) None of above 6. Mean Kinetic energy of a particle is  $\langle E \rangle$ .....
- (B)  $\frac{5}{2}$  KT
- (C)  $\frac{1}{2} KT$  (D)  $\frac{7}{2} KT$
- 7. Which of the three physical parameters remain constant in the system of interest in canonical ensemble
  - (A) N, V, T
- (B) E,P, $\mu$
- (C)N,P,S
- (D)None of above
- 8. In Fermi-Dirac system constituent particle of the gas are ...... From one another
  - (A) Distinguishable

- (C) Fixed
- (B) Indistinguishable
- (D) None of above
- 9. When constituent particle of a gas ...... With one another, it terms as an ideal gas
  - (A) Interact

(C) Highly interact

(B) Do not interact

- (D) None of above
- 10. Most probable velocity of particle is......

(A) 
$$\vartheta_{mp} = \sqrt{\frac{2KT}{m}}$$
 (B)  $\vartheta_{mp} = \sqrt{\frac{3KT}{m}}$  (C)  $\vartheta_{mp} = \sqrt{\frac{5KT}{m}}$  (D) None of above

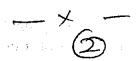
### Q-2. Short Questions (Any ten)

- 1. Explain ferromagnetic material.
- 2. Give the relation between Helmholtz function, Gibbs function & Enthalpy.
- 3. State second law of thermodynamic in terms of entropy.
- 4. Define : μ-space & Γ-space.
- 5. State Nernst's heat theorem.

(P.T.02)

6. Explain Macroscopic and microscopic state. 7. Define: Degeneracy & chemical potential. 8. Show that mean kinetic energy of a particle per degree of freedom is  $\frac{1}{2}$  KT. 9. Explain canonical ensemble. 10. State the Pauli exclusion principle. 11. What is Fermi Dirac gas and Einstein Boltzmann gas. 12. In which category, constituent particle of the gas are distinguishable from one another Q-3(a) Obtain the clausius Clapeyron's latent heat equation for first order phase transition.  $\frac{dP}{dT} = \frac{L}{T(V_2 - V_1)}$ (b) Explain first and third law of Thermodynamics. 3 (a) Obtain Maxwell's thermodynamical equations. (b) Explain Enthalpy with necessary formula. **Q-4(a)** State and prove Liouville's theorem (b) Show that in a steady state probability density is independent of the coordinates of the phase space OR (a) What is Gibbs paradox in microcanonical ensemble? How it is removed? 7 (b) Show that Sackur-Tetrode formula for entropy of a perfect gas is (1) Free frome the Gibbs paradox and (2) It violates the third law of thermodynamics. 3 Q-5(a) Derive formulas for thermodynamical quantities for an ideal gas in canonical ensemble. (b) Discuss equivalence of microcanonical and canonical ensemble. 3 (a) Obtain an expression for Maxwell's distribution of velocities of the particle. (b) Obtain the formula for the most probable velocity of particle. **Q-6(B)** For the M.B. distribution of the particles obtain formula  $ni = gi e^{-\alpha - \beta Ei}$ . In M.B. distribution gas has two particles in the ith state whose degeneracy is three. Find out the number of independent ways of selecting the particles in the state. OR Obtain the expression for the F.D. distribution of particle among various

Obtain the expression for the F.D. distribution of particle among various states. A Fermi-Dirac gas has two particles in the i<sup>th</sup> state whose degeneracy is three. Find the number of independent ways of selecting the particle in the state.



No. of Printed Pages: 03 Sardar Patel University

[43/A-17]

Vallabh Vidyanagar - 388120

BSc [Semester-V]

Subject Physics Course Code No: US05CPHY05

CBCS (Regular and NC All)

Subject: Physics Wednesday, Date 15-11-2017

Title of the Paper: Analog Devices and Circuits

Time: 10.00 am to 01.00 pm **Total Marks-70** 

#### Multiple Choice Questions: [Attempt all] Q-1

A JFET has  $I_{DSS}$  = 10 mA and  $V_P$  = 4 V, then  $V_{GS (off)}$  = \_ (1)

(a) 4 V

(b) -4 V

(c) 40 V

A JFET has  $I_{DSS}$  = 10 mA and  $V_P$  = 4 V, then  $R_{DS}$  = (2)

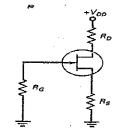
(a)  $400 \Omega$ 

(b)  $500 \Omega$ 

(c)  $600 \Omega$ 

(d) 700 Ω

(3)



This circuit diagram shows \_\_\_\_\_\_of FET.

- (a) Gate bias
- (b) voltage divider bias
- (c) Two supply source bias
- (d) Self bias
- (4)The correct sentence is
  - (a) For CE configuration, he is always negative and hie is always positive.
  - (b) For CE configuration, hie is always negative and hie is always positive.
  - (c) For CE configuration, hie and hie both are always positive.
  - (d) For CE configuration, hie and hie both are always negative.
- The lower cut off frequency of the CE transistor amplifier (f1) will be the (5) highest if we select  $C_e = \mu F$ .

