[S8]

Sardar Patel University



B. Sc. (Semester – III) Examination

Date: 15-6-2022, Saturday

Time: 12:00 - 2:00 Pm

Industrial Chemistry Vocational

COURSE NO: US03CICV22 (Fundamentals of Organic Chemistry)

Notes: Figures to the right indicate full marks.

Total marks: 70

Q.1 Answer the following Multiple-Choice Questions. (All are compulsory)

(10)

- 1. Which molecule from the below give intermolecular H- bond?
 - A. R-OH

C. R-O-R

B. R-SH

D. R-CO-R

- 2. Arrange the following atoms in decreasing order of electro negativity. 1. Cl, 2. Br, 3. I
 - A. 3>2>1

C. 3>1>2

B. 1>2>3

D. 2>1>3

- 3. Which of the following is most stable carbocation?
 - A. CH3+

C. (Ph)3C+

B. (H3C)3C+

- D. CH3-CH2+
- 4. Which of the following has the highest boiling point?
 - A. CH₂CH₂CH₂CH₂Cl

C. CH₃CH₂CH₂CH₂OH

B. OHCH2CH2CH2CH2OH

- D. CH₃CH₂OCH₂CH₃
- 5. Which of the following compound has the highest boiling point?
 - A. CH₃CH₂CHO

C. CH₃COCH₃

B. CH₃CH₂CH₂OH

- D. CH₃CH₂OCH₃
- 6. Which of the following has the highest boiling point?
 - A. CH₃CH₂OH

C. CH₃OH

B. CH₃CH₂CH₂CH₂OH

D. CH₃CH₂CH₂OH

- 7. Acetone contains.....
 - A. Nine σ bond plus one π bond
 - B. Ten σ bonds
 - C. Eight σ bond plus two π bond
 - D. Nine π bond plus one σ bond
- 8. Which of the following will have the highest boiling point?
 - A. Methanal

C. Propanal

B. Ethanal

- D. Butanal
- 9. Consider a 1°, 2° and 3° amine, all of equivalent molecular weight. Which amine is most likely to have lowest boiling point
 - A. 1º amine
 - B. 2° amine
 - C. 3° amine
 - D. Not enough information to determine
- 10. Which of the following will have highest boiling point?
 - A. Methylamine

C. Ethylamine

B. Diethylamine

D. Ammonia

Q.2 Are the following statements true or false? (All are compulsory	Q	.2 Are the	following	statements	true or	false? ((All a	are com	oulsor	y
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(08)

- 1. A free radical is Neutral in character. True / False?
- 2. Hemolytic cleavage of a bond generates Free radical. True / False?
- 3. Grain alcohol is the other name for "Ethyl alcohol". True / False?
- 4. Lucas reagent is HCl/ZnCl2. True / False?
- 5. The melting points aldehydes and ketones tend to "Increase with increasing molecular weight". True / False?
- 6. Ketones are prepared by oxidation of "Secondary alcohol". True / False?
- 7. Aliphatic primary amines react with cold nitrous acids to form "Diazonium salts". True / False?
- 8. Hinsberg's reagent are "Benzene-sulfonyl chloride". True / False?

Q.3 Answer the following short questions (Attempt Any 10 out of 12)

(20)

- 1. Acetic acid is a weaker acid than formic acid? Explain.
- 2. 2-chlorobutanoic acid is a stronger acid than butanoic acid, explain.
- 3. p-nitro phenol has higher melting point than o-nitro phenol.
- 4. m-& p- Nitrophenol have high boiling point and higher water solubility than o- Nitrophenol. Why?
- 5. o-Nitrophenol is steam volatile. Explain.
- 6. Give reaction for synthesis of Alcohols.
- 7. Discuss the structure and physical properties of ketones.
- 8. Carboxylic acids are more acidic than Alcohols, explain.
- 9. Why acid derivatives are hydrolyzed more readily in acidic/alkaline than neutral medium?
- 10. An aliphatic amine is more basic than ammonia, explain.
- 11. Explain, effect of substituent on basicity of aromatic amines.
- 12. Write synthesis of "p- bromo aniline from aniline".

Q.4 Answer the following Long questions (Attempt Any 04)

(32)

- 1. Write note on "Resonance effect".
- 2. Discuss the "Theories of acids and bases".
- 3. Compare Oxymercuration-demercuration and Hydroboration-Oxidation in detail.
- 4. Discuss acid catalyzed dehydration of alcohol.
- 5. Write a note on "Transesterification reaction".
- 6. Discuss the "Grignard synthesis of a Carboxylic Acid".
- 7. Discuss the reactions of Amine with Nitrous Acid.
- 8. Write a note on "Hinsberg Test".

