Seat No.:	
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[10]

SARDAR PATEL UNIVERSITY

	Bachelor of Science (Semeste US03CICV52: Fundamentals	·
Date:	: 21st November 2022, Monday	•
	: 10:00 AM to 01:00 PM	Total: 70 Marks
NOT	E: Figure to the right indicate full marks of the	e questions. 3.
Q-1	Multiple Choice Questions	[10]
1. W	hich of the following is less stable carbocation	n?
	(a) CH ₃ ⁺	
	(b) (CH ₃) ₃ C ⁺	
	(c) CH ₃ CH ₂ ⁺	
	(d) (CH ₃) ₂ CH ⁺	
2. Th	e decreasing order of "s" character in the three	e hybrid orbitals is
	(a) $sp>sp^2>sp^3$	(c) $sp^2 > sp > sp^3 d$
	(b) $sp^3 > sp^2 > sp$	(d) None of these
3. A	Nucleophile is	(-)
	(a) Lewis acid	
	(b) Electron deficient species	
	(c) An electron rich speciesd.	
	(d) None of these	
4. W	hat does CH3-CO-CH2-CH2-CH2-CH3's name	e stand for?
	(a) Hexan-1-one	(c) Pentan-1-one
	(b) Hexan-2-one	(d) Pentan-2-one
5. Re	sorcinol on distillation with zinc dust gives	
	(a) Benzene	(c) Toluene
	(b) Cyclohexane	(d) m-xylene
6. W	hich of the following is the strongest acid?	•
	(a) Formic acid	(c) Acetic acid
	(b) Dichloroacetic acid	(d) Triflouroacetic acid
7. Co	onc.HNO3 and conc.H2SO4 react with benzoic	
	(a) o-nitrobenzoic acid	(c) m-nitrobenzoic acid
	(b) p-nitrobenzoic acid	(d) o,p-dinitrobenzoic acid
8. WJ	hich of the following ketones has the common	
	(a) Dimethyl ketone	
	(b) Ethyl methyl ketone	
	(c) Diethyl ketone	
	(d) N-propyl ketone	
9. WI	hich of the following compound is a secondar	ry amine?
	(a) CH ₃ -CH(NH ₂)-CH ₃	(c) CH ₃ CH ₂ NHCH ₃
	(b) H ₂ NCH ₂ CH ₂ NH ₂	(d) (CH ₃) ₃ N
10. A	primary amine can be identified using	
	(a) HCl	(c) NaOH
	(b) CHCl2+ KOH	(d) CHCl ₂

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1. Explain the boiling point Dimethyl ether and ethanol. 2. Define term steric effect. 3. Define term electronic effect. 4. Which do you expect to have the higher boiling point: o-chloro phenol or m-chloro phenol? Why? 5. Why 2-chlorobutanoic acid is a stronger acid than butanoic acid? 6. Explain homolytic cleavage of covalent bond. 7. Why p-nitrophenol have high boiling point and higher water solubility than onitrophenol? 8. Write a reaction of alcohols. 9. Explain the acidity of dicarboxylic acid. 10. Differentiate a direct and indirect method of diazotization reaction. 11. Write a synthesis of 1,2,3-tri bromo benzene. 12. Why amines are weaker basic? Write a short note on Inductive effect & Steric effect. [05] Q-3 (A)

Short Answer Question (Attempt TEN out of TWELVE)

[20]

Q-2

	(B)	Write a short note on Carbanions & Carbocation. OR	[05]
	(A)	Write a short note on resonance effect.	[05]
	(B)	Explain Hydrogen bonds in details.	[05]
Q-4	(A)	Write a note on 'Ethers'.	[05]
	(B) OR	Write a note on "Oxymercuration-demercuration" reaction.	[05]
	(A)	Write a note on acid catalyzed dehydration of alcohol.	[05]
	(B)	Write a note on the addition of Grignard reagent in carbonyl compound.	[05]
Q-5	(A)	Write a short note on Cross-aldol condensation.	[05]
	(B) OR	Discuss the Transesterification reaction in details.	[05]
	(A)	Explain the synthesis of a carboxylic acid in details.	[05]
	(B)	Discuss the Hell-Volhard-Zelinsky reaction.	[05]
Q-6	(A)	Write a short note on methods of diazotization reaction.	[05]
-	(B) OR	Discuss the reactions of amine with nitrous acid.	[05]
	(A)	Write a note on analysis of amines.	[05]
	(B)	Write a synthesis of 1,2,3 tri bromo benzene from p-nitro aniline.	[05]
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