(a) 10

(b)

(c) 30

(d) 40

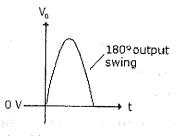
If  $r_{bb}\!=\!800~\Omega$  and  $r_{b'e}\!=\!200~\Omega$  then according to hybrid  $\pi\text{-model}$  to study (6)the high frequency response of CE amplifier  $h_{ie}\text{=---}\Omega$ 

(a) 700

(c) 900

1000

**(7)** 



This is an example of the output swing for a \_\_\_\_\_ push pull

amplifier.

Class A (a)

(b) Class B (c) Class C

(d) Class AB

(8)	In push pull power amplifiers, the output signal varies for a full 360° of the cycle.					
	(a) Class A (b) Class B (c) Class C (d) Class AB					
(9)	For the ideal inverting amplifier using OpAmp given that $R_i$ =2K $\Omega$ and $R_i$ =20K $\Omega$ . Voltage gain of the OpAmp is					
in extra en Mala	(a) 10 (b) -10 (c) 11 (d) -11					
(10)	For the ideal non-inverting amplifier using OpAmp given that $R_1$ =2 $K\Omega$ and $R_1$ =20 $K\Omega$ . Voltage gain of the OpAmp is					
	(a) 10 (b) -10 (c) 11 (d) -11					
Q-2	Answer any TEN questions in short.					
(1)	Draw schematic symbols of (i) n-channel JFET and (ii) p-channel					
	JFET.					
(2)	Compare construction of the depletion-mode MOSFET and the					
	enhancement-mode MOSFET.					
(3)	What are the advantages of JFET compared to a BJT?					
(4)	What are the factors on which high frequency response of transistor					
	amplifier depend?					
(5)	What is the difference between ordinary amplifier and tuned amplifier?					
	Discuss classification of small signal tuned amplifier.					
(6)	Discuss the effect of an emitter bypass capacitor on low frequency					
	response of transistor amplifier.					
(7)	What is the main drawback of class B amplifier? How it is going to					
	overcome using class AB push pull amplifier?					
(8)	Define conversion efficiency of an amplifier. What is the maximum					
	conversion efficiency of class B push pull amplifier?					
(9)	What is complementary symmetry?					
(10)	Why an Op-amp is called as operational amplifier? Draw the					
nargaans Bartijoonis	schematic symbol for Op-amp.					
(11)	Calculate the output voltage of an OpAmp inverting adder for the					
	following sets of input voltages and resistors. In all cases $R_f$ = 500 k $\Omega$					
	$V_1$ =3 $V$ , $V_2$ = 1 $V$ , $R_1$ = 250 $k\Omega$ and $R_2$ = 500 $k\Omega$ .					
(12)	Draw the diagram of integrator and differentiator using OpAmp.					

Q-3 (a) Draw and discuss the drain curves and transconductance 7 curves of JFET. (b) Define transconductance of FET. Calculate transconductance of 3 the FET in the following cases: (i) If  $i_d$  = 0.2 mA pp when  $v_{gs}$  = 0.1 V pp and (ii) If  $i_d = 1$  mA pp when  $v_{gs} = 0.1 \text{ V pp}$ . Comment on the result. OR (a) Discuss two types of JFET analog switch. (b) Draw the circuit diagram of current-source biasing of JFET and describe it. (a) Derive the following amplifier equations: 7 (1) Current gain  $A_i = \frac{-h_f}{1 + h_0 R_L}$  (2) Input resistance  $R_i = h_i + h_r A_i R_L$ (3) Voltage gain  $A_i = \frac{A_i R_L}{R_i}$ . (b) Discuss effect of coupling capacitor on low-frequency response 3 of CE transistor amplifier. OR Q-4 (a) Discuss high frequency response of CE transistor amplifier. 7 Draw diagram, which shows (i)  $\alpha$  cut of frequency (ii)  $\beta$  cut off frequency and (iii) gain bandwidth product (fr). (b) List four h-parameters. Define and explain any one. 3 (a) Write a note on harmonic distortion. How even harmonics is 7 Q-5 eliminated using Class A push-pull circuit, derive the (b) What is the importance of transistor phase inverter? Draw the circuit 3 diagram and describe it. OR (a) Explain the classification of push pull power amplifiers based 7 Q-5 on class of operation and compare them. (b) List the criteria for designing power amplifier. 3 Q-6 Drawing AC equivalent circuit of differential amplifier and 10 hence derive expressions for gain of the amplifier in difference and common mode configurations. Define and explain the following Op-Amp parameters and 10 Q-6 describe universal balancing techniques to determine such parameters: (i) Input offset voltage (ii) PSRR and (iii) Input bias current.

.

•

en degree in die de land de land de land de land de land de land de land de land de land de land de land de la Her groupe de land de land de land de land de land de land de land de land de land de land de land de land de Land de land de land de land de land de land de land de land de land de land de land de land de land de land de

.

. []

No. of pages: 2

## SARDAR PATEL UNIVERSITY B. Sc, 5th Semester (Under CBCS) Friday, 17th November 2017

Session: Morning, Time: 10:00 AM to 01:00 PM

Subject Code: (PHYSICS) US05CPHY06 Course Title: Astronomy and Astrophysics

N.	B: i All symbols have their usual meaning. ii Figure at the right side of the question	Max Ma	rks: 70
W	rite correct answer for each of the follow	ng MCQs.	[10]
1	Diffraction grating is a surface consisting of lines.	f large number of finely placed closed	d
	a) Transmission	b) Equidistance	
	c) Dispersion	d) None of above	
2	Photometry is used for recording theo		
	a) Image	b) Brightness	
	c) Spectrum	d) Photograph	
3	Projected chromospheric streamers are norm		1
	a) Spicules	b) Faculae	
	c) Plages	d) Filaments	
4	prominence is not a type of promine	•	2.7
	a) Quiescent		
	c) Inactive	b) Eruptive	
5	is not a type of binary stars.	d) Sunspot type	
		en en et en en en en en en en en en en en en en	
	a) Eclipsing	b) Spectroscopic	
,	c) Visual	d) Spectrometric	
6	The thermal ionization process in stars does r	ot depend on	
	a) Mass	b) Density	
	c) Pressure	d) Temperature	
7	The Hertzsprung-Russell diagram represent v	ariation of .	
	a) Temperature vs. Intensity	b) Intensity of balmer lines vs. Temperature	
	c) Absolute magnitude vs. Spectral	d) none of these	
	type for all stars	es are	
8	The principal 'actors' of highly interesting ga	lactic drama are	
	a) Galaxies	b) Comets	
	c) Stars	d) Cosmic rays	
9	The line absorption is produced by the distant	interstellar gas in the spectra of	
	a) Hot stars	b) Galaxies	
	c) Granules	d) Clusters	
10	Positive radial velocities are observed in stars		

b) 45°and 255°

d) 315°

a) 90° and 270°

c) 135°

Que 2	Write	e answers of any ten questions in brief.	[20]
•		What is f/a ratio of telescope?	
	$\hat{\overline{2}}$	What is stellar magnitude sequence?	
	3	What do you mean by astronomical spectrograph?	
-	4	With the help of graph explain the temperature distribution near the	
		photosphere boundary.	
	5	Define Plages and Filaments.	
	6	Define solar wind.	
	7	Write notes on optical double stars and binary stars.	
	8	Draw light curves from different binaries.	
	9	Write a note on Sirius.	
	10	Define 'hyperfine splitting'.	
	11	Draw a diagram explaining the structure of our galaxy.	
	12	What is REDDENING EFFECT of distant stars in the plane of galaxy?	
Que 3	[A]	What is optical telescope? Discuss various factors of optical telescope.	[06]
Que 3	[B]	Write a note on Photoelectric photometry.	[04]
	[ոյ	OR See	
Que 3	[A]	Write short notes on earth's atmosphere and electromagnetic radiation.	[06]
Que 3	[A] [B]	Explain only construction and working of slit spectrograph and slit less spectrograph.	[04]
		gu de la company de la company truo prominences in detail	[06]
Que 4	[A]	Classify prominence and discuss any two prominences in detail.	[04]
	[B]	Write a detail note on 'Chromosphere'.  OR	[]
		,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的,我们就是一个大型的	[06]
Que 4	[A]	Write a note on 11 year cycle of sunspots.  How are various forms of solar activities related to sun's magnetic field?	[04]
	[B]	How are various forms of solar activities related to sun a magnetic related	
		D. 1 A.4-11 II D. Alegaram of stars	[10]
Que 5		Explain in detail H-R diagram of stars.  OR	1 3
_			[10]
Que 5		Derive Boltzmann's formula in logarithmic form for rth level.	_
Que 6	[A]	Explain in brief the radio observation of galaxy at 21-cm wavelength.	[06]
<b>Q</b>	[B]	C 1 0	[04]
	[~]	OR OR	
	[A]	Enlist the rotational parameters of galaxy and discuss them.	[06]
Que 6			[04]



No. of Printed Pages: 02

## SARDAR PATEL UNIVERSITY

### B. Sc. (ZOOLOGY) - Fifth Semester Examination

Tuesday, 7<sup>th</sup> November, 2017 10:00 a.m. to 1:00 p.m.

US05CZOO01: INVERTEBRATA

	1) Figures to th 2) Draw diagra	=					. :
	2) Diaw Giagia	iii wherever her	005541 y			Total marl	cs: 70
Q-1	Multiple Cho	ice Questions:			,		10
1. Aris	stotle is the fath	er of	classificatio	on?		•	
	a) biological	b) animal	c) plant	d) none			
2. The	Greek word -t	axi means					
	a) taxology	b) taxies	c) taxonomy	d) All			
3. Scie	entific name of	plant and anima	al is written in.	language	e		
	a) English	b) French	c) Latin	d) All			
4. Wha	at type of asexu	al reproduction	n is found in po	rifera?		-	
	a) by budding	b) Fert	ilization	c) a & b	d) none	•	
5. Cor	al polyps secret	e					
	a) Mgso4	b) H <sub>2</sub> SO <sub>4</sub>	c) CaCO <sub>3</sub>	d) CuSO4			
6. Whi	ich of the follow	ving is a acoelo	mate and trible	oblstic animal?			
	a) porifera	b) protozoa	c) coelentrata	d) all of these			
7	is found	in male Ascari	S.				
	a) copulatory	setae b) pect	tin c) seta	ae d) none	;		,
8. Boo	k lung is found	in	· ,				
	a) ascaris	b) sea anemor	ne c) scor	pion d) none	<b>;</b>	÷	
9. Ink	gland is found	in					
	a) Limulus	b) Spider	c) Starfish	d) Sepia			
10. To	rsion found in	phy	ylum				
	a) protozoa	b) mammals	c) arthropod	d) mollusks			

Q-2	Short Questions: [any TEN]					
	1. Write the contributions of Lamarck in animal taxonomy					
	2. Draw Linnaeus animal classification					
	3. Write importance of classification					
	4. Draw diagramme of coral polyps					
	5. Write characters of metazoan					
	6. Define coelenterate with example					
	7. Write only name of sensory organs of scorpion					
	8. Write characters of Annelid					
	9. Classify scorpion giving reasons					
	10. Write characters of crustaceans.					
	11. Classify Limulus giving reasons					
	12. Write characters of mollusk					
Q-3	a) Explain animal Taxonomic Hierarchy with suitable example.	06				
	b) Write characters of protozoa.	04				
	OR					
Q-3	a) Write detail note on history of animal taxonomy.	07				
	b) Use of Electron microscopy in taxonomy.	03				
Q-4	Describe Polymorphism in coelenterate.	10				
	OR					
Q-4	a) Reproduction in Porifera.	05				
	b) Economic importance of corals.	05				
Q-5	Explain reproductive system of Ascaris.	10				
	OR					
Q-5	Describe digestive system of scorpion with digestion.	10				
Q-6	Write detail note on Digestive system of Sepia	10				
	OR					
Q-6	a) Write a note on zoological importance of peripatus.	05				
	b) Write note on pedicelaria in star fish.	05				

\*\*\*\*\*\*\*\*\*\*\*\*\*\*\*\*

## [58]

### SARDAR PATEL UNIVERSITY

### **B.Sc. SEMESTER-V, EXAM-2017**

## **ZOOLOGY, US05CZOO02 (COMPARATIVE ANATOMY OF CHORDATES)**

DATE	: 9-11-2017		TIME: 10 am TO 1pm			n		
DAY:	THURSDAY			MARI	<b>∢S: 70</b>			
Q-1.	MULTIPLE CHOICE Q	UESTIONS.			(1	LO)		
1.	The cells die & are di	scharged their secretion	on is known as _		glands.			
	(a) Holocrine (b) Ap	ocrine (c) Me	erocrine (d) End	locrine				
2.	Most teleost fishes co	ontained which type o	f scales?					
	(a) Placoid	(b) Ganoid	(c) Cycloid	•	(d) Ctenoid			
3.	Which feather usually	Which feather usually forms a worm insulating layer?						
	(a) Contour	(b) Down	(c) After shaft	(d) Hai	r			
4.	In, the ton	igue is attached anteri	orly at the marg	gin of th	ne jaw.			
	(a) Frog	(b) Snakes	(c) Turtle		(d) Salamander			
5.	Oral hood is the char	acteristic of						
	(a)Amphibians	(b) Amphioxus	(c) Fishes		(d) Birds			
6.	The crop glands are p	resent in						
•	(a) Cyclostomes	(b) Fishes	(c) Amphibians	5.	(d) Birds			
7.	The ruminant stomac	h is the characteristic	of					
	(a) Amphibians	(b) Reptiles	(c) Aves	(d) Her	bivorous mamma	3		
8.	In thepylor	ic caeca is present.						
	(a) Shark	(b) Mackerel	(c) Sting ray		(d) Sea horse			
9.	Azygos vein is presen	t in						
	(a)Amphioxus	(b) Cyclostomes	(c) Reptiles		(d) Mammals			
10.	Name the second aor	tic arch						
	(a) Systemic	(b) Mandibular	(c) Hyoid		(d) Pulmonary			

Q-2.	ANSWER IN SHORT (ANY TEN).	(20)
1. 2. 3. 4. 5. 6. 7. 8.	Classify the glands according to type of secretions they produce. Write about the carapace and plastron of Turtle. Write about unicellular glands. Comment upon the epidermal teeth. Draw and labeled the diagram of Premolar tooth. Write about the lips of vertebrate. Write about intestine of Amphibians. Explain the structure of pancreas. Write about the stomach of Fishes.	
10.	Explain the working of heart in Human.	
11. 12.	Write about veins of Amphioxus.  Explain the terms hepatic portal and renal portal circulation.	
Q-3.	(a) Write about types of dermal scales found in Fishes.	(5)
	(b) Describe the structure of contour feather.	(5)
	OR	
Q-3.	(a) Write about the comparative account of glands of Mammals.	(5)
	(b) Describe the types of beaks.	(5)
Q-4.	(a) Describe the comparative anatomy of glands of oral cavity.	(7)
	(b) Describe the tongue of Reptiles.	(3)
	OR	
Q-4.	(a) Describe the comparative account of teeth of Mammals.	(7)
	(b) Explain the types of attachment of teeth.	(3)
Q-5.	(a) Describe the comparative anatomy of Mammalian stomach.	(7)
	(b) Describe the structure of Human liver.	(3)
	OR	
Q-5.	(a) Write about the comparative anatomy of vertebrate oesophagus.	(7)
	(b) Describe the intestine of Fishes.	(3)
Q-6.	Describe the comparative anatomy of heart with labeled diagrams.	(10)
	OR	
Q-6.	Describe the comparative anatomy of aortic arches with labeled	
	diagrams.	(10)



No. of Printed Pages: 2

### SARDAR PATEL UNIVERSITY

B.Sc. - I Semester, NOVEMBER-2017

CELL BIOLOGY US 05 CZOO 03

TIME: 10.00 AM -1.00 PM

DATE: 11/11/2017, SATURDAY

**MARKS: 70** 

		•		
Q-1 MU	LTIPLE CHOICE QUESTIO	NS:		[10]
1.	GENERALLY LIGHT MICI	ROSCOPE IS USE TO OBSERVE		
	A, CELL	B, CELL ORGANELLES	C, CELL DIVISION	D, ALL
2.	ACIDOPHILIC TISSUE HA	AVE AFFINITY TO	+	
	A, BASIC STAIN	B, ACIDIC STAIN	C, NEUTRAL STAIN	D, NONE
3.	FLUID- MOSAIC MODEL	. OF PLASMA MEMBRANE CONST	ITUENTS	
	A, LIPID	B, EXTRINSIC PROTEIN	C, INTRINSIC PROTEIN	D, ALL
4.	ENDOPLASMIC RETICU	LUM IS ABSENT IN		
	A, RBC	B, WBC	C, PLATELETS	D, NEURON
5.	WHICH IS GIANT CHRO	MOSOME		
	A, LAMP SHADE	B, LAMP GLASS	C, LAMP STAND	D, LAMP BRUSH
6.	ANEUPLOIDY IS THE VA	ARIATION OF CHROMOSOMES IN		
	A, n	B, n-n	C, n+2n	D, n+2
7.	CISTERNAE IS A PART C	)F		
	A, RIBOSOME	B, NUCLEUS	C, GOLGI BODY	D, CENTRIOLE
8.	CILIA ORIGINS FROM			
	A, CENTRIOLE	B, BASAL BODY	C, RIBOSOME	D, MICROBODIES
9.	PROKARYOTIC RIBOSO	MES ARE KNOWN AS		
	A, 50S	B, 60S	C,70S	D,80\$
1.0.	RIBOSOME IS MAINLY			
	A, DNA	B, PROTEIN	C, RNA	D, LIPID
			•	

- 1. WRITE DEHYDRATION METHOD IN MICROTOME
- 2. WRITE ABOUT LEEUWENHOEK MICROSCOPE
- 3. WRITE ABOUT HPLC

Q-2 ANSWER IN SHORT [ANY TEN]

- 4. DRAW A LABELED DIAGRAM OF PLASMA MEMBRANE
- 5. STRUCTURE OF CISTERNAE IN ENDOPLASMIC RETICULUM
- 6. WRITE ABOUT ACTIVE PERMEABILITY
- 7. DRAW A STRUCTURE OF METACENTRIC CHROMOSOME
- 8. WRITE ABOUT TRISOMY IN HUMAN
- 9. WRITE ABOUT DELETION IN CHROMOSOMAL VARIATIONS
- 10. STRUCTURE OF FLAGELLA
- 11. WRITE ABOUT NUCLEUS
- 12. WRITE FUNCTIONS OF PEROXISOMES

-X-



[20]

Ų-3	B, DESCRIBE TYPES OF LIGHT MICROSCOPES	[05] [05]
	OR	
Q-3	A, EXPLAIN RADIO INIMUNO ESSAY B, EXPLAIN METHODS OF CELL FRACTIONATION	[05]
Q-4	A, EXPLAIN STRUCTURE AND FUNCTIONS OF LYSOSOME B, EXPLAIN FUNCTIONS OF PLASMA MEMBRANE OR	[05]
Q-4	A, EXPLAIN FUNCTIONS OF MITOCHONDRIA B, EXPLAIN STRUCTURE AND FUNCTIONS OF ENDOPLASMIC RETICULUM	[05] [05]
Q-5	DESCRIBE GIANT CHROMOSOMES	[10]
	OR	,
Q-5	DESCRIBES NUMERIC ABNORMALITIES IN CHROMOSOMES	[10]
Q-6	EXPLAIN STRUCTURE OF BASAL BODY AND CILIA	[10]
	OR	
Q-6	DESCRIBE STRUCTURE OF RIBOSOMES AND ITS FUNCTIONS	[10]

-X-(2)

No. of Printed Pages: 2

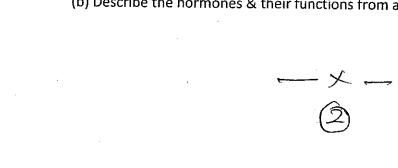
## SARDAR PATEL UNIVERSITY

### **B.Sc., SEMESTER-V EXAMINATION-2017**

### **ZOOLOGY, US05CZOO04 (ANIMAL PHYSIOLOGY)**

DATE:	: 13-11-2017	,	TIME: 10am to 1pm			
DAY:	MONDAY		MARKS: 70			
Q-1.	MULTIPLE CHOICE QU	JESTIONS.			(10	))
1.	The structural and fur	nctional unit of skeleta	l muscle is know	wn as		
	(a) Sarcolema	(b) Sarcomere	(c) Sarcoplasm		(d) SR	
2.	Name the sheath of c	onnective tissue that v	vraps around ir	dividua	il skeletal muscle	
	(a)Endomysium	(b) Epimysium	(c) Perimysium	1	(d) Tendon	
3.	Intercalated discs are	present in	_ muscle fiber.			
	(a) Smooth	(b) Skeletal	(c) Cardiac		(d) All of these	
4.	Gaps in the myelin sh	eath on an axon is Kno	wn as			
	(a) Neurolemma	(b) Nissl bodies	(c) Synapse	(d) No	de of Ranvier	
5.	Which component of	the brain regulates po	sture & balanco	e of bod	ly?	
	(a)Pons	(b) Cerebellum	(c) Hypothalar	nus	(d) Diencephalon	
6.	Pancreas is a	gland.				
	(a) Endocrine	(b) Exocrine	(c) Heterocrine	2	(d) None of these	ž
7.	The main function of	bile juice is				
•	(a)Dig.of lipids	(b) Emulsification of li	pids (c)Dig.of	orotein	(d) All of these	
8.	Which one of the foll	owings salivary gland i	s present in Hu	man?	·	
	(a)Parotid	(b)Submaxillary	(c) Submandib	ular	(d)All of these	
9.	Which hormone is re-	sponsible for maintain	calcium & pho	sphate l	level in the blood?	?
	(a) Parathormone	(b)Thyroid	(c) Epinephrin	e	(d)GH	
10.	Oxytocin hormone is	Synthesis from				
	(a) Neurohypophysis	(b) Adenohypophysis	(c) Thyroid		(d) Pancreas	

Q-2.	ANSWER IN SHORT (ANY TEN).	(20)
1.	Name the properties of muscle fiber.	
2.	Explain the term oxygen debt.	
3.	Draw & labeled the diagram of cardiac muscle fiber.	
4.		
	,	
12.		
	•	
Q-3.	(a) Write a note on proteins of skeletal muscle fiber.	(5)
	(b) Explain the neuromuscular junction (NMJ) with labeled diagram.	(5)
	OR	` ,
Q-3.	Describe the muscle metabolism.	(10)
Q-4.	(a) Explain the impulse transmission on non-myelinated nerve fiber.	(6)
	(b) Write about structure and functions of fore brain.	(4)
	OR	
Q-4.	(a) Write about number, name & function of human cranial nerves.	(6)
	(b) Describe the structure of reflex arc.	(4)
Q-5.	(a) Write a note on composition & functions of gastric juice.	(5)
	(b) Describe the functions of liver.	(5)
	OR	
Q-5.	Explain the absorption of food from gastrointestinal tract.	(10)
Q-6.	Describe the secretion, functions & abnormalities of thyroid hormone.	(10)
	OR	. ,
Q-6.	(a) Explain the action of lipid soluble hormone.	(5)
	(b) Describe the hormones & their functions from adrenal cortex.	(5)
	1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12.  Q-3.  Q-4.  Q-5.  Q-6.	<ol> <li>Name the properties of muscle fiber.</li> <li>Explain the term oxygen debt.</li> <li>Draw &amp; labeled the diagram of cardiac muscle fiber.</li> <li>Write about the morphological classification of nerve fiber.</li> <li>Name the different neurotransmitters.</li> <li>Explain the term conditional reflex with example.</li> <li>Write about any four functions of saliva.</li> <li>Explain the phases of gastric juice secretion.</li> <li>Write about the composition of bile.</li> <li>Define the term tropic hormone and name the pituitary tropins.</li> <li>Write about the functions of oxytocin hormone.</li> <li>Explain the role of hormone receptors.</li> <li>Q-3. (a) Write a note on proteins of skeletal muscle fiber.</li> <li>(b) Explain the neuromuscular junction (NMJ) with labeled diagram.         <ul> <li>OR</li> </ul> </li> <li>Q-3. Describe the muscle metabolism.</li> <li>Q-4. (a) Explain the impulse transmission on non-myelinated nerve fiber.</li> <li>(b) Write about structure and functions of fore brain.         <ul> <li>OR</li> </ul> </li> <li>Q-4. (a) Write about number, name &amp; function of human cranial nerves.</li> <li>(b) Describe the structure of reflex arc.</li> <li>Q-5. (a) Write a note on composition &amp; functions of gastric juice.</li> <li>(b) Describe the functions of liver.             <ul> <li>OR</li> </ul> </li> <li>Q-5. Explain the absorption of food from gastrointestinal tract.</li> <li>Q-6. Describe the secretion, functions &amp; abnormalities of thyroid hormone.</li> <li>OR</li> <li>Q-6. (a) Explain the action of lipid soluble hormone.</li> </ol>



SEAT No.\_\_\_\_

Se

No. of Printed Pages: 02

[44]

### SARDAR PATEL UNIVERSITY

B.S.C. (V-Sen) November-2017

Genetics

US 05 CZOO 05

TIME: 10:00 AM-1.00 PM

**DATE: 15/11/2017 WEDNESDAY** 

MARKS: 70

### Q-1 MULTIPLE CHOICE QUESTIONS:

[10]

1.	IN GENETIC INTERACTION	I F IS STAND FOR.		
	A, PARENT	B, GENERATION	C, GAMETE	D, ALLELE
2.	1:2:1 IS CRO	SS		
	A, DIHYBRID	B, MONOHYBRID	C, BACK	D, TEST
3.	WHICH CHARACTER IS CH	IOSEN BY MENDEL IN MONOHYBRI	D CROSSES?	
	A, SEED COLOUR	B, ROOT COLOUR	C, STEM COLOUR	D, NONE
4.	ALLELES FOR BLOOD GRO	OUP "B" ARE		
	A, I <sup>A</sup> I <sup>O</sup>	B, I <sup>B</sup> I <sup>O</sup>	C, I <sup>A</sup> I <sup>B</sup>	D, I <sup>o</sup> I <sup>o</sup>
5.	ANTIGENS ARE PRESENT	IN BLOOD GROUP O		
	A, NONE	B, AB	C, A	D, B
6.	BACK CROSS IS THE CROS	S WITH		
	A, FIRST GENERATION	B, ANY ONE PARENT	C, RECESSIVE PARENT	D, DOMINANT PARENT
7.	CHIASMATA TAKES PLACE	E DURING		
	A, PROPHASE	B, METAPHSE	C, ANAPHASE	D, TELOPHASE
8.	IN PEDIGREE O SYMBOL	IS FOR		
	A, MALE	B, FEMALE	C, CARRIER	D, CHILD
9.	FRATERNAL TWINS ARE P	RODUCED FROM		
	A, TWO ZYGOTE	B, TWO SPERMATOZOON	C, ONE ZYGOTE	D, ALL POSSIBILITIES
10.	DIAGRAM CHART OF CHI	ROMOSOMES IS		
	A, PEDIGREE	B, IDEOGRAM	C, KARYOTYPE	D, NONE

### Q-2 ANSWER IN SHORT [ANY TEN]

[20]

- 1. DEFINE LAW OF SEGREGATION
- 2. WRITE REASONS FOR SELECTION OF PEA PLANT BY MENDEL
- 3. WRITE ABOUT F1 FOR AA X AA
- 4. DEFINE EPISTASIS?
- 5. WHAT IS MONO HYBRID TEST CROSS
- 6. WRITE ABOUT CONGENITAL ICHTHYOSIS
- 7. WRITE ABOUT COMPLETE LINKAGE
- 8. WRITE ABOUT CHROMOSOMAL MECHANISM IN DROSOPHILA
- 9. WRITE ABOUT DOUBLE CROSSING OVER
- 10. EXPRESS MUTATION
- 11. HARMFUL EVOLUTIONARY EFFECT OF EUPHENICS
- 12. WRITE ABOUT DELETION MUTATION

Q-3	A, WHAT IS CO- DOMINANCE? EXPLAIN B, EXPLAIN MONOHYBRID CROSS IN DROSOPHILA	[05] [05]
	OR	
Q-3	A, EXPLAIN 9:3:3:1 RATIO B, EXPLAIN LETHAL GENES AND SUB LETHAL GENES IN HUMAN	[05] [05]
Q-4	EXPLAIN NON EPISTATIC GENETIC REACTION	[10]
	OR	
Q-4	EXPLAIN ALLELES OF SKIN COLOUR AND EXPLAIN WITH SUITABLE EXAMPLE	[10]
Q-5	A, EXPLAIN HETEROGAMETIC SEX DETERMINATION B, EXPLAIN EXPRESSIONS OF X LINKED GENES IN MAN OR	[06] [04]
Q-5	A, EXPLAIN MECHANISM ABOUT CROSSING OVER B, EXPLAIN ENVIRONMENTALLY SEX DETERMINATION	[06] [04]
Q-6	DESCRIBE SUBSTITUTION MUTATION OR	[10]
Ω-6	DESCRIBE SEX CHROMOSOMAL ABNORMALITIES	[10]

S

SEAT No.

No. of Printed Pages : 02

[37]

Sardar Patel University

### $B. \ Sc. \ Semester \ V \ Examination \ 2017$

### **ZOOLOGY**

US05CZOO06 (Environmental Biology) 17<sup>th</sup> November 2017, Friday

10:00 am to 1:00 pm

Q I. Multiple Choice Qu	estions.	Total Marks: 70 [10]
1. Hyena plays the role of	in ecosystem.	
A. Producers	B. Decomposers	
C. Scavengers	D. All of these	
2. In ecology, human is		·
A. Herbivorous	B. Carnivorous	
C. Omnivorous	D. None of these	
3 are primary	producers.	
A. Phytoplanktons	B. Zooplanktons	
C. Grass	D. Both A & C	
4. Which pyramid is not in	ncluded in ecological pyramids?	
A. Pyramid of numb		у
C. Pyramid of biom	ass D. Pyramid of territo	ory
5. Which of the following	is the full name of GAP?	
A. Ganga agricultur	al plan B. Ganga action plar	1
C. Ganga actual plan	D. none	
6. Due to agricultural deve	elopment, we lost our	
A. Forest	B. Food	
C. Rocks	D. None of these	
7. Which type of adaptation	ns are shown by camel?	
A. Desert	B. Arboreal	
C. Terrestrial	D. Aquatic	
8. Which of the following	shows hibernation behavior?	
A. Lion	B. Rat	
C. Fly	D. Snake	
9. Which of the following	is an air pollutant?	
A. Sulfur	B. Lead	
C. Carbon dioxide	D. All of these	
10. The animal responsible	for overgrazing is	
A. Tiger	B. Lion	•
C. Rabbit	D. None of these	

(	) II.	Answer the following in short.(Attempt any Ten)	[20
		Give the importance of ecology.	
	2.	the trace of faire ecosystem.	
	3. 4.	remainded with suitable examples.	
	4. 5.	and the short of t	
	5. 6.	The state of the s	
		Define overgrazing. Write names of ecosystems.	
	8.		
		and the state of t	
	10	List the important causes of ecosystem degradation.  What is radioactive pollution? Write pages of a line with the page of a line with the l	
	11	What is radioactive pollution? Write names of radioactive pollutants.  Draw a neat and labeled diagram of pyramid of biomass.	
	12	Write a short note on noise pollution.	
		a short hote on horse polition.	
Q	Ш	(a) Explain fresh water ecosystem with a labeled diagram.	<u>የ</u> ሰድነ
		(b) Describe the role of bacteria in ecosystem.	[05] [05]
		OR	[US]
Q	III.	What is ecosystem? Mention the types of ecosystems and describe forest ecosystem	
		in detail.	[10]
_	****		1
Q	1V.	(a) Describe water cycle in detail.	[07]
			[03]
_	<b>TT</b> 7	OR	
Ų	IV.	Explain ecological pyramids with suitable examples.	[10]
Ω	V.	Write about notes and	
V	٧.	Write short notes on: (a) Prevention of deforestation	
		(h) Effects of mining on natural helitate	[06]
		OR	[04]
0	V.	Describe intra and interpretation in the state of the sta	[ d
Ψ.		best recentification interspectific animal relationships.	[10]
Q	VI.	How can we save our rivers against pollution? Explain.	1101
		OR	[10]
Q	VI.	Write an essay on effects of global warming on ecosystem.	[10]
		-	L J
		X